CHEMICAL STUDIES OF THE HOST ROCKS OF THE LAKE DUFAULT MINE, QUEBEC

bу

Herbert Challes

A thesis submitted to the faculty of Graduate Studies and Research in partial fulfillment of the requirements for the degree of Doctor of Philosophy.

Department of Geological Sciences McGill University Montreal

August, 1966

TABLE OF CONTENTS

	Page
INTRODUCTION	1
PREVIOUS WORK	2
GEOLOGICAL SETTING	4
NORANDA AREA	4
INTRODUCTION	4
EFFUSIVE ROCKS	4
Andesite Rhyolite	4 5
INTRUSIVE ROCKS	6
Flavrian and Powell Granite Lake Dufault Granodiorite Diorite Dike System	6 6 7
STRUCTURE	8
Folding and Faulting Lineaments and Domes	8
MASSIVE SULPHIDE DEPOSITS	9
ALTERATION	10
LAKE DUFAULT MINE AREA	14
INTRODUCTION	14
ORE BODIES	14
HOST. ROCKS	15

	TABLE OF CONTENTS - (cont'd)	
		Page
	Amulet Andesite Bedded, Cherty Tuff Waite Rhyolite	16 17 17
	STRUCTURE	20
	Faulting Lineaments and Domes Newbec Breccia	20 20 21
GEOCHEMIS	STRY OF THE HOST ROCKS	22
GENE	ERAL APPROACH	22
	SAMPLING PROCEDURE	22
	Rhyolite and Andesite Tuff	22 23
	SAMPLE REDUCTION	23
	SAMPLE PREPARATION (BEYOND REDUCTION)	23
	ANALYTICAL PROCEDURES	23
	REPLICATE ANALYSES: ESTIMATION OF ANALYTICAL AND SAMPLE PREPARATION ERROR	27
•	ERROR IN SAMPLE REDUCTION	30
	METHOD OF PRESENTATION OF RESULTS	31
·.	Data Charts Frequency Distribution Diagrams	31 33
	Type I : Unimodal symmetrical curves	33
	Type II : Unimodal asymmetrical curves with positive skewness Type III : Bimodal or Multimodal	35
	curves	36

TABLE OF CONTENTS - (cont'd)	
e sette de la companya de la compan La companya de la co	Page
Type IV: Asymmetrical curves with a frequency greater than zero in the lowest valued cell of the histogram Type V: J-shaped distributions	36 36
Tables Summarizing the Vertical Distribution of Elements	37
METHOD OF PRESENTATION OF RESULTS FOR SAMPLES OF TUFF	37
ACCURACY OF RESULTS	40
BACKGROUND VALUES FOR ELEMENTS IN THE HOST ROCKS AND THEIR SIGNIFICANCE	49
' MAJOR ELEMENTS	49
TRACE ELEMENTS	58
Waite Rhyolite	58
Cobalt, nickel and chromium Lead and tin Zirconium, strontium and	58 58
yttrium Gallium Silver, copper and zinc	59 60 60
Amulet Andesite	60
Cobalt, chromium, nickel and tin Lead Zirconium Strontium Yttrium and gallium Silver, copper and zinc	62 62 63 63 63
Conclusions Bedded Charty Tuff	64 65

TABLE OF CONTENTS - (cont'd)	
	Page
DISTRIBUTION OF ELEMENTS IN THE HOST ROCKS	65
LATERAL DISTRIBUTION OF ANOMALOUS VALUES OF ELEMENTS IN RHYOLITE AND ANDESITE	65
Rhyolite Andesite Summary of Anomalous Values	65 66 66
LATERAL DISTRIBUTION OF ELEMENTS IN BEDDED, CHERTY TUFF	67
VERTICAL DISTRIBUTION OF ELEMENTS IN RHYOLITE AND ANDESITE	67
DISCUSSION OF RESULTS	98
Major Elements	101
Silicon Aluminum Sodium Potassium Magnesium Calcium Iron (total as Fe ₂ 0 ₃) Manganese Titanium	101 102 103 104 107 107 108 108
Trace Elements	108
Group A: Elements "added" to altered zones (Cu, Zn, Pb, Ag, Sn, Cr, Zr, Y) Group B: Elements "removed" from altered zones (Sr, Ga) Group C: Elements "added" to or "removed" from altered zones (Co, Ni)	109 110
Summary and Conclusions	111

TABLE OF CONTENTS - (cont'd)	
	Page
ORE DEPOSITS SIMILAR TO THOSE OF THE NORANDA AREA	116
GEOCHEMICAL PROSPECTING APPLICATIONS	117
SUMMARY	118
CLAIM OF ORIGINAL WORK AND CONTRIBUTION TO	
KNOWLEDGE	123
ACKNOWLEDGMENTS	124
BIBLIOGRAPHY	126
APPENDIX A: Frequency Distribution Diagrams of the Elements in Rhyolite and Andesite	139
APPENDIX B : Data Charts of the Elements in Rhyolite and Andesite	140
Abbreviations Used in Presentation of	
Results B-1 SiO ₂ B-2 TiO ₂ B-5 Al ₂ O ₃ B-8 Fe (total as Fe ₂ O ₃) B-11 MnO B-14 CaO B-17 K ₂ O B-20 MgO B-23 Na ₂ O B-23 Co B-29 Co B-32 Cr B-35 Cu B-38 Ga B-41 Ni B-44	

TABLE OF CONTENTS - (cont'd)	_
	Page
Pb B-47 Sn 8-50 Sr B-53 Y 8-56 Zn B-59 Zr B-62	
APPENDIX C : Analytical Results for Bedded Cherty Tuff	1 41
APPENDIX D : Summary of Anomalous Values of the Elements in Rhyolite and Andesite	142
APPENDIX E : Test for Systematic Vertical	
Variation of Elements in Rhyolite and Andesite	1 43
APPENDIX F : Use of the I.B.M. 7040 Computer	
in Handling Data	144
APPENDIX G : Description of Sample Preparation and Analytical Procedures	145
APPENDIX H : Photographs of the Host Rocks	146
APPENDIX I : Sample Locations	147

LIST OF FIGURES

•			Page
Figure	1.	Geological Setting of the Noranda Area	3
Figure	2.	Generalized Cross Section, Lake Dufault Mines Ltd.	13
Figure	3.	Types of Frequency Distributions of Concentrations of Elements in Geochemical Systems	32
Figure	4.	Formulae for Data Chart Lines (for Normal Distributions)	34
Figure	5.	Peacock's Lime : Alkali Index for Noranda Area Rocks	45
Figure	6.	Variation Diagrams for Rocks of the Noranda Area	46
Figure	7.	Plot of Niggli alk-si Values for Noranda Area Rocks	48
Figure	8.	Plot of Niggli alk-si Values for Control Groups of Lavas : Oceanic Alkaline; Orogenic Calc-alkaline	48
Figure	9.	Comparison of the Major Element Composition of Andesites, Basalt, Granite and Continental Crust	50
Figure	10.	Comparison of the Waite Rhyolite with the Continental Crust and the Average Felsic Rock	57
Figure	11.	Comparison of the Trace Element Composition of Andesites, Baselt, Granite and Continental Crust	61
Figure	12.	Distribution of ${\rm SiO}_2$ in Rhyolite at the Main Contact	68
Figure	13.	Distribution of Na ₂ O in Rhyolite at the Main Contact	69

	LIST OF FIGURES - (cont'd)	
		Page
Figure 14.	Distribution of Ni in Rhyolite at the Main Contact	70
Figure 15.	Distribution of Fe in Rhyolite at the Main Contact	71
Figure 16.	Distribution of MnO in Rhyolite at the Main Contact	72
Figure 17.	Distribution of MgO in Rhyolite at the Mein Contact	73
Figure 18.	Distribution of Ag in Rhyolite at the Main Contact	74
Figure 19.	Distribution of Cu in Rhyolite at the Main Contact	75
Figure 20.	Distribution of Pb in Rhyolite at the Main Contact	76
Figure 21.	Distribution of Sn in Rhyolite at the Main Contact	77
Figure 22.	Distribution of Zn in Rhyolite at the Main Contact	78
Figure 23.	Distribution of Al ₂ O ₃ in Andesite at the Main Contact	79
Figure 24.	Distribution of CaO in Andesita at the Main Contact	80
Figure 25.	Distribution of K ₂ O in Andesite at the Main Contact	81
Figure 26.	Distribution of Co in Andesite at the Main Contact	82
Figure 27.	Distribution of Cr in Andesite at the Main Contact	83
Figure 28.	Distribution of Ni in Andesite	84

	LIST OF FIGURES - (cont'd)	Page
Figure 29.	Distribution of Sn in Andesite at the Main Contact	85
Figure 30.	Distribution of Ag in Andesite at the Main Contact	86
Figure 31.	Distribution of Cu in Andesite at the Main Contact	87
Figure 32.	Distrit tion of Pb in Andesita at the Main Contact	88
Figure 33.	Distribution of Zn in Andesite at the Main Contact	89
Figure 34.	Distribution of Anomalous Values for Fe, MnO and MgO in Rhyolite at the Main Contact	90
Figure 35.	Distribution of Anomalous Values for Ag, Cu, Pb, Sn and Zn in Rhyolite at the Main Contact	91
Figure 36.	Distribution of Anomalous Values for Cu, Pb, Ag and Zn in Andesite at the Main Contact	92
Figure 37.	Distribution of Pb in Tuff at the Main Contact	93
Figure 38.	Distribution of Cd in Tuff at the Main Contact	94
Figure 39.	Distribution of Zn in Tuff at the Main Contact	95
Figure 40.	Systematic Vertical Variation of the Elements in Waite Rhyolite	96
Figure 41.	Systematic Vertical Variation of	97

		LIST OF FIGURES - (cont'd)	
			Page
figure	A1 .	Elements Normally Distributed in Rhyolite: TiO2, Al203, CaO, K2O	A-1
Figure	A2.	Elements Normally Distributed in Rhyolite: Ga, Sr, Y	A-2
Figure	A3.	Elements Normally Distributed in Andesite : SiO ₂ , TiO ₂ , Fe, CaO, MnO, MgO	A-3
Figure	A4.	Elements Normally Distributed in Andesite: Na ₂ O, Ga, Y, Zr	A-4
Figure	A5.	Elements with Frequency Distributions Other than Normal in Rhyolite : Cu, Ni, Ag, Sn, Zn, Pb	A-5
Figure	A6.	Elements with Frequency Distributions Other than Normal in Rhyolite : SiO ₂ , MnO, Fe, Na ₂ O, MgO, Zr	A-6
Figure	A7.	Elements with Frequency Distributions Other than Normal in Andesite : Al ₂ O ₃ , K ₂ O, Ag, Co, Cr	A-7
Figure.	A8.	Elements with Frequency Distributions Other than Normal in Andesite : Cu, Pb, Ni, Sn, Sr, Zn	A-8
Figure	F1.	Computer Program Used in Data Handling; Fortran Statements	F-2
Figure	F2.	Special Data Sheet Used in Data Handling	F - 5
Figure	F3.	Example of Output of the Computer Program	F-6
Figure	G1.	Sample Reduction and Sample Preparation for the Major Element Procedure, X-ray Fluorescence	G - 1

	LIST OF FIGURES - (cont'd)	Page
Figure G2.	Sample Preparation, Major Element Procedure, Optical Spectrograph	G-2
Figure G3.	Sample Preparation, Involatile Trace Element Procedure, Optical Spectrograph	G-3
Figure G4.	Sample Preparation, Volatile Trace Element Procedure, Optical Spectrograph	G-4

LIST OF TABLES

			Page
Table	1.	Sample Preparation and Analytical Error	26
Table	2.	Analysis of Errors of Manipulation	29
Table	3.	Analytical Data for Standard Samples, G-1, W-1, CAAS Syenite, CAAS Sulphide	39
Table	4.	Major Element Content of the Host Rocks	41
Table	5.	Published Analyses of Noranda Area Rocks	42
Table	6.	Some Analyses of Low Potash, High Silica (>70%), Volcanic Rocks	52
Table	7.	Trace Element Content of the Host Rocks	56
Table	8.	Summary of Chemical Variations in the Waite Rhyolite	97
Table	9.	Summary of Chemical Variations in the Amulet Andesite	98
Table	10.	FeO Content of the Waite Rhyclite	103
Table	11.	FeO Content of the Amulet Andesite	106
Table	C1.	Analytical Results for Tuff : $5i0_2$, $Ti0_2$, Al_20_3 , Fe_20_3	C-1
Table	C2.	Analytical Results for Tuff : MnO, CaO, K ₂ O, MgO	C-2
Table	C3.	Analytical Results for Tuff : Na ₂ O, Ag, Bi, Cd	C - 3
Table	C4.	Analytical Results for Tuff : Co, Cr, Cu, Ga	C-4
Table	C5.	Analytical Results for Tuff : Ge, Ni, Pb, Sn	C-5
Table	C6.	Analytical Results for Tuff : Sr, Y, Zn, Zr	C-6

	LIST OF TABLES - (cont'd)	
		Pag
Table D1.	Summary of Anomalous Values for Major Elements in Rhyclite : SiO_2 , TiO_2 , Al_2O_3 , CaO , K_2O , Na_2O	D-1
Table D2.	Summary of Anomalous Values for Trace Elements in Rhyolite : Ga, Ni, Sr, Y, Zr	D-2
Table D3.	Summary of Anomalous Values for "Ore-Major" Elements in Rhyolite : Fe ₂ O ₃ , MnO, MgO, Sum	D-3
Table D4.	Summary of Anomalous Values for "Ore- Trace" Elements in Rhyolite : Ag, Cu, Pb, Sn, Zn, Sum	D-4
Table D5.	Summary Anomalous Values for Major Elements in Andesite : SiO ₂ , TiO ₂ , Fe ₂ O ₃ , MnO, CaO, K ₂ O, MgO, Na ₂ O	D - 5
Table D6.	Summary of Anomalous Values for Trace Elements in Andesite : Co, Cr, Ni, Ga, Sn, Y, Zr	D - 6
Table D7.	Summary of Anomalous Values for "Ore- Trace" Elements in Andesite : Ag, Cu, Pb, Zn, Sum	D-7
Table E1.	Number of Diamond Drill Holes Showing Systematic Vertical Variation in Waite Rhyolite (away from ore)	E-1
Table E2.	Number of Diamond Drill Holes Showing Systematic Vertical Variation in Waite Rhyolite (near ore)	E-2
Table E3.	Number of Diamond Drill Holes Showing Systematic Vertical Variation in Amulet Andesite (away from ore)	E-3
Table E4.	Number of Diamond Drill Holes Showing Systematic Vertical Variation in Amulet Andesite (near ore)	E-4

·	LIST OF TABLES - (cont'd)	Page
Table G1.	Time Required for an Inexperienced Technician to Prepare Approximately Sixty Samples for All Procedures	G-5
Table G2.	Details of Analytical Procedures Using the Optical Spectrograph	G - 6
Table I1.	Sample Locations	I-1

LIST OF DATA CHARTS

		Page
Chart 1.	SiO ₂ in Rhyolite	B - 3
Chart 2.	SiO ₂ in Andesite	B-4
Chart 3.	TiO ₂ in Rhyolite	B - 6
Chart 4.	TiO ₂ in Andesite	B-7
Chart 5.	Al ₂ O ₃ in Rhyolite	8-9
Chart 6.	Al ₂ O ₃ in Andesite	8-10
Chart 7.	Fe (Total as Fe ₂ 0 ₃) in Rhyolite	B -1 2
Chart 8.	Fe (Total as Fe ₂ 0 ₃) in Andesite	B -1 3
Chart 9.	MnO in Rhyolite	B-15
Chart 10.	MnO in Andesite	B -1 6
Chart 11.	CaO in Rhyolite	B -1 8
Chart 12.	CaO in Andesite	B-19
Chart 13.	K ₂ O in Rhyolite	B-21
Chart 14.	K ₂ O in Andesite	B-22
Chart 15.	MgD in Rhyolite	B-24
Chart 16.	MgO in Andesite	B-25
Chart 17.	Na ₂ O in Rhyolite	B-27
Chart 18.	Na ₂ O in Andesite	B - 28
Chart 19.	Ag in Rhyolite	B-30
Chart 20.	Ag in Andesite	B-31
Chart 21.	Co in Rhyolita	8-3:

•	LIST OF DATA CHARTS - (cont'd)	
		Page
Chart 22.	Co in Andesite	B-34
Chart 23.	Cr in Rhyolite	B - 36
Chart 24.	Cr in Andesite	B - 37
Chart 25.	Cu in Rhyolite	B - 39
Chart 26.	Cu in Andesite	B-40
Chart 27.	Ga in Rhyolite	B-42
Chart 28.	Ga in Andesite	B-43
Chart 29.	Ni in Rhyolite	B - 45
Chart 30.	Ni in Andesite	8-46
Chart 31.	Pb in Rhyolite	B-48
Chart 32.	Pb in Andesite	B-49
Chart 33.	Sn in Rhyolite	B - 51
Chart 34.	Sn in Andesite	B-52
Chart 35.	Sr in Rhyolite	B - 54
Chart 36.	Sr in Andesite	8-55
Chart 37.	Y in Rhyolite	8-57
Chart 38.	Y in Andesite	8-58
Chart 39.	Zn in Rhyolite	B-60
Chart 40.	Zn in Andesite	B-61
Chart 41.	Zr in Rhyolite	B-63
Chart 42.	Zr in Andesite	B-64

LIST OF PLATES

			· · · · · · · · · · · · · · · · · · ·	
				Page
Plate 1.		•		H-1
		Waite Rhyolite, Waite Rhyclite,		
Plate 2.		•		H-2
Fig.	1.		deformed chloritic ase siliceous matrix.	
Plate 3.				H - 3
		Waite Rhyolite, Waite Rhyolite,		
Plate 4.				H-4
_		Waite Rhyolite, Waite Rhyolite,	• •	\$
Plate 5.				H - 5
Fig. Fig.			scoriaceous fragment. tuffaceous matrix of	
Plate 6.				H-6
Fig.	1.	fractured and m	, altered, from the mineralized zone below	٠.
Fig.	2.		, altered, from the mineralized zone below	
Plate 7.			·	H-7
Fig.	1.	Amulet Andesite		

LIST OF PLATES - (cont'd)						
						Page
Plate 8.						H-8
Fig. Fig.			Rhyolite, Rhyolite,	pyrocla	astic. astic.	
Plate 9.						H - 9
. Fig. Fig.	1a. 1b.	Waite Waite	Rhyolite, Rhyolite,	finely pyrocl:	pyroclasti astic.	C.
Plate 10.						H-10
Fig. Fig.	1a. 1b.	Waite Waite	Rhyolite, Rhyolite,	finely finely	pyroclasti pyroclasti	c.
Plate 11.						H-11
Fig. Fib.	1a. 1b.	Waite Waite	Rhyolite, Rhyolite,	pyrocla	astic. astic.	
Plate 12.					•	H-12
Fig.	1.		Andesite i Cherty Tu		act with	
•				•	•	
Map 1						•

Geological Compilation -- Noranda Area

pocket

INTRODUCTION

This thesis is the outcome of a proposal made in the spring of 1962 by W. C. Martin to Falconbridge Nickel Mines Limited and Lake Dufault Mines Limited. Mr. Martin suggested the application of geochemical prospecting techniques to the problem of finding blind ore bodies on a property of Lake Dufault Mines Limited, the site of a new base metal discovery. The writer was employed to make preliminary studies for this purpose. He spent the first month of the summer of 1962 absorbing facts and ideas about the geology of the area. The remainder of the summer was spent collecting diamond drill core samples for chemical analysis. Chemical analyses were performed in the Department of Geological Sciences, McGill University.

It is the purpose of this thesis to report the results of this geochemical study and to interpret the chemical features of the host rocks of the Lake Dufault Mine on the basis of what is known of the geology of the Noranda area in general and of the Lake Dufault Mine in particular.

PREVIOUS WORK

Wilson (1910) and Cook, James and Nawdsley (1931) provided the first summary reports on the geology of the Noranda area. Wilson's (1941) memoir remains a standard reference. More recently, Robinson (1951), Dugas and Hogg (1962), Gilmore (1965) and Roscoe (1965) have contributed to the general knowledge of the Noranda area.

The first chemical data from the Noranda area were published by Wilson (1941) who noticed, among other things, that the volcanic rocks had a "...large proportion of soda in contrast with the small content of potash...", a relation first suggested by Schindler (1934) from petrographic data. Additional chemical data were presented by Riddell (1952) who established the chemical features of wallrock alteration in the district. Lickus (1965) gives many chemical analyses of rock on the property of Vauze Mines Limited. Roscoe (1965) published chemical analyses for major units of the Noranda volcanic rocks plus some chemical analyses of minerals.

Many studies of individual mines have been recorded, the most detailed of which include that of the Horne Mine by Price (1933, 1934, 1949) and that of the Vauze Mine by Lickus (1965).

Wilson (1941, p. 60) lists the following structural features associated with the massive sulphide ore deposits of the Noranda district: 1. faults, 2. pyroclastic and tectonic breccia in volcanic rock, 3. impermeable rock "barriers" above the deposits and, in some cases, 4. anticlinal folds. Wilson also reviews conflicting arguments

which bear on the relationship of "late" diabase dikes to ore. Wilson favored a post-diabase age for the Noranda deposits and hence, like many of his contemporaries who studied the geology of the Noranda district, believed that the sulphide ore bodies were formed after their host rocks had been deformed and metamorphosed.

Since Wilson's memoir was published, additional information has become available and many geologists, perhaps influenced by the work of such men as Oftedahl (1958) and Stenton (1960), have come to realize that an alternate hypothesis is applicable to the problem of ore genesis in the Noranda district. They propose that the Noranda ore deposits and the volcanic rock in which they occur are contemporary products of Precambrian volcanic activity centered in the Noranda area. This idea was under serious consideration at the time of the C.I.M. Symposium on Massive Sulphide Deposits in 1959 (Gill, 1959). It has been advocated strongly during the past year by Gilmore (1965), Lickus (1965), and Goodwin (1965).

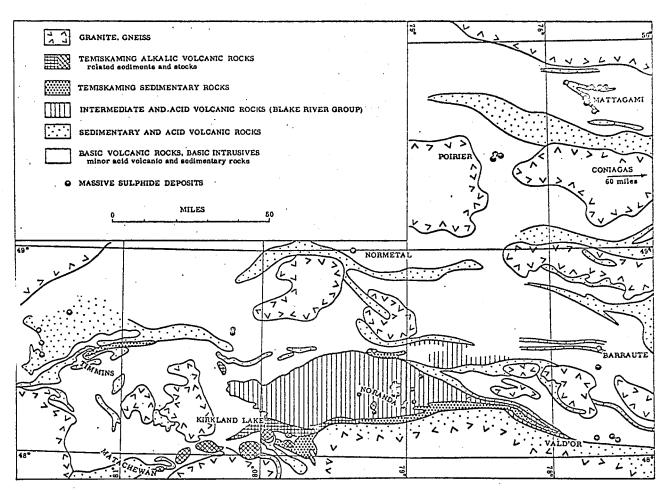


Fig. 1 Geological Setting of the Noranda Area. (from Roscoe, 1965)

GEOLOGICAL SETTING

NORANDA AREA

INTRODUCTION

The Noranda ore deposits are all located in an elliptically shaped area (80 x 20 miles) of intermediate and siliceous volcanic rocks of Precambrian age. These rocks were called the "Blake River Group" by Gunning and Ambrose (1939) (fig. 1). To the south and parallel to the long axis of this area lies the east-trending Cadillac-Bouzan fault zone. South of this major "break" lie greywacke and conglomerate which give way still further south to metamorphosed sedimentary rocks mixed with granite and gneiss. All of these rocks are covered in places to the south and west by conglomerate and greywacke. The area to the north, east and west of the Noranda area consists mainly of mafic volcanic rocks and intrusives with minor bands of siliceous volcanic and sedimentary rocks.

EFFUSIVE ROCKS

Andesite

The rocks in the Noranda area that can be definitely recognized as flows range in chemical composition from basalts to dacites. Roscoe (1965) notes that the less mafic flows occur higher up in the sequence. Pillows and amygdales are common in these rocks, and minor flow breccia

and chert have been reported. Microscopic examination of thin sections of andesite usually reveals clouded plagioclase feldspar with the alteration products, sericite, chlorite, actinolite, calcite and epidote, plus some quartz. Therefore the rocks of the Noranda area are properly placed in the greenschist facies of regional metamorphism. Schists are found only in fault zones.

Rhyolite

The only siliceous rocks in the area that might be classed as flows are small irregularly-shaped bodies of quartz-feldspar porphyry. One of these bodies has a flat base, a dome-like top and a long axis which trends N. 70° E. (staff, Lake Dufault Mines Ltd.). This "dome" is located 4,000 feet south of "the Lower A" ore body of the Amulet mine but is exposed only in a small area on the surface. It is suggested that these small siliceous bodies are the so called "endogenous domes" found so often in areas of siliceous volcanic activity (Rittman, 1962).

The remainder of the siliceous rocks have been called rhyolite, rhyolite flow breccia, rhyolite breccia and rhyolite tuff. In the opinion of the writer essentially all of this rock is pyroclastic. Most of the rhyolite units are thin relative to their areal extent and thus have a general form inconsistent with the shape of true siliceous lava flows. Often the most massive-appearing rhyolite can be seen, on closer examination, to consist entirely of fragments. The writer is convinced that detailed petrographic examination of the siliceous rocks of the Noranda area would, in most cases, reveal pyroclasts. A minor

amount of bedded tuff occurs with much of the pyroclastic rock.

INTRUSIVE ROCKS

The intrusive rocks of the area include a myriad of dikes and sills of widely varying size, shape and composition. Some sills are closely related to the effusive rocks in the Noranda area (e.g., quartz porphyry, "metadiabase") and some dikes cut across all rock structures (e.g., "late" quartz-diabase). Many of the massive sulphide ore bodies are cut by dikes and probable sills (Lickus, 1965). Only the larger intrusive bodies for which some chemical data are available are described here.

Flavrian and Powell Granite

The Flavrian and Powell granite bodies are similar in many respects. Both intrusive masses contain abundant inclusions of rhyolite and other rocks, both are altered to a moderate degree with the development of chlorite, sericite, epidote and carbonate, and both cut andesite and rhyolite. On the basis of limited chemical work both have the same chemical composition (table 5, p.42). Both are also similar, chemically, to the rhyolite which they cut. This evidence suggests that the Flavrian and Powell granite and the rhyolite of the Noranda area all were crystallized from magmas of similar composition.

Lake Dufault Granodiorite

The Lake Dufault granodiorite is located to the east of the Flavrian mass, cutting andesite, rhyolite and diorite. The western

half of the stock is relatively free of inclusions and is less altered than the Flavrian and Powell granite. The eastern half of the stock is a mixed rock (Webber, 1962) and contains a possible pendant of rhyolite in which the MacDonald ore body occurs. One analysis of a dacite flow at Vauze Mine corresponds to the few chemical analyses given for the granodiorite (Lickus, 1965) (table 5, p.42). The texture of granodiorite apophyses in country rock is as coarse as that of the main body (Wilsen, 1941). It would seem therefore that the granodiorite intruded an already hot country rock.

Diorite Dike System

A map by J. Boldy (map I) shows a dike system around the Flavrian and Powell granite masses which, according to Wilson (1941), is composed of quartz diorite, diorite, gabbro, diorite porphyry and albite granite. The variety of rock types plus the fact that internal chilled contacts have been observed in one of the larger dikes (Lickus, 1965), suggest that the rocks of the dike system have been intruded in a series of pulses. "Flat" or sill-like diorite is known to occur east of the Amulet mine, east of the Lake Dufault mine, at the Vauze ore body and near the Francoeur mine. Irregularly shaped areas of diorite are probably the "flat" or sill portions of the system. The dike portions of the system have steep dips so far as is known and actually lie alone

Some of Wilson's rock types may actually be unrelated to the diorite dike system. They may be massive portions of andesite flows or sills closely associated with the flows.

faults. These faults have a smaller apparent dip slip away from the Flavrian granite. The impression is gained that the diorite of the dike system encloses volcanic rock which was broken into huge blocks, possibly as a result of subsidence in the area of the Flavrian and Powell granite bodies.

STRUCTURE

Folding and Faulting

The structure of the Noranda area is relatively simple. The only strong deformation has occurred along the southern border of the area where the volcanic rocks became involved in the deformation along the Cadillac-Bouzon fault zone. Here, faults are numerous and the volcanic formations are folded. North of this deformed area the rocks are only gently folded (if at all) and the greatest deformational event seems to have been the faulting referred to in connection with the emplacement of the diorite dike system.

"Late" diabase dikes are offset by faults but are otherwise undeformed (map I), (Wilson, 1941, p. 47).

<u>Lineaments</u> and Domes

Many air photograph lineaments mark major faults, and post-ore faults have been noted at the Vauze (Lickus, 1965) and Lake Dufault mines (staff, Lake Dufault Mines Ltd.). Some lineaments have been shown to have great significance for it is at intervals along them that thickening of the rhyolite occurs. The Lake Dufault, Vauze and

Waite ore bodies are found at the top of locally thickened sections of rhyolite (Lickus, 1965; staff, Lake Dufault Mines Ltd.), a fact which suggests that rhyolite and ore may be genetically related.

MASSIVE SULPHIDE DEPOSITS

Gold-bearing quartz veins are found in most of the rock types present in the Noranda erea, including the Flavrian and Powell granite. Minor, chalcopyrite bearing, quartz veinlets are present in the Powell granite on the Don Rouyn property and in extrusive rocks throughout the district.

All of the massive sulphide ore bodies in the Noranda area are found at the stratigraphic top of a siliceous pyroclastic rock unit except the upper ore bodies of the Old Waite mine and the Amulet "Upper A" ore bodies. The host rock for these ore bodies is andesite. The hanging wall rock for many ore bodies is bedded tuff or andesite. The ore bodies are tabular and their larger dimensions are essentially parallel to the nearest stratification or flow contact. In recent years many geologists have considered these features, which are common to many ore deposits, as suggestive of a close relationship between ore formation and an early volcanic environment (Thomson, 1960).

The massive sulphide ore bodies are easily grouped into combinations of (1) pyrite-sphalerite ore and (2) pyrite-chalcopyrite-pyrrhotite ore with or without sphalerite. Roscoe (1965) describes a large scale stratigraphic zoning in the Noranda district sulphide deposits. Sphalerite, often interlayered with pyrite and pyrrhotite (Lickus, 1965), is concentrated at the top of the massive sulphide ore

and chalcophrite is more common near the base and stratigraphically below massive ore in altered, mineralized rhyolite. This zoning holds true for the Horne deposits even though the strata are tilted to a vertical attitude. Roscoe suggests that such zoning in vertical strata must have been developed when the strata were in a sub-horizontal position. Interesting sulphide breccias have been noted at the Vauze mine (Lickus, 1965), at the West MacDonald deposit, and at one of the Horne ore bodies (Roscoe, 1965). Roscoe states that the breccies "contain displaced and corroded sulphide fragments and siliceous fragments that can be envisaged as bounced about during explosive fumarolic activity and recemented by sulphides and some siliceous material. These features suggest deposition at the surface, perhaps within loose tuffaceous materials." At the Vauze mine, Lickus describes sulphide fragments of contrasting mineral composition side by side in agglomerate. He suggests that fragmentation of sulphides must have been contemporaneous with the formation of the agglomerate. In summary, this evidence suggests that a genetic relationship exists between the Noranda area sulphide deposits and the volcanic rocks in which they are enclosed.

ALTERATION

The most striking feature of alteration in the Noranda area is the occurrence, with some of the ore bodies, of "Dalmatianite," a spotted rock containing biotite, cordierite and anthophyllite.

"Dalmatianite" is not found at the largest group of ore bodies in the district (the Horne mine) where only chlorite and sericite

alteration is present. The significant facts of wallrock alteration in the Noranda area have been given by Riddell (1952). He shows that two styles of wallrock alteration corresponding to two types of massive sulphide ore deposits are present in the district. Sodium, magnesium and calcium have been leached, potassium has been fixed or added, and sericite has been formed in altered zones near massive sulphide bodies which contain pyrite and sphalerite but only minor copper. Where copper is present in the massive sulphides, magnesium has been added and chlorite has been formed. Sodium, calcium and, in some places, potassium have been removed from this last type of altered zone.

The chemical composition of "Dalmatianite" zones is similar to that of the chlorite-sericite zones (Riddell, 1952; Lickus, 1965).

At the Vauze mine an altered zone containing an unusually high amount of magnesium occurs stratigraphically below the massive sulphide ore body in fractured, mineralized rhyolite (Lickus, 1965). No magnesium seems to have been added to andesite which forms the hanging wall of massive ore. This is also a feature of the Lake Dufault deposit as described in a forthcoming section. It is the opinion of some geologists who have studied the Noranda district, and the writer concurs, that the andesite which now forms the hanging wall of massive sulphides in these deposits should contain some added magnesium if the flows were present at the time the sulphides were emplaced. A "Dalmatianite pipe" within andesite connects the "Lower A" and "Upper A" ore bodies in the Amulet mine area. Otherwise, magnesium metasomatism in andesite is rare.

"Dalmatianite" alteration zones have been reported only in areas near the Lake Dufault, Waite and Amulet ore bodies, all of which are

close to the Lake Dufault granodiorite stock. It seems likely that biotite and "Dalmatianite" develop from the isochemical contact metamorphism of chlorite-sericite alteration. Riddell (1952) describes a concentric mineral and chemical zoning in "Dalmatianite" at the "Lower A" ore body, Amulet mine. The mineral zoning is, from the center outward, anthophyllite (gedrite)-biotite, cordierite-anthophyllite (gedrite)-biotite, cordierite-biotite and biotite and is, in the opinion of the writer, most probably caused by chemical zoning in a pre-existing chlorite-sericite alteration zone. Magnesium and iron are present in large amounts in the altered rock and increase as potassium decreases towards the center of the zone. Thermal metamorphism of a chloritesericite mass having this chemical zoning would result in the formation of biotite until potassium is used up, cordierite until aluminum is used up, and anthophyllite (gedrite) if there is enough magnesium, iron and silica left to satisfy its requirements. Lickus (1965) reports a similar type of chemical zoning in a chlorite-sericite "pipe" at the Vauze mine.

The existence of a contact metamorphic aureole near the Lake

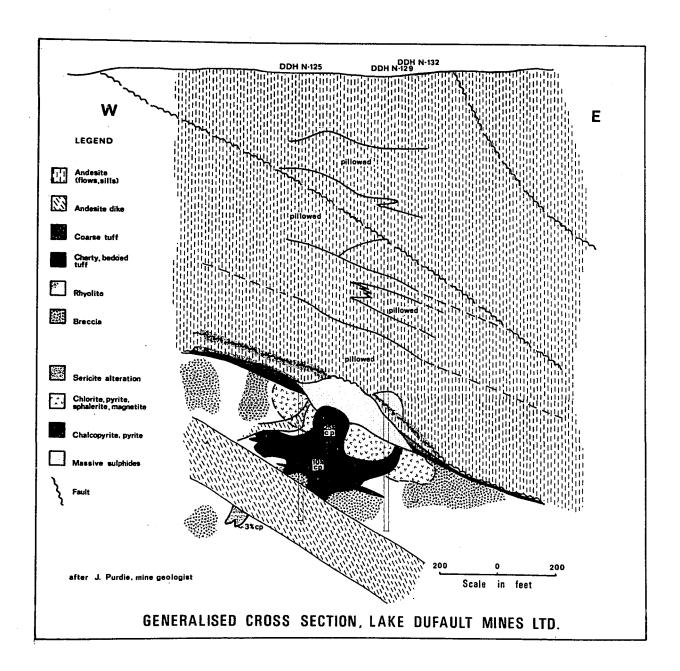
Dufault granodiorite stock is suggested by the work of Schindler (1934)

on the old Waite-Ackerman-Montgomery property. He described pyroxene

in andesite up to 200 feet away from granodiorite and biotite in andesite

as far as 1.200 feet away from the granodiorite-andesite contact.

The anthophyllite-cordierite assemblage of the "Dalmatianite pipes" belongs to the hornblende-hornfels facies of contact metamorphism (Turner and Verhoogen, 1960, p. 511-520). These authors "tentatively correlate the hornblende-hornfels facies with temperatures of about 550° to 700° C. in the pressure range $PH_{20} = 1,000 - 3,000$ bars."



LAKE DUFAULT MINE AREA

INTRODUCTION

The large Noranda properties of Lake Dufault Mines Limited are located mainly in Dufresnoy and Duprat townships approximately five miles directly north of Rouyn-Noranda (map I). The properties can be subdivided into two areas, one south and one north of the Lake Dufault granodiorite stock. The southern area is near the mined-out Amulet ore bodies. The northern area lies east of the mined-out Waite and Vauze ore bodies and is known as the Norbec area. The Norbec area consists of three faulted blocks of andesite and rhyolite separated by diorite of the dike system. These blocks are bounded at the north by a northeast trending fault and at the south by the Lake Dufault granodiorite stock. The Lake Dufault ore body is located in the central block. Geochemical samples were taken from the entire Norbec area.

ORE BODIES

The Lake Dufault ore body, actually two ore bodies, is a typical Noranda-style occurrence having a lens-shaped massive sulphide portion underlain by a fractured zone containing vein and disseminated sulphides (fig. 2). The massive body measures 150 feet by 400 feet by 550 feet. The longest dimensions lie in the plane of the contact between hanging wall andesite and foot wall rhyolite. Massive ore contains pyrite, pyrrhotite, sphalerite, chalcopyrite, magnetite and traces of galena and constitutes 1,644,400 tons of ore grading 9.3% zinc, 4.6% copper, 2.6oz.

silver and 0.04oz. gold. The low grade fracture zone is located stratigraphically below the massive body. The economically important mineral here is chalcopyrite, but pyrite, pyrrhotite, sphalerite and magnetite are common. That part of the fractured section which makes ore has a crude elliptical shape. To date, 545,000 tons of ore grading 1.6% copper, 0.1% zinc, 0.2oz. silver and 0.003oz. gold has been proved in this location. The long axis of the fractured and mineralized zone is parallel to the long axis of the massive sulphide body and is connected to massive ore by two pipe-like fracture zones. Layered sphaleritepyrrhotite-pyrite ore is found at the top of the massive sulphide body. Sheared and structureless areas are common in massive ore. Coarse pyrite "nodules," similar to those reported by Kinkel (1963) in recrystallized ore, are present. A large scale zoning is present in massive ore as zinc increases and copper decreases toward the stratigraphic top away from the chalcopyrite-filled fractures below (staff, Lake Dufault Mines Ltd.).

HOST ROCKS

Amulet andesite forms the hanging wall of the massive sulphide ore body. Waite rhyolite forms the foot wall of massive ore and is host to the ore of the fractured zone. Bedded, cherty tuff is found mainly between andesite and rhyolite. Dugas and Hogg (1962) and Lickus (1965) review the stratigraphy of the general area.

^{1.} These units were called Waite Hills andesite and Waite Lake siliceous rhyolite by Wilson (1941, p. 7). The terms "Amulet andesite" and "Waite rhyolite" are used by the staff of Lake Dufault Mines Ltd. and by Lickus (1965) for the same units in the Vauze mine area.

Amulet Andesite

The Amulet andesite is characteristically pillowed and amygdaloidal (Plate 7, fig. 1; Appendix H). Massive sections are common and some of these are almost certainly dikes and sills.

Alteration which is common throughout all of the andesite consists of three main types: (1) chlorite, actinolite, epidote and sericite which have developed from the breakdown of plagioclase feldsper and mafic minerals during regional metamorphism; (2) patches of epidote, quartz and calcite, and (3) white, sericitic patches. Small laths and phenocrysts of altered plagioclase feldspar are common.

Biotite has formed in the andesite in places close to diorite and granodiorite.

Chlorite, actinolite, sericite, quartz, calcite, pyrite, and pyrrhotite are often found in amygdales, small fractures and spots throughout the andesite. Away from the contact with massive sulphides only scattered bits of chalcopyrite with pyrrhotite and, very rarely, sphalerite have been reported. Chalcopyrite seems to have a much wider distribution in andesite than does sphalerite.

A major type of alteration of andesite is present mainly at its base (Plate 7, fig. 2; Appendix H). Here, the development of quartz-carbonate veinlets has accompanied the large scale development of sericite which changes the greenish andesite to a tan colored rock. Shearing always accompanies this alteration which is seen elsewhere, well away from the base of the andesite in other shear zones and, on a small scale, adjacent to some fractures. This alteration is commonly absent where andesite forms the foot—wall of massive ore.

Bedded, Cherty Tuff

Bedded, cherty tuff occurs mainly at the contact of andesite and rhyolite but it also has been seen well within the rhyolite. It is characterized by numerous, thin, alternating layers of light to dark gray chert and green chlorite (Plate 9, fig. 1; Appendix H). The bedded tuff is usually from two to twenty feet thick and seems to be thickest near ore. Pyrite and pyrrhotite are the most common sulphides present and are usually concentrated in thin layers. Chalcopyrite and sphalerite are often present. The contact between the bedded, cherty tuff and the underlying rhyolite is vague and unmarked in some places. Tuff at the main contact is sometimes sheared almost beyond recognition.

Waite Rhyolite

Microscopic and quantitative X-ray diffraction examination of the rhyolite has revealed the presence of sodic plagioclase feldspar both as phenocrysts and cryptocrystaline intergrowths with quartz in the matrix (A. R. Graham, personal communication). Flakes of chlorite and sericite, specks of magnetite and mossics of quartz are common.

Far from being massive and uniform, the rhyolite exhibits a host of internal structures (Plates 1 to 5; Appendix H). The most common structure is a vague breccia, often barely discernible on a fresh surface, consisting of small (1" diameter) gray fragments in a slightly lighter matrix (Plates 8, 9; Appendix H). Other light colored, large, often angular fragments or blocks of highly vesicular rhyolite occur in a matrix which varies from a vague breccia through uniform and massive gray

rhyolite to a relatively soft chloritic and sericitic rock (Plate 5; Appendix H). Another uncommon but interesting variety of breccia consists of small (about 1"), chloritic and sericitic, often deformed (stretched, flattened) fragments in a very dense, laminated, cream, lavender, pink or gray, cherty-appearing matrix (Plate 2; Plate 9, figs. 2a, 2b; Appendix H). Other less common breccias occur, as well as much rhyolite which appears massive or, at the most, vesicular or laminated. It is probable that almost all of the Waite rhyolite is a complex of very fine to coarse tuff, welded and nonwelded, which was deposited, at least for the most part, subaqueously. This latter condition is suggested by the presence of bedded, cherty tuff at the top and, in some places, within the rhyolite.

Like other rhyolite units in the Noranda area the Waite rhyolite is thin relative to its areal extent and like other deposits of the area the Lake Dufault ore body is located on a thickening of the rhyolite.

Sericite has formed along fractures, in spots, or over large areas throughout the rhyolite. Quartz, calcite, chlorite and epidote often occur in the rhyolite in spots and fractures. The borders of magnetite-ilmenite grains are altered to leucoxene.

The only large scale local alteration in the rhyolite is below the massive ore body in a fractured and mineralized zone (Plate 6; Appendix H). Here, the rhyolite has been completely reconstituted and recrystallized. A relatively narrow zone, present along the base of the massive ore, consists of pyrite, sphalerite, chlorite, sericite, garnet and magnetite in spots and fractures in the rhyolite. Below this zone and extending through it in places, is an extensive zone

characterized by the presence of chalcopyrite, which is disseminated and in fractures. The amount of chalcopyrite in this zone increases upward as a function of the size and frequency of fractures. A major feature of this alteration zone is the development of a mosaic of clear, strain-free quartz grains. Sericite is present as flakes and chlorite appears as rosettes as well as scattered flakes. Cordierite, anthophyllite, biotite, chlorite and garnet have been found in this zone. It appears that the copper-rich zone is later than and cuts across the narrower pyrite-sphalerite-magnetite-chlorite zone beneath massive ore.

Apart from the fractured and altered zones beneath the ore no locallized alteration zone was seen in the rhyolite. Pyrite and pyrrhotite are disseminated throughout the rhyolite in scattered specks and are very rarely present in veinlet form in spite of the fact that the rhyolite is well fractured in many places. Quartz-carbonate-chlorite veinlets and fractures containing essentially no sulphides often are found in rhyolite containing relatively large amounts of disseminated sulphides (Plate 9, fig. 1a; Appendix H). Sulphides occurring in distinctly fragmental rhyclite are almost entirely confined to the matrix. Three of four polished sections of pyroclastic rhyolite, examined with a reflecting microscope, revealed very fine disseminated sphalerite with coarser pyrite and pyrihotite only in the matrix of the breccia. The fourth section revealed no sphalerite but a few grains of pyrite and pyrrhotite were noticed in the breccia matrix. Outside of the ore zone, sphalerite is more noticeable and chalcopyrite is less noticeable in the rhyolite than they were in the overlying andesite.

Magnetite is a conspicuous accessory mineral.

STRUCTURE

The most evident structural features in the Norbec area are faults, lineaments, thickened rhyolite and the Newbec breccia. Evidence of folding has not been mapped although Wilson (1941) placed fold axies to the west and northwest of the Norbec area. The contact between Amulet andesite and Waite rhyolite, often the locus of bedded, cherty tuff, is commonly referred to as the "main contact."

Faulting

The Norbec area is relatively undeformed. The main epigenetic structures are large northwest trending, steep faults now occupied by diorite dikes. Movement along these faults has formed three fault blocks and caused the raising of the volcanic rock in the eastern part of the Norbec area. The "main contact" and bedding in tuff in each of these fault blocks dip moderately (30°) to the east. Northwest and northeast trending faults are found in the Norbec area. Some of the faults have shallow dips and one offsets part of the Lake Dufault massive sulphide ore body (fig. 2). Some movement may have occurred along the "main contact."

Lineaments and Domes

Northeast (N. 70° E.) and northwest (N. 20° W.) trending air photo lineaments, some of which are known to be expressions of fault zones, cut across the Norbec area. The Lake Dufault ore body occurs on

a broadly thickened section of rhyolite near the intersection of two of these lineaments (staff, Lake Dufault Mines Ltd.).

Newbec Breccia

A circular exposure of breccia, the Newbec breccia is located in the southeast corner of the Norbec area. Wilson (1941) compares this breccia to breccia pipes in Leadville, Colorado, which are said to be of volcanic origin.

GEOCHEMISTRY OF THE HOST ROCKS

GENERAL APPROACH

The "main contact" in the Norbec area of the Lake Dufault
Mines Ltd. property is a moderately dipping plane pierced by several
vertical diamond drill holes. Usually four samples from andesite
immediately above the contact and three to four samples from rhyolite
immediately below the contact were taken from the core of each drill
hole available for sampling. Each sample represents approximately
twenty five feet of drill core. Cherty, well-bedded tuff was also
sampled where it occurred in drill core at the main contact. All the
samples were analysed for as many elements as time and opportunity
permitted.

SAMPLING PROCEDURE

Rhyolite and Andesite

The object of sampling was to obtain a sample representing twenty five feet of rhyolite or ancesite in a drill core. This was accomplished by a systematic sampling procedure which was suggested by Dr. A. R. Graham of Falconbridge Nickel Mines Limited. Two-inch long pieces of core were collected at twenty-inch intervals over twenty five feet of core, giving fifteen pieces in one sample bag. Each sample weighed approximately three pounds. Dikes were excluded from the sample

and where necessary the length of the sampled core was adjusted slightly so that the sample would at least appear to be homogeneous.

Tuff

Because of the heterogeneity of the bedded, cherty tuff it was necessary to split in half the complete diamond drill core and to use one of the halves as a sample. Only the bedded variety of tuff was sampled, and, for each drill hole, one sample represents the total amount at the main contact (one exception, three samples in DDH #121).

SAMPLE REDUCTION

The entire content of one sample bag was crushed to one quarter inch size in a steel plated jaw crusher. This material was split down to a large handful using a Jones splitter. The sample was then brought to minus one hundred mesh using a Braun disk pulverizer with ceramic plates. The fine product was split down using a Jones splitter so that a portion would half fill a small vial. All of the crushed and pulverized sample was saved. 1

SAMPLE PREPARATION (BEYOND REDUCTION)

The rest of the sample preparation involved weighing, mixing, fusing, crushing, grinding and filling of sample containers depending on which of the four analytical procedures was being followed. 2

^{1.} Appendix G, figure G1.

^{2.} Appendix G, figures G1, G2, G3, G4.

During the course of the three years of analytical work some of the techniques of mixing and crushing were changed and many different workers were involved in sample preparation.

ANALYTICAL PROCEDURES

Four analytical procedures were employed using a G.E. model XRD 3 X-ray emission spectrograph and a JACO 3.4 meter plane grating spectrograph.

The SiO₂, Al₂O₃, TiO₂, MnO, Total Fe as Fe₂O₃, K₂O and CaO contents of the samples were determined by an X-ray fluorescence method. The powdered rock sample was mixed with a flux (lithium tetraborate), fused, pulverized and packed by hand into a plastic sample holder. Natural materials analysed by the National Bureau of Standards (N.B.S.) and the Geological Survey of Canada (G.S.C.) were used as standards.

Na₂O and MgO in the samples were determined with an optical spectrograph using material prepared for the X-ray method but mixed with graphite.² The lithium of the flux acted as an internal standard.

The N.B.S. and G.S.C. materials were used as standards.

The Co, Cr, Cu, Ga, Ni, Sr, Y and Zr contents of the samples were determined by an optical spectrographic method. The powdered rock sample was mixed with a flux (lithium carbonate), fused, pulverized, combined with graphite and packed in cupped graphite electrodes. Indium was added as an internal standard. Salts and oxides of the metals were added to an artificial base for standards.

Appendix G, figure G1.

^{2.} Appendix G, figures G1, G2, table G2.

^{3.} Appendix G, figure G3, table G2.

The Ag, Bi, Cd, Ge, Pb, Sn, and Zn contents of the samples were determined by an optical spectrographic method. The already powdered sample was ground to an extra fineness (about two hundred mesh), mixed with graphite and packed in cupped graphite electrodes. Antimony was added as an internal standard and the sample in the electrode was covered by a graphite electrode cap. Salts and oxides of the metals were added to an artificial base for standards.

All analyses using the optical spectrograph were done in duplicate.

^{1.} Appendix G, figure G4, table G2.

Table 1
Sample Preparation and Analytical Error

		Andesite #970			Rhyolite #965	•
	N	×± CA	<u>R</u>	N	x± CA	<u>R</u>
SiO _{2%}	18	51.2 ± 8.6%	1.059	16	74.3 ± 5.0%	.9178
Ti02%	18	1.26 ± 5.9%	. 4471	16	0.26 ± 9.1%	4755
A1 203%	18	$20.0 \pm 7.7\%$. 7912	16	13.9±14.2%	1.350
Fe ₂ 03%	17	8.7±3.0%	.1978	16	5.4 ± 2.3%	.0861
Mn 0%	.18	0.11 ± 5.6%	. 305	16	0.49± 3.0%	.0497
Ca0%	18	7.3 ± 2.9%	.1938	16	3.4± 4.0%	.1326
K20%	18	0.47±14.9%	. 4562	16	1.40±6.6%	.2008
Mg0%	17	6.2±11.1%	•5529	16	1.20 ± 9.4%	.1132
Na20%	- 17	3.5 ± 11.1%	.5528	16	1.35 ± 20.9%	.9002
Ag ppm	17	0.34±53.7%	1.119	19	1.05 ± 24.8%	. 41 78
Co ppm	17	72 ± 13.9%	. 4805	15	5.7±20.9%	
Cr ppm	18	227 ± 13.6%	.4619	8	51 ± 27.5%	1.34
Cu ppm	18	72 ± 30.5%	.3593	16	312 ± 20.4%	.2915
Ga ppm	18	26 ± 10.5%	•6322	16	30±9.3%	.6026
Ni ppm	18	167± 30.3%	.7218	15	3.6 ± 78.7%	.8420
Pb ppm	16	1.4 ± 66.4%	• 7 5 5 4	19	15.6±16.4%	.1683
Sn ppm	16	5.2± 63.3%	.997 8	19	1.3 ± 27.8%	. 31 42
Sr ppm	16	260 ± 30.2%	•9265	16	414±25.3%	.9843
Y ppm	18	20 ± 17.1%	.6528	16	62±19.3%	. 8961
Zn ppm	18	227±19.5%	• 4579	19	357±10.3%	.1167
Zr ppm	18	176 ± 17.5%	•9994	16	416 = 27.9%	.7564
		/•				

$$C_A = \frac{\sigma}{R} \times 100$$

$$R = \frac{C_A \text{ (analytical)}}{C \text{ (total)}}$$

CA = relative deviation (analytical)

or = sample standard deviation

x = mean of N replicate analyses

N = number of analyses

R = ratio of analytical deviation to total sample deviation

C(total) = sample deviation calculated from all samples except

those near the ore body

REPLICATE ANALYSES: ESTIMATION OF ANALYTICAL AND SAMPLE PREPARATION ERROR

One sample of andesite (#970) and one sample of rhyolite (#965) were analysed repeatedly throughout the entire period of analytical work. Variations in the results are caused by errors in the analytical and sample preparation stages.

The sample relative deviation (C_A) due to analytical and sample preparation errors for each of the two replicate samples was calculated according to formula (1):

$$C_{A} = \frac{\sigma}{x} \times 100 \qquad \sigma = \frac{1}{N} \sqrt{N \sum_{i=1}^{N} x_{i}^{2} - (\sum_{i=1}^{N} x_{i})^{2}} \quad (1)$$

Where o = sample standard deviation 1

N = number of analyses per sample

 $x_i = an individual analysis$

 \bar{x} = arithmetic mean of N analyses

CA = relative deviation (analytical)

The results are given in Table 1.

Presentation of C_A (relative deviation attributable to analytical error) helps to point out that any one analysis represents a range of possible values. In this report, the value in question is actually the average of N analyses $(\bar{\mathbf{x}})$, and the analytical error in the average² is C_A/\sqrt{N} . For example if there are four individual analyses in one drill

best estimate of 1. To convert sample standard deviation to population standard deviation multiply by N/N-1.

^{2.} C_A/\sqrt{N} might be called the "relative" error of the mean.

hole, sixty-eight percent of the time the real value of their average should be within the range represented by $x \pm C_A/\sqrt{4}$ or $x \pm C_A/2$. If there were no real variations in the concentration of an element in four samples from a drill hole the analytical error would result in a mean value having a 95% chance of being in the range $x \pm C_A$. When C_A is large (that is, greater than 15 or 20%) this expression of confidence cannot be used. For discussion of this point see Ahrens and Taylor (1961, p.116-118).

CA was calculated from the repeated analysis of one sample and cannot be strictly applied to all samples. For example, CA may vary as a function of the level of measurements. Also the 95% limits are based on a normal distribution. It is therefore an approximate figure and as such is only a guide (the best one we have) to establish confidence in the analytical results.

The ratio, R, between the analytical relative deviation and the total relative deviation indicates approximately how much of the total variation in the samples can be attributed to errors in sample preparation and analysis. Since C, the total relative deviation, is not accurate for all but normal, unimodal distributions, R is only an approximate figure for many elements. Nevertheless, only greatly anomalous values would seem significant for an element having R close to unity while analytical error would seem relatively insignificant when dealing with anomalous values for an element having R less than 0.2.

^{1.} e.g. if x = 10.0% and $C_A = 5\%$, $x + C_A = 10.0\% + \frac{5}{\sqrt{100}} \times 10.0\% = 10.5\%$.

^{2.} All samples close to the ore body have been removed from the calculations (i.e., those from diamond drill holes 135, 151, 163, 132, 133, 142, 149, 127, 128, 129, 125).

Table 2

Analysis of Errors of Manipulation

	020	Tio2	Fe ₂ 0 ₃	MnO	K20
_ 2					
ca ² Analytical	.00864	.00264	.00291	.00649	.00676
2			•		
cp ² Sample Preparation	.00559	.00042	.00099	.00483	.00170
er ² Sample Reduction	.00067	.00114	.00008	.00154	.00035
c 2					
Total	.0149	.00420	.00398	.01286	.00881

1. relative variance; formula (2)

ERROR IN SAMPLE REDUCTION

In order to estimate the error created by pulverizing and splitting and to compare this error with the error in sample preparation (beyond reduction), a hierarchical model of replicate analysis was used.

The relative variance (c^2 ; formula 2) of a small number of replicate analyses was calculated at three levels: (1) analytical, (2) sample preparation, and (5) sample reduction (Shaw, 1961).

$$c^{2} = \frac{N^{2}}{N-1} \left[\sum_{i=1}^{N} x_{i}^{2} + \left(\sum_{i=1}^{N} x_{i} \right)^{2} \right] \left(\sum_{i=1}^{N} x_{i} \right)^{2}$$
 (2)

c² = relative variance
x_i = individual analysis
N = number of analyses

Since the factor $\frac{N}{N-1}$ is included in formula (2) c^2 is an

"unbiased estimate" of the population relative variance. The total relative variance is the sum of the relative variances caused by analytical error (c_a^2) , sample preparation error (c_p^2) and sample reduction error (c_r^2) .

$$c^2 = c_a^2 + c_p^2 + c_r^2$$

Table 2 gives the results of this analysis and shows how little of the total relative variance is attributable to errors in sample reduction.

^{1.} For definitions see p.23

METHOD OF PRESENTATION OF RESULTS

Data Charts

All the analytical data are summarized on charts (charts 1 to 42; Appendix B). By means of three graphs, a single chart shows for any one element and rock type: (1) the average value, \bar{x} , of the individual analyses for each drill hole, (2) the sample standard deviation of those values $(\sigma)^1$, and (3) the individual values themselves (x). Thus much pertinent information about the distribution of one element in rhyolite or andesite can be obtained by looking at one chart. The individual analyses are plotted from left to right in the order in which the samples were taken. Thus, for andesite, the last sample in a drill hole is closest to the main contact. For rhyolite, the first sample is closest to the main contact. Note that the numbers on the vertical scale of the graphs of average and individual values do not always start at zero. This fact should be kept in mind when comparing the values appearing on one graph with those on another graph.

The horizontal lines on these graphs are an attempt to designate background and anomalous values. The values which the horizontal lines represent were chosen after a study of the form of the frequency distribution of the element in question in andesite or rhyolite.

^{1.} The symbol o is used to designate sample standard deviation in the ASTM Manual on Quality Control of Materials referred to in this report. A more commonly used symbol is "s."

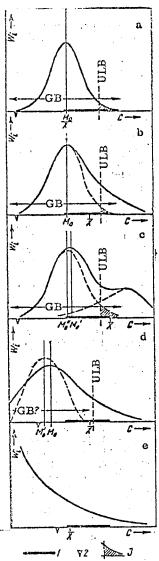


Fig. 1. Types of frequency distributions of concentrations of elements in geochemical systems

C-concentration of element; W₁-frequency; X-arithmetic mean of concentration; M₀-mode; ULB-upper limit of background concentration; GB-geochemical background; 1-concentration range which may contain average content of element; 2-sensitivity of analytical method; 3-region of large absolute deviations

Fig. 3 From Smirnov, 1963

Frequency Distribution Diagrams

All the analytical data are summarized on columnar frequency distribution diagrams (histograms) (Appendix A). These diagrams provide the easiest and clearest way to illustrate the behavior of an element in a group of samples or, as in this case, in rhyolite or andesite.

The anomalous values of an element were separated from background values on the basis of the form of its frequency distribution. The frequency distributions obtained can be placed in five groups:

I - Unimodal (one peak) symmetrical curves (fig. 3a); II - Unimodal asymmetrical curves with positive (or negative) skewness (fig. 3b);

III - Bimodal or multimodal curves (fig. 3c); IV - Symmetrical or asymmetrical curves with some samples containing less than detectable amounts of an element (a frequency much greater than zero in the lowest valued class interval of the histogram) (fig. 3d); and V - the so-called J-shaped curves without a peak (fig. 3e) (Smirnov, 1963).

Elements which show a frequency distribution close to being unimodal and symmetrical (Type I) are referred to as being <u>normally</u> <u>distributed</u>.

Type I. Unimodal symmetrical curves (fig. 3a)

When values for an element in a system (rhyolite or andesite) are close to being normally distributed, anomalous values are chosen by a method outlined in the ASTM Manual on Quality Control of Materials, 1951. Part 3, Section 9b, p. 61.

^{1.} A curve is drawn by smoothing a histogram.

Formulae for Data Chart Lines (for Normal Distributions)

i -

Where: \overline{x} = the grand average of observed values of x for all samples 2

o = the weighted average drill hole standard deviation
N = number of samples per drill hole

 $c_2 = 0.5642$ for N = 2 $c_2 = 0.7236$ for N = 3 $c_2 = 0.7979$ for N = 4 $c_2 = 0.8407$ for N = 5

 c_2 is a factor that is a function of N and expresses the ratio between the expected value of $\overline{\sigma}$ for a large number of samples of N observed values each and the true standard deviation (σ) of the universe sampled.

Values of $c_2 = \overline{\sigma}$ approach a maximum of 1 as the

number of observed values (N) increases. The expected value of $\overline{\sigma}$ is based on a <u>normal</u> distribution.

^{1.} ASTM Manual on Quality Control of Materials, 1951, Part 3, Section 9, p. 61.

^{2.} except those from drill holes close to the orebody

For normally distributed elements, the graphs for averages (\bar{x}) and individuals (x) are marked with a straight dashed line (central line) and two symmetrical solid lines (limits). The central line marks the grand average content of an element in the rock (\bar{x}) and all values outside the limiting lines are anomalous. Figure 4 gives the formulae which were used to calculate the limiting lines. 2

The spread of the background limiting lines for averages and standard deviations should depend on the number of samples per diamond drill hole (N). The limiting lines for averages are approximately 3.8 standard deviations if N=2, 2.4 standard deviations if N=3 and 1.9 standard deviations if N=4. The limiting line for σ does not vary significantly with the number of samples (N). The limiting line on the graph of individuals is at 3σ .

All the formulae for calculating limits are based on the choice of three o's as a limit. This means that, assuming the background values have a normal distribution, 0.3% of the anomalous values are actually background.

Type II - Unimodal asymmetrical curves with positive skewness (fig. 3b)

Background and anomalous values forming distributions of Types II, III, IV, and V were separated using judgment only.

^{1.} ULB = upper limit of background; LLB = lower limit of background.

Values for samples close to the known ore body were not included in the calculations because of the good chance these values could belong to another distribution. If this proved to be the case these values would show up as anomalous on the graph. If not, it should be clear that no real harm was done by removing these samples. The samples removed are from drill holes 125, 127, 128, 129, 132, 133, 135, 142, 149, 151, and 163.

In this case, Type II, we can make an assumption to aid in our choice of a limit to background. The assumption is this: background values form a normal distribution, therefore the left branch of a positively skewed, unimodal, asymmetrical curve corresponds to the left branch of the symmetrical curve of the background values (Smirnov, 1963). It is then possible to sketch in the right branch of the symmetrical background curve. All values above the highest value falling on the normal curve would then be anomalous.

Because of the uncertainty involved in designating anomalous values in distributions of Types II, III, IV, and V the range of anomalous values was split up into two or three groups as shown on the histograms. Range a to b could be called possibly or probably anomalous while ranges greater than b might be called anomalous.

Type III - Bimodal or multimodal curves (fig. 3c)

Anomalous values were distinguished in multimodal distributions by assuming, as was done for Type II distributions, that the highest peak represented a symmetrical background distribution. Other lower peaks represent distributions of anomalous values.

Type IV - Asymmetrical curves with a frequency greater than zero in the lowest valued cell of the histogram (fig. 3d)

Type V - J-shaped distributions (fig. 3e)

Choosing anomalous values for these two classes of frequency distributions is strictly guess-work. One can only designate the most commonly occurring values as background and pick an upper limit to the background using good judgment.

As mentioned previously, the anomalous values have been classified further into two or three ranges marked a-b, b-c,>c on the histograms presented here.

Tables Summarizing the Vertical Distribution of Elements (Appendix E)

If the top half of the samples of rhyolite from a drill hole shows higher concentrations of an element than the bottom half, this drill hole was included in the first column of Tables E-1 and E-2; if not, a record was made in the second column. Groups of samples from drill holes were split into categories of "two samples only" and "three or more samples."

The same was done for the bottom of the andesite (Tables E-3, E-4).

METHOD OF PRESENTATION OF RESULTS FOR SAMPLES OF TUFF

The results for tuff were treated differently because of the small number of samples. In Tables C-1 through C-6 (Appendix C) the analyses of tuff samples are arranged in columns in order of increasing values. The middlemost value of each column is the <u>median</u>. Samples of tuff from holes near the known ore body are noted with an asterisk.

If the chemical compositions of these samples were not related to the presence of ore, the values would occur ideally in equal numbers on either side of the median. If, then, the ratio (r) of the number of "near ore body" samples in the 0 - 50% range to the number of them in the

Diamond drill holes cutting tuff near the ore body: 163, 135, 190, 139, 181, 159, 164, 144, 174.

^{2.} r = # in 0 - 50% range # in 50 - 100% range

50 - 100% range is 4/5 or 5/4 no significant variation in the tuff is demonstrated. Only the values for those elements which look interesting (r = 3/6 or better) are plotted on a plan of diamond drill holes (figs. 38, 39, 40).

Analytical Data for Standard Samples, G-1, W-1, CAAS Syenite, CAAS Sulphide

Element	G-1		พ-1		CAAS s	yenite	CAAS sulphide		
•	Fleischer ¹	Sakrison	Fleicher	Sakrison	Webber ²	Sakrison	Webber	Sakrison)
S102 A1 ₂ 03	72.64% 14.04	78.3% 25.4	52.64% 14.85	61.3% 18.2	59.45% 9.58	64.3% 15.4	34.52% 9.46	44.7% 15.8	
Fe ₂ 03 Fe0	0.87 0.98	1.783	1.41 8.74	10.8 ³	2.27 5.44	7.8 ³	~-	25.0 ³	
Fe	·	PP PA		•••	==		22.87	~-	
MgO	0,41	0.38	6.62	7.8	4.07	4.5	3.93	4.41	
CaO	1.39	1.36	10.96	10.7	10.32	9.1	3,96	3.9	
Na ₂ 0	3.32	3.52	2.07	2.30	3.24	3.8 -	1.03	2.65	
K20	5.45	5.32	0.64	0.65	2.75	2.41	0.63	0.72	
TiO ₂	0.26	0.28	1.07	1.06	0.49	0.45	0.81	0.86	
MnO	0.03	0.03	0.16	0.17	0.40	0.38	0.11	0.14	
Ag	0.04ppm	<0.3ppm	0.06ppm	0.26ppm	1.6ppm	0.5ppm	3.9ppm	0•5ppi	m
Co	2.3	1.0	52	61	19	29	546	510	
Cr	22	30	120	242	58	125	368		
Cu	13	15	110	119	25	37	B291	~~~	
Ca	18	30	16	22	18	31	13		
Ni	1-2	2	7 8	7 8	42	68	13103		
PЬ	49	44	8	17	516	485	248	265	
Sn	4		3	11	12	8	11	4	
Sr	250	534	180	252	320		110	168	⊣
Υ	13	12	25	20	447	670	. 21	15	Table
Zn	45	200	82	192	200	192	298	1640	6
Zr	210	427	100	1 43	3048		108	253	C1

fleicher (1965)
 Webber (1965)
 Fe (total) as Fe₂0₃

ACCURACY OF RESULTS

Each of the standard samples, G-1, W-1, CAAS syenite and CAAS sulphide, was analysed once. Table 3 presents the results of these analyses along with the "preferred values" for the elements from many analyses of the standard samples as given by Fleicher (1965) and Webber (1965).

Large, systematically high discrepancies are noted for SiO₂, Al₂O₃, Cr, Ga, Sr, Zn and Zr. Conclusions which must be tempered by consideration of these discrepancies are noted in the forthcoming sections.

Analyses for some of the above-mentioned elements have relatively low precisions (Table 1), a fact which may account for a part of the inaccuracies. The background value given for each element in andesite and rhyolite in this work is the most common value (statistical mode) of more than two hundred analyses and thus is free from error caused by low precision.

Many of the above-mentioned elements are present in the standard samples in amounts near the lower detection limits of the analytical procedures used in this work. This is true of chromium in G-1 and CAAS syenite, strontium in W-1 and CAAS sulphide and zinc in G-1 and W-1. This may account for some of the inaccuracies noted for these elements.

Table 4
MAJOR ELEMENT CONTENT OF THE HOST ROCKS

	AMULET AND Mode (%)	DESITE mean (%)	WAITE RH mode (%)	Mean (%)	TU median (%)	range (%)
SiO ₂	52.5	52.4	77.0	75.5	61.0	15.4
A1 ₂ 0 ₃	16.4	18.3	13.6	14.2	15 . B	8.4
Fe as Fe ₂ 0 ₃	9.4	9.3	4.5	4.9	13.2	15.8
Mg8	5.2	4.9	0.5	1.1	1.5	2.6
CaO	10.3	8.9	2.5	2.6	5.5	8.0
Na ₂ 0	2.9	2.8	2.7	3.1	2.8	2.6
K ₂ 0	0.5	1.1	1.3	1.5	1.7	2.0
TiO ₂	1.11	1.20	0.23	0.32	0.68	0.68
MnO	0.16	0.19	0.10	0.16	0.15	0.14

	N	S10 ₂	A1203	FeC	Fe ₂ 0 ₃	MgO	Ca0	Na ₂ 0	к ₂ 0	TiO ₂	MnO	Rittman's ¹ o (1962)
1	12	48.52	12.90	11.31	3.53	6.25	10.11	1.94	0.40	1.42	0.27	1.0
2	3	49.1	15.3	13.7		6.7	11.2	2.0	0.2	1.5	0.23	• B
3	4	49.42	14.64	9.12	2.11	6.80	10.17	2.32	0.30	0.94	0.22	1.07
	4	50.1	14.1	13.6		7.6	9.1	3,5	0.1	1.8	0.27	1.8
. 4 5	1	51.75	17.47	9.16	2.96	9.40	9.16	5.20	1.37	1.15	ቹ	.17
5 6	1	53.86	14.56	9.16	1.36	7.44	2.49	1.38	1.01	1.04	0.10	•5
7	1	53.91	14.72	7.12	2.53	5.00	6.50	3.40	0.26	1.48	0.16	1.2
8	1	54.02	15.67	8.46	2.38	4.12	7.14	4.84	0.64	1 ₀ 63	0.01	2.7
9	3	54.4	17.1	9.9		4.0	9.1	3.5	0.3	1.6	0.17	1.3
10	1	55.35	15.0	7.42	1.55	5.03	5.72	3,82	0.85	1.18	0.16	1.8
11	6	56.5	15.6	9.7		6.3	5.6	4.4	0.3	1.3	0.20	1.6
12	1	57 . 87	14.53	6.16	2.04	4.62	6.12	3.23	0.69	1.11	0.16	1.0
13	8	58.2	17.0	7.4		3.8	7.8	4.0	$0.\epsilon$	1.3	0.17	1.4
14	1	58.3	16.63	5.86	1.42	4.18	8.31	3.08	0.62	1.12	0.16	1.0
15	1	59.56	14.18	5.06	1.85	4.27	4.77	3.11	1.66	0.93	0.17	1.4
16	1	60.28	15.08	7.42	1.60	2.79	3.77	5.52	0.21	1.60	0.12	1.9
17	1.	62.66	13.80	6.90	1.73	2.31	2.48	3.84	1.29	0.85	0.14	1.3
18	4	66.0	16.7	4.0		2.4	4.B	4.2	1.4	0.5	0.07	1.4
19	1	68.72	14.33	2.32	0.99	0.83	2.64	6.80	0.55	0.70	0.08	2.1
20	1	72.72	11.00	2.71	1.22	0.30	2.46	2.10	2.52	0.24	0.09	• 7

^{1.} $\sigma = \frac{(Na_20 + K_20)^2}{5i0_2 - 43}$; a σ of less than 1.8 demonstrates a strong to extreme calc-alkaline affinity.

^{2.} N = number of analyses.

^{3.} Except for Roscoe (1965), the analyses originally were presented along with analyses for water and other minor constituents.

DURITSHED ANALYSES (nF	NORANDA	AREA	ROCKS
----------------------	----	---------	------	-------

	N	si0 ₂	A1203	Fe0	Fe ₂ 0 ₃	MgO	CaO	Na ₂ 0	K ₂ 0	TiO ₂	ทีกปี	Rittman's ¹ o (1962)
21 22 23 24 25 26 27 28 29 30 31 32	1 1 1 1 1 3 1 1 1 1 1 6	72.90 73.79 74.76 74.60 74.89 75.3 76.00 76.42 78.7 78.78 85.11 60.5	10.90 11.83 11.50 11.43 13.27 12.5 11.27 11.08 11.5 11.21 7.97	2.86 4.18 2.05 1.54 3.65 2.5 1.73 1.54 2.2 2.77 0.98 4.5	2.51 0.72 1.77 1.84 0.36 0.67 1.22	1.02 0.62 0.77 0.66 0.40 1.3 0.73 0.60 0.5 1.33 0.05 2.1	1.64 2.21 0.81 1.42 1.15 1.6 1.36 1.61 1.7 0.21 0.77	3.83 3.94 4.89 3.87 4.25 5.4 4.71 4.37 3.4 3.13 2.68 5.1	1.09 0.85 1.18 1.72 0.37 1.0 0.50 1.15 1.6 0.80 0.89 6.8	0.25 0.39 0.20 0.25 0.22 0.52 0.19 0.2 0.28 0.25 0.6	0.04 0.07 0.03 0.03 TR 0.12 0.06 0.04 0.07 TR TR	.8 .75 1.2 1.0 .67 1.3 .9 .9 .7 .43 .30 2.5

^{1.} $\sigma = \frac{\left(\text{Na}_2\text{O} + \text{K}_2\text{O}\right)^2}{\text{SiO}_2 - 43}$; a σ of less than 1.8 demonstrates a strong to extreme calc-alkaline affinity.

PUBLISHED ANALYSES OF NORANDA AREA ROCKS

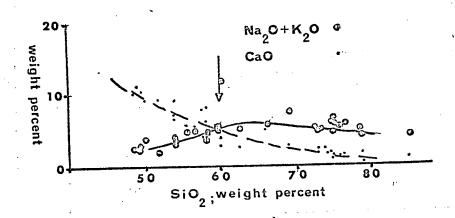
- "Flat" diorite, Waite Lake Area, Duprat Twp., Lickus (1965).
- 2. Gabbro in Waite and Amulet Andesites, Roscoe (1965).
- "Steep" diorite, Waite Lake Area, Dufresnoy Twp., Lickus (1965).
- 4. Basalts, pre-Blake River Group, Kirkland Lake and Noranda Areas, Roscoe (1965).
- 5. Pyroxene gabbro, Dufresnoy Twp., Wilson (1941).
- 6. Waite andesite, Waite Lake Area, Duprat Twp., Lickus (1965).
- "Metadiabase," Waite Lake Area, Dufresnoy Twp., Lickus (1965).
- 8. Massive andesite, Dufresnoy Twp., Wilson (1941).
- 9. Diorite in Waite and Amulet andesite, Roscoe (1965).
- Waite andesite, Waite Lake Area, Dufresnoy Twp., Lickus (1965).
- 11. Waite andesite, Blake River Group, Vauze Drill hole, Roscoe (1965).
- 12. Amulet andesite, Waite Lake Area, Duprat Twp., Lickus (1965).
- 13. Amulet andesite, Vauze drill hole, Roscoe (1965).
- 14. Amulet andesite, Waite Lake Area, Dufresnoy Twp., Lickus (1965).
- 15. "Metadiabase," Waite Lake Area, Dufresnoy Twp., Lickus (1965).
- 16. Unaltered andesite, Amulet Mine, Riddell (1952).
- 17. "Transitional" dacite, Waite Lake Area, Dufresnoy Twp., Lickus (1965).
- 18. Lake Dufault granodiorite, Roscoe (1965).
- 19. Rhyolite (trachyte), Amulet Mine, Riddell (1952).
- 20. Waite rhyolite, Waite Lake Area, Dufresnoy Twp., Lickus (1965).
- 21. Waite rhyolite, Vauze Mine breccia member, Dufresnoy Twp.. Lickus (1965).
- 22. Flavrian granite, Duprat Twp., Wilson (1941).
- 23. Waite Rhyolite, lower member, Waite Lake Area, Dufresnoy Twp., Lickus (1965).
- Waite rhyolite, upper member, Waite Lake Area,
 Dufresnoy Twp., Lickus (1965).

Table 5 (cont'd)

PUBLISHED ANALYSES OF NORANDA AREA ROCKS

- 25. Powell granite, Rouyn Twp., Wilson (1941).
- 26. Flavrian Lake sodic granite, Roscoe (1965).
- 27. Unaltered rhyolite, Amulat Mine, Riddell (1952).
- 28. Waite rhyolite, Waite Lake Area, Dufresnoy Twp., Lickus (1965).
- 29. Waite rhyolite, Roscoe (1965).
- 30. Rhyolite, Duprat Twp., Wilson (1941).
- 31. Spherulitic, micropegmatitic rhyolite, Duprat Twp., Wilson (1941).
- 32. Syenites, Matachewan (Ontario), Duparquet, Dasserat, Noranda, Roscoe (1965).

figure 5. Peacock's alkali : lime index for rocks from the Noranda area.



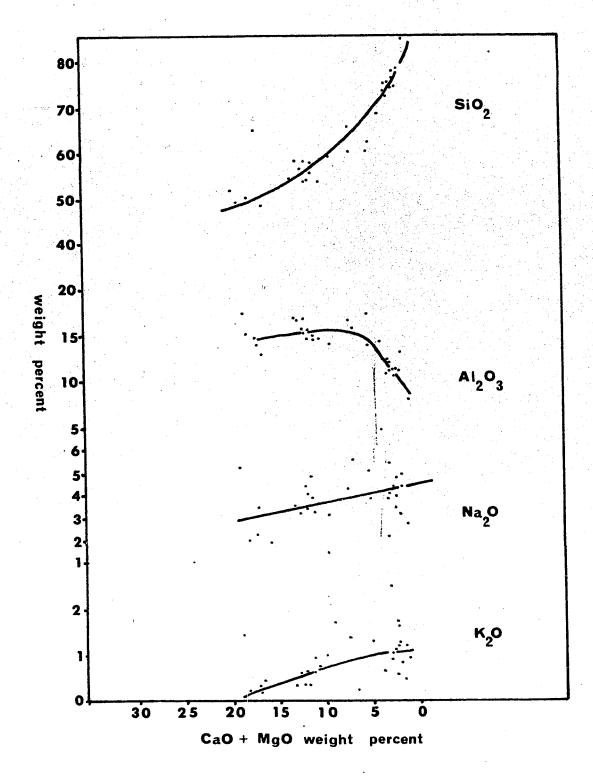


Fig. 6. Variation Diagrams for Rocks of the Noranda Area.

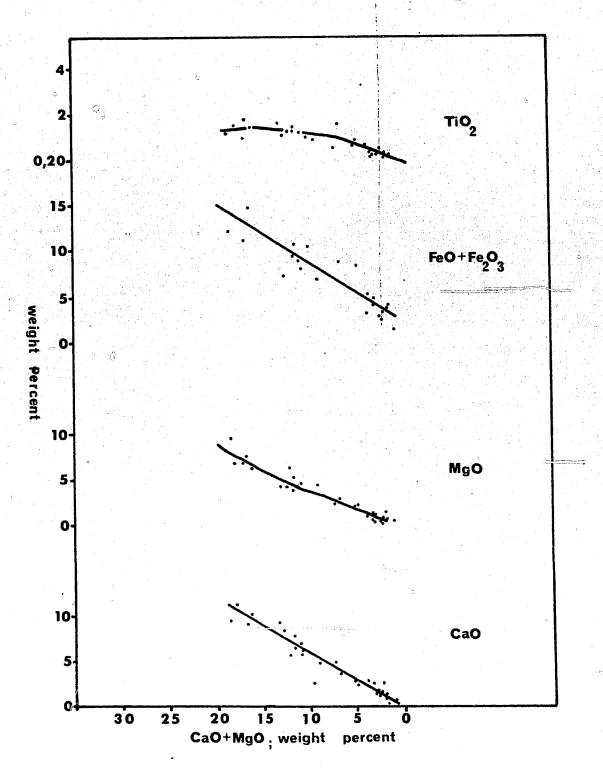


Fig. 6 cont'd. Variation Diagrams for Rocks of the Noranda Area.

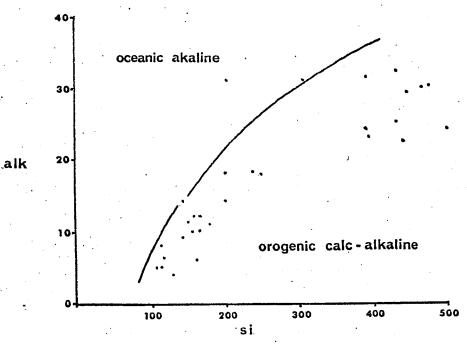


Fig. 7 Plot of Niggli alk-si values for Rocks of the Noranda Area.

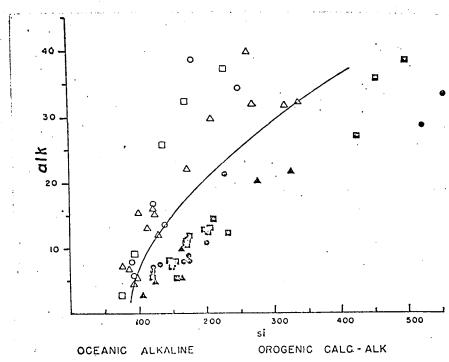


Fig. 5. Plot of Niggli alk si values for control groups of lavas, Oceanic alkaline: \triangle Truk, \square Tahiti, O Kerguelen; orogenic calc alk: \blacktriangle Fuji, \square Izu, O Mariana.

Fig. 8 Figure 5 from Wilson, H. D. B., 1965.

BACKGROUND VALUES FOR ELEMENTS IN THE HOST ROCKS AND THEIR SIGNIFICANCE

MAJOR ELEMENTS

Background values (statistical modes) for major elements in Amulet andesite, Waite rhyolite and tuff obtained in this work are presented in table 4. Analyses of Noranda area rocks were collected from the literature and are presented in table 5.

Variation diagrams for the Noranda rocks are presented in fig. 6. The Niggli values for "alk," plotted against "si," are presented in fig. 7. The resulting graph can be compared with a similar plot of oceanic alkaline and orogenic calc-alkaline rocks (fig. 8) (Wilson, H.D.B., 1965). Peacock's alkali-lime index is given in fig. 5 and Rittman's (1962, p. 110) sigma values are presented in table 5. These parameters show that rocks from the Noranda area have no alkaline affinity and that, with the exception of minor syenite, all the rocks belong to a single, highly differentiated, calc-alkaline volcanic suite. The distinction between the Noranda volcanic suite and the spilite suite is less clear. The average spilite is a basalt with high contents of soda (Na₂0 - 4.9%) and titania (TiO₂ - 3.2%) (Turner and Verhoogen, 1960, p. 271). Roscoe's average basalt, from the Kirkland Lake and Noranda areas, contains much less of these constituents (Roscoe, 1965). On the basis of 261 analyses, H.D.B.

Chemical analyses obtained for Waite rhyolite and Amulet andesite
in this work have come from samples within one hundred feet of the
"main contact" and therefore do not represent the whole of the units.

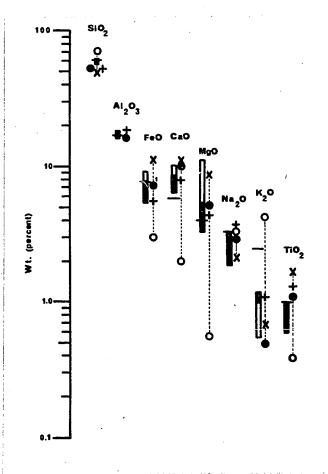


figure 9. COMPARISON OF THE MAJOR ELEMENT COMPOSITION OF ANDESITES, BASALT, GRANITE AND CONTINENTAL CRUST. Filled rectangles, andesite from Japan and New Zealand; unfilled rectangles, mafic andesite from Japan and New Zealand; 6, Amulet andesite; +, Nockolds (1954) average andesite; 0, granite; X, basalt; -, continental crust (after Taylor and White, 1965)

Analyses of New Zealand and Japanese andesite are by Taylor and White (1965).

The data for granite and basalt were compiled by Taylor and White from Daly (1933), Kolbe (1965), and Turekian and Wedepohl (1961).

The data for continental crust are based on a 1:1 ratio of mafic and siliceous rocks and were compiled by Taylor and White from Poldervaart (1955) and Taylor (1964).

^{1.} Average of 11 wet chemical analyses by H. Soutar, McGill University.

^{2.} Kolbe, P., Ph.D. thesis, Australian National University, 1965.

Wilson concluded that the Archean volcanic rocks (mostly from Ontario) belong not to the alkaline or spilite suites but to the basalt-andesiterhyolite association typical of continental and island arc areas (Wilson, H.D.B., 1965). In fact, it is not so much a high soda content but a distinctly low potash content which characterizes the Noranda volcanic suite. This deficiency in potash is discernable in the Amulet andesite (fig. 9) but is particularly noticeable in the Waite rhyolite which contains 1.3 per_cent K20 compared to 5.35 per_cent K20 in Nockolds' (1954) average calc-alkalic rhyolite. The Flavrian and Powell granites are chemically very similar to the Waite rhyolite and it is probable that both rocks were formed from the same magma. If this is correct, the original alkali content of the Walte rhyolite must not have been altered by a process such as alkali transfer during devitrification. Twelve of forty eight samples of Archean rhyolite analyzed by Wilson are chemically similar to the Noranda rhyolite (Wilson, H.D.B., et al, 1965). Except for increased amounts of iron and possibly potassium, the median contents of major elements in tuff are intermediate when compared with the compositions of Amulet andesite and Waite rhyolite.

Recent work suggests that potash deficient rhyolite is one of the end products of an extreme differentiation of a relatively uncontaminated, oceanic, tholeiltic magma (Engel, 1965). This idea is supported by the fact that practically all reported analyses of relatively young, potash-poor rhyolites come from volcanic island areas

^{1.} Although "rhyo-dacite" is a more appropriate term for this unit, the term "rhyolite" is retained because of its common usage. Nockolds' (1954) average rhyo-dacite contains 3.01 per cent K_2O .

SOME ANALYSES OF LOW POTASH, HIGH SILICA (> 70%), VOLCANIC ROCKS

	NO. OF	Na ₂ 0	K ₂ 0	•	
LOCATION	ANALYSES	PER CENT	PER CENT	REFERENCE	
SAIPAN AND GUAM	4	3.5	1.7	Schmidt, 1952	
JAPAN					
Northeast	6	3.8	1.6	Yagi, 1962	
Izu-Hakone Area	4	3.7	1.3	Isshiki, 1964	
Kamitaga Dacite Dome	1	3.9	1.8	Kuno, 1950	
Shikotsu District	1	3.6	2.0	Ishikawa et al, 1957	
Aira District	1	4.0	2.9	Ishikawa et al, 1957	
Usu Volcano	2 .	4.2	1.1	Minikami et al, 1950	
Hakone Volcano	1	4.1	1.4	Kuno, 1962	
Asama Volcano	4	4.3	2.2	Kuno, 1962	
Agagi Volcano	1 .	2.9	2.2	Kuno, 1962	
Narugo Volcano	2	3.3	1.6	Kuno, 1962	
Akita-Yake-Yama Volcano	1	3.4	2.1	Kuno, 1962	
Kōzu-sima	4	4.4	2.3	Kuno, 1962	
Nii-Zima Volcano	1	4.3	2.9	Kuno, 1962	
Atosanupuri Volcano	1	4.0	1.6	Kuno, 1962	
NEW ZEALAND	8	4.6	3.0	Ewart, 1963	_
SUMATRA					Table
Krakatau Group	4	4.6	2.6	Westerveld, 1952	0
SAKHALIN ISLAND	1	2.0	1.0	Verokhov, 1952	
EASTER ISLAND GROUP	. 5	5.3	3,3	Bandy, 1937	

SOME ANALYSES OF LOW POTASH, HIGH SILICA (> 70%), VOLCANIC ROCKS

<u>LOCATION</u>	NO. OF ANALYSES	Na ₂ 0 PER CENT	K20 PER CENT	REFERENCE
KAMCHATKA	11	3.7	3.1	Tomkeiff, 1949
EAST KAMCHATKA Uzon	1	4.4	2.4	Vlodavetz and Piip, 1959
ALASKAN-ALEUTIAN ARC	3	4.2	2.7	Tomkeiff, 1949 .
WASHINGTON	1	3.2	1.0	Coombs, 1952
ALASKA Katmai	2	4.2	2.8	Senner, 1926
YELLOWSTONE	1	4.6	2.5	Coombs and Howard, 1960
CHILE	2	2.9	1.3	Casertano, 1963
SOUTH SANDWICH ISLANDS	· 1	4.7	0.7	Gass, 1963
ICELAND	4	4.7	3.2	Carmichael, 1962

judged to be at the edge of the sial of the continents (table 6).

Potash deficient Cenozoic volcanic rocks of the western United States occur in a narrow zone parallel to the Pacific Coast. Increasing potash content in volcanic rocks eastward from the coast was correlated with a thickening of the sial as demonstrated by the decreasing Bouger gravity (Moore, 1965). Relatively young volcanic rocks become progressively more potash rich toward the Asian continent across the Japanese and Indonesian Islands (Ishikawa and Katsui, 1959), (Kuno, 1960), (Rittman, 1962, p. 159).

Vast areas of relatively young, siliceous volcanic rocks occur closer to the center of continents but these rocks almost always contain abundant potash and correspond to Nockolds' (1954) averages for rhyolites. Engels (1965) states that "...tholeiltic lavas, contaminated with Si, K and the chemically coherent trace elements and radiogenic isotopes from the sial, also have been the predominant or only magma generated in the mantle under the continents."

If Engels' suggestion is correct we are presented with two major mechanisms for the generation of siliceous rocks. One involves the assimilation by a tholeiitic magma of sial under the continents and, presumably, differentiation which produces rhyolite containing potash and soda in roughly equal, and substantial, amounts. The other mechanism involves the extreme differentiation of a tholeiitic magma, with a minimum of assimilation of sial, probably in a wet environment (Osborn, 1959), at the edges of continents or in island arc areas and produces a potash deficient, siliceous rock referred to most often in the literature as soda - rhyolite. Accumulation of volatile substances during such

a differentiation process would result in much explosive volcanic and hydrothermal activity.

The major element analyses of the Noranda area rocks support two ideas. First, the Noranda rocks belong to a single highly differentiated volcanic rock suite which is similar to younger extremely differentiated rock suites found at the edges of continents, especially along island arc systems. Second, the close association of both the Noranda ores and much recent solfataric activity to the pyroclastic varieties of this suite of rocks suggests that the Noranda ores and the recent sulphur and sulphide deposits of the volcanic islands may share a similar origin.

The Canadian shield, locale of many of the Noranda-type of ore bodies is also, on the basis of a few analyses, the site of some potash-poor rocks (Engels, 1965; Wilson, H.D.B., 1965; Maxwell et al, 1965).

Other examples of this association are the Mattagami area, Quebec (Sharp, 1964), the North Coldstream Mine, Port Arthur, Ontario (Giblin, 1964) and the East Buchans Mine, Newfoundland (Relly, 1960). Outside Canada | examples are the Jerome area, Arizona (Anderson and Creasy, 1958) and the Shasta district, California (Kinkel et al, 1957).

Other areas in the world where soda-rich (and therefore possibly potash-poor) rocks have been reported associated with massive pyritic deposits include the Philippines, Alaska, Scandinavia, Central Europe, Cyprus, Turkey and the Ural Mountains in Russia (Kinkel, 1962).

^{1.} For an excellent review of young sulphur and sulphide deposits in Japan see Mukaiyama (1959).

Table 7
TRACE ELEMENT CONTENT OF THE HOST ROCKS

•	WAITE RHYOLITE		AMULET ANDESITE		TI	TUFF	
•	(ppm) mode	(ppm)	(ppm)	(bbw) wesu	median (ppm)	range (ppm)	
Ag	.17	•38	•18	•24	1.25	4.36	
Co .	mostly	<1	43	53.8	85	71.3	
Cr	mostly	< 40	155	197	150	176	
Cu	20	104	70	119	700	2213	
Ga	27	30	24.5	24.5	25 .	19	
Ni	1	7.8	50	104	60	77.9	
РЬ	0.7	2.4	. 0.3	1.0	4.6	24.8	
Sn	0.5	1	2.5	. 1. 4	4	21.1	
Sr	370	418	150	197	340	285	
Υ	50	70	20	24	32	· 23	
Zn ·	180	394	160	240	800	3925.	
Zr	260	405	180	180	260	223	

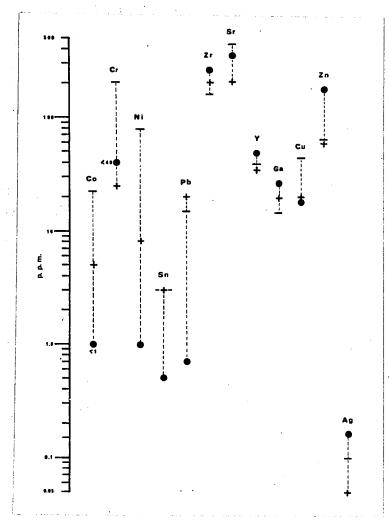


figure 10. COMPARISON OF THE WAITE RHYOLITE WITH THE CONTINENTAL CRUST AND THE AVERAGE FELSIC ROCK. 8, Waite rhyolite; +, average felsic rock; -, continental crust.

The data for the average felsic rock are from Vinogradov (1962).

The data for continental crust are based on a 1:1 ratio of mafic and siliceous rock and are from Meson (1960) (cf. figs. 9 and 11).

TRACE ELEMENTS

The background values (statistical modes) of many trace elements in the host rock of the Lake Dufault ore body are assumed to represent the original contents of these elements in the rock. Other trace elements have been added or removed to such an extent that rock containing original amounts of these elements may have been sampled only in a few places. Background values of the trace elements are presented in table 7.

Waite Rhyolite

The background values of trace elements in Waite rhyolite are plotted in fig. 10 along with abundance data for the earth's crust and felsic rocks. The earth's crust is assumed to be composed of equal portions of siliceous and mafic rock (Mason, 1960).

Cobalt, nickel and chromium

The Waite rhyolite, when compared with the average felsic rock, contains very small amounts of Co, Ni, and Cr. Wager and Mitchell (1951) in their study of the contents of trace elements in the rocks of the Skaergaard intrusion, showed that Cr, Ni and to some extent Co, are concentrated in early rocks so that later rocks contain little or none of these elements. These elements are concentrated in early-formed minerals in a crystallizing magma and, therefore, become depleted in residual melts produced by fractional crystallization.

Lead and tin

The small amount of lead in the Waite rhyolite may be related to its low potassium content. As reported in an earlier section,

Engel (1965) suggests that magma generated in the mantle under the continents becomes contaminated with radiogenic isotopes as well as potassium from the sial. Unlike the Amulet andesite (fig. 12), the Waite rhyolite has a low tin content.

Zirconium, strontium and yttrium

The background zirconium¹ content of the Waite rhyolite is less than that of granophyres associated with differentiated basic bodies like the Skaergaard intrusion (300-500ppm or more) but greater than that of the majority of rhyolites or quartz-monzonites and granites of the orogenic belts (200ppm or less) (Chao and Fleicher, 1960). The zirconium content of the Waite rhyolite is similar to that of sodarhyolites and quartz latites reported by Chao and Fleicher from the Aleutian Islands. Alkalic volcanic rocks contain much greater amounts of zirconium than volcanic rocks of the calc-alkalic or tholeiitic series.

The Waite rhyolite has a high strontium to calcium ratio. High ratios of strontium to calcium in many rhyolites and pegmatites have been reported by Turekian and Kulp (1956) who suggest that these ratios may be a consequence of magmatic differentiation by fractional crystallization.

Wager and Mitchell (1951) showed that yttrium tended to substitute for calcium in "late" pyroxene in rocks of the Skaergaard intrusion. The Waite rhyolite contains slightly more yttrium than the average felsic rock (fig. 10).

^{1.} Analyses of standard samples (table 3) show that the background values of zirconium and strontium given in this work may be erroneously high.

Gallium

The gallium¹ content of the Waite rhyolite is slightly higher than that reported in felsic rocks (fig. 10) and in siliceous volcanic rocks from Kamchatka (Borisenok, 1959). Gallium shows no tendency to become concentrated in rest magmas and becomes separated from aluminum only during supergene alteration.

Silver, copper and zinc

The background contents of Ag and Zn¹ in the Waite rhyolite are higher than their contents in the average felsic rock. The background content of copper in the Waite rhyolite is similar to the average copper content of felsic rocks (Vinogradov, 1962; Wedepohl, 1953, 1962).² Variations of these elements in the Waite rhyolite clearly show that large amounts have been added during the ore-forming process.

Amulet Andesite

The background trace element composition of the Amulet andesite is plotted with analyses of andesite from New Zealand and Japan, estimates of the trace element content of the average granite, basalt and continental crust (Taylor and White, 1965), and abundance data for andesite and other intermediate rocks given by Vinogradov (1962) (fig. 11).

Analyses of standard samples (table 3) show that the background values of gallium and zinc given in this work may be erroneously high.

The value for copper is higher than Wedepohl's average for felsic rocks.

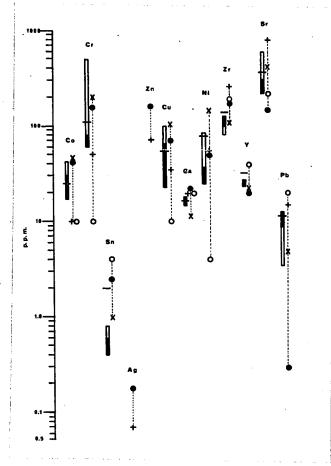


figure 11. COMPARISON OF THE TRACE ELEMENT COMPOSITION OF ANDESITES, BASALT, GRANITE AND CONTINENTAL CRUST. Filled rectangles, andesite from Japan and New Zealand; unfilled rectangles, mafic andesite from Japan and New Zealand; 8, Amulet andesite; +, Vinogradov's (1962) average andesite and intermediate rock; 0, granite; X, basalt; -, continental crust (after Taylor and White, 1965).

Analyses of New Zealand and Japanese andesite are by Taylor and White (1965).

The data for granite and basalt were compiled by Taylor and White from Daly (1933), Kolbe (1965) and Turekian and Wedepohl (1961).

The data for continental crust are based on a 1:1 ratio of mafic and siliceous rocks and were compiled by Taylor and White from Poldervaart (1955), Taylor (1964).

^{1.} Kolbe, P., Ph.D. thesis, Australian National University, 1965.

Cobalt, chromium, nickel and tin

The background values for Co and Cr¹ in Amulet andesite are higher than their usual contents in andesites whereas that for nickel is normal. The same is true of andesites from Japan and New Zealand. Analyses of pyrite from the Noranda area given by Roscoe (1965) showed unusually high amounts of tin and cobalt. The background tin content of the Amulet andesite is significantly higher than the tin content of andesite from Japan and New Zealand (fig. 11).

Lead

The background content of lead in the Amulet andesite is lower than the lead content of the average andesite. New Zealand and Japanese andesites show the same feature (fig. 11). This deficiency in lead parallels the potassium deficiency in these rocks.

Zirconium

The background content of zirconium² in the Amulet andesite is lower than Vinogradov's (1962) average and higher than the zirconium content of andesites from Japan and New Zealand (fig. 11) and the Aleutian Islands (65ppm, 13 analyses) (Chao and Fleicher, 1960). The zirconium content of the Amulet andesite corresponds to that of the andesite from the Cascade Mountains in California and Oregon. What is perhaps more significant is the fact that the content of zirconium in the Amulet andesite (180ppm) differs relatively little from that in the Waite rhyolite (260ppm). This minimal variation is typical of the

Analyses of standard samples (table 3) show that the background values of chromium given in this report may be erroneously high.

Analyses of standard samples (table 3) show that the background values of zirconium given in this work may be erroneously high.

basalt-andesite-dacite-rhyolite series of the Circum-Pacific region (Chao and Fleicher, 1960). These authors show that the alkalic rock series is higher in zirconium and shows a more pronounced increase of zirconium from mafic to siliceous composition. The calc-alkalic rocks of the Southern California batholith and the Caledonian rocks of Scotland have irregularly varying zirconium contents.

Strontium

The Amulet andesite has a much lower background content of strontium than is usual in andesites. It is quite possible that this was caused by the leaching of some of the original strontium, especially at the base of the andesite (fig. 42).

Yttrium and gallium

The yttrium and gallium¹ contents of the Amulet andesite are essentially the same as that reported in other andesite.

Silver, copper and zinc

The background contents of Ag and Zn¹ in the Amulet andesite are very similar to those in the Waite rhyolite and are higher than the Ag and Zn contents of the average intermediate rock. The background copper content of the Amulet andesite is higher than that of the Waite rhyolite and higher than the average copper contents of intermediate rocks (Vinogradov, 1962). It will be shown in forthcoming sections that certain amounts of these elements have been added to the Amulet andesite.

Analyses of standard samples (table 3) show that the background values of gallium and zinc given in this thesis may be erroneously high.

Conclusions

It has been suggested in a previous section that the rocks of the Noranda area belong to a single volcanic rock suite which is similar to suites of younger, extremely differentiated volcanic rock found at the edges of continents, especially along island arc systems. A comparison of the contents of trace elements in the Waite rhyolite and Amulet andesite with the small amount of data available for volcanic rocks from volcanic islands lends some support to this idea. Of special significance in this regard may be the unusually low lead content of the Waite rhyolite and Amulet andesite. Engel (1965) suggests that magma generated under the continents would become contaminated with radiogenic elements as well as with potassium.

The contents of other elements in the host rocks suggest a similarity between these rocks and rocks from volcanic island areas.

These elements are Zr in the Waite rhyolite and Co, Cr, and Zr in the Amulet andesite.

The unusually low contents of Co, Cr and Ni in the Waite rhyolite compared with the average felsic rock may be explained by the tendency of these elements to become depleted in residual melts produced by fractional crystallization.

The Waite rhyolite and the Amulet andesite contain unusually high background amounts of Ag and Zn and the Amulet andesite has a high background copper content. As shown in forthcoming sections, these elements have been added to the host rocks during the ore-forming process.

Bedded, Cherty Tuff

The median values in tuff of elements found concentrated in ore (Ag, Co, Cu, Ni, Pb, Sn, Zn) are higher than the background values of these elements in either Waite rhyolite or Amulet andesite. The range of values for each of these elements in tuff is also greater than that of any of the other elements (Cr, Ga, Sr, Y, Zr). This feature of the tuff implies a relationship between it and ore.

DISTRIBUTION OF ELEMENTS IN THE HOST ROCKS

LATERAL DISTRIBUTION OF ANOMALOUS VALUES OF ELEMENTS IN RHYOLITE AND ANDESITE

Perusal of the data charts (Appendix B) shows that the average value (\bar{x}) for an element in the rock sampled from one drill hole is in many cases beyond the limits of background. The significance of these "anomalous" values is established by plotting them on diamond drill hole plans (figs. 12 to 33). One plan is given for each element with more than just a few anomalous values in rhyolite or andesite. Those drill hole locations marked with open circles denote rhyolite or andesite which contains background amounts of an element.

Rhyolite

Rhyolite which contains SiO_2 , Fe (total as Fe_2O_3), Mno, MgO, Na₂O, Ag, Co (?), ¹ Cu, Ni, Pb, Sn and Zn in anomalous amounts was

Because most samples contain less than detectable amounts of Co, results for this element are not plotted on a plan of diamond drill holes.

sampled from several diamond drill holes. Rhyclite directly below the massive sulphide ore body, mainly in an altered and mineralized zone, is deficient in SiO₂ and Na₂O (figs. 12 and 13). Rhyclite containing abnormally high amounts of nickel was cut by several drill holes scattered throughout the property (fig. 14). The "ore-major" and "ore-trace" elements, Fe (total as Fe₂O₃), MnO, MgO, Ag, Co (?), Cu, Pb, Sn and Zn occur in abnormally high amounts close to the sides of the massive sulphide ore body and in the altered and mineralized zone beneath it (figs. 15 to 22).

Andesite

Andesite containing abnormally high amounts of Al₂O₃, CaO, K₂O, Ag, Co, Cr, Cu, Ni, Pb, Sn and Zn was cut by several diamond drill holes in the Norbec area. Anomalous values for Al₂O₃, CaO, K₂O, Co, Cr, Ni, and Sn are scattered throughout the area (figs. 23 to 29). The "ore-trace" elements in andesite (Ag, Cu, Pb, Zn) occur in amounts greater than background over the massive sulphide ore body (figs. 30 to 33).

Summary of Anomalous Values

On tables in Appendix D the average value for each element from each drill hole is designated as background (equal to "0"), probably anomalous (equal to "1")¹ and anomalous (equal to "2").²
For normally distributed elements, values outside background limits

^{1.} Values between "a" and "b" on the histograms (Appendix A).

^{2.} Values outside of "b" on the histograms (Appendix A).

(ULB, LLB) are weighted as anomalous (equal to "2" or "-2" respectively). Results for the "ore-elements" are summed and the sum is plotted on diamond drill hole plans (figs. 34 to 36). These figures confirm the contention that the "ore-trace" and "ore-major" elements are present in anomalous amounts in the host rock over, under and close to the sides of the massive sulphide ore body. The northwest Norbec section is the only other sampled part of the property which may have been similarly effected. The results are further summarized on tables 8 and 9, p.99,100.

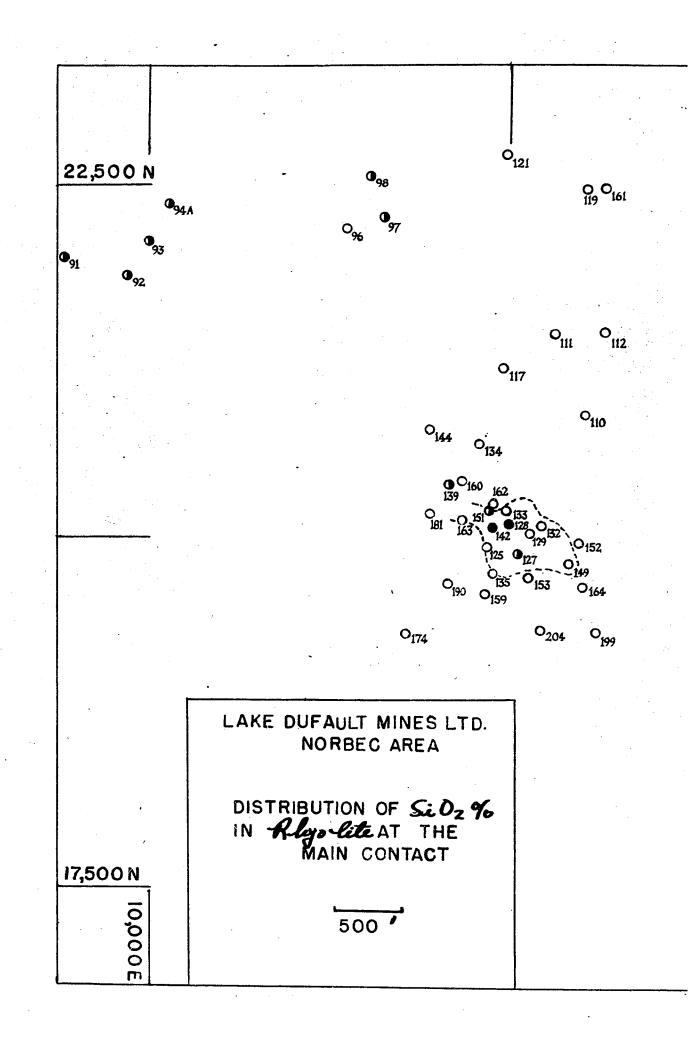
LATERAL DISTRIBUTION OF ELEMENTS IN BEDDED, CHERTY TUFF

Analyses of tuff samples are treated as described previously (p.37). The only elements which seem to have a lateral distribution related to the massive sulphide ore body are Pb, Cd, and Zn (figs. 37, 38 and 39).

VERTICAL DISTRIBUTION OF ELEMENTS IN RHYOLITE AND ANDESITE

Very simple graphs constructed from the data of tables E1, E2, E3 and E4 (Appendix E) show which elements tend to be present in greater or lesser amounts near the main contact (figs. 40, 41). For rhyolite, the ordinate of the graphs is the number of holes in which the top half of the sampled section contains a higher concentration of the element than does the bottom half. The abcissa is the number of holes in which the top half of the sampled section is deficient in an element. The base of the andesite is tested in the same way. If, for a given element, there is no variation at all in many drill holes, the plot of that element would fall close to the origin. If a given element has a random

SiO₂ rhyolite



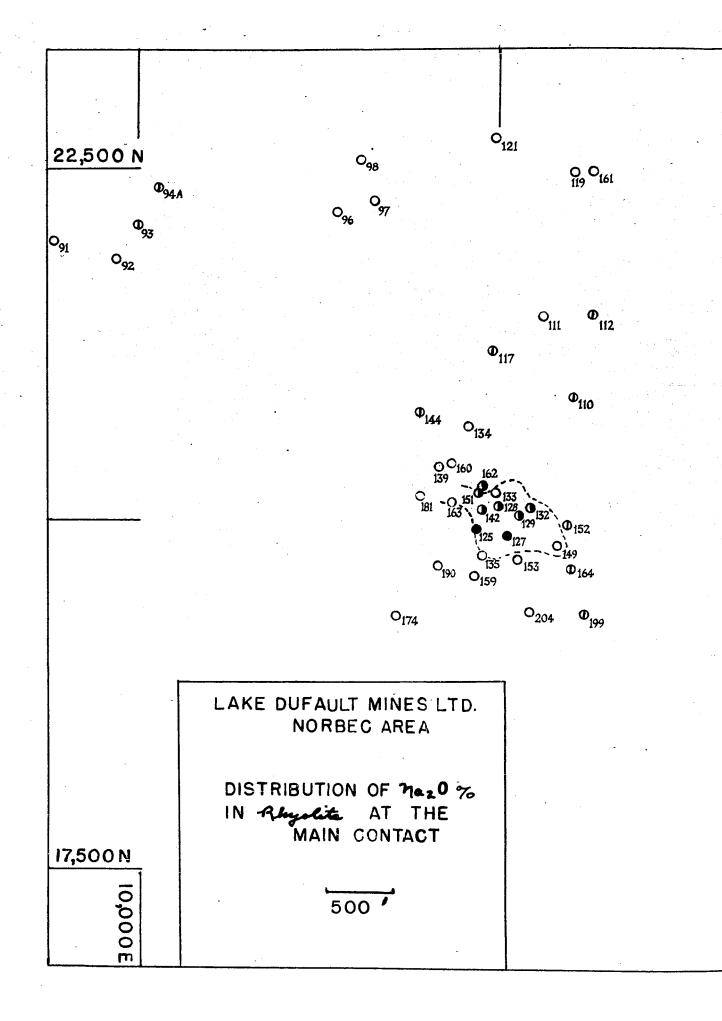
O₁₅₇ 0113 O₁₅₅ O₁₅₀ O₁₂₂ O₁₂₀ · 0116 O₁₅₈ 0118 O₁₂₃ LEGEND Outline of ore O Diamond drill hole 0 > 71.0 71.0 -65-0

< 65.0

15,000 E

l

Na₂O rhyolite

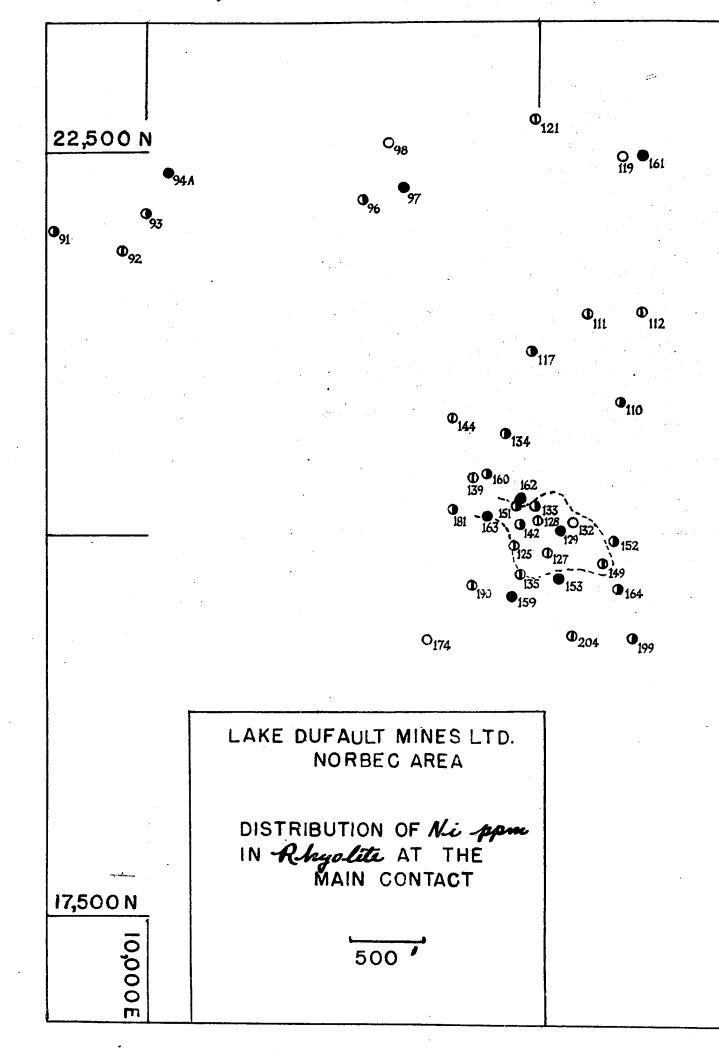


0161		
0 112		
	0 ₁₅₇	
	O ₁₁₃	
	Φ ₁₅₅ Φ ₁₅₀	
	0 ₁₁₄	
	O ₁₂₂	
	144	
	O ₁₂₀ O ₁₁₆	
•		0 ₁₅₈
99	$oldsymbol{\Phi}_{113}$	
		÷
	•	
	O ₁₂₃	
	LEGEND	,
	Outline of ore	
		·
	O Diamond drill hole O < 3.3 > 1.7	
	Ø 3.3-4.4	
	⊕ >4.4	
	9 1.7~0.5	·
	● < 0.5	.
15,000 E	₩ ₹0.3	
0		ļ
\sim 1		
0		l l

Ni rhyolite

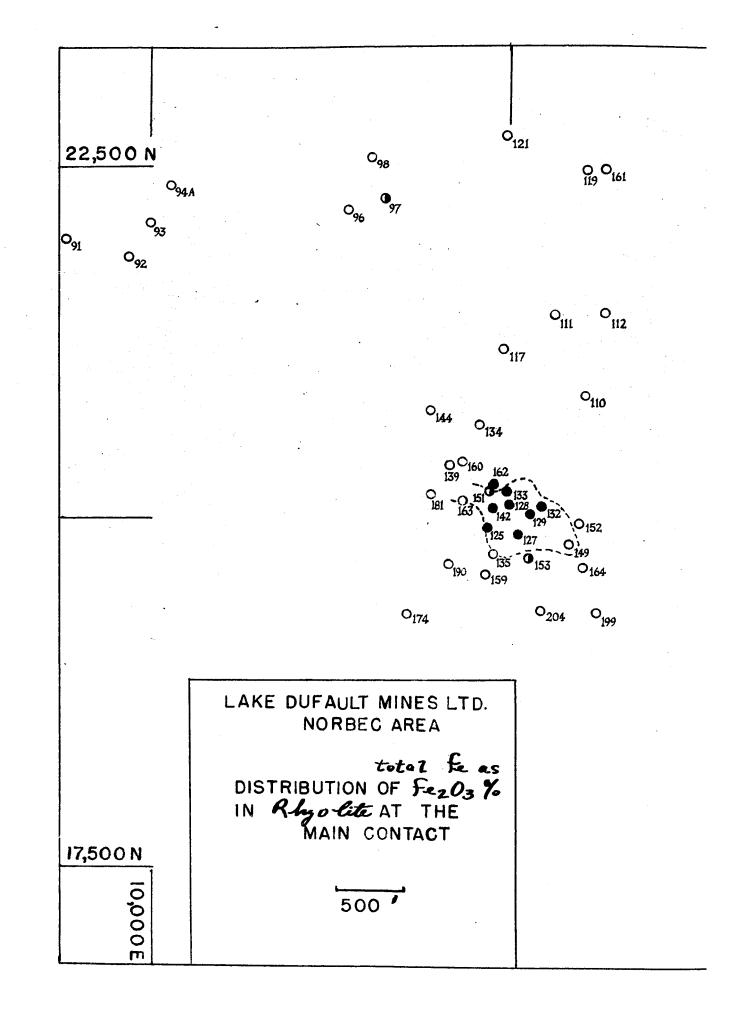
Figure 14

17



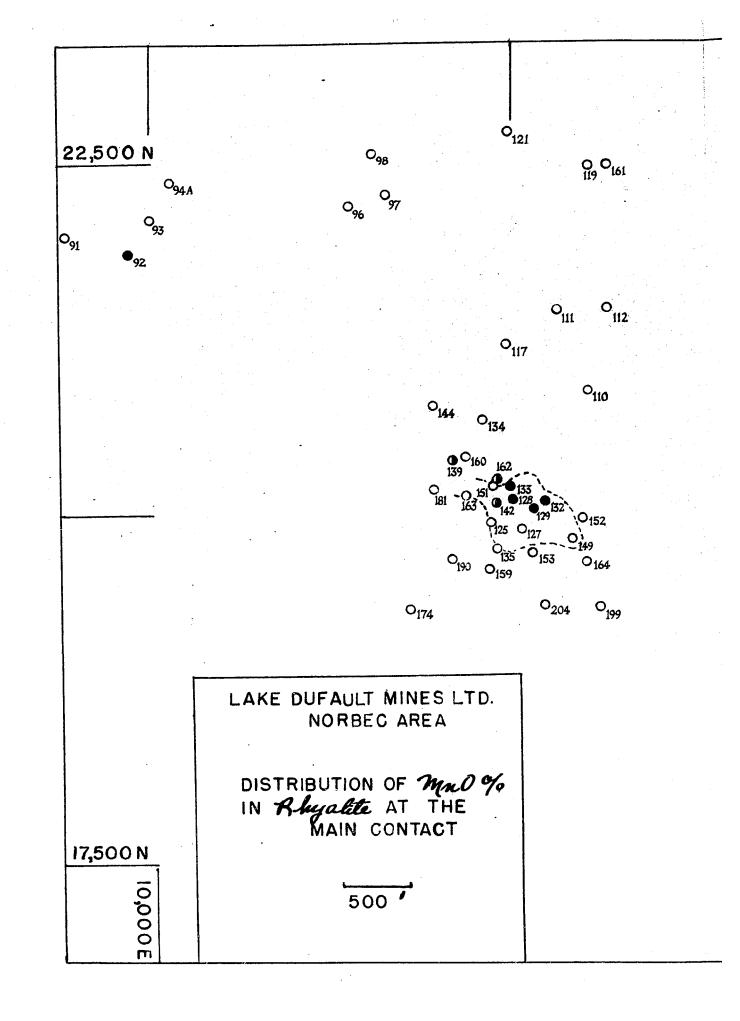
1 Φ114 **o**₁₂₂ **D**120 . Ф116 **●**118 **123** LEGEND Outline of ore O Diamond drill hole ₹ 2.5 0 25-50 5.1-10.0 > 10.0 15,000 E

Fe rhyolite



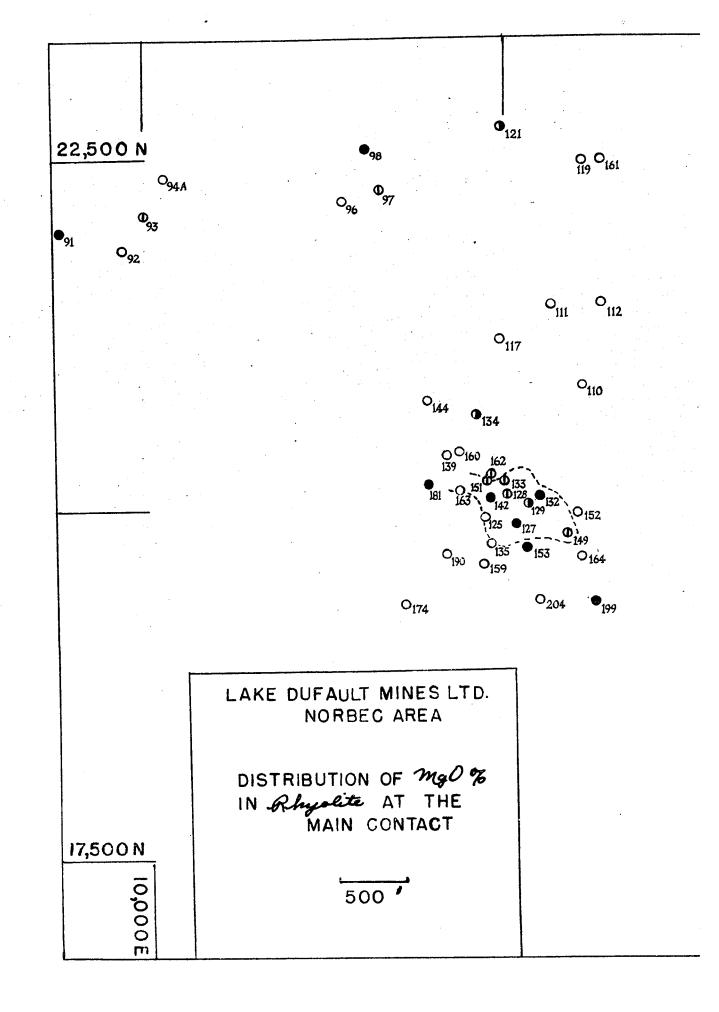
		•		
•				
O		•		
, O ₁₆₁				
	esta-EÁ			
•	gere			
0112				
112				
	0			
	O ₁₅₇			
10	O ₁₅₅	O ₁₁₃		
	O ₁₅₅	150		
•				
	en e		0 ₁₁₄	
i2		O ₁₂₂		
•		122		
64				
·		O ₁₂₀	O ₁₁₆ .	O ₁₅₈
٥ ₁₉₉			Oile	. 128
199			- 110	
	- A second con-			
		O ₁₂	3	
	·	LEGEND		
	en e			
		Outline		
		0 Diamone	d drill hole	
		0 (6.0		,
		6.0-8	.0	
		> 8.0		
	_			
	15,000 E			
	8			
	ŏ			
	m			

MnO rhyolite



)161)112 O₁₅₇ O₁₅₅ O₁₅₀ 0114 O₁₂₂ 0120 O₁₅₈ 0116 0113 199 O₁₂₃ LEGEND Outline of ore O Diamond drill hole < 0.2 0.2 - 0.35 > 0.35 15,000 E

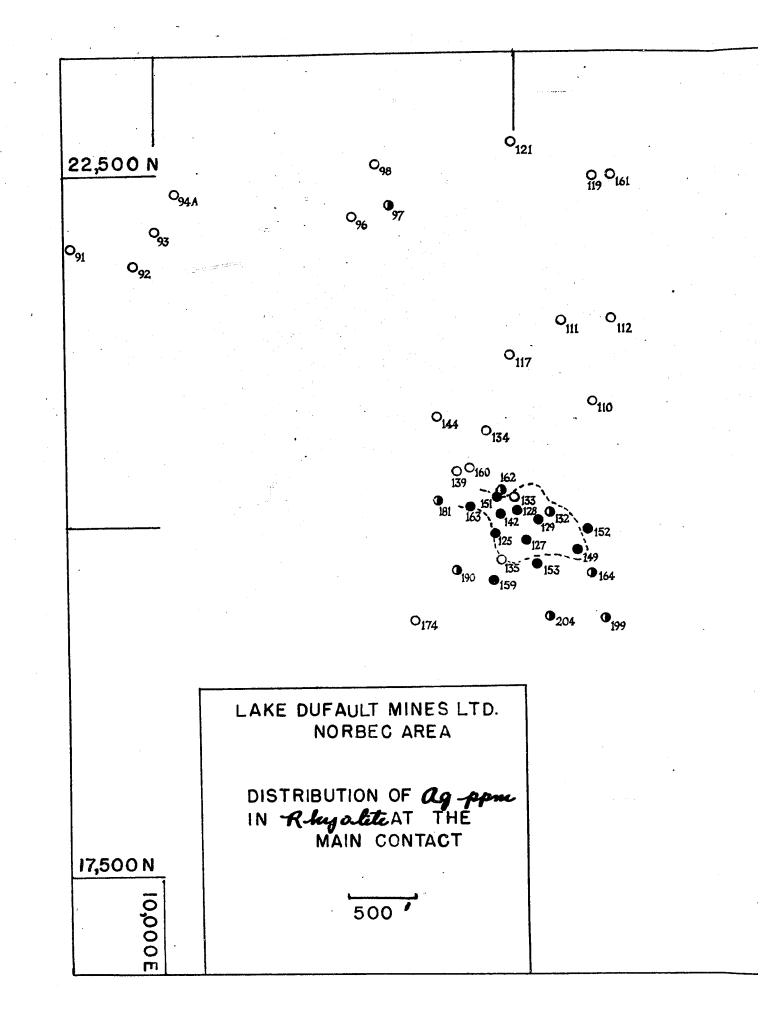
MgO rhyolite



161 112 O₁₅₇ O₁₅₅ 0114 Ф₁₂₂ O₁₂₀ O₁₅₈ 0116 Oil8 199 O₁₂₃ LEGEND Outline of ore O Diamond drill hole < 1.0 0 1.0-1.50 Φ 1.51-2.0 > 2.0 15,000 E

÷

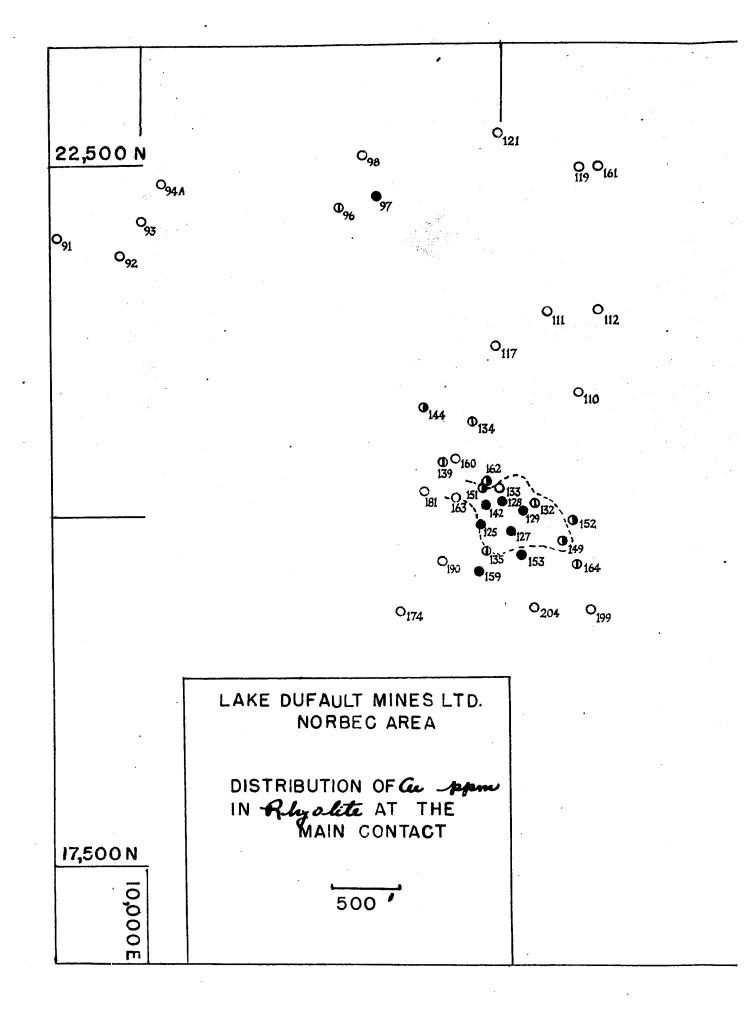
Ag rhyolite



0 0 161 0 0₁₅₇ 0113 O₁₅₅ 0110 0114 O₁₂₂ 0120 **O**116 O₁₅₈ 0118 O₁₂₃ LEGEND Outline of ore O Diamond drill hole < 0.4 0.4-0.8 > 0.8 15,000 E

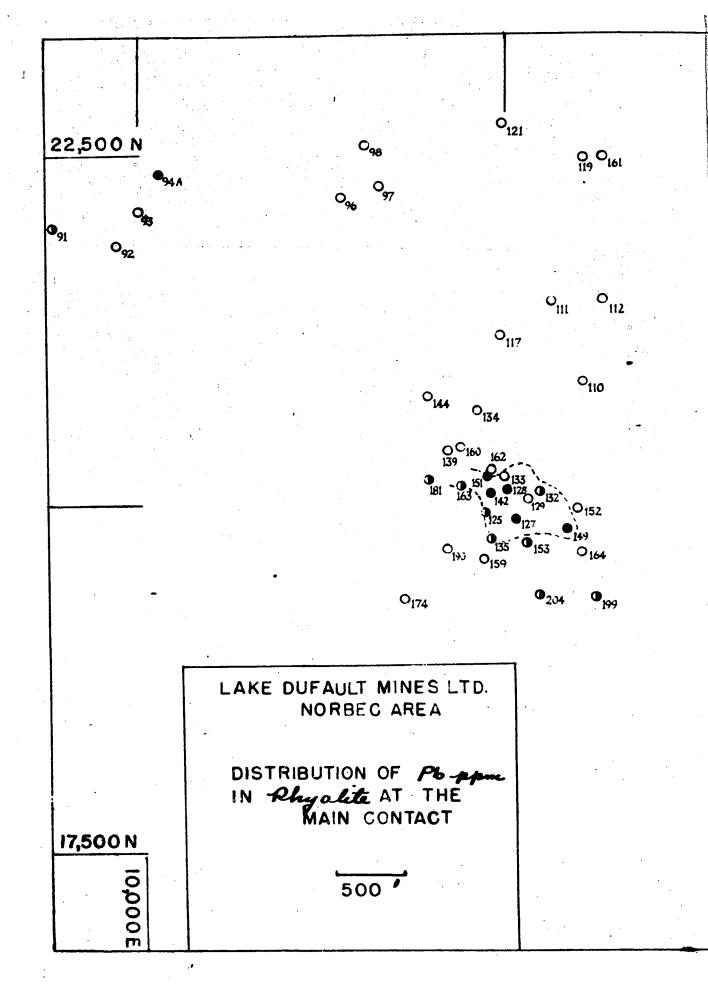
Cu rhyolite

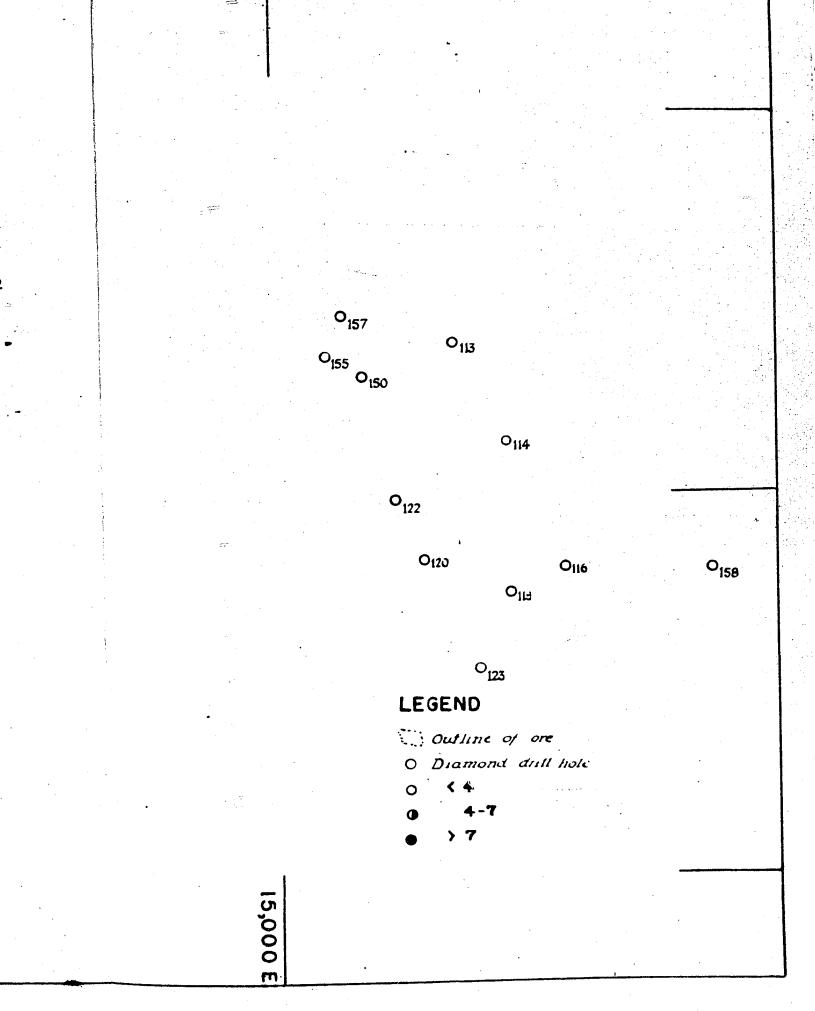
Flgure 19



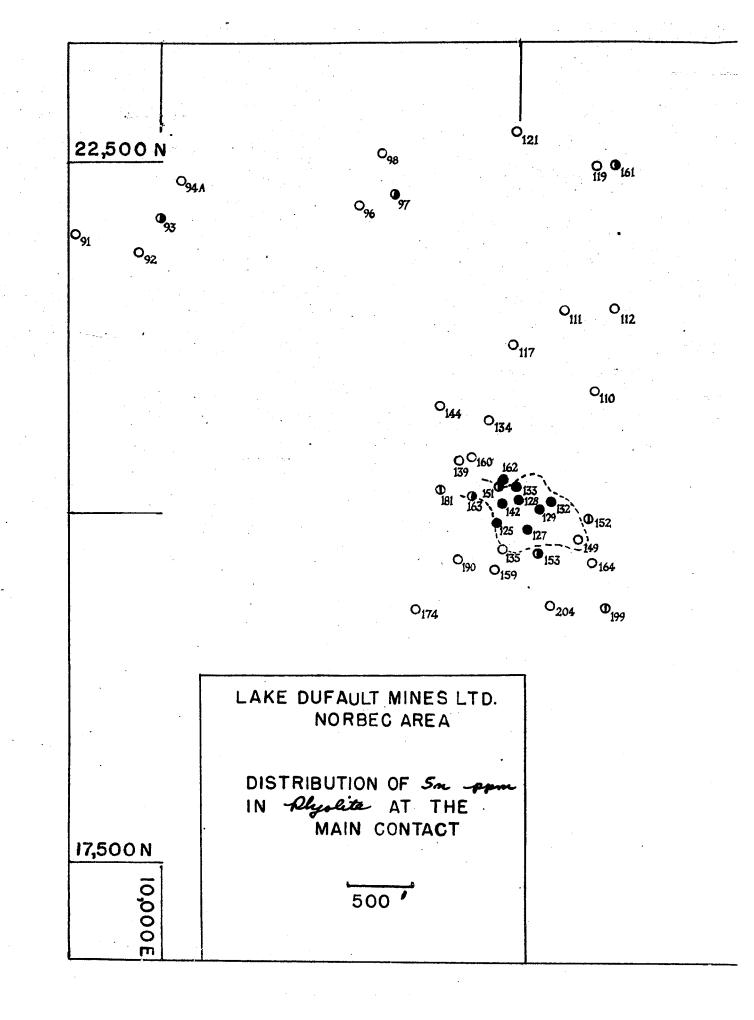
	Learn Committee			
•				
00161		•		
•		·		
		•		
				1
				1.
0112				1
112	• Company of the com			
		0		
		0 ₁₅₇		
		0113		1
0110		O ₁₅₅		
		O ₁₅₅ O ₁₅₀		
• "			P ₁₁₄	
D ₁₅₂ 49 Ф ₁₆₄		O ₁₂₂		
1		122		
49	•			
Ψ ₁₆₄		O ₁₂₀	O ₁₁₆ O ₁₅₈	
		· ·		3
O ₁₉₉	e vizir	·	Oile	1
177			•	- 1
		Φ ₁₂₃		
				l
•		LEGEND	·	
		Outline of	n.f. aa	1
	-			
	•	O Diamond	drill hole	
		0 < 80	•	
		D 80 -11	50 .	- 1
	•			1
		0 150-2	225	l
	•	> 225		
	, ·	_		
		<u> </u>		Ì
	ž	o l		
				1
		^ !		1
	,	7		ŀ

Pb rhyolite Figure 20



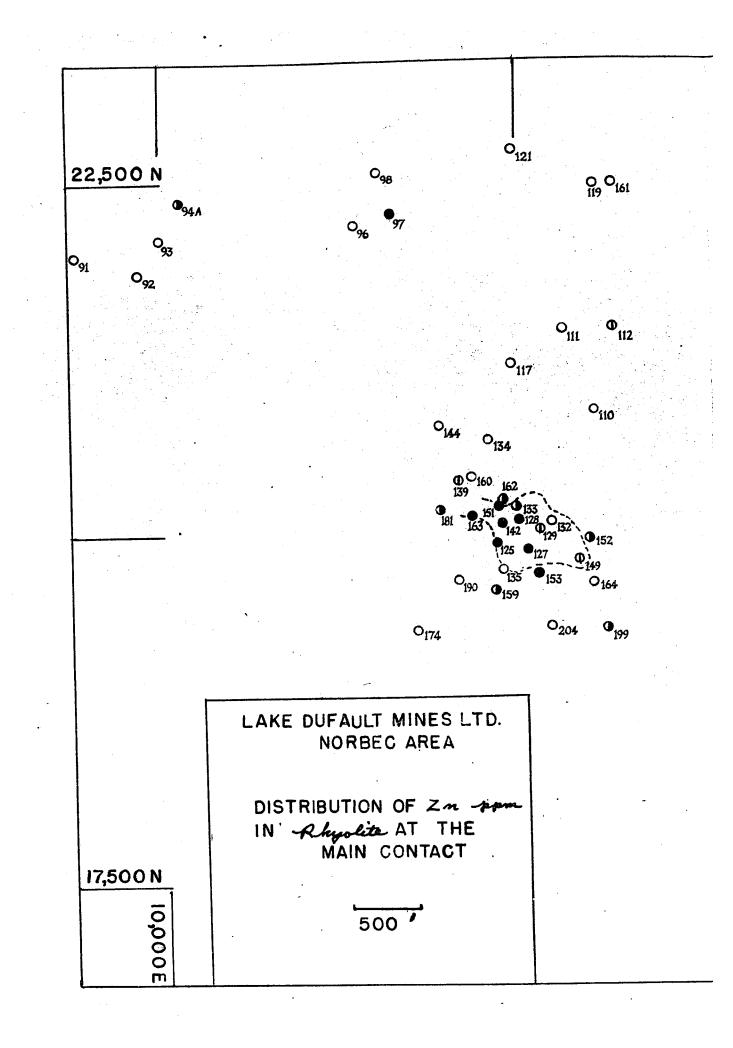


Sn rhyolite Figure 21



the state of the s			
			•
) 157		
	O _{1!3}		
	O ₁₁₃ 5 O ₁₅₀		
	0150		
	O ₁₁ ,	4	
	• ₁₂₂		
	Φ ₁₂₀	0116	0.
	$\mathbf{\Phi}_{11}$		O ₁₅₈
	· · · · · · · · · · · · · · · · · · ·		
	•	•	
	Ф ₁₂₃		•
	LEGEND		*
	Outline of	one	
	O Diamond a		•
	0 (1.0		- '
	. Ф 1.0-1.5		
	0 1.51 - 2.5		
1	> 2.5		·
5			
			•
0			
15,000 E			

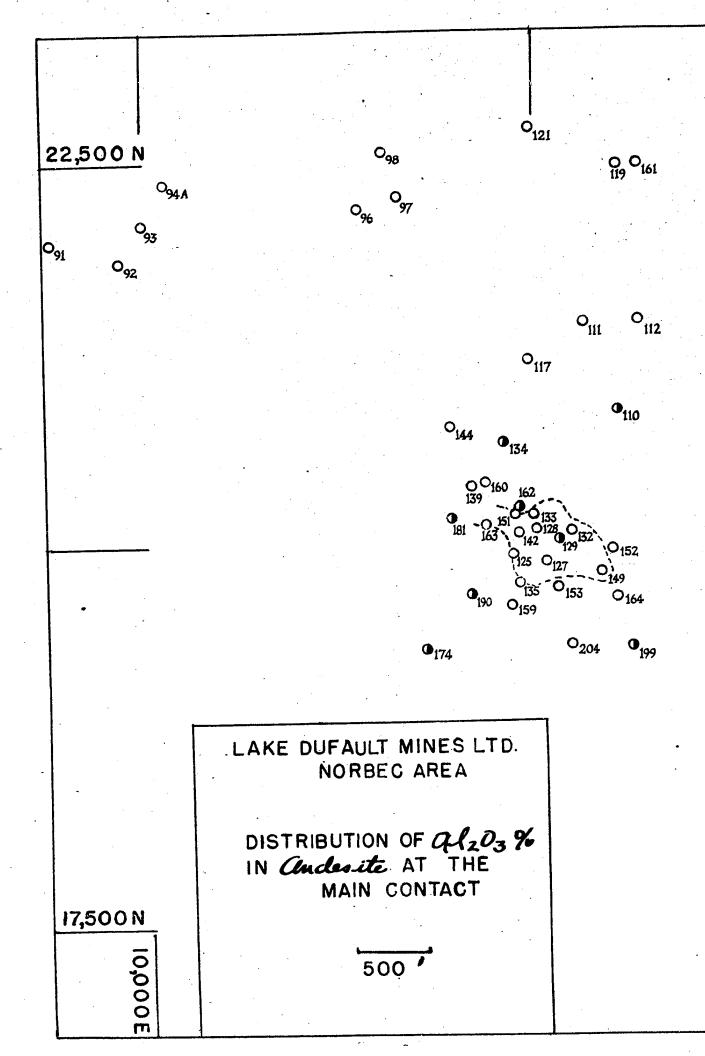
Zn rhyolite



• 157 • 155 • 150 • 150 • 114

..

 Al_2o_3 andesite



_^

O₁₅₇
O₁₅₅
O₁₅₀
O₁₁₄
O₁₂₂
O₁₂₀
O₁₂₈
O₁₂₈

LEGEND

O .

Outline of ore

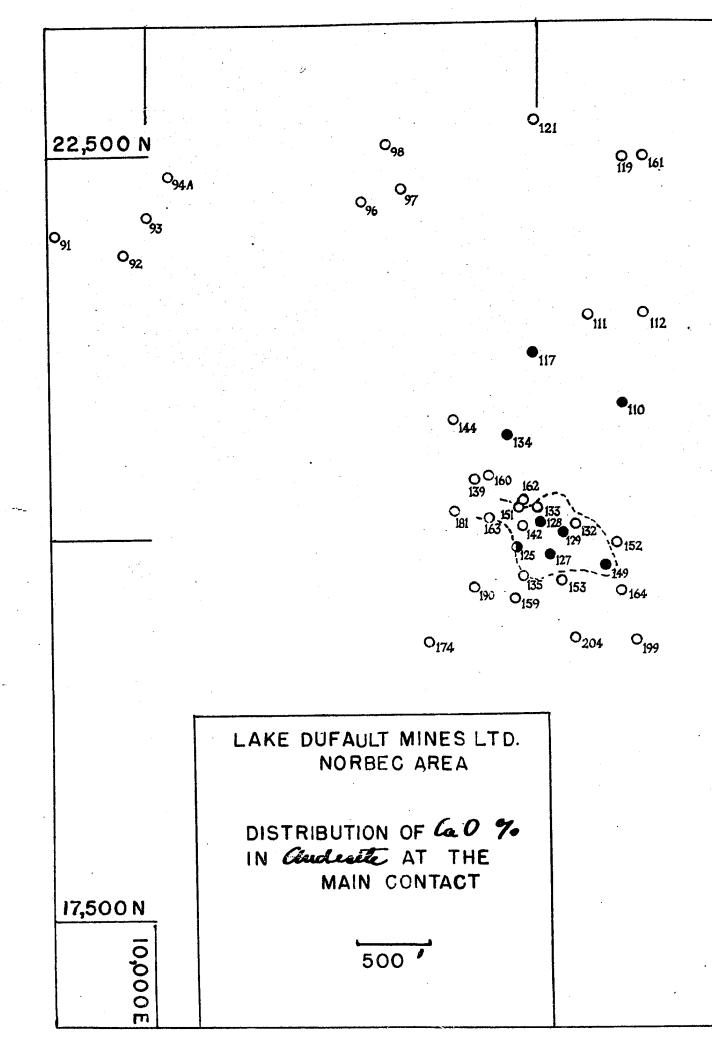
₹ 19.0

> 21.5

O Diamond drill hole

19-0-21.5

CaO andesite



O₁₅₇
O₁₅₅
O₁₅₀

114.

122

●120

●158

Offi

116

O₁₂₃

LEGEND

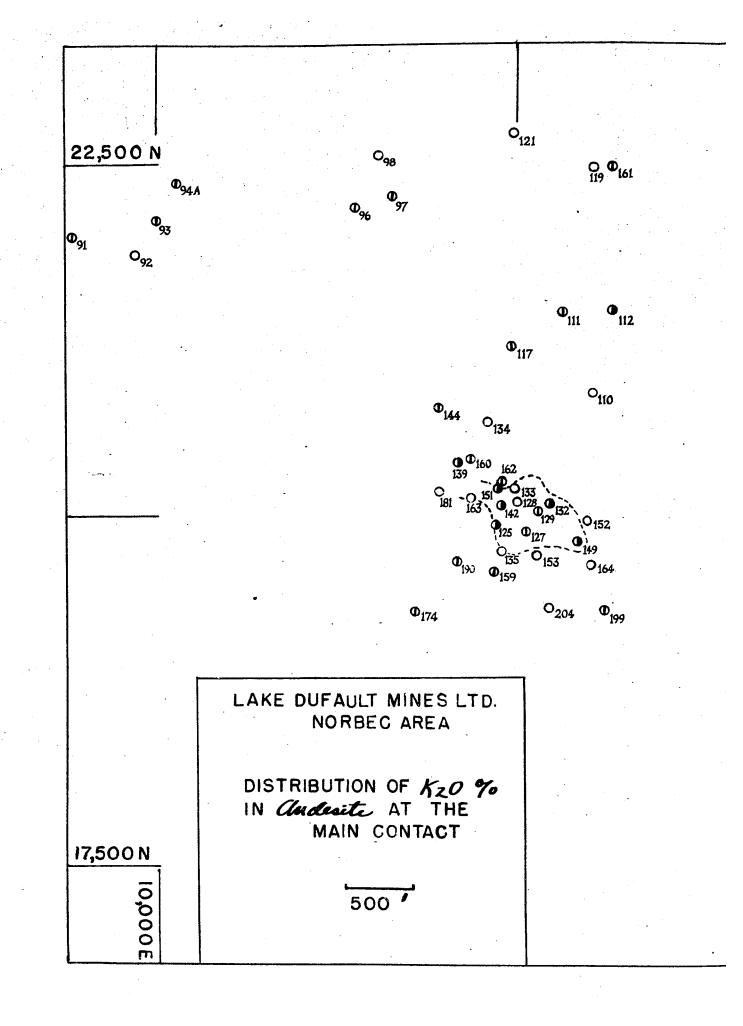
Outline of one

O Diamond drill hole

• > ULB

0 < LLB

K₂0 andesite



Ο₁₅₇
Φ₁₅₅
Ο₁₅₀
Φ₁₁₄
Ο₁₂₂
Φ₁₂₀
Φ₁₆
Φ₁₅₈

LEGEND

0

Outline of ore

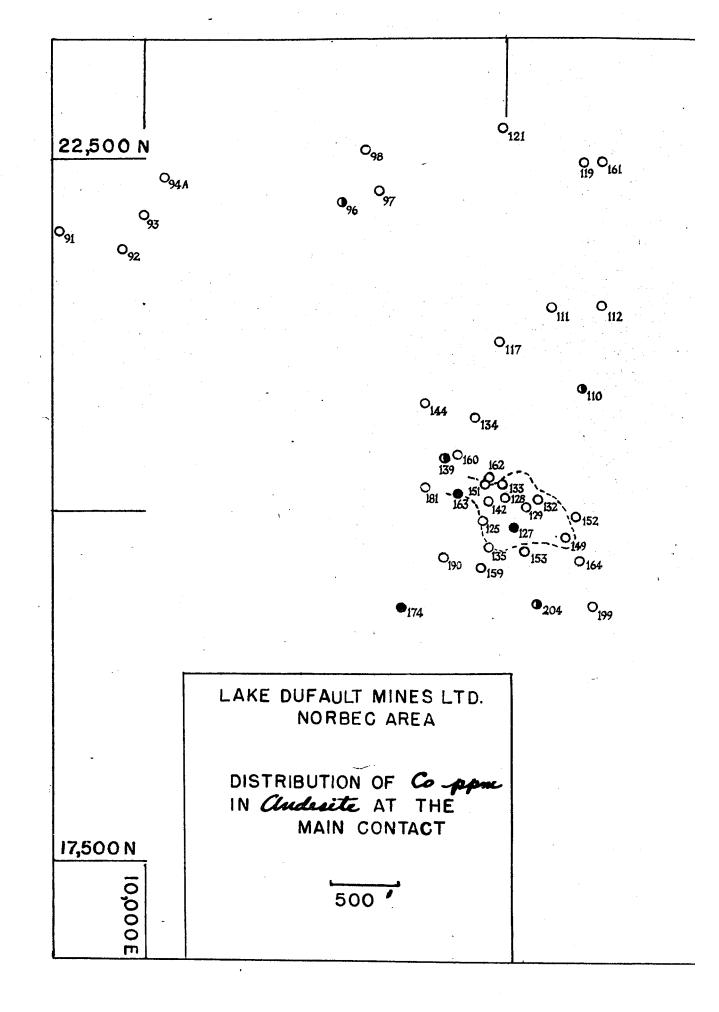
ሩ 0.75

> 1.25

O Diamond drill hole

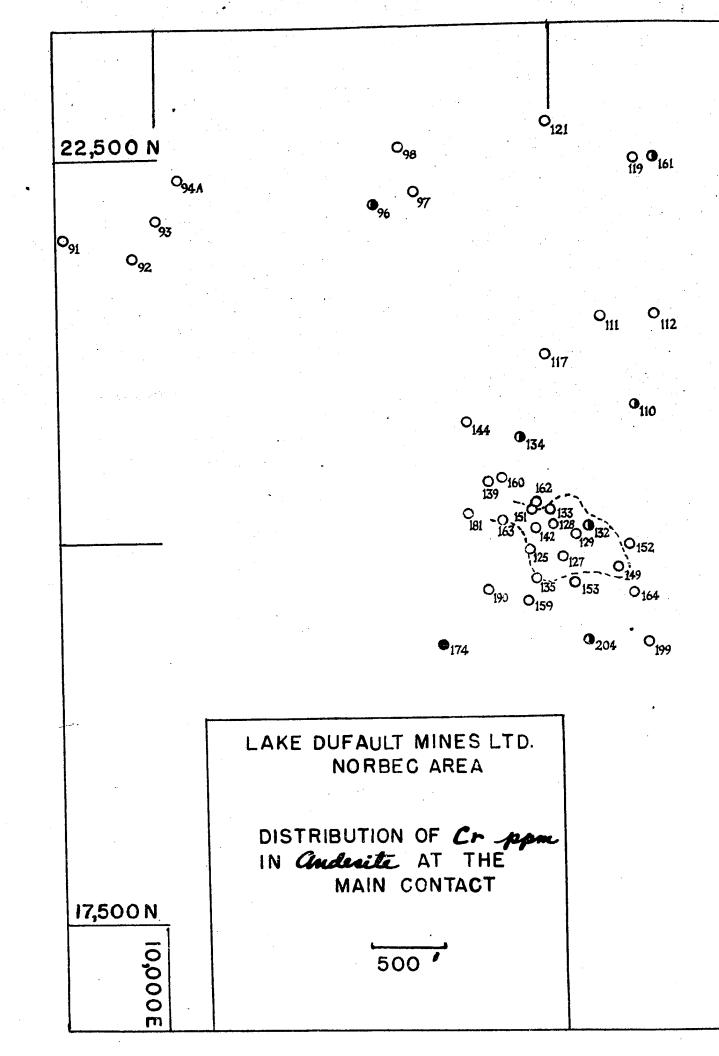
0.75-1.0 1.01-1.25

Co andesite



0161 O₁₅₇ 0113 O₁₅₅ ●₁₅₀ 0114 O₁₂₂ **1**20 0116 **●**158 Oil8 199 O₁₂₃ LEGEND Outline of ore O Diamond drill hole < 63 63-80 > 80 15,000 E

Cr andesite



O₁₅₇
O₁₅₅
O₁₅₀
O₁₂₂

O_{ii6}

•ii8

O114

158

123

LEGEND

120

Outline of ore

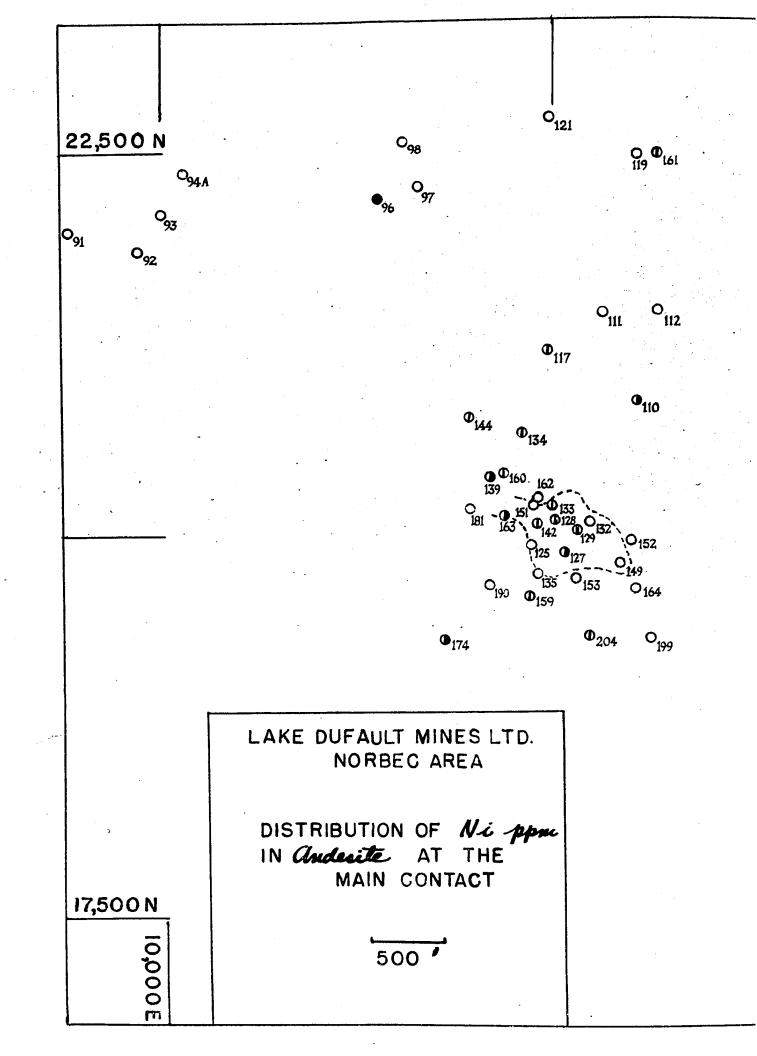
O Diamond drill hole

0 **< 240**

240-300

• > 300

Ni andesite



O₁₅₇
O₁₅₅
O₁₅₀

●120

Ф116

o₁₅₈

●118

0₁₁₄

•₁₂₃

LEGEND

Outline of ore

O Diamond drill hole

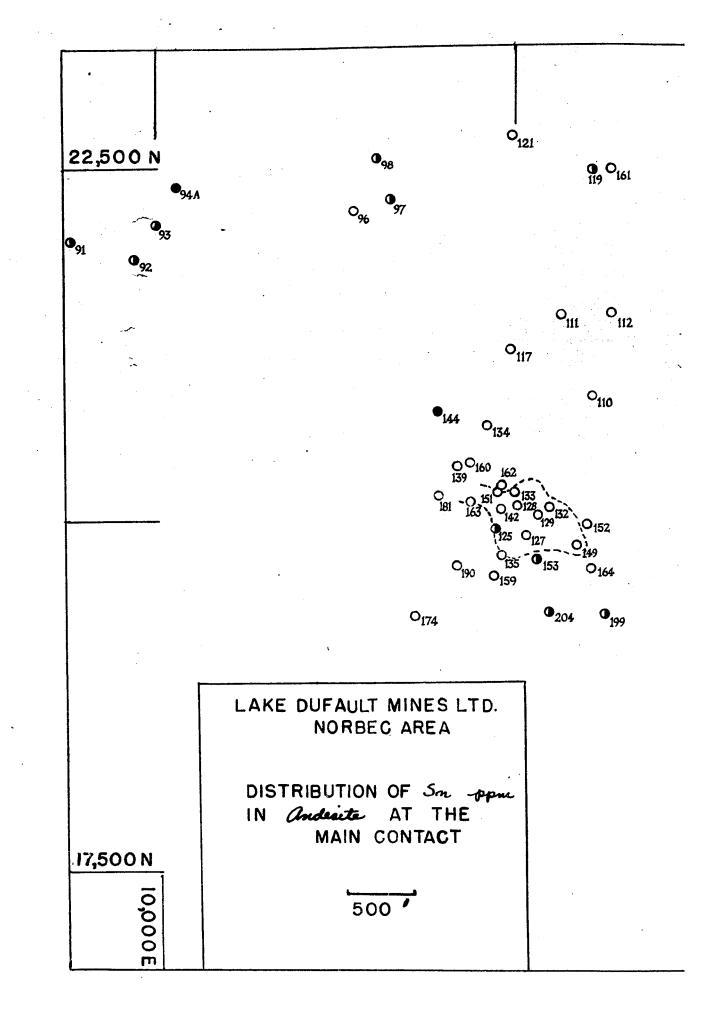
0 (100

Φ 100-150

0 151-200

> 200

Sn andesite



61

12

9

157

O₁₅₅ O₁₅₀

0114

O₁₂₂

0120

0116

0158

Oil

O₁₂₃

LEGEND

Outline of ore

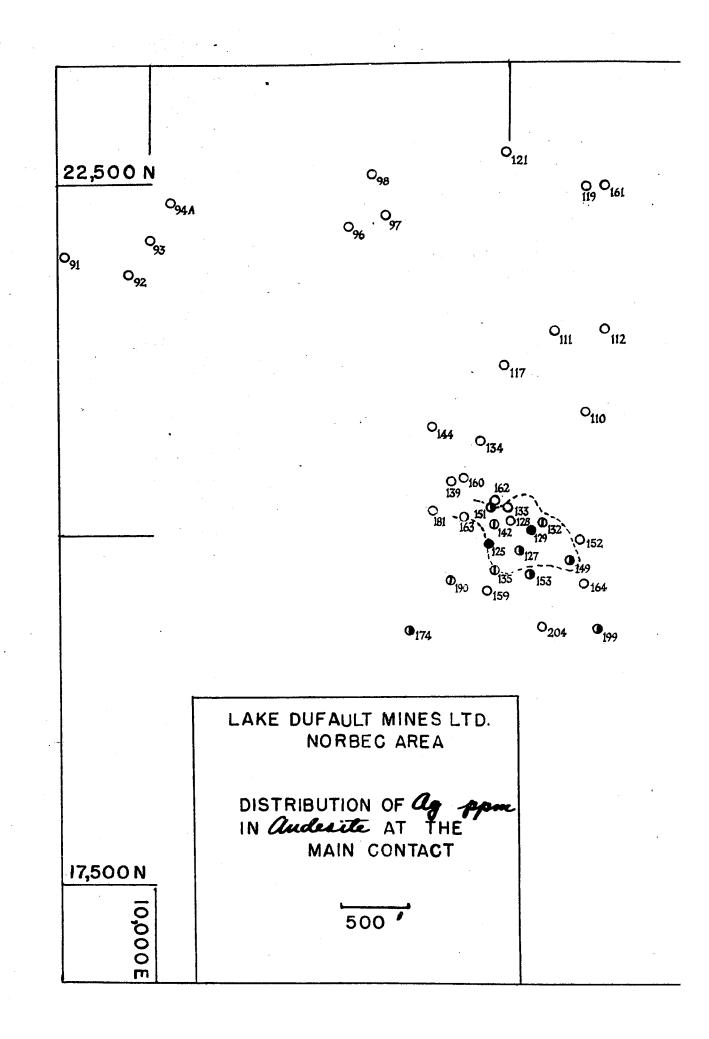
O Diamond drill hole

< 5.0

5.0-7.5

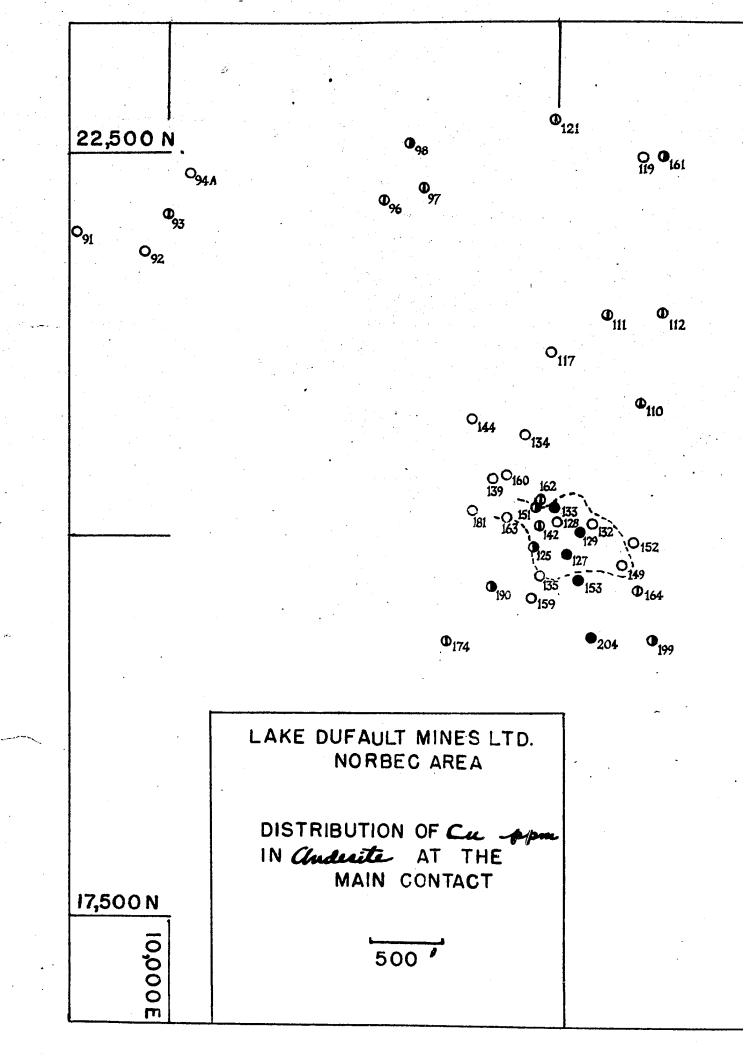
> 7.5

Ag andesite



161 112 O₁₅₇ 0113 O₁₅₅ 0114 · 0₁₂₂ 0120 O₁₅₈ 0116 Oin **>9** O₁₂₃ LEGEND Outline of ore O Diamond drill hole < 0.35 0 0.35 - 0.50 Φ 0.51- 1.0

Cu andesite



O₁₅₅ Φ₁₅₀
O₁₁₄
O₁₂₂
O₁₂₂
O₁₁₆
Φ₁₅₈
O₁₂₃

LEGEND

Outline of one

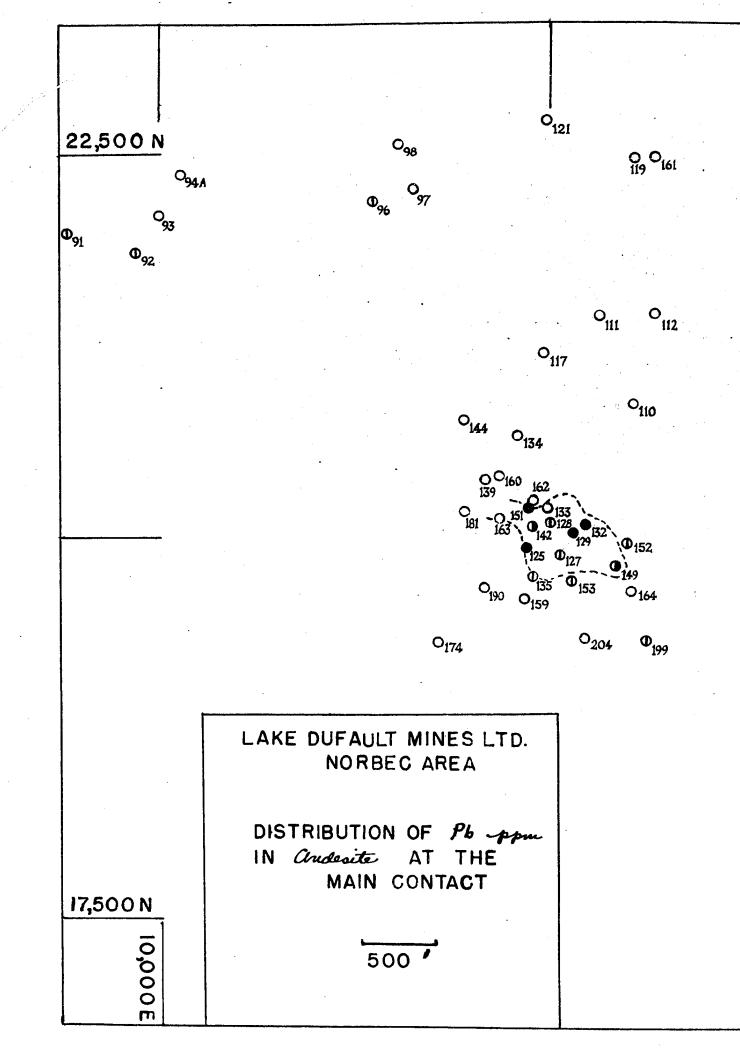
< 100

250

O Diamond drill hole

100 - 150 150.1 - 250

Pb andesite



O₁₅₇
O₁₅₅
O₁₅₀

0114

0122

0120

0116

0₁₅₈

Oiia

O₁₂₃

LEGEND

Outline of ore

O Diamond drill hole

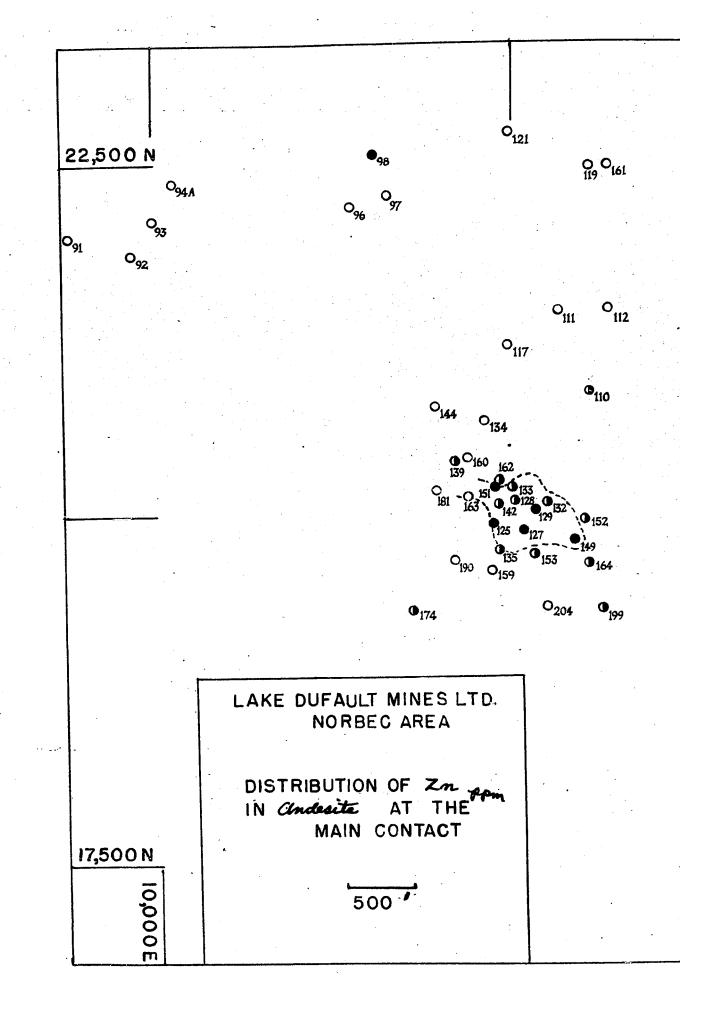
0 > 1.5

① 1.5 - **3.0**

3.01-6.0

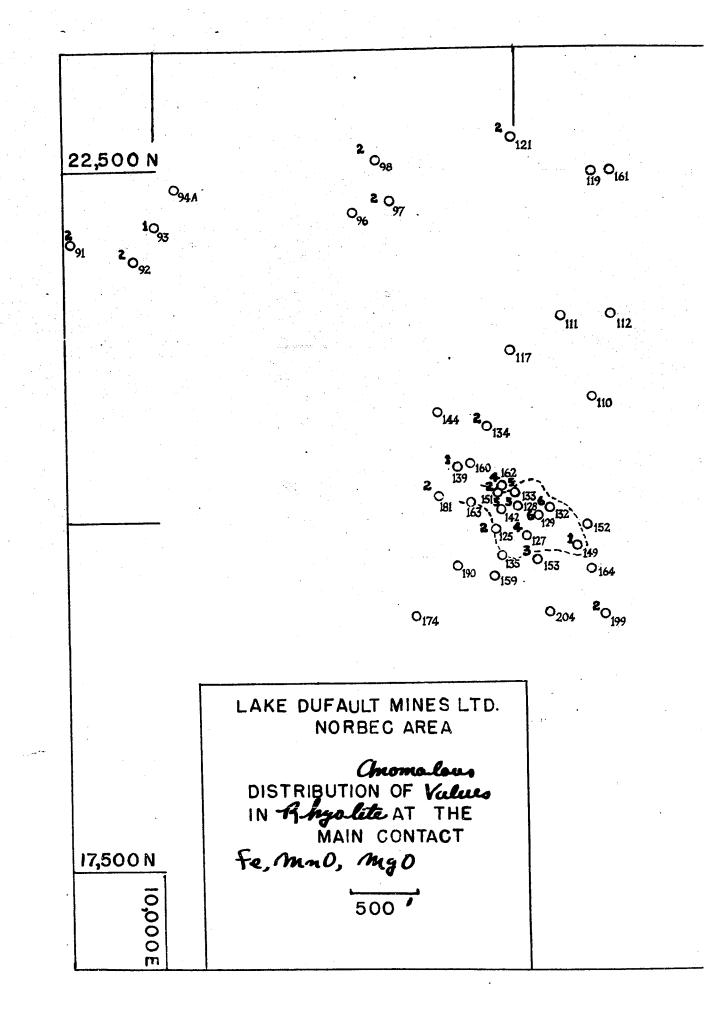
● 〈 6.0

Zn andesite



0₁₅₇ 0113 0114 O₁₂₂ O₁₂₀ 0116 0₁₅₈ 0118 O₁₂₃ LEGEND Outline of ore O Diamond drill hole < 250 250-500

Anomalous Values
Fe, MnO, MgO
rhyolite
Figure 34



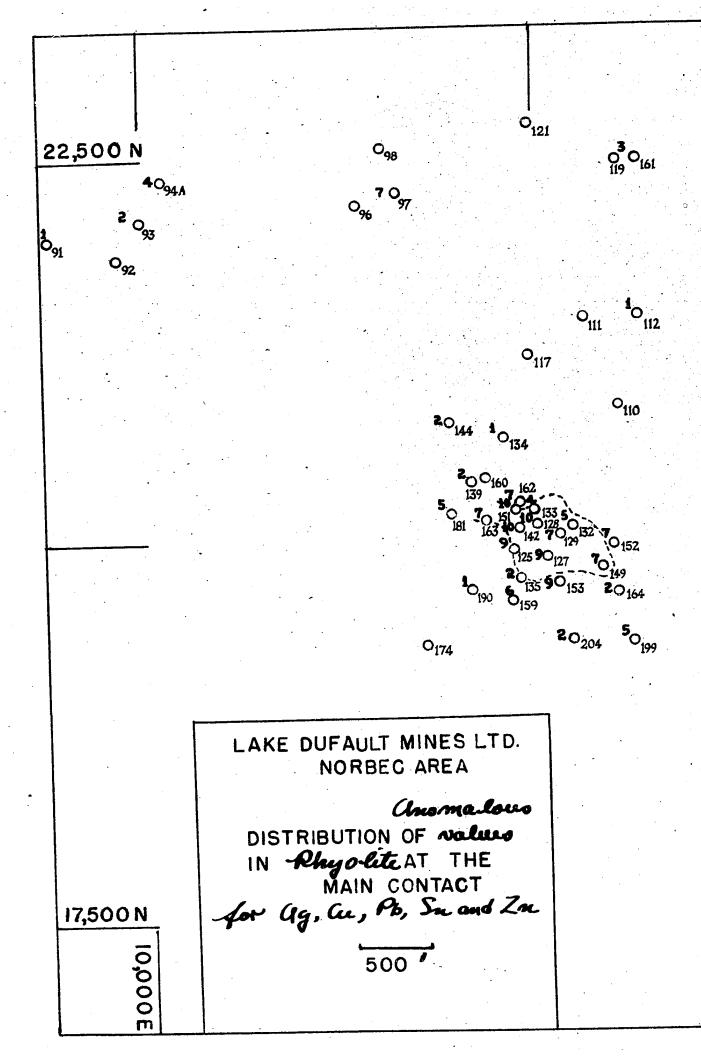
O₁₅₇
O₁₅₅
O₁₅₀
O₁₁₄

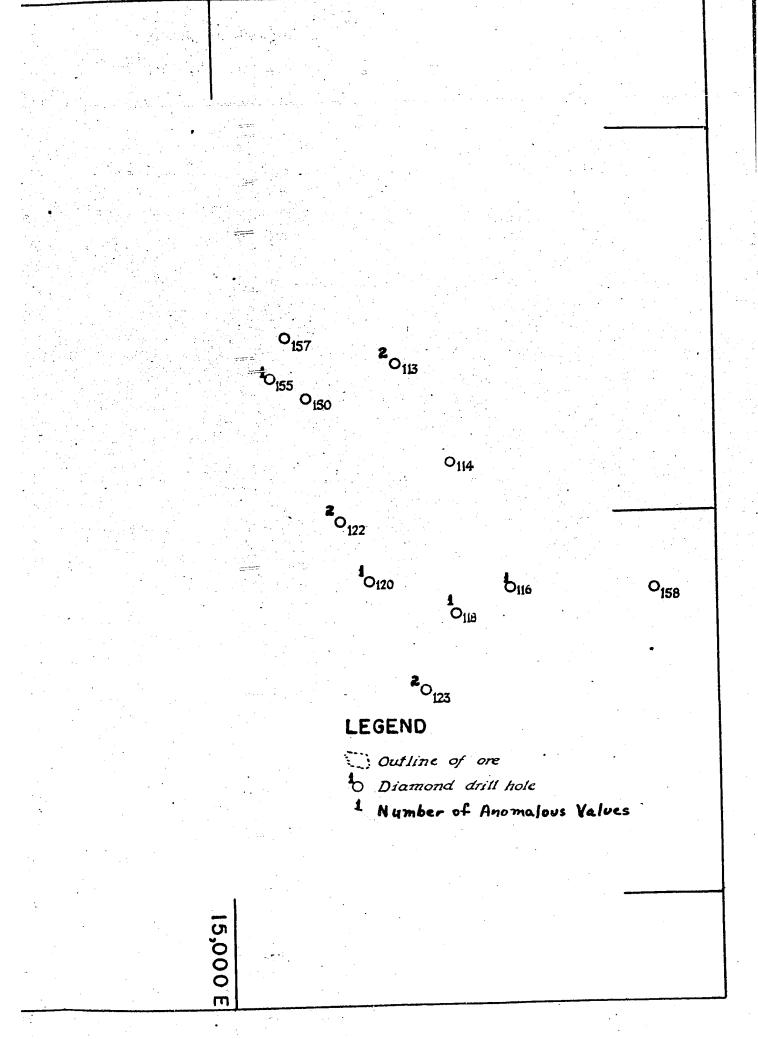
!O₁₂₂
O₁₂₀
O₁₁₆
O₁₅₈
O₁₁₉
!O₁₂₃
LEGEND
Outline of ore

10 Diamond drill hole

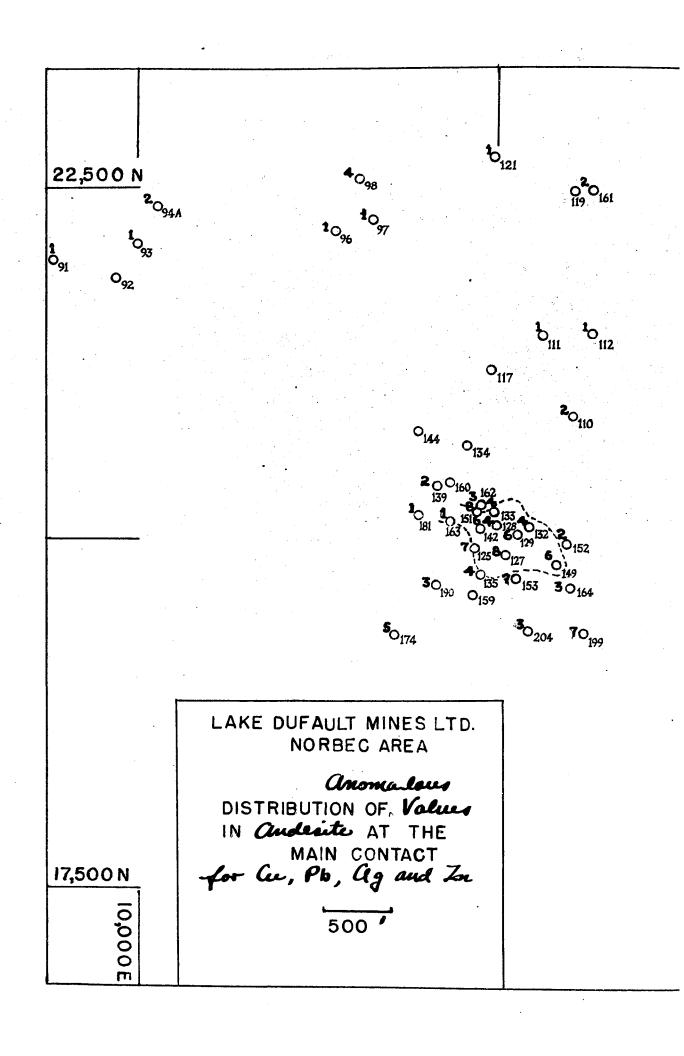
Number of Anomalous Values

Anomalous Values
Ag, Cu, Pb, Sn, Zn
rhyolite
Figure 35





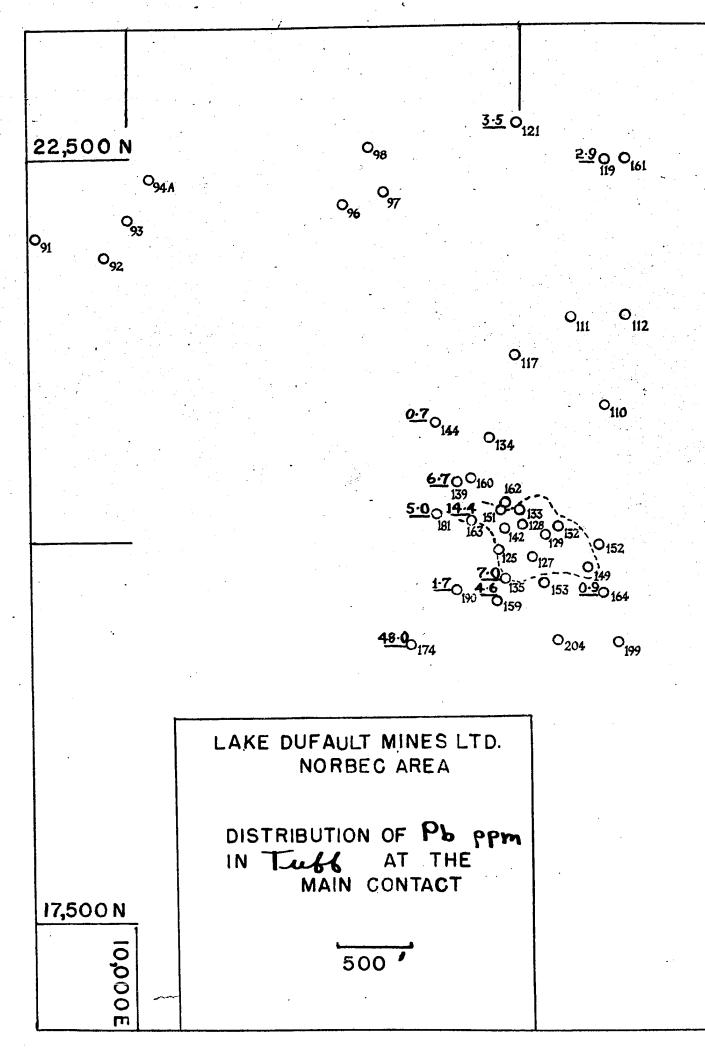
Anomalous Values
Cu, Pb, Ag, Zn
andesite
Figure36



0114 O₁₂₂ O₁₂₀ 0116 O₁₂₃ LEGEND Outline of ore

O Diamond drill hole 1 Number of Anomalous Values

Pb tuff



5:3 O₁₅₇ 25:3 O₁₅₅ 1:8 O₁₅₀

O₁₂₂

O₁₂₀ 2.8_{O₁₁₆}

0114

9<u>·3</u> O₁₅₈

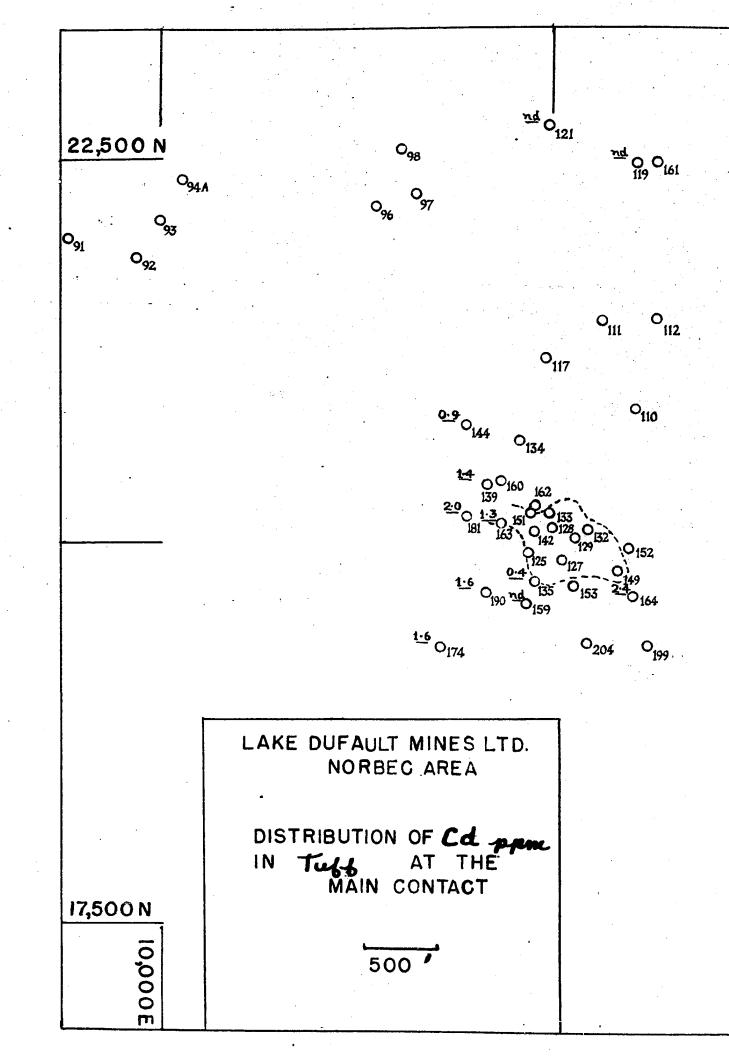
O₁₂₃

LEGEND

Outline of one

O Diamond drill hole

Cd tuff



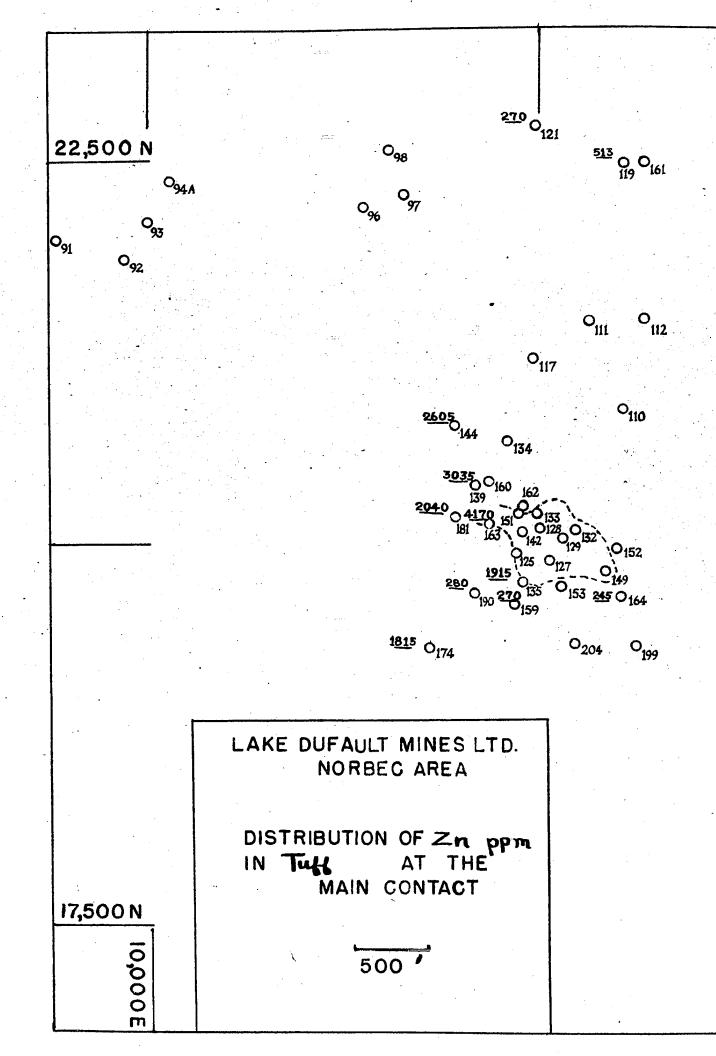
O-T O157
O-T O155
O100
O114
O122
O120
O120
O123
LEGEND
O123
LEGEND
Outline of ore

× O Diamond drill hole
× = Cd ppm

15,000 E

1·1 O₁₅₈

Zn tuff Figure 39



2140 O₁₅₇ 1945 O₁₅₅ 438 O₁₅₀

0122

O₁₂₀ 1225 O₁₁₀ O₁₁₀

0114

2280 O₁₅₈

O₁₂₃

LEGEND

Outline of ore

X
O Diamond drill hole
X = Zn ppm

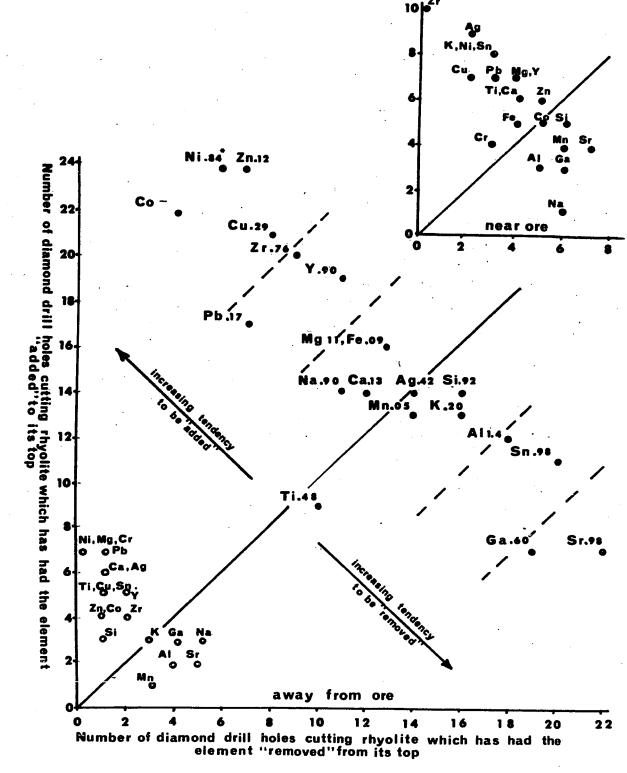


Fig. 40. Systematic Vertical Variation of the Elements in Waite Rhyolite. (o - 2 samples only)*R

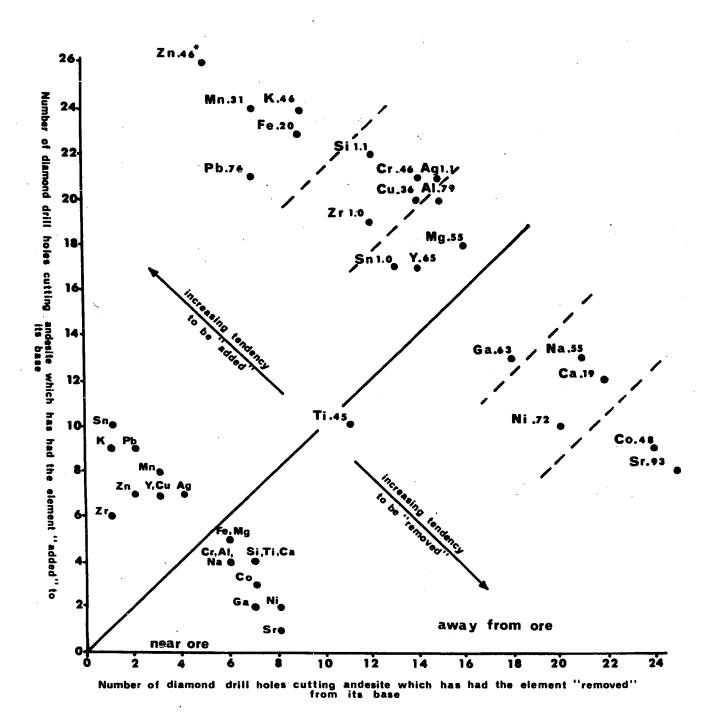


Fig. 41. Systematic Vertical Variation of the Elements in Amulet Andesite. ${}^*\!R$

variation, the plot for that element would fall close to the forty-five degree line which extends from the origin. Results from drill holes near to the ore body are plotted separately as are results from drill holes represented by only two samples.

The proportion of analytical error in a total variation, R, is given for each element on the graphs. Even though R might be large (0.7-1.0) the fact that the highest or lowest values are biased with respect to the main contact still has some significance as analytical errors in analyses performed in no special order (haphazardly) should be randomly distributed (not in a strict, statistical sense).

Results of these tests are summarized on tables 8 and 9.

DISCUSSION OF RESULTS

A review of the chemical data (tables 8 and 9) shows that there are three areas where systematic variations were detected in the contents of elements in andesite and rhyolite. These areas are:

- 1. An area of lateral variations close to the ore body, principally in rhyolite but also in andesite (figs. 12 to 36).
- 2. An area of systematic vertical variations near to and away from the ore body at the top of the rhyolite (fig. 40).
- 3. An area of systematic vertical variations near to and away from the ore body at the base of the andesite (fig. 41).

Area 1 overlaps areas 2 and 3.

Diamond drill hole numbers 125, 127, 128, 129, 132, 133, 135, 142, 149, 151, 163.

Table 8
SUMMARY OF CHEMICAL VARIATIONS IN THE WAITE RHYOLITE

LATERAL VARIATION

VERTICAL VARIATION TOP OF RHYOLITE

close to ore body and in fractured and mineralized zone		away from ore	obscure	near ore (may be obscured by lateral changes toward ore) ²	
Added	Removed	Added Removed	Added	Removed	
Fe (total as Fe ₂ 0 ₃) Mn Mg Zn Cu Pb Ag Sn Co ¹	Na (Si?) (Ca?) (Sr?)	Ni Ga Zn Sr Co (Al?) Cu (Sn?) (Zr?) (Y?) (Pb?)	Zr Ag K Ni Sn (Cu?) (Pb?) (Mg?)	Na (Ga?) (Sr?)	

^{1.} Most of the sampled rhyolite contains less than detectable amounts of Co.

Because of the small number of holes involved less reliability should be placed on these results.

Table 9
SUMMARY OF CHEMICAL VARIATIONS IN THE AMULET ANDESITE

LATERAL VARIATION

VERTICAL VARIATION BASE OF ANDESITE

over the ore body		away from ore		near ore (may be obscured by lateral changes toward ore)	
Added	Removed	Added	Removed	Added	Removed
Ag Cu Pb Zn		Zn Mn K Fe Pb (Si?) (Cr?) (Ag?) (Cu?)	Co Sr (Ni?) (Ca?) (Na?)	Sn K Pb (Zr?) (Zn?) (Mn?) (Y?) (Cu?) (Ag?)	Sr Ni Ga (Co?) (Si?) (Ca?) (Ti?)

Because of the small number of holes involved less reliability should be placed on these results.

Major Elements

All analyses for major elements in this report are expressed as oxide weight per_cent. Major element analyses in other studies of rock alteration have been converted from oxide weight per_cents to ion grams per cubic centimeter, standard cells, etcetera, but Tooker (1963) shows that, regardless of the conversion, the data show the same gross relationships. No large scale change in porosity has been observed in the sampled volcanic rocks. Tooker (1963) gives a review of pertinent literature on the subject of hydrothermal wallrock alteration.

Silicon

No major mobilization of silicon was detected in the Lake
Dufault rocks but small scale additions or subtractions could have
gone undetected because of insufficiently precise analyses. Lower
amounts of silicon are found in fractured rhyolite beneath the massive
ore body but the sulphide-filled fractures which cut the rock could
easily be the result of the filling of open spaces rather than the
replacement of silicate minerals. The presence of mosaics of unstrained,
clear quartz suggests, however, that a breakdown of silicate structures
has occurred in this fractured zone. The data suggest a possible
addition of silicon to the base of the Amulet andesite.

Aluminum

No major mobilization of aluminum was detected but, as was the case with silicon, a small scale variation might be masked by analytical errors. Aluminum is mobile in strongly acid thermal waters commonly found at the surface in volcanic areas (Zelenov, 1965) and the degree to which aluminum has been leached gives an idea of the extent of acid

alteration in the rock. It is possible that aluminum was removed slightly from the top of the Waite rhyolite. That extreme acid leaching conditions existed at a few places in the Noranda volcanic district is suggested by the occurrence of talc in the alteration zones of the Quemont Mine (Riddell, 1952) and the Horne Mine (Price, 1933).

Aluminum is stable in areas of <u>subaerial</u> solfataric-hydrothermal alteration only as alumite, a mineral which has not been found in the Noranda area.

Sodium

Sodium practically has been eliminated from rhyolite in the altered zone under the massive sulphide ore body. The removal of sodium from the top of the rhyolite occurred only near the ore body. Sodium may have been removed from the alteration zone at the base of the andesite away from the known ore body. The loss of sodium from the Lake Dufault rocks follows the pattern of sodium loss in many other examples of wall rock alteration (Tooker, 1963) as well as the weathering of soils and rocks and accompanies the alteration of these materials to clay minerals. Sodium is one of the first elements to be leached from wall rock undergoing argillic alteration by acid solutions (Mukaiyama, 1959) and it is possible that at least the rhyolite below massive ore has been altered in this way.

Areas of present day <u>subaerial</u> hydrothermal activity have been described in the Kamchatka and Kurile Islands (Naboko, 1962; Zelenov, 1965), Japan (Mukaiyama, 1959), the Valley of the 10,000 Smokes (Lovering, 1957) and at Wairakei (Steiner, 1953). Surface rocks in these areas are characterized by the acid leaching of alkali metal

and alkali earth elements, and, in some places iron, silicon, aluminum and titanium.

Ames et al (1958) describes the <u>subaqueous</u> alteration of a tuff in which the alkali metal and alkali-earth elements were not leached.

Potassium

Except for a possible addition to the top of the rhyolite near ore, potassium metasomatism does not seem to have occurred in the rhyolite even under the extreme conditions found in the altered and mineralized zone beneath the massive ore body. On the other hand, the base of the andesite near to and away from the ore body is characterized by sharp increases in potassium. Not all of the andesite which forms the foot wall of massive ore has an unusually high content of potassium.

Potassium in rocks is concentrated mainly in potassic feldspars and micas which are much more stable in altered zones than sodic rockforming minerals. However, even potassic feldspars break down under extreme alteration, releasing potassium to the traversing solutions. The fact that it was not removed from the altered and mineralized zone below the Lake Dufault massive ore body requires a special explanation. A very reasonable one is offered by Hemley (1964). The mobility of potassium is directly related to the stability of sericite. The stability of sericite is controlled largely by the K+ /H+ activity ratio, and the relative activities of Ca++, Na+, and Mg++ as well as K+ ions. If the activities of the Ca++, Na+, and Mg++ ions are high enough, the stability field of sericite may be displaced by that of a clay mineral or chlorite and K+ may be leached from the rocks. If these

suggestions are correct, the fact that a large amount of sodium is leached in the altered rocks (particularly beneath the massive ore) while potassium is fixed means that the activity of K⁺ relative to H⁺, Na⁺, Ca⁺⁺, and Mg⁺⁺ ions in solution was great enough to allow the formation of sericite.

Magnesium

Large amounts of magnesium have been added to the rhyolite beneath the massive sulphide ore body while smaller amounts may have been added to the top of the rhyolite near ore. The andesite shows no effect of magnesium metasomatism. The addition of magnesium in the altered zone beneath the ore body is marked by the development of chlorite, biotite, cordierite and anthophyllite. As suggested in a previous section, the assemblage cordierite-anthophyllite-biotite was probably produced by the thermal metamorphism of a pre-existing chlorite-sericite alteration zone. If the explanation offered for the distribution of sodium and potassium beneath the massive ore is correct, the presence of large amounts of magnesium presents a problem.

In recent years much theoretical and experimental work has been applied to the problem of the transportation and deposition of ore elements. Helgeson (1964) presents a good review of the various mechanisms which have been suggested to account for certain features common to many ore deposits. In spite of recent advances in the chemistry of hydrothermal solutions, no single mechanism has been suggested which would explain magnesium metasomatism as typified by the Lake Dufault deposit and others in the Noranda area. It is not known if a solution capable of leaching large amounts of sodium while

Table 10
FeO CONTENT OF WAITE RHYOLITE

DDI	H Fe0(%) ¹	۶_{е2}0₃(%)²	Fe ₂ 0 ₃ /(Fe ₂ 0 ₃ +Fe0)
t 164	4 3.33 3.74 1.59 2.00	1.4 0.3 0.9 1.1	.298 .075 .360 .355
t 13	4 4.08 3.54 3.04 3.03	0.0 0.9 5.8 1.1	.00 .205 .659 .268
t 13 o p	2 3.50 10.09 4.33 7.23	1.7 4.2 2.0 2.7	.327 .294 .316 .273
t 15 o p	3 3.52 2.90 3.31 5.54	3.1 2.5 1.4 6.2	•470 •463 •290 •30

^{1.} Analyses by H. Soutar, Dept. Geol. Sciences, McGill University.

^{2.} Calculated from total Fe as Fe203 by X-ray fluorescence.

Table 11
FeO CONTENT OF AMULET ANDESITE

	DDH	FeO(%) ¹	Fe ₂ O ₃ (%) ²	Fe ₂ 0 ₃ /Fe ₂ 0 ₃ .	+Fe0)
b a s	164	7.21 7.30 6.91	1.5 3.1 1.4	.172 .298 .169	
8 8	134	5.56 6.26 7.61 8.73	1.8 1.5 2.2 2.6	.243 .192 .224 .230	
b a s	132	7•86 7•95	2.0 2.4	.202 .231	
8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	153	7•41 6•75	3.6 1.8	.327 .209	

^{1.} Analyses by H. Soutar, Dept. Geol. Sciences, McGill University.

^{2.} Calculated from total Fe as Fe203 by X-ray fluorescence.

favoring the development of sericite could also, at the same time, be introducing large amounts of magnesium into the wall rocks. At the present time the writer favors the idea that the rhyolite beneath the massive ore has undergone two stages of metasomatism, the first characterized by the leaching of sodium and the second characterized by the addition of magnesium.

Magnesium metasomatism, marked by the development of montmorillonite and/or chlorite, has been described from areas of recent hydrothermal activity (Steiner, 1953, Wairakei; Naboko, 1962, Kamchatka and Kurile Islands). This type of alteration, which occurs below zones of acid leaching, is generated by near neutral sodium-chloride hydrothermal solutions. According to Ames (1958) a magnesium-rich bentonite from California was formed by additions of magnesium from the water of an alkaline lake.

Calcium

The result of calcium metasomatism in the Waite rhyolite is too complicated to be outlined by the sampling scheme followed in this work. Calcium has been leached and, in adjacent areas, added to the rhyolite under the massive sulphide ore body (chart 11, Appendix B). Calcium may have been removed in some places at the bottom of the andesite.

Iron (total as Fe203)

Iron has been added in large amounts to fractured rhyolite, beneath the massive sulphide ore body and most likely is present in chlorite, biotite, cordierite and anthophyllite as well as in sulphides. A small amount has been added to the base of the andesite. A few wet chemical analyses of FeO are presented in tables 10 and 11. These

analyses suggest that Fe++ may be concentrated in places at the top of the rhyolite. Away from the altered zone below the massive ore body, iron in rhyolite is most likely present in pyrite, pyrrhotite, magnetite, and chlorite. The same is probably true of extra amounts of iron in the base of the andesite.

Manganese

Rhyolite below the massive sulphide ore body has received additional amounts of manganese. Manganese has also been added to the base of the andesite near to and away from ore. Garnets have been identified from the alteration zone beneath the ore body. Sphalerite, which often contains significant amounts of manganese, is distributed widely in the altered zone and close examination probably would result in the finding of manganese oxides or carbonates.

Titanium ·

Titanium was immobile in all alteration zones in the Lake
Dufault rocks. Titanium is found mostly in refractory minerals such
as ilmenite, sphene, rutile and leucoxene but it also replaces iron in
magnetite, and silicon, aluminum, iron and magnesium in biotite and
amphibole. Ultra-acid thermal waters in volcanic areas can leach titanium
from rocks (Zelenov, 1965) and the fact that titanium, like aluminum,
was immobile in the alteration zones of the Lake Dufault mine shows that
the solutions causing alteration were not extremely acid.

Trace Elements

Because analyses of mineral separates were not attempted in this work, statements as to the mode of occurrence of trace elements

(as well as major elements) in the Lake Dufault rocks are speculative and subject to test. For convenience, the trace elements are split into three groups: A. trace elements "added" to altered zones, B. trace elements "removed" from altered zones, and C. trace elements "removed" from one zone and "added" to others.

Group A: Trace elements "added" to eltered zones (Cu. Zn. Pb. Aq. Sn. Cr. Zr. Y)

Of the group A elements only copper, zinc and lead are added consistently to all three alteration zones. Silver has been added to all zones except the top of the rhyolite away from the ore body. these four elements zinc is added most consistently and in the greatest amounts, copper less so, lead much less so and silver barely so. It is interesting to note that both the amount and consistency of metasomatism involving zinc, copper, lead and silver are proportional to their concentration in the Lake Dufault ore body. Fractured rhyolite under the ore body contains the largest concentrations of these elements whereas much smaller amounts are found at the top of the rhyolite and the bottom of the andesite. Sphalerite is the most important carrier of zinc in these altered zones. The zinc spinel, gahnite, has been reported in an alteration zone at the Amulet mine (Riddell. 1952). Sphalerite, chalcopyrite, pyrite and pyrrhotite are disseminated and in fractures in rhyolite beneath the massive ore body and at the base of the andesite. The writer wishes to emphasize the previously noted fact that, although the rhyolite is fractured in many places, sulphides are disseminated widely and finely and are present very rarely in veinlet form in rhyolite away from the altered and mineralized zone below massive When breccias are noticeable in the rhyolite, sulphides can be

seen to be confined to the breccia matrix. Quartz-carbonate-chlorite veinlets, essentially barren with respect to sulphides, are found often in rhyolite which contains relatively large amounts of finely disseminated sulphides.

Tin was added to the rhyolite and to the bottom of the andesite near the ore body but possibly was removed from the top of the rhyolite away from the ore.

Chromium is present in detectable amounts in the rhyolite mainly near the ore body and may have been added to the base of the andesite.

Zirconium may have been concentrated at the bottom of the andesite and the top of the rhyolite near to and away from the ore.

As most zirconium in rocks is present in the refractory accessory mineral zircon, it is difficult to account for its apparent enrichment in these altered zones. Griffitts and Rader (1963) may have found a similar situation in Utah where beryllium and fluorine (as well as Fe, Mn, Zn, Pb, Sn and the rare earths) are concentrated in the top twenty five feet of a 150 foot thick tuff.

Yttrium may have been added to the base of the andesite and the top of the rhyolite.

Group B: Trace elements "removed" from altered zones (Sr. Ga)

Strontium was leached in places from all three alteration zones. Strontium is known to accompany calcium in minerals, especially feldspar and mica, and would be leached along with calcium during the alteration of these minerals.

Gallium has been removed from the top of the rhyolite near to and away from ore and from the base of the andesite near ore. Gallium ions may replace Al⁺⁺⁺ in minerals having a mica structure as well as Fe⁺⁺⁺ and Zn⁺⁺ in magnetite and sphalerite. Tooker (1963) reports that gallium generally is leached from sericitized rock of the Front Range Mineral Belt, Colorado. Gallium becomes separated from aluminum during supergene alteration in both acid and alkaline environments (Borisenok, 1959).

Group C: Trace elements "added" to or "removed" from altered zones (Co. Ni)

Cobalt has been added to the rhyolite beneath the ore body.

Both nickel and cobalt have been added to the top of the rhyolite and have been removed from the base of the andesite.

Cobalt and nickel can occur in hornblende and biotite as well as sulphides and their behavior in the alteration zones of the Lake Dufault mine probably is a reflection of this dual nature. Cobalt and nickel are removed readily from host minerals during alteration. It is probable that the same solutions which added potassium to the base of the andesite removed cobalt and nickel along with sodium, calcium, strontium and gallium. Originally containing very low amounts of cobalt and nickel, the top of the rhyolite probably received additions of these elements in the form of sulphides along with the sulphides of iron, zinc, copper, lead, and silver.

Summary and Conclusions

Stratigraphically below and adjoining the Lake Dufault massive sulphide ore body is a highly altered, fractured and mineralized zone

that most certainly was the channel for the ore-forming solutions. The ore elements plus tin, lead, cobalt, iron, manganese and magnesium have been added in large amounts to this zone; sodium and possibly calcium and strontium have been removed and, significantly, aluminum, titanium and potassium were immobile. These chemical features place certain limits on the nature of the ore-forming solution.

The immobility of eluminum, titenium and potassium in the altered zone indicates that the ore-forming solutions were never extremely acid or alkaline. The leaching of sodium and the addition of magnesium is explained best by assuming two stages of hydrothermal activity or at least a separation of early acid and late, near neutral or alkaline, magnesium-rich solutions, possibly as suggested by Korzhinskii (1963) (the "acid-base filtration effect"). The separation of copper and zinc on both mine and district scales and the association of magnesium metasomatism only with copper mineralization strongly suggests that some mechanism of hydrothermal differentiation involved both ore and major elements and played a significant part in the formation of the sulphide ores in the Noranda area. The mineral assemblage now present in the highly altered channel is probably the result of the thermal metamorphism of a pre-existing chlorite and sericite alteration.

Strontium, gallium and, possibly, slight amounts of aluminum have been removed from the top of the rhyolite in places throughout the Norbec area. Sodium has been removed from the top of the rhyolite only near ore. Iron, magnesium, potassium, silicon and titanium have not been removed from the top of the rhyolite.

Studies of areas of <u>subaerial</u> hydrothermal activity show that surface rocks in these areas are characterized by the acid leaching of

alkali metal and alkali earth elements, and, in some places, iron, silicon, aluminum and titanium. This may not be true of weakly acid, neutral, or alkaline, <u>subaqueous</u> environments. Sulphur-rich solutions passing into a water cover instead of the atmosphere would be less likely to leach the acid-solumble elements from the rocks because the restricted availability of oxygen would inhibit the production of acid sulphate ions.

The writer concludes that the aforementioned chemical features of the top of the Weite rhyolite, along with the presence of bedded, cherty tuff within and on top of the rhyolite and the presence of pillows at the base of the andesite, all suggest that the deposition of the Waite rhyolite and any solfataric-hydrothermal activity at its surface occurred under water. According to Mukaiyama (1959), accumulations of sulphur which form one of the major classes of recent Japanese sulphur deposits are formed in crater lakes or ponds. Sulphur sinks to the bottom of the lake to form a "sedimentary" deposit.

Zirconium and yttrium may be concentrated at the top of the rhyolite but neither the mode of occurrence of these elements nor the reason for their concentration is apparent.

Nickel, cobalt, copper, zinc and possibly lead, elements which are present in the Lake Dufault ore body, have been added to the top of the rhyolite in many places throughout the Norbec area. The greatest amounts of these elements, in addition to certain amounts of iron, manganese, magnesium, silver and tin, occur close to the Lake Dufault ore body.

It has been emphasized in previous sections that, excepting the highly mineralized zone below the massive sulphide ore body, sulphides

are present in the Waite rhyolite as fine disseminations in the matrix of pyroclastic breccias and very rarely are found in fractures. Sulphide minerals which have been identified in the rhyolite are pyrite, pyrrhotite, sphalerite and chalcopyrite.

The writer concludes that the elements which have been added to the top of the rhyolite either are present in or are associated closely with sulphides which are genetically related to sulphide ore and which were introduced before fractures were developed in the rhyolite, possibly before the complete lithification of water-saturated pyroclastic breccias.

The Amulet andesite is altered at its base but this alteration neither is confined to nor always present in the area where andesite forms the hanging wall of massive ore. The alteration is characterized by the formation of sericite, the addition of potassium and the removal of cobalt, nickel and, possibly, strontium, calcium and sodium.

Sericite alteration is developed on a smaller scale well within the andesite. The presence of a "Dalmatianite pipe" in andesite between the "Upper A" and "Lower A" ore bodies of the Amulet mine shows that the andesite is susceptible to magnesism metasomatism.

The lack of added magnesium in altered andesite which is in contact with massive ore and the complete absence of altered andesite in places directly over massive ore leads the writer to conclude that it is possible that the massive sulphide ore body was formed prior to the formation of the overlying andesite.

Minor amounts of silver, copper, lead and zinc have been added to the base of the andesite over the massive sulphide ore body and in some places away from ore. This feature of the andesite does not conflict with the writer's interpretation, as a certain amount of contamination of andesite lava would be expected where it came into contact with concentrations of sulphides.

Bedded, cherty tuff contains higher amounts of zinc, cadmium, and lead near the massive ore body. The writer suggests that the tuff was deposited in the later stages of mineralization, possibly at the same time as layered pyrite-pyrrhotite-sphalerite was being formed at the top of the massive sulphides.

In summary, the writer feels that the distribution of the chemical elements in the host rocks of the Lake Dufault mine can be explained most satisfactorily on the basis of the following assumptions. Sulphur-rich hydrothermal solutions ascended through a local channel in the Waite rhyolite and into a water cover whereupon sulphur and other elements associated with ore were deposited. The character of the solutions changed from an early acid stage to a later magnesium-rich, near neutral or alkaline stage. In any case, the solutions were never extremely acid or alkaline. Certain amounts of the elements were dispersed in the water cover and deposited in the water-saturated top of unconsolidated pyroclastic breccia. Cherty, bedded tuff was formed at the later stages of hydrothermal activity along with layered ore at the top of massive sulphides. Andesite lava passed into the water cover and onto the massive sulphide body. Where andesite lava came into contact with sulphides the base of the flow was contaminated with the ore elements. A later alteration of the andesite is not directly related to the presence of ore.

ORE DEPOSITS SIMILAR TO THOSE OF THE NORANDA AREA

Many ore deposits other than those of the Noranda district have been described as having "strata-bound," tabular, zoned ore bodies associated with pyroclastic rock and magnesium metasomatism. Of these, ore bodies of the Mattagami, Buchans, North Coldstream, Shasta and Jerome areas have been noted as occurring with potassium-poor siliceous rock.

Most Japanese copper-pyrite deposits, including the "Kuroko type," are closely related to Mesozoic and Cenozoic pyroclastic rock ("greentuff," "green rock") and are very similar to the Noranda deposits (Hayashi, 1961, 1962a, 1962b; Hashimoto et al, 1962; Ishikawa, 1960).

"Noranda-style" ore deposits occurring in metamorphic terrains are associated with spotted rocks containing biotite, cordierite and anthophyllite. This association has been reported in Paleozoic rocks in Japan (Hitachi mine, Kuroda, 1961), in Precambrian "leptite" rocks in Scandinavia (Skelefte and Orijarvi districts, Geizer, 1962, 1963; Gavelin, 1939; Eskola, 1963) and in Precambrian, high grade metamorphic rocks in Canada (Coronation mine, Froese and Whitmore, 1964).

GEOCHEMICAL PROSPECTING APPLICATIONS

A systematic geochemical survey of the Lake Dufault mine area would aid in the search for new, blind ore bodies. Elements which occur in large amounts in the massive sulphide ore body are present in higher than normal amounts in the wall rocks next to ore. It is difficult to judge from this work the exact shape or extent of this anomalous zone but figure 35 suggests that it extends at least five hundred feet laterally away from ore in rhyolite.

Perusal of the data charts (Appendix B) shows that anomalous values for "ore-trace" elements can be present at least fifty or seventy-five feet into the andesite above massive ore.

This project serves to emphasize an important point regarding the use of dispersion patterns around ore bodies as prospecting aids; that is, there must be a strong geological control over sampling and interpretation of results.

SUMMARY

In the last decade, and particularly in the last few years, some geologists have come to regard certain features of the Noranda massive sulphide deposits as indicative of a close time-space relationship between the massive ore bodies and their host rocks. These features are:

- 1. Most massive sulphide bodies are located at the top of a siliceous pyroclastic unit, often where the pyroclastic rock is unusually thick;
- 2. The ore bodies are tabular and their largest dimensions are essentially parallel to the nearest stratification or flow contact:
- 3. A large scale stratigraphic zoning is present in many deposits regardless of their present attitude;
- 4. Sulphide fragments with contrasting mineral compositions are found side by side in agglomerate; and
- 5. Some deposits exhibit alteration zones containing large amounts of iron and magnesium stratigraphically below but not above massive sulphide ore bodies.

"Dalmatianite" alteration, found near the Lake Dufault, Waite and Amulet ore bodies, is suggested to be the result of the thermal metamorphism of pre-existing chlorite-sericite alteration, the source of heat being the nearby Lake Dufault granodiorite stock.

The Lake Dufault ore deposit is typical of the massive sulphide deposits in the Noranda area.

Variation diagrams and various indices obtained from published chemical analyses of the igneous rocks in the Noranda area show that essentially all of the units sampled belong to one, calc-alkalic, volcanic suite derived from a common parent magma by extreme fractional crystallization. Basaltic rock from the Noranda area contains less sodium and titanium than the average spilite.

The most significant chemical characteristic in the Noranda volcanic rocks is a deficiency of potassium. This deficiency is most apparent in the Waite rhyolite. Young volcanic rock suites having the same characteristic have been reported almost exclusively in the Circum-Pacific region, most notably on the ocean side of island arcs. Some similarity in trace element composition, particularly in the case of lead, has been noted between these low potassium rocks and the Waite rhyolite and Amulet andesite. In volcanic areas at the edges of continents a parental tholeiitic magma could differentiate by fractional crystallization with a minimum of contamination with material from the sial and yet with a sufficient quantity of water to allow the differentiation process to produce extremely siliceous rock. Accumulation of volatile substances during such a differentiation process would result in much explosive volcanic and hydrothermal activity. It is suggested that the Noranda massive sulphide deposits were formed in an environment similar to that of the relatively young sulphur and sulphide deposits of the Circum-Pacific volcanic islands.

Unusually low amounts of cobalt, chromium and nickel in the Waite rhyolite compared with the average felsic rock may be explained by the tendency of these elements to become depleted in residual melts

produced by fractional crystallization. Unusually high amounts of silver, copper and zinc have been added to the Waite rhyolite and Amulet andesite.

A highly altered zone below the massive sulphide orn body marks the channel of the ore-forming solutions. The ore elements plus tin, lead, cobalt, iron, manganese and magnesium have been added to rhyolite in this zone; sodium and possibly calcium have been removed and aluminum, titanium, potassium and possibly silicon were stable. The chemical features of this zone are explained best by assuming two stages of hydrothermal activity or at least a separation of early acid and late, near neutral or alkaline, magnesium rich solutions. The solutions were neither extremely acid nor extremely alkaline. The original alteration minerals were probably chlorite and sericite, the present mineral assemblage being the result of a later thermal metamorphism.

Strontium, gallium and, possibly, slight amounts of aluminum have been removed from the top of the rhyolite in places throughout the Norbec area. Sodium has been removed from the top of the rhyolite only near ore. Iron, magnesium, potassium, silicon and titanium have not been removed from the top of the rhyolite. Nickel, cobalt, copper, zinc and possibly lead, elements which are present in the Lake Dufault ore body, have been added to the top of the rhyolite in many places throughout the Norbec area. The greatest amounts of these elements, in addition to certain amounts of iron, manganese, magnesium, silver and tin, occur close to the Lake Dufault ore body.

The Amulet andesite is altered at its base but this alteration neither is confined to nor always present in the area where andesite forms the hanging wall of massive ore. The alteration is characterized by the formation of sericite, the addition of potassium and the removal of cobalt, nickel and, possibly, strontium, calcium and sodium. Minor amounts of silver, copper, lead and zinc have been added to the base of the andesite over the massive sulphide ore body and in some places away from ore.

Bedded, cherty tuff contains higher amounts of zinc, cadmium and lead near the massive ore body.

After consideration of the geological features of the Noranda district and the Lake Dufault mine area, the writer feels that the distribution of chemical elements in the host rocks of the Lake Dufault mine can be explained most satisfactorily on the basis of the following assumptions. Sulphur-rich hydrothermal solutions ascended through a local channel in the Waite rhyolite and into a water cover whereupon sulphur and other elements associated with ore were deposited. Certain amounts of the elements were dispersed in the water cover and deposited in the water-saturated top of unconsolidated rhyolite breccia. In the later stages of hydrothermal activity, the formation of cherty, bedded tuff in many places at the top of the rhyolite was accompanied by the formation of layered ore at the top of massive sulphides. Andesite lava passed into the water cover and onto the massive sulphide body. Where andesite lava came into contact with sulphides, the base of the flow was contaminated with the ore elements. A later alteration of the andesite is not directly related to the presence of ore.

The presence of certain elements in higher amounts at the top of the rhyolite and to a lesser extent at the base of the andesite near the Lake Dufault ore body suggests that a systematic geochemical survey of the Lake Dufault mine area would be useful in the search for new, blind ore bodies.

CLAIM OF ORIGINAL WORK AND CONTRIBUTION TO KNOWLEDGE

The candidate claims that the following is original work:

- 1. The synthesis of published chemical analyses and geological facts regarding the Noranda area;
- 2. The examination and sampling of thousands of feet of diamond drill core from the property of Lake Dufault mines, examination of surface and underground exposures, and the study of polished sections and thin sections of samples selected from the Lake Dufault mine area;
- 3. The basic design of the sampling procedure and methods of presentation of results; and
 - 4. The performance of all unpublished chemical analyses.

The candidate claims that this work has resulted in the following contributions to knowledge:

- 1. "Dalmationite" alteration zones in the Noranda area are the products of the thermal metamorphism of pre-existing chlorite-sericite alteration zones;
- 2. The rocks of the Noranda area belong to a single volcanic suite which is similar to younger, distinctive suites of potassium-poor volcanic rock which have been found only at the edges of continents and in island arcs; and
- 3. Elements in the host rocks of the Lake Dufault mine are distributed in a manner which, to the best of the candidate's knowledge, has never been previously described and interpreted and which can be explained most satisfactorily by assuming that the host rocks and ore are contemporaneous products of Precambrian volcanic activity.

ACKNOWLEDGMENTS

The writer is grateful to W. C. Martin for suggesting this study.

Collecting samples for chemical analysis was accomplished with the help of D. Quimet and his assistants. The first chemical analyses were produced in the spring of 1963 with the help of R. Beckett and R. Webster. Because of financial support from Falconbridge Nickel Mines Limited and Lake Dufault Mines Limited, the writer was able to spend the summer of 1963 at McGill University, where, assisted by D. Pilbeam and with the advice of J. Jellema, the bulk of the major element analyses and some trace element analyses were completed. By February, 1964, the first report to the companies showed that major element analyses alone would not be useful in finding new ore. As a result of this finding, the companies consented to having the remainder of funds spent on a detailed study of the trace element content in the host rocks of the new discovery in the Norbec section of the property. During the winter of 1963-1964, T. Frizzell and H. Wiggett prepared samples for trace element analyses.

The writer spent the summer of 1964 in the employ of Dr. W. G. Robinson and J. Boldy of the Noranda office of Falconbridge Nickel Mines Limited and gained, through association with these men, much valuable experience in the Noranda area.

During the winter of 1964-1965, R. Bogoch and Mrs. H. Sakrison prepared the bulk of the Norbec samples for trace element analysis.

With more assistance from the two companies and with the help of Mrs. W. Beaton and Miss N. Gill, the writer was able to finish analytical work and begin a compilation of results during the summer of 1965. By the end of August, 1965, about 450 samples were analysed for each of twenty four elements.

Financial aid in the form of the J. Williamson Memorial and the David Stewart Memorial Fellowships was received by the writer during the course of his studies at McGill University.

The writer wishes to thank the officers of Falconbridge Nickel Mines Limited and Lake Dufault Mines Limited for financial support.

Deserving special thanks are: Dr. A. R. Graham of Falconbridge Nickel Mines Limited for his continuous support and helpful suggestions;

J. Boldy of Falconbridge Nickel Mines Limited for sharing his detailed knowledge of the Noranda area with the writer, for reading appropriate parts of the manuscript and for contributing a map of the Noranda area; and D. H. Brown and J. J. Purdie of Lake Dufault Mines Limited for many hours of discussion with regard to the geology of the Lake Dufault Mine, for reading appropriate sections of the manuscript, for guided underground tours and for maps and samples.

The writer wishes to thank his advisors, Dr. J. E. Gill and Dr. G. R. Webber for much support and helpful criticism.

BIBLIOGRAPHY

- Ahrens, L. H., and Taylor, S. R., 1961, Spectrochemical analysis.

 2nd edition, Addison-Wesley, Reading, Mass., pp. 454.
- Ambrose, J. W., 1941, Clericy and La Pause Map Areas, Quebec. Can. Dept. Mines Tech. Surv., Geol. Surv. Can. Mem. 233, pp. 86.
- Ambrose, J. W., 1944, Duparquet-Larder Lake-Rouyn region, Ontario and Quebec. Can. Dept. Mines Tech. Surv., Geol. Surv. Can. Paper 44-29.
- Ames, L. L., Sand, L. B., and Goldich, S. S., 1958, A contribution on the Hector, California bentonite deposit. Econ. Geol., v. 53, p. 22-37.
- Anderson, C. A., and Creasey, S. C., 1958, Geology and ore deposits of the Jerome area, Yavapai County, Arizona, with sections on the United Verde Extension Mine by G. W. H. Norman and on the Cherry Creek Mining district by R. E. Lehner. U. S. Geol. Surv. Profes. Paper 308, pp. 185.
- Am. Soc. Testing Mater., 1951, ASTM manual on quality control of materials. ASTM, Spec. pub. 15-C, pp. 136.
- Auger, P. E., 1942, Advance report of Desvaux Lake Area, Dasserat township, Temiscamingue Co., Quebec Dept. Mines, Geol. Rept. 27, pp. 24.
- Bandy, M. C., 1937, Geology and petrology of Easter Island. Bull. Geol. Soc. Am., v. 48, p. 1589-1610.
- Barth, T. F. W., 1962, Theoretical petrology. 2nd edition, J. Wiley and Sons, N.Y., pp. 416.
- Bass, M. N., 1956, An interpretation of the geologic history of part of the Timiskaming subprovinces, Canada. Unpublished Ph. D. thesis, Princeton Univ., pp. 326.
- Bass, M. N., 1961, Regional tectonics of part of the southern Canadian Shield. J. Geol., v. 69, No. 6, p. 668-702.
- Borisenok, C. A., 1959, Distribution of gallium in the rocks of the Soviet Union. Geochemistry (USSR), No. 1, p. 52-70.
- Bray, R. C. E., 1940, A comparison of the non-opaque minerals of certain parts of the Waite-Amulet area, Que. M.Sc. thesis, McGill Univ., pp. 56.

- Brown, W. L., 1948, Normetal Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 683-691.
- Bruce, E. L., 1932, Arntfield-Aldermac Mines map area, Beauchastel Twp. Quebec Bur. Mines, Ann. Rept., Pt. C, p. 29-91.
- Buffam, B. S. W., 1925, Destor area, Abitibi Co., Que., Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., Pt. C, p. 82-104.
- Campbell, F. A., 1962, Age of mineralization at Quemont and Horne Mines. Bull. Can. Inst. Mining Met., v. 55, No. 605, p. 627-630.
- Campbell, F. A., 1963, Sphalerite-pyrrhotite relationships at Quemont Mine. Can. Mineralogist, v. 7, p. 367-374.
- Carmichael, I. S. E., 1962, A note on the composition of some natural acid glasses. Geol. Mag., v. 99, p. 253-264.
- Casertano, L., 1963. Chilean continent. Pt. XV, Catalogue of the Active Volcanoes of the World Including Solfatara Fields, Intern. Assoc. of Volcano., pp. 55.
- Chao, E. C. T., and Fleischer, M., 1960. Abundance of zirconium in ingeous rocks. 21st, Copenhagen, Rept. Session, Norden, Pt. 1, p. 106-131.
- Cooke, H. C., 1922, Opasatika map area, Timiskaming Co., Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., Pt. D, p. 19-74.
- Cooke, H. C., 1923, Some gold deposits of western Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., Pt. CI, p. 76-125.
- Cooke, H. C., 1925, Gold and copper deposits of western Quebec. Can.
 Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., Pt. C,
 p. 28-151.
- Cooke, H. C., 1930, The Amulet Mine, Que. Trans. Can. Inst. Mining Met., v. 33, p. 398-408.
- Cooke, H. C., 1930, The compound laccolith of Lake Dufault, Que. Trans. Roy. Soc. Can. Sect. 24, Ser. 3, v. 24, p. 89-98.
- Cooke, H. C., James, W. F., and Mawdsley, J. B., 1934, Geology and ore deposits of Rouyn-Harricanaw region. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Mem. 166, pp. 314.
- Coombs, H. A., 1952, Spherulitic breccias in a dome near Wenatchel, Washington. Am. Mineralogist, v. 37, No. 3-4, p. 197-206.

- Coombs, H. A., and Howard, A. P., 1960, United States of America. Catalogue of the Active Volcanoes of the World, including Solfatara Fields, Pt. IX, Intern. Assoc. Volcano., pp. 68.
- Copeland, J. G., 1951, Alteration associated with sulphide mineralization at Waite Amulet Mine, Que. Unpublished M. Sc. thesis, Univ. of Toronto.
- Daly, R. A., 1933, Igneous rocks and the depths of the Earth. McGraw-Hill, N.Y., pp. 598.
- Davidson, S., 1948, Beattie Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 692-700.
- Denis, F., 1933, Composition of Noranda Mines ores. Unpublished M.Sc. thesis, McGill Univ., pp. 34.
- Dresser, J. A., and Denis, T. C., 1944, Geology of Quebec, vol. II, Descriptive Geology. Quebec Dept. Mines, pp. 544.
- Dresser, J. A., and Denis, T. C., 1949, Geology of Quebec, vol. III, Economic Geology. Quebec Dept. Mines, pp. 562.
- Dugas, J. and Hogg, W. A., 1962, An outline of the Rouyn-Noranda area, N. W. Quebec. Can. Mining J., v. 83, No. 4, p. 101-104.
- Dugas, J. and Hogg, W. A., 1962, Geological compilation, Rouyn-Noranda area. Prelim. Map U-265, Quebec Dept. Mines.
- Engel, A. E. J., Engel, C. G., and Havens, R. G., 1965, Chemical characteristics of oceanic basalts and the upper mantle. Geol. Soc. Am. Bull., v. 76, No. 7, p. 719-734.
- Eskola, P., 1963, The Precambrian of Finland. in the Precambrian, ed. Rankama, K., J. Wiley and Sons, N.Y., p. 145-263.
- Ewart, A., 1963, Petrology and petrogenesis of the Quaternary pumice ash in the Taupo area, New Zealand. J. Petrol., v. 4, No. 3, p. 392-431.
- Fenner, C. N., 1926, The Katmai magmatic province. J. Geol., v. 34, No. 7, Pt. 2, p. 673-772.
- Fiske, R., 1963, Subaqueous pyroclastic flows in the Ohanapecosh formation, Washington. Bull. Geol. Soc. Am., v. 74, No. 4, p. 391-406.
- Fleicher, M., 1965, Summary of new data on rock Samples G-1 and W-1, 1962-1965. Geochim. Cosmochim. Acta, V. 29, p. 1263-1284.
- Froese, E., and Whitmore, D. R. E., 1964, Cordierite-Anthophyllite rocks of the Coronation Mine area. Bull. Can. Inst. Mining Met., v. 57, No. 623, p. 318.

- Gass, I. G., Harris, P. G., and Holdgate, W. M., 1963, Pumice eruption in the area of the South Sandwich Islands. Geol. Mag., v. 100, p. 321-330.
- Gavelin, S., 1939, Geology and ores of the Malanas district, Vasterbotten, Sweden. Sveriges Geol. Undersokn Arsbok, Ser. C., 0424, 33.
- Geijer, P., 1963, The Precambrian of Sweden. in the Precambrian, v. I, ed. Rankama, K., J. Wiley and Sons, N.Y., p. 81-143.
- Giblin, P. E., 1964, Geology of the Burchell Lake area. Ontario Dept. Mines Geol. Rept., No. 19, p. 1-39.
- Gilbert, J. E., 1960, Distribution and general characteristics of the massive sulphide deposits of the Province of Que. Trans.,
 Can. Inst. Mining Met., v. LXVIII, p. 69-76.
- Gill, J. E., 1948, Wasa Lake Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 730-734.
- Gill, J. E., 1960, The occurrence of massive sulphide deposits in Canada, Part 1. Introduction. Trans. Can. Inst. Mining Met., v. 63, p. 37-38.
- Gill, J. E., and Schindler, N. R., 1932, Geologyof the Waite-Ackerman-Montgomery property, Deprat and Dufresnoy townships, Que. Trans. Can. Inst. Mining Met., v. 35, p. 398-416.
- Gilmore, P., 1965, The origin of the massive sulphide mineralization in the Noranda district, Northwestern, Que. Proc. Geol. Assoc. Can., v. 16, p. 63-81.
- Goodwin, A. M., 1965, Mineralized volcanic complexes in the Porcupine-Kirkland Lake-Noranda region, Canada. Econ. Geol., v. 60, No. 5, p. 955-971.
- Graham, E. P., 1964, Exploiting a small ore body Vauze Mines Ltd. Bull. Can. Inst. Mining Met., v. 57, No. 624, p. 409-412.
- Graham, R. B., 1954, Parts of Hebecourt, Duparquet and Destor townships. Quebec Dept. Mines, Geol. Rept., No. 61, pp. 64.
- Griffitts, W. R., and Rader, L. F., Jr., Beryllium and fluorine in mineralized tuff, Spor Mountain, Juab County, Utah. U. S. Geol. Surv., Profes. Paper, 475-B, p. 16-17.
- Guimond, R., 1964, Lake Dufault Mines, Mining in Canada, Oct., p. 15-29.
- Gunning, H. C., 1927, Syenite Porphyry of Beauchastel Twp., Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Geol. Sur. Bull., 46, p. 31-41.

- Gunning, H. C., and Ambrose, T. W., 1939, Temiskaming-Keewatin problem.
 Trans. Roy. Soc. Can., v. 33, Sec. 4, p. 19-49.
- Gussow, W. C., 1937, Petrogeny of the major acid intrusives of N. W. Que. Trans. Roy. Soc. Can., Sec. IV, v. 31, p. 129-161.
- Hall, J. D., 1939, Geology of the Lower "A" ore body, Waite Amulet. Unpublished M.Sc. thesis, McGill Univ., pp. 51.
- Hart, E. A. and Gill, J. E., 1948, Arntfield Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 711-718.
- Harvie, R., 1923, Dufresnoy may area, Abitibi district, Quebec. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., Pt, CI, p. 145-150.
- Hashimoto, K., Kamono, H., and Hayashi, S., 1962, On the Uchinotai Kuroko (black ore) deposits, Kosaka Mine. Mining Geology (Japan), v. 12, No. 51, p. 129-142.
- Hawley, J. E., 1931, Granada Gold Mine and vicinity, Rouyn Twp., Temiscamingue Co., Que. Quebec Bur. Mines, An. Rept., Pt. B, p. 3-58.
- Hawley, J. E., 1933, McWatters Mine Gold Belt, E. Rouyn and Joannes Twps. Quebec Bur. Mines, An. Rept., Pt. C, p. 3-74.
- Hawley, J. E., 1948, The Aldermac copper deposit. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 719-730.
- Hawley, J. E., 1948, Francoeur Mines. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 701-710.
- Hayashi, S., 1961, On the mode of occurrence of Kuroko (black ore) in the Motoyama ore deposits of the Kosaka Mine, Akita prefecture, Japan. Mining Geology (Japan), v. 11, No. 47, p. 442-443.
- Hayashi, S., 1962a, On some ore deposits of the Kuroko (black ore) type in the western part of the Kosaka mining district. Mining Geology (Japan), v. 12, No. 51, p. 35-38.
- Hayashi, S., 1962b, On the genetic relation between the Baramori volcanic rocks and ore deposits of the Kosaka Mine, Akita prefecture, Japan. Mining Geology (Japan), v. 12, No. 51, p. 84-92.
- Helgeson, H. C., 1964, Complexing and hydrothermal ore deposition.

 Macmillan, New York, pp. 128.
- Hemley, J. J., and Jones, W. R., 1964, Chemical aspects of hydrothermal alteration with emphasis on hydrogen metasomatism. Econ. Geol., v. 59, No. 4, p. 538-569.

- Ishikawa, T., 1960, Some petrological considerations on the Miocene volcanic activities in green-tuff regions in Japan. J. Fac. Sci. Hokkaido Univ., Ser. IV, v. 10, p. 471-480.
- Ishikawa, T., Minato, M., Kuno, H., Matsumoto, T., and Yagi, K., 1957, Welded tuffs and deposits of pumice flow and nues ardente in Japan. Proc. Inter. Geol. Congr., Mexico, Sect. I, v. 1, p. 137-150.
- Ishikawa, T., and Katsui, Y., 1959, Some considerations on the relation between the chemical character and the geographical position of the volcanic zones in Japan. J. Fac. Sci., Hokkaido Univ., Ser. IV, v. 10, p. 163-182.
- Isshiki, N., 1964, Mode of eruption of Miyake-jima volcano in historic times. Bull. Volcano., v. 27, p. 29-48.
- James, W. F., 1922, Deparquet map area, Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., Pt. D, p. 75-96.
- James, W. F., and Mawdsley, J. B., 1924, Clericy and Kinojevis map areas, Temiscamingue and Abitibi Counties, Quebec. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., Pt. C, p. 99-125.
- Kerr, P. F., 1955, Hydrothermal alteration and weathering, in Poldervaart, A., ed., Crust of the earth - a symposium. Geol. Soc. Am. Spec. Paper 62, p. 525-543.
- Kindle, E. D. 1941, Northeast part Beauchastel Township, Temiscamingue, Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Paper. 45-7.
- Kinkel, A. R., Jr., Hall, W. E., and Albers, J. P., 1957, Geology and base metal deposits of West Shasta, copper-zinc district, Shasta Co., Calif. U.S. Geol. Surv., Profes. Paper 285, pp. 156.
- Kinkel, A. R., Jr., 1962a, The One Knob massive sulphide copper deposit, North Carolina; an example of recrystallized ore. Econ. Geol., v. 57, No. 7, p. 1116-1121.
- Kinkel, A. R., 1962b, Observations on the pyrite deposits of the Huelva district, Spain, and their relation to volcanism. Econ. Geol., v. 57, No. 7, p. 1071-1080.
- Korzhinskii, D. S., 1963, Hypothesis of the advancing waxe of acidic components in postmagmatic solutions. Symposium, Problems in Post-Magmatic Ore Deposition, Geol. Surv. of Czechoslovakia, p. 157-160.
- Kuno, H., 1950, Petrology of Hokone volcano and adjacent areas of Japan. Bull. Geol. Soc. Amer., v. 61, p. 957-1020.

- Kuno, H., 1960, High alumina basalt. J. Petrol., v. 1, p. 121-145.
- Kuno, H., 1962, Japan, Taiwan and Marianas. Catalogue of the Active Volcanoes of the World, including Solfatara fields, Part XI, Intern. Assoc. Volcano., pp. 332.
- Kuroda, Y., 1961, Minor elements in a metasomatic zone related to a copper-bearing pyrite deposit. Econ Geol., v. 56, No. 5, p. 847-854.
- L'Esperance, R. L., 1951, The geology of Duprat township and some adjacent areas, N.W. Que. Unpublished Ph.D. thesis, McGill Univ., pp. 316.
- Lickus, R. J., 1965, Geology and Geochemistry of the ore deposits at the Vauze Mine, Noranda district, Quebec. Unpublished Ph.D. thesis, McGill Univ., pp. 134.
- Lovering, T. S., 1957, Halogen-acid alteration of ash at fumarole #1, Valley of Ten Thousand Smokes, Alaska. Bull. Geol. Soc. Am., v. 68, p. 1585-1604.
- MacGregor, J. G., 1928, Structural features of certain Rouyn ore bodies. Can. Mining J., v. 49, p. 456-460.
- MacGregor, J. G., 1929, Exploration in Rouyn camp. Trans. Can. Inst. Mining Met., v. 32, p. 41-50.
- MacKenzie, G. S., 1941, Halliwell Mine map area, Timiskaming County. Quebec Bur. Mines., Geol. Rept. 7, map 492.
- Mason, B., 1960, Principles of geochemistry. 2nd edition, J. Wiley, pp. 310.
- Maxwell, J. A., Dawson, K. R., Tomilson, M. E., Pocock, D. M., and Tetreault, D., 1965, Chemical analyses of Canadian rocks, minerals and ores. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Geol. Surv. Bull., 115, pp. 476.
- McKenzie, G. S., 1940, Fortune Lake and Wasa Lake map area, Dasserat and Beauchastel townships. Quebec Bur. Mines, Geol. Rept. No. 5, pp. 24.
- McMurchy, R. C., 1948, Powell Rouyn Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 739-747.
- McDuat, W., 1872-73, Report of an examination of the country between Lakes Temaskaming and Abitibi. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Prog. Rept., p. 112-135.
- Minakami, T., Ishikawa, T., and Yagi, K., 1950, The 1944 eruption of volcano Usu in Hokkaido, Japan. Bull. Volcano., v. 11, p. 45–157.

- Moore, J. G., 1962, K/Na ratio of Cenozoic igneous rocks of the western United States. Geochim. Cosmochim. Acta, v. 26, p. 101-130.
- Mukaiyama, H., 1959, Genesis of sulfur deposits in Japan. J. Fac. Sci. Univ. Tokyo, Sect. II, Supplement to vol. XI, pp. 148.
- Naboko, S. I., 1963, Conditions of the present-day hydrothermal metamorphism of volcanic rocks. Intern. Geol. Rev., v. 5, No. 7, p. 850-858.
- Nockolds, S. R., 1954, Average chemical compositions of some igneous rocks. Bull. Geol. Soc. Am., v. 65, p. 1007-32.
- Norman, G. W. H., 1942, The Cadillac belt of Northwestern Quebec. Trans. Roy. Soc. Can., Sec. 4, v. 36, p. 89-98
- Norman, G. W. H., 1948, Major faults, Abitibi region, Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 822-839.
- Oftedahl, C., 1958, A theory of exhalative-sedimentary ores. Geol. For. Forh. Stockholm, v. 80, No. 492, p. 1-19.
- O'Neill, J. J., 1932, The Beatie Gold Mine, Deparquet Twp. Que. Bur. Mines, An. Rept., Pt. C, p. 3-28.
- O'Neill, J. J., 1933, Beatie-Galatea map area. Quebec Bur. Mines, An. Rept., Pt. C, p. 75-114
- Osborn, E. F., 1959, Role of oxygen pressure in the crystallization and differentiation of basaltic magma. Am. J. Sci., v. 257, p. 609-647.
- Parks, W. A., 1904, The geology of a district from Lake Temiskaming northward. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., v. 16, p. 198-228A.
- Peale, R., 1930, Some ore deposits of the Rouyn district, Northwestern Que. Can. Unpublished Ph.D. thesis, Harvard Univ.
- Peale, R., 1931, Geology of the Waite-Ackerman-Montgomery deposit. Trans. Can. Inst. Mining Met., v. 34, p. 198-215.
- Poldervaart, A., 1955, Chemistry of the earth's crust. Crust of the Earth, Geol. Soc. Am., Spec. Paper, 62, p. 119-144.
- Price, P. R., 1933, Geology of the Horne Mine, Noranda, Que. Unpublished Ph.D. thesis, McGill Univ., pp. 288.
- Price, P., 1934, The Horne Mine. Trans. Can. Inst. Mining Met., v. 37, p. 108-140.
- Price, P., 1949, Waite Amulet Mines Limited. Geology of Quebec by J. A. Dresser and T. C. Denis, Quebec Dept. Mines, Geol. Rept. 20, v. 3, p. 361-383.

- Price, P., 1953, Noranda Mines Ltd. (Horne Mine). Geology and Mineral Deposits of Northwestern Quebec, Geol. Am. and Geol. Assoc. of Can. Guide Book for Field Trip No. 10, November 12-14, p. 13-14.
- Price, P., and Bancroft, W. L., 1948, Waite Amulet Mine: Waite Section. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 748-756.
- Putnam, H. M., 1943, Paragenesis of the ore of the Normetal Mine, Abitibi Co., Que. Econ. Geol., v. 38, p. 313-322.
- Relly, B. H., 1960, The geology of Buchans Mine, Newfoundland.
 Unpublished Ph.D. thesis, McGill Univ., pp. 281.
- Riddell, J. E., 1952, Wall rock alteration around base metal sulphide deposits of northwestern Que. Unpublished Ph.D. thesis, McGill Univ., pp. 220.
- Rittmann, A., 1962, Volcanoes and their activity. Interscience Pub., N.Y., pp. 305.
- Robinson, W. G., 1940, Preliminary report on Flavrian Lake district, Beauchastel Twps. Quebec Bur. Mines, Prelim. Rept., No. 145, pp. 7.
- Robinson, W. G., 1941, The Flavrian Lake map area and the structural geology of the surrounding district. Unpublished Ph.D. thesis, McGill Univ., pp. 99.
- Robinson, W. G., 1943, Flavrian Lake area. Quebec Dept. Mines, Geol. Rept., No. 13, pp. 21.
- Robinson, W. G., 1948, Part of northwest quarter of Beauchastel Twp., Rouyn-Noranda Co. Quebec Dept. Mines, Geol. Rept., No. 30, pp. 19.
- Robinson, W. G., 1951, Structural geology and ore deposits of the Rouyn-Noranda district. Proc. Geol. Assoc. Can., v. 4, p. 61-73.
- Roscoe, S. M., 1965, Geochemical and isotopic studies, Noranda and Mattagami areas. Bull. Can. Inst. Mining Met., v. 58, No. 641, p. 965-971.
- Schindler, N. R., 1933, Geology of the Waite-Ackerman-Montgomery property, Duprat and Dufresnoy Twp. Que. Unpublished M.Sc. thesis, McGill Univ., pp. 108.
- Schindler, N. R., 1934, Igneous rocks of Duprat Lake and Rouyn Lake areas, Que. Unpublished Ph.D. thesis, McGill Univ., pp. 165.

- Schmidt, R. G., 1957, Geology of Saipan, Marianas Islands, petrology of the volcanic rocks. U.S. Geol. Surv. Profes. Paper, 280B, pp. 127.
- Schwartz, G. M., 1956, Hydrothermal alteration as a guide to ore in Econ. Geol., 50th An. vol., Pt. 1, p. 300-323.
- Scott, J. S., 1948, Quemont Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 773-775.
- Seigel, H. D., Winkler, H. A., and Boniwell, J. B., 1957, Discovery of the Mobrun Copper Ltd. sulphide deposit, Noranda mining district, p.Q. Methods and Case Histories in Mining Geophysics, Can. Inst. Mining Met., p. 327-245.
- Sharp, J. I., 1964, Precambrian geology and sulphide deposits of the Mattagami area, Quebec. Unpublished Ph.D. thesis, McGill Univ., pp. 202.
- Shaw, D. M., 1961, Element distribution laws in geochemistry. Geochem. Cosmochim. Acta, v. 23, p. 116-134.
- Shiozaki, H., 1961, On the keratophyric rocks associated with bedded cupiferous pyrite deposit of the Nako Mine, Japan. J. of Earth Sci., Nagoya Univ., v. 9, No. 1, p. 173-184.
- Smirnov, S. I., 1963, Statistical estimation of the geochemical background in geochemical prospecti**rg**. Geochemistry (USSR), No. 3, p. 353-364.
- Smirnov, V. I., 1965, Convergence of sulphide deposits. Intern. Geol. Rev., v. 7, No. 6, p. 978-983.
- Stanton, R. L., 1960, General features of conformable "pyrite" ore bodies, parts I and II, Trans. Can. Inst. Min. Met., v. 63, p. 22-36.
- Steiner, A., 1953, Hydrothermal rock alteration at Wairakei, New Zealand. Econ. Geol., v. 48, p. 1-13.
- Stumpfl, E. F., and Clark, A. M., 1964, A natural occurence of CogSg, identified by X-ray microanalysis. Neues Jahrb. Mineral., Monatsh., v. 8, p. 240-245.
- Suffel, G. G., 1935, Relations of later gabbro to sulphides at the Horne Mine, Noranda, Que. Econ. Geol., v. 30, p. 906-907.
- Suffel, G. G., 1948, Waite Amulet Mine: Amulet section. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 757-763.
- Tanton, T. L., 1919, The Harricana-Turgeon Basin, northern Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Mem. 109.

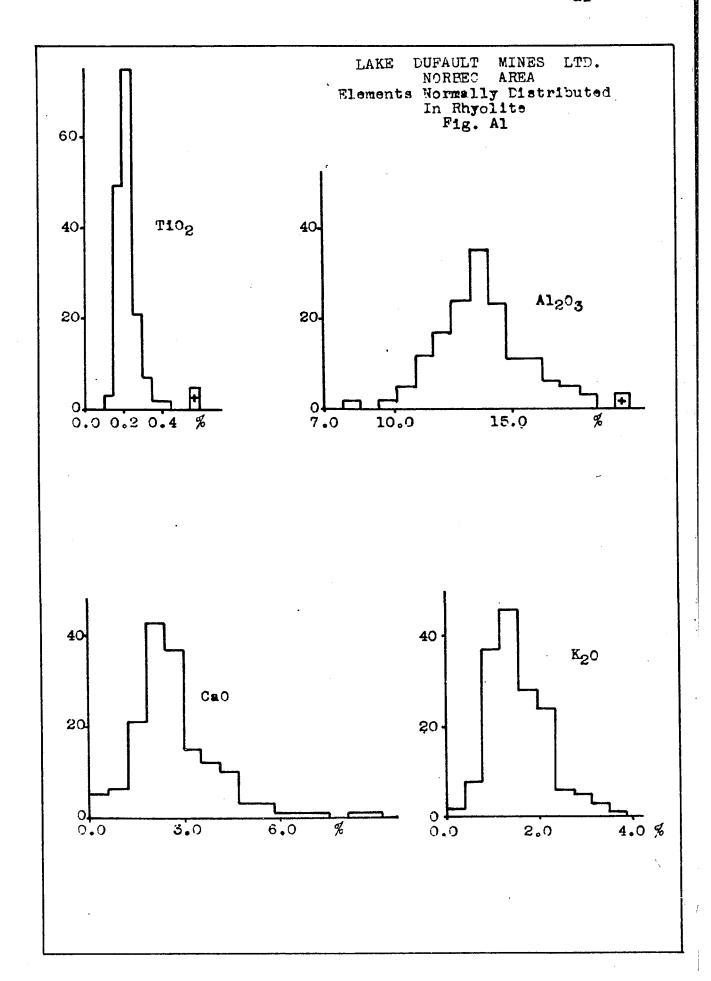
- Taylor, 8., 1957, Quemont Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 405-412.
- Taylor, S. R., 1964, Abundance of chemical elements in the continental crust. Geochim. Cosmochim. Acta, v. 28, p. 1273-1285.
- Taylor, S. R., and White, A. J. R., 1965, Geochemistry of andesites and the growth of continents. Nature, v. 208, No. 5007, p. 271-273.
- Thomson, J. E., 1960, Massive sulphide occurrences in Ontario. Trans. Can. Inst. Mining Met., v. 63, p. 77-81.
- Tolman, C., 1951, Normetal Mine area, Abitibi County. Quebec Dept. Mines, Geol. Rept., No. 34.
- Tomkeieff, S. I., 1949, The volcanoes of Kamchatka. Bull. Volcano., v. 8, p. 87-113.
- Tooker, E. W., 1963, Altered wall rocks in the central part of the Front Range mineral belt, Gilpin and Clear Creek Counties, Colorado. U.S. Geol. Surv. Profes. Paper, 439, pp. 102.
- Turekian, K. K., and Kulp, J. L., 1956, The geochemistry of strontium. Geochim. Cosmochim. Acta, v. 10, p. 245-296.
- Turekian, K. K., and Wedepohl, W. H., 1961, Distribution of the elements in some major units of the earth's crust. Bull. Geol. Soc. Am., v. 72, p. 175-192.
- Turner, F. J., and Verhoogen, J., 1960, Igneous and metamorphic petrology. McGraw-Hill, N.Y., pp. 694.
- Verokhov, V. F., 1962, New data on the composition of the products of Middle Miocene volcanism in the Makarov region (Sakhalin Island). Dokl. Acad. Nauk SSSR, Earth Sci. Sect., v. 137, p. 307-308.
- Vinogradov, A. P., 1962, Average contents of chemical elements in the principle types of igneous rocks of the earth's crust. Geochemistry (USSR), No. 7, p. 641-664.
- Vlodavetz, V. I., and Piip, B. I., 1959, Kamchatka and continental areas in Asia. Catalogue of the Active Volcanoes of the Word Including Solfatara Fields, Intern. Assoc. Volcano., Part VIII, pp. 110.
- Wager, L. R., and Mitchell, R. L., 1951, The distribution of trace elements during strong fractionation of basic magma a further study of the Skaergaard intrusion, East Greenland. Geochim. Cosmochim. Acta, V. 1, p. 129-208.
- Walker, T. L., 1930, Dalmatianite, the spotted greenstone from the Amulet Mine, Noranda, Que. Univ. of Toronto Studies, Geol. Ser., No. 29, p. 9-12.

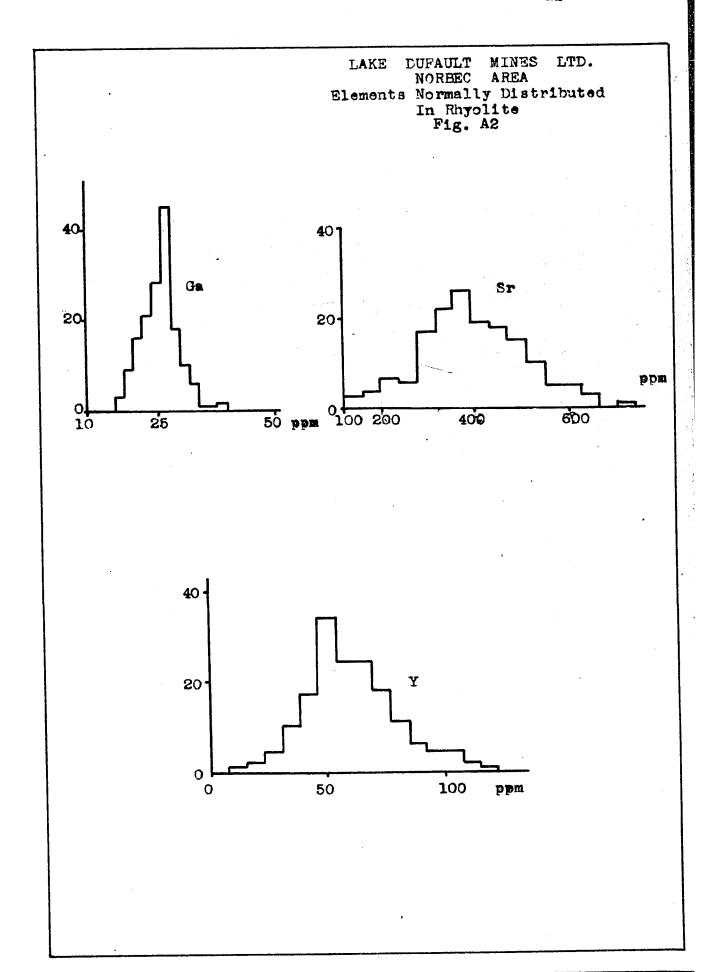
- Webber, G. R., 1962, Variation in the composition of the Lake Dufault granodiorite. Bull. Inst. Mining Met., v. I, p. 84-91.
- Webber, G. R., 1965, Second report of analytical data for CAAS syenite and sulphide standards, Geochim. Cosmochim. Acta, v. 27, No. 4, p. 229-248.
- Wedepohl, K. H., 1953, Untersuchungen zur geochemie des Zinks. Geochim. Cosmochim. Acta, v. 3, p. 93-142.
- Wedepohl, K. H., 1963, Deitrage zur geochemie des Zupfers. Geol. Rundschau, v. 52, p. 492-504.
- Westervelt, J., 1952, Quarternary volcanism on Sumatra. Bull. Geol. Soc. Am., v. 63, p. 561-594.
- Wilson, H. D. B., Andrews, P., Moxham, P. L., and Ramal, K., 1965, Archean volcanism in the Canadian shield. Can. J. Earth Sci., v. 2, No. 3, p. 161-175.
- Wilson, M. E., 1908, Lake Oposatika and the height of land. Geol. Surv. Can., Sum. Rept., p. 121-123.
- Wilson, M. E., 1910, Northwestern Quebec adjacent to the interprovincial boundary and the national transcontinental railway. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., p. 203-207.
- Wilson, M. E., 1911, Kewagama Lake map area, Pontiac and Abitibi, Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., p. 273-279.
- Wilson, M. E., 1913, Kewagama Lake map area, Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Mem. 39, pp. 139.
- Wilson, M. E., 1918, Temiskaming Co., Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Mem. 103, pp. 197.
- Wilson, M. E., 1933, Amulet Mine, Noranda dist., Que. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Sum. Rept., Pt. D, p. 83-120.
- Wilson, M. E., 1934, The multiple and complementary sills and dykes of the Waite-Ackerman-Montgomery Mines: Noranda dist., Que. Trans. Roy. Soc. Can. 3rd Series, v. 28, Sec. 4, p. 65-74.
- Wilson, M. E., 1935, Rock alteration at the Amulet Mine, Noranda district, Quebec. Econ. Geol., v. 30, p. 478-492.
- Wilson, M. E., 1938, The Keewatin lavas of the Noranda districts, Que. Univ. of Toronto Studies, Geol. Ser. No. 41, p. 75-82.
- Wilson, M. E., 1941, Noranda district, Quebec. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Mem. 229, pp. 162.

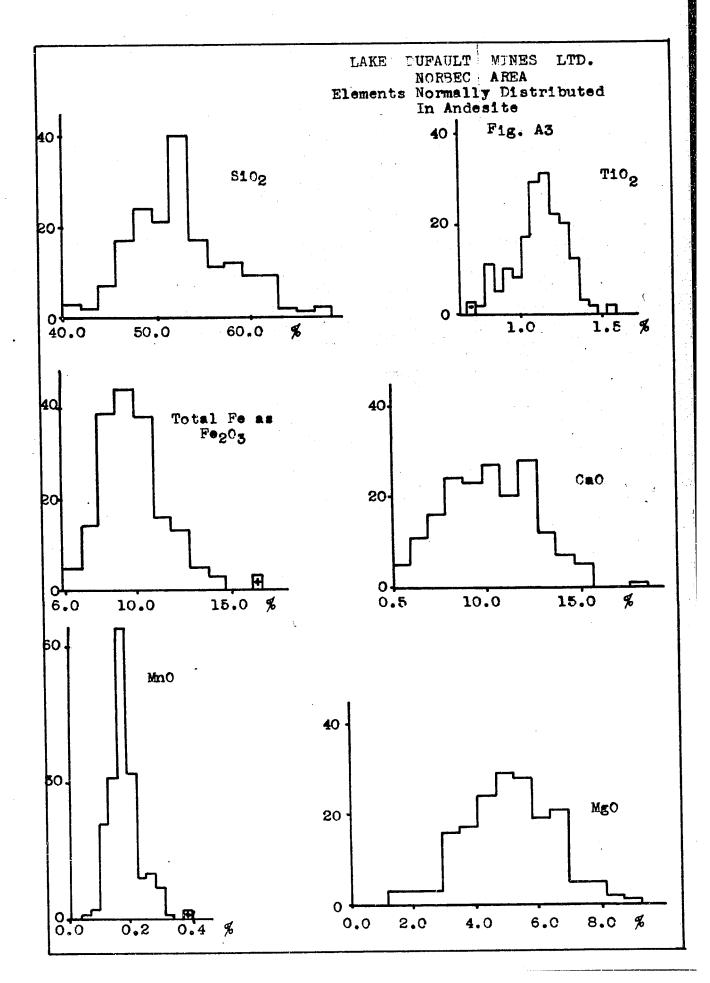
- Wilson, M. E., 1943, The early Precambrian succession in western Quebec. Trans. Roy. Soc. Can., Sec. 4, v. 37, p. 118-138.
- Wilson, M. E., 1948, McWatters Mine. Structural geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 783-789.
- Wilson, M. E., 1962, Rouyn-Beauchastel map-areas. Can. Dept. Mines Tech. Surv., Geol. Surv. Can., Mem. 315, pp. 140.
- Wilson, M. E., Hopper, R. V., and Trenholme, L. S., 1948, Rouyn Merger Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 787-796.
- Wilson, M. E., and Lee, A. C., 1948, Senator-Rouyn Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 735-738.
- Wilson, M. E., and McQuarry, W. R., 1948, Stadacona Mine. Structural Geology of Canadian Ore Deposits, Can. Inst. Mining Met., p. 776-782.
- Yagi, K., 1962, Welded tuffs and related pyroclastic deposits in northeastern Japan. Bull. Volcano., v. 24, p. 109-128.
- Zelenov, K. K., 1965, Geochemistry of aluminum and titanium in volcanic areas of island arcs. Intern. Geol. Rev., v. 7, No. 3, p. 526-539.

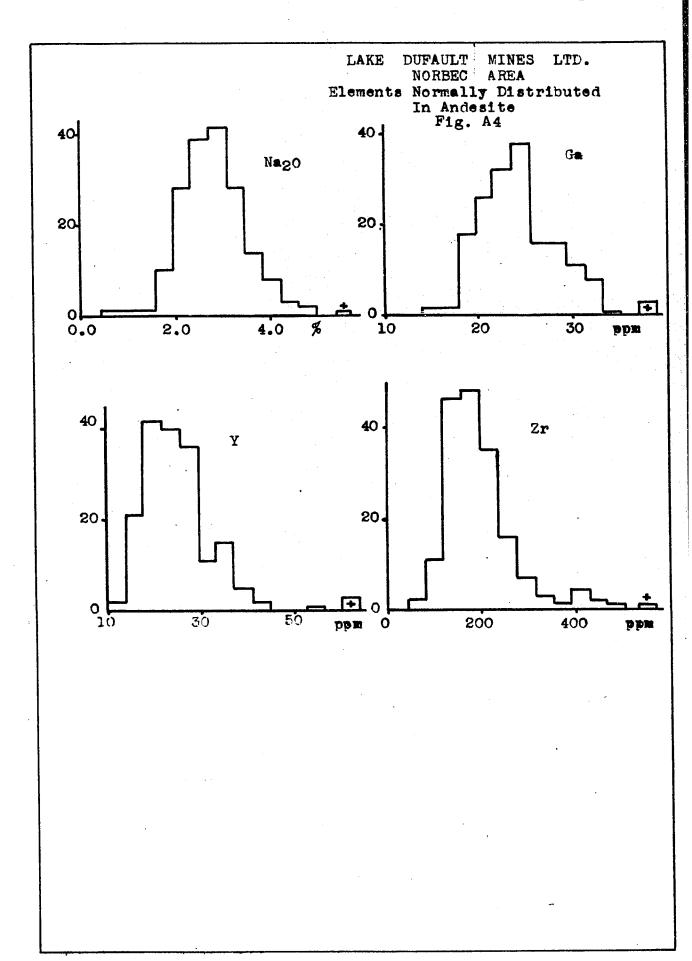
APPENDIX A

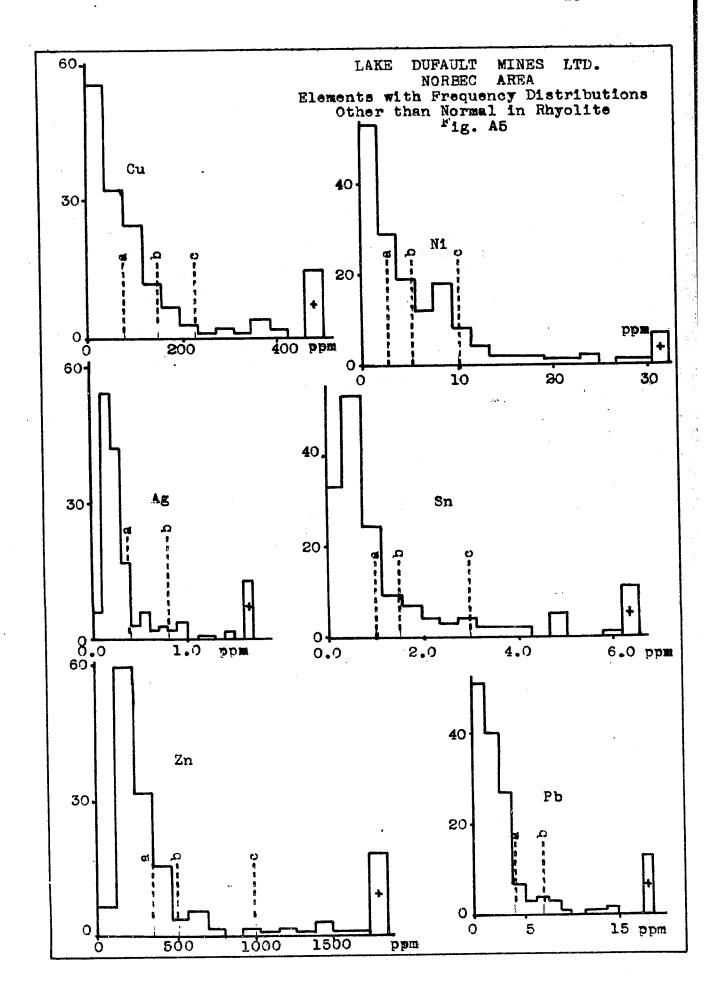
Frequency Distribution Diagrams
of the Elements in Rhyolite
and Andesite

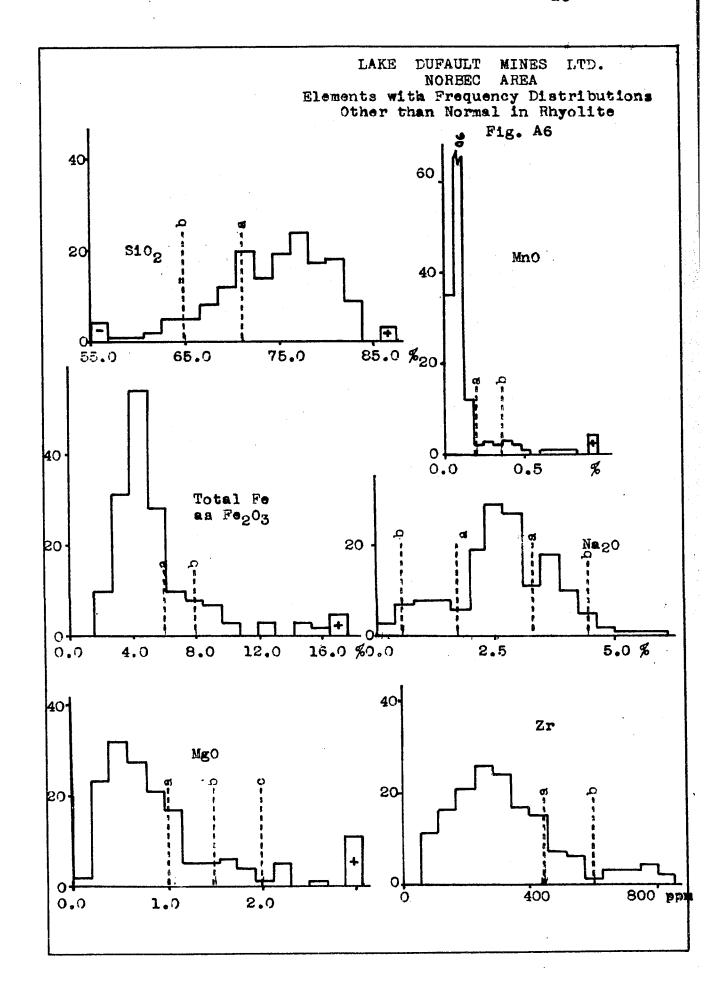


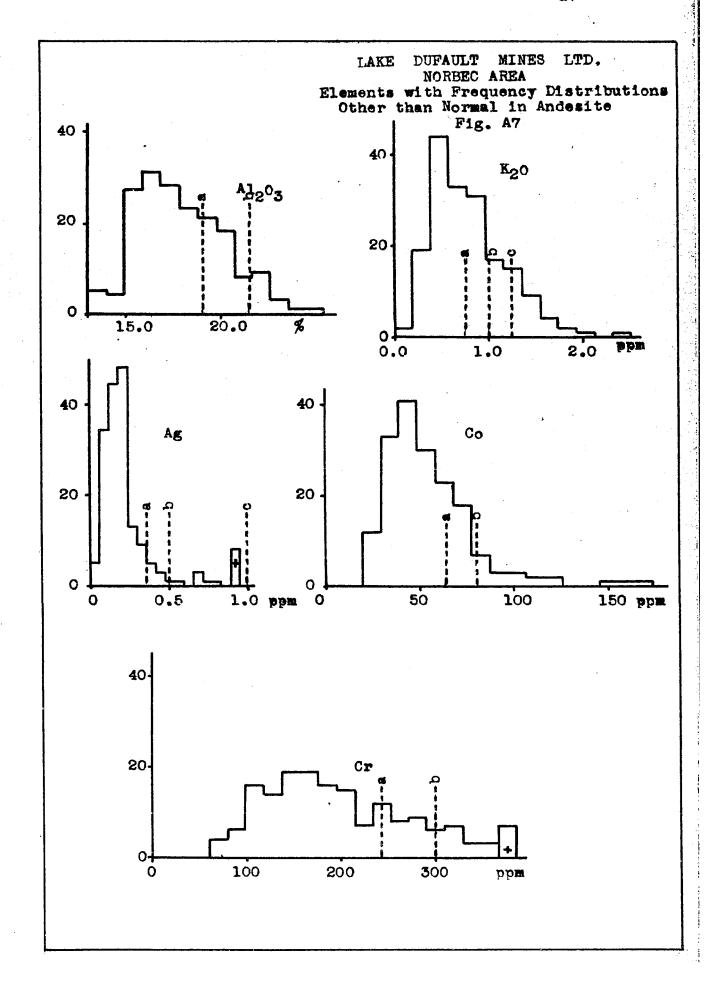


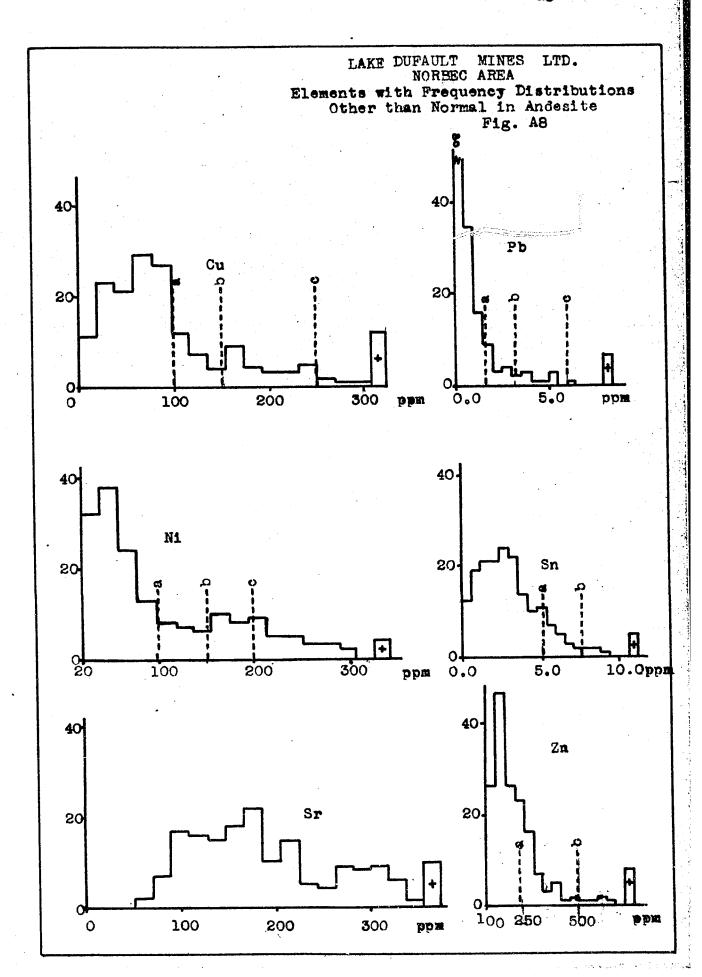












APPENDIX B

Data Charts of the Elements
in Rhyolite and Andesite

Abbreviations Used in Presentation of Results

*= - grand mean

Mo - mode (most common value)

*o - sample standard deviation (weighted average of o)

FD - type of frequency distribution

a,b,c - limits of anomalous ranges

*ULB - upper background limits (for normal distribution only)

LLB - lower background limits (for normal distribution only)

CA - relative deviation of replicate analyses

R - CA/C* total

r - # samples 0 - 50% / # samples 50 - 100% (for tuff samples)

^{*}All samples close to the ore body removed from calculations

Silicon (oxide) (SiO₂) Rhyolite chart 1

\$\frac{1}{x}\$ 75.5%

= Mo 77.0%

\$\tilde{\tau}\$ 4.1%

FD Type II (except negatively skewed)

a 71.0%

b 65.0%

CA 5.0% of \$\frac{1}{x}\$

R .92

Most of the variation of SiO₂ in rhyolite can be attributed to analytical error, yet two anomalous conditions can be discerned. SiO₂ content is quite naturally low where the rhyolite is mineralized near the ore body. Several drill holes slightly low in SiO₂ are grouped together in the northwest corner of the property. The grouping of these low values is the best evidence that the low SiO₂ values are anomalous.

SiO₂ Andesite chart 2

\$\bar{x}\$ 52.4%

Mo 52.5%

\$\bar{x}\$ 4.2%

FD Type I

*ULB 52.4 + 6.4%

LLB 52.4 - 6.4%

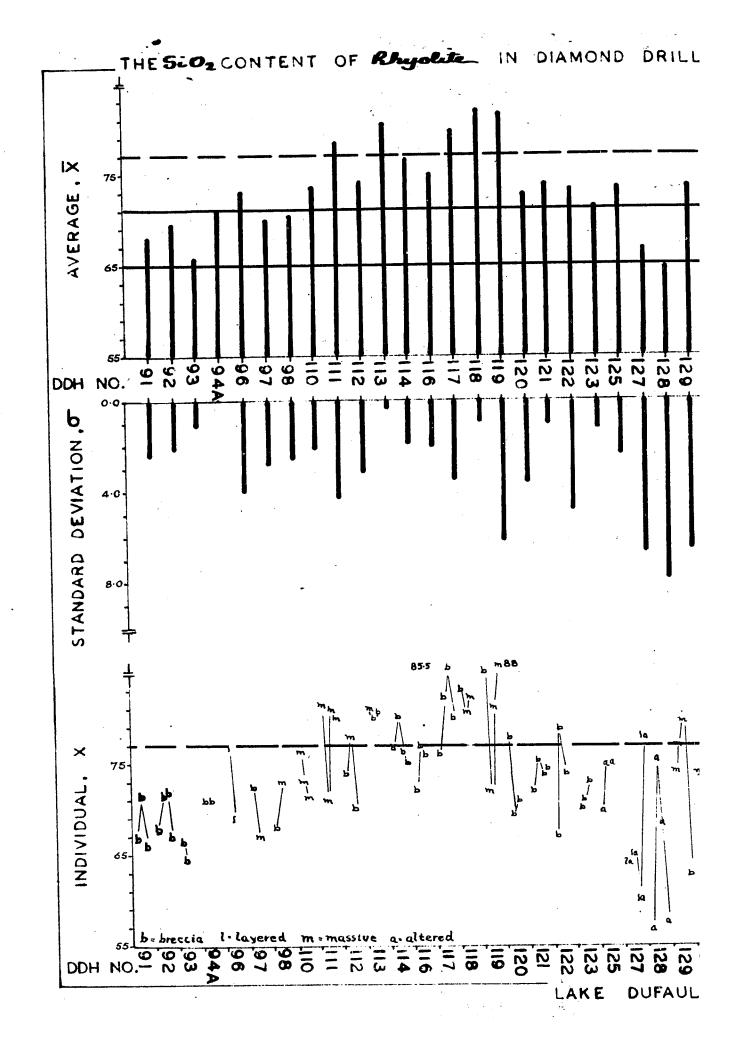
CA 8.6% of \$\bar{x}\$

R 1.06

Diamond drill hole 125 is anomalous because rhyolite was mistakenly logged as andesite. Samples from diamond drill holes 174, 181, 190, 199 and 204 were the only ones which were analysed in one group at one time. Because of this it is highly probable that the high values for SiO₂ in these holes are due to analytical error.

^{*}Background limits vary slightly as a function of N, the number of samples. All values given are for N=4; see the data charts for effect of changing N.

Chart 1 SiO₂ rhyolite



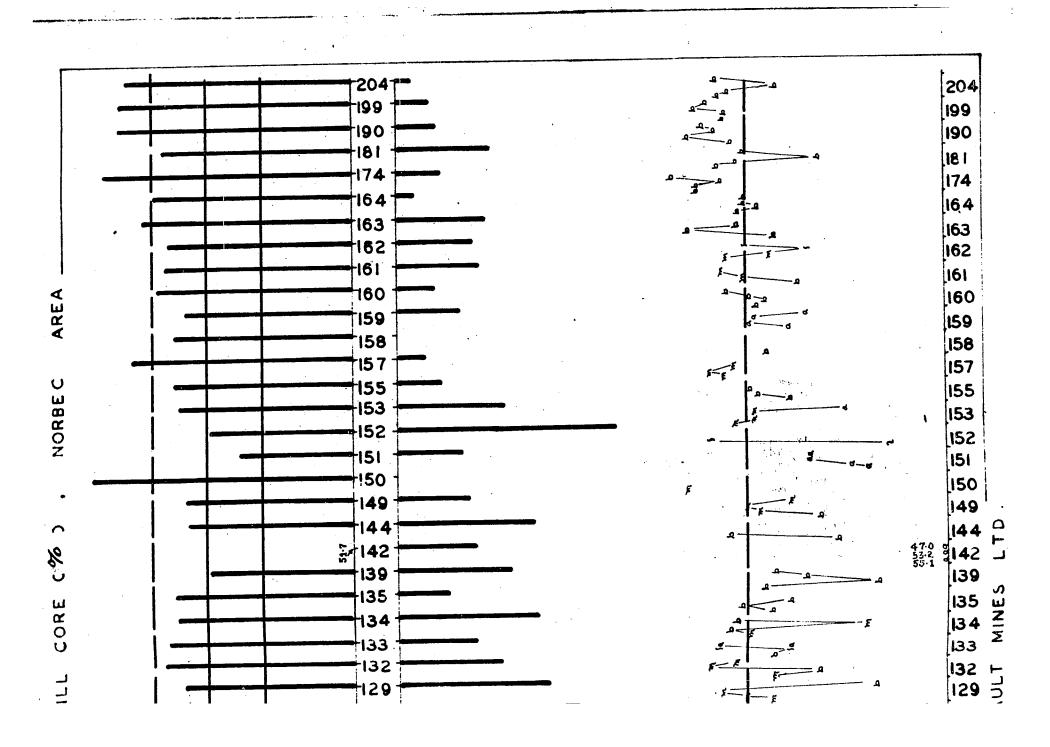
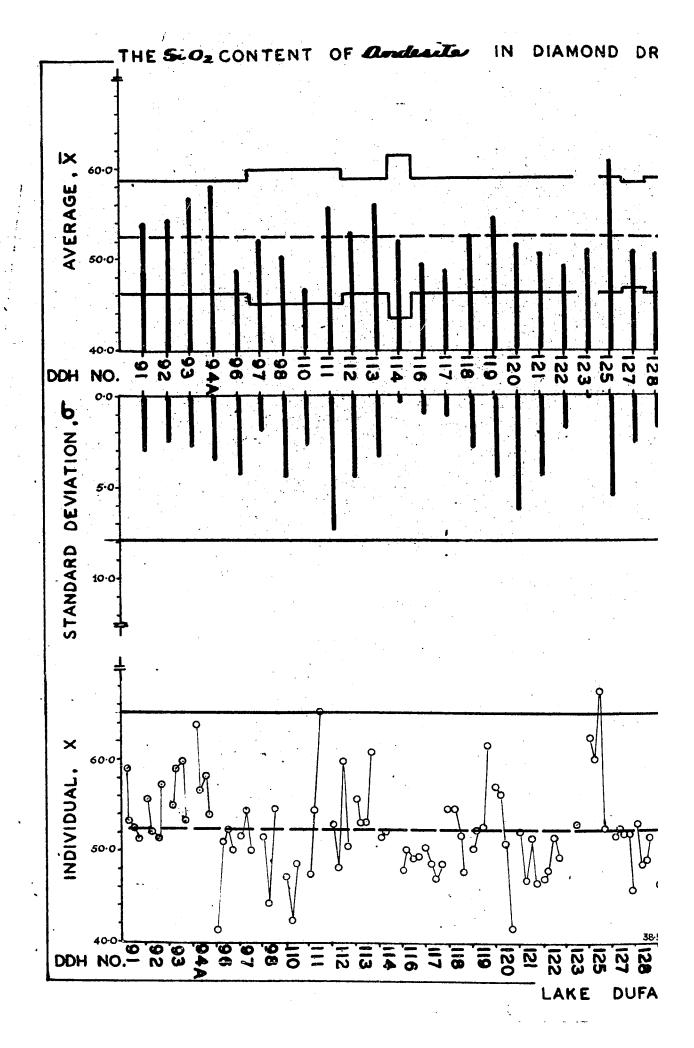
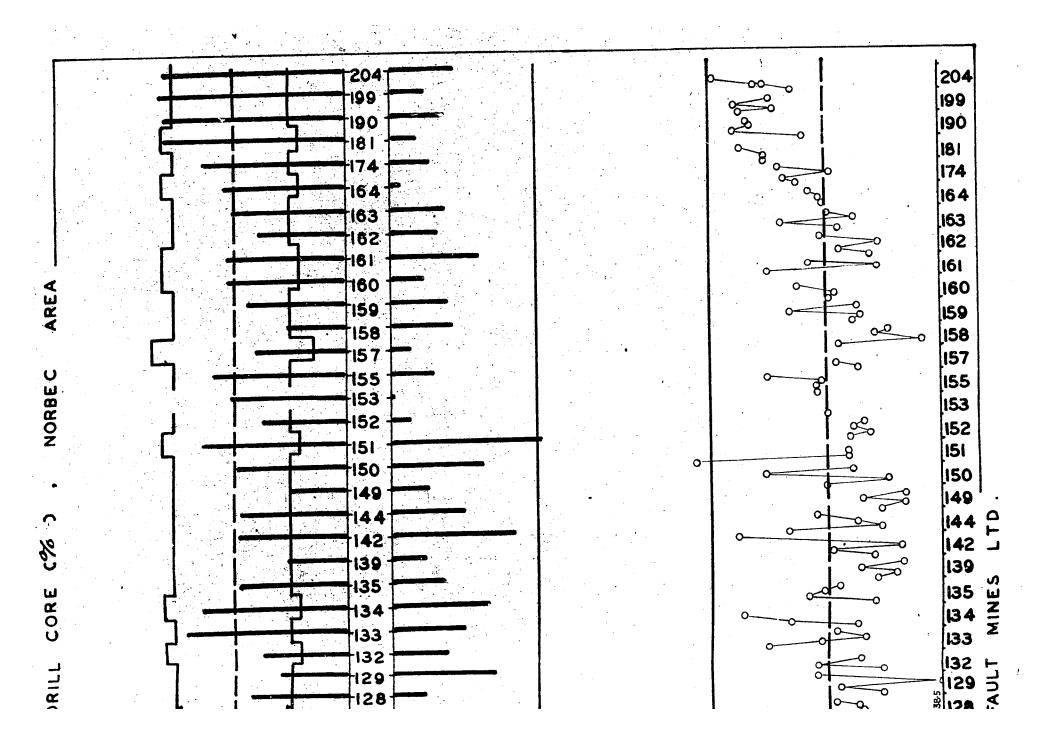


Chart 2 SiO₂ andesite





Titanium (oxide) (TiO2) Rhyolite chart 3

\(\bar{x}\) 0.323%

Mo 0.23%

\(\bar{x}\) 0.1%

FD Type I

ULB 0.32 + 0.16%

LLB 0.32 - 0.16%

CA 9.1% of \(\bar{x}\)

R 0.48

Samples from hole 125 have an extra high TiO_2 content because they are actually andesite. Other anomalous samples were taken from holes 134, 139 and 153 near the ore body. Hole 96 in the Northwest Norbec section cut rhyolite with a high content of TiO_2 .

TiO₂ Andesite chart 4

\$\bar{z}\$ 1.20%

Mo 1.11%

\$\bar{c}\$.159%

FD Type I

ULB 1.20 + .24%

LLB 1.20 - .24%

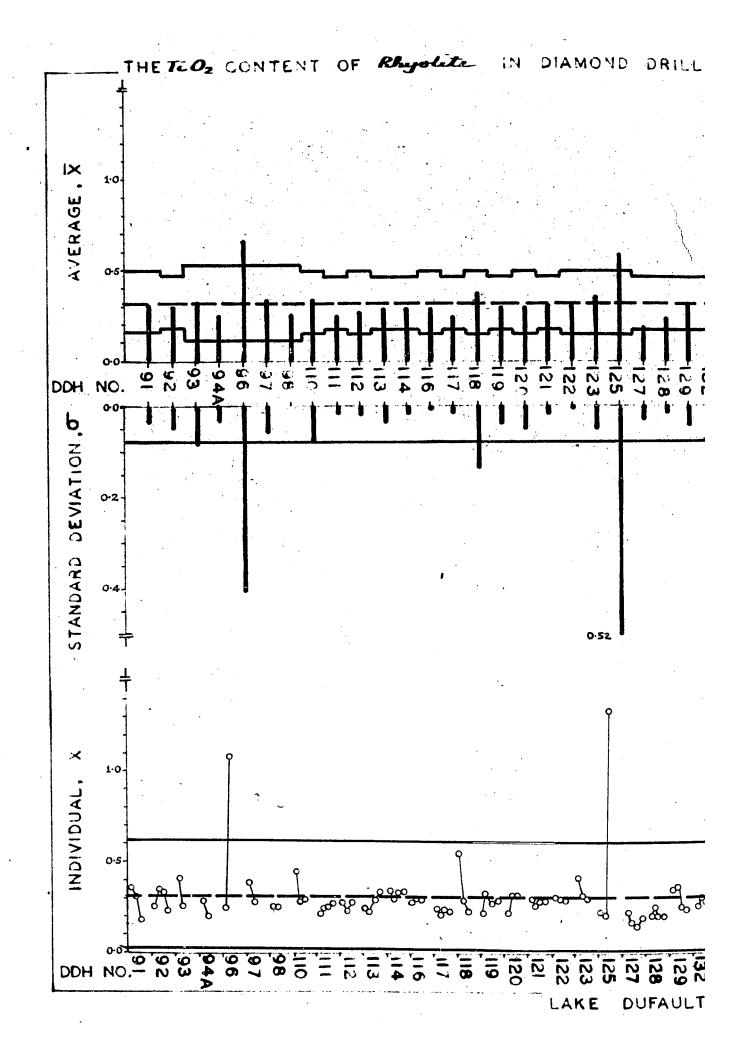
CA 5.9% of \$\bar{x}\$

R .45

Sections of andesite cut by four holes in the northwest corner of the property are abnormally low in TiO₂. Two samples from diamond drill hole 125 are actually rhyolite, not andesite. Andesite cut by diamond drill hole 161 is low in TiO₂. Diamond drill holes 190 and 199 are abnormally high, possibly because of analytical error.

Chart 3

TiO2 rhyolite



)

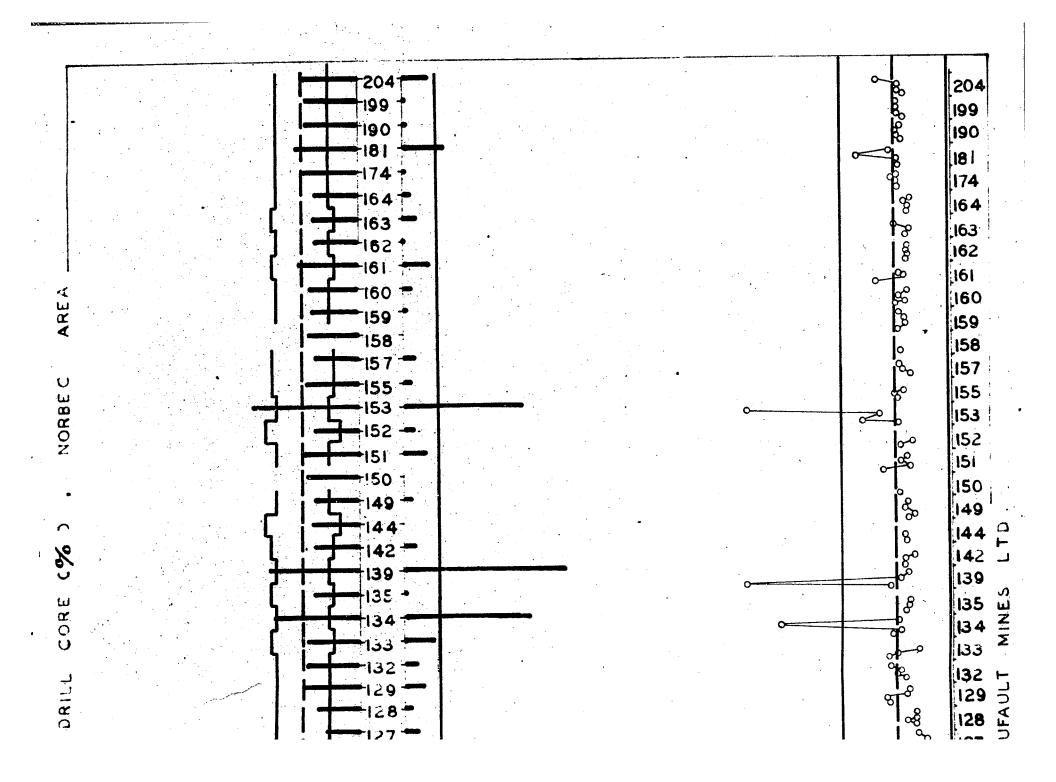
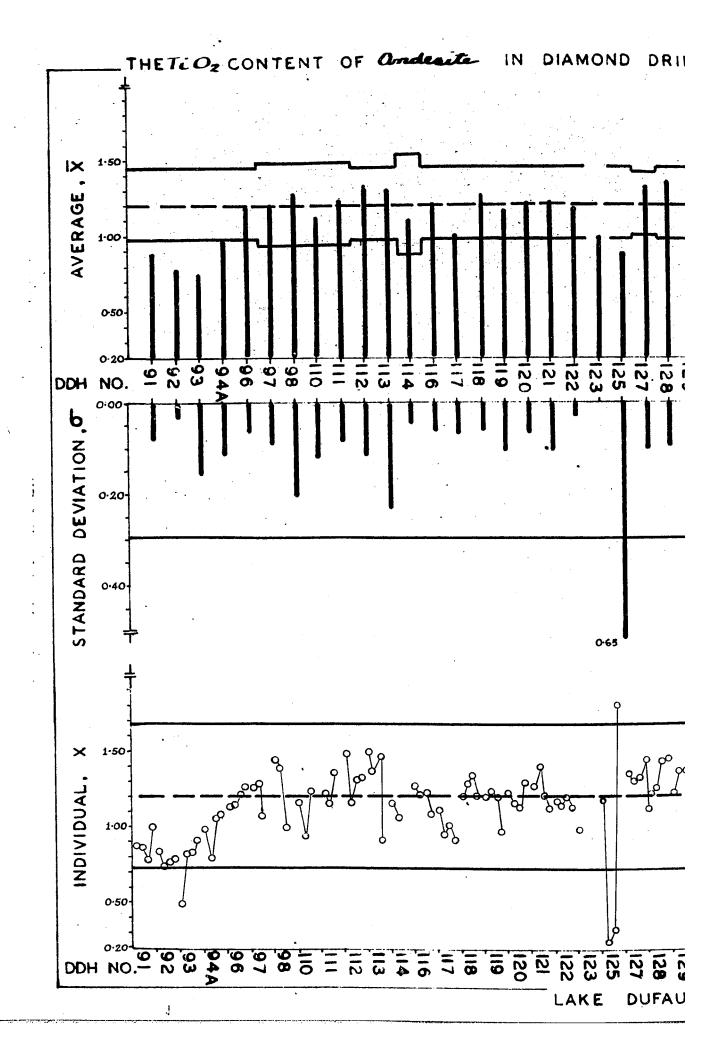
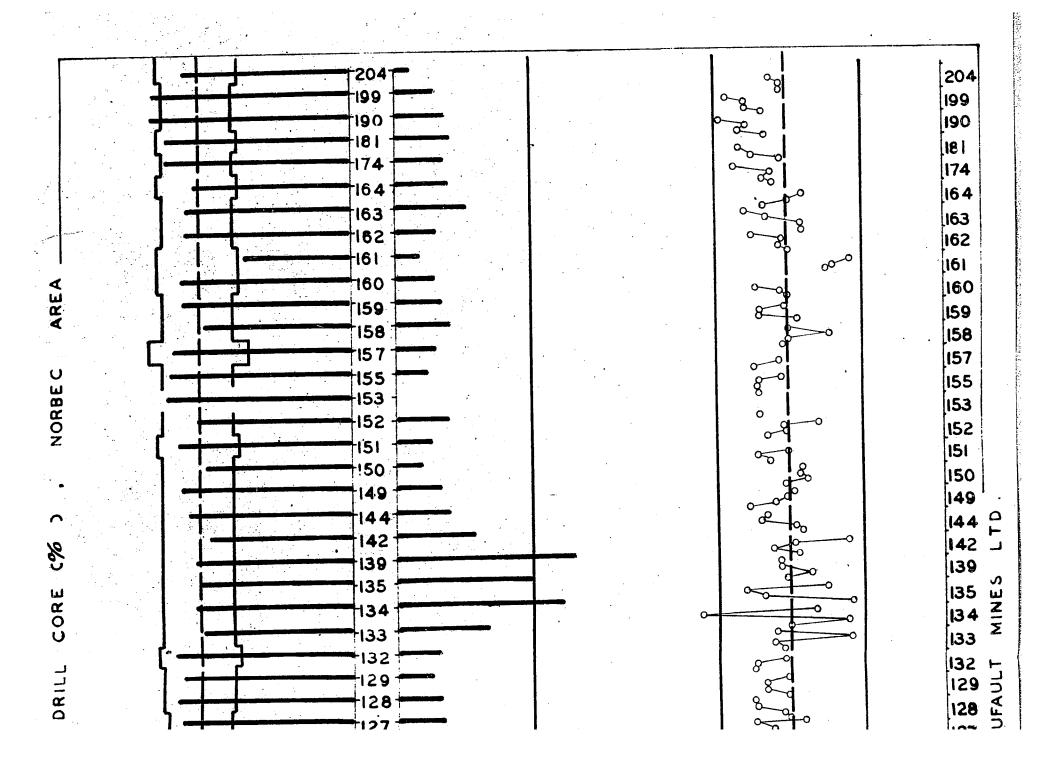


Chart 4

TiO2 andesite





Aluminum (oxide) (Al₂O₃) Rhyolite chart 5 \$\overline{\pi}\$ 14.2% Mo 13.6% \$\overline{\pi}\$ 1.6% FD Type I ULB 14.2 + 2.4% LLB 14.2 - 2.4%

CA 14.2% of X

R 1.35

Near the ore body, mineralized rhyolite cut by holes 127, 128 and 142 is diluted with respect to Al_2O_3 as it was with SiO_2 . Holes 181 and 204 cut rhyolite containing greater than normal amounts of Al_2O_3 .

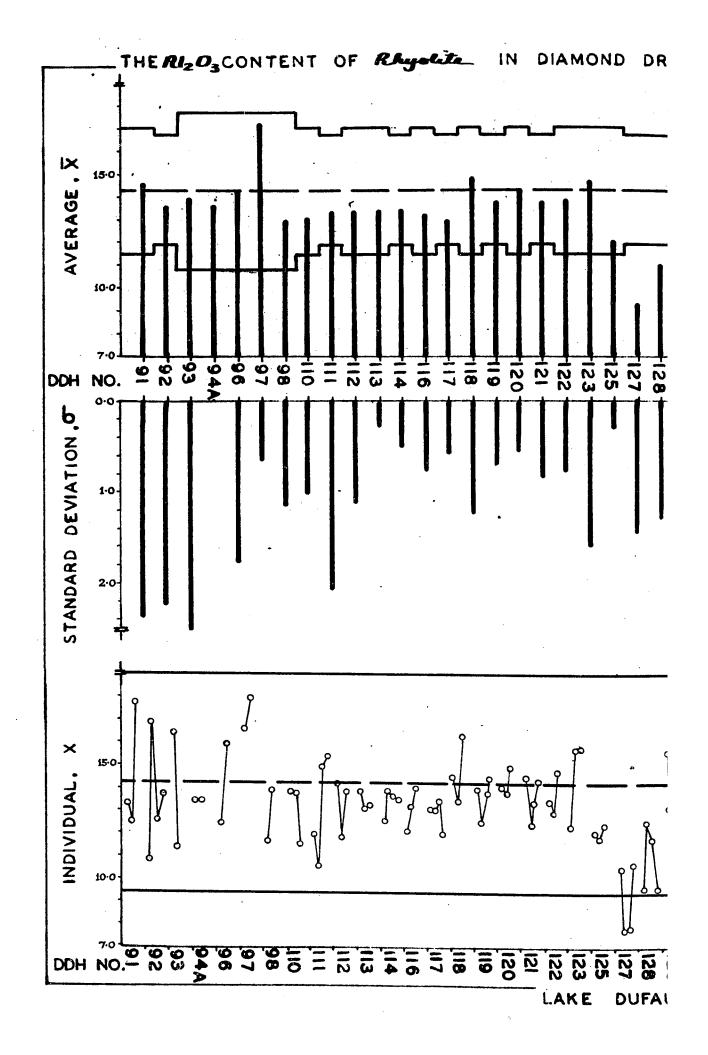
Al_2O_3 Andesite chart 6

18.3% Mo 16.4% To 1.6% FD Type II a 19.0% b 21.5% CA 7.7% of X R .79

Anomalous sections of ${\rm Al}_2{\rm O}_3$ are scattered and apparently are not related to the presence of ore.

Chart 5

Al₂0₃ rhyolite



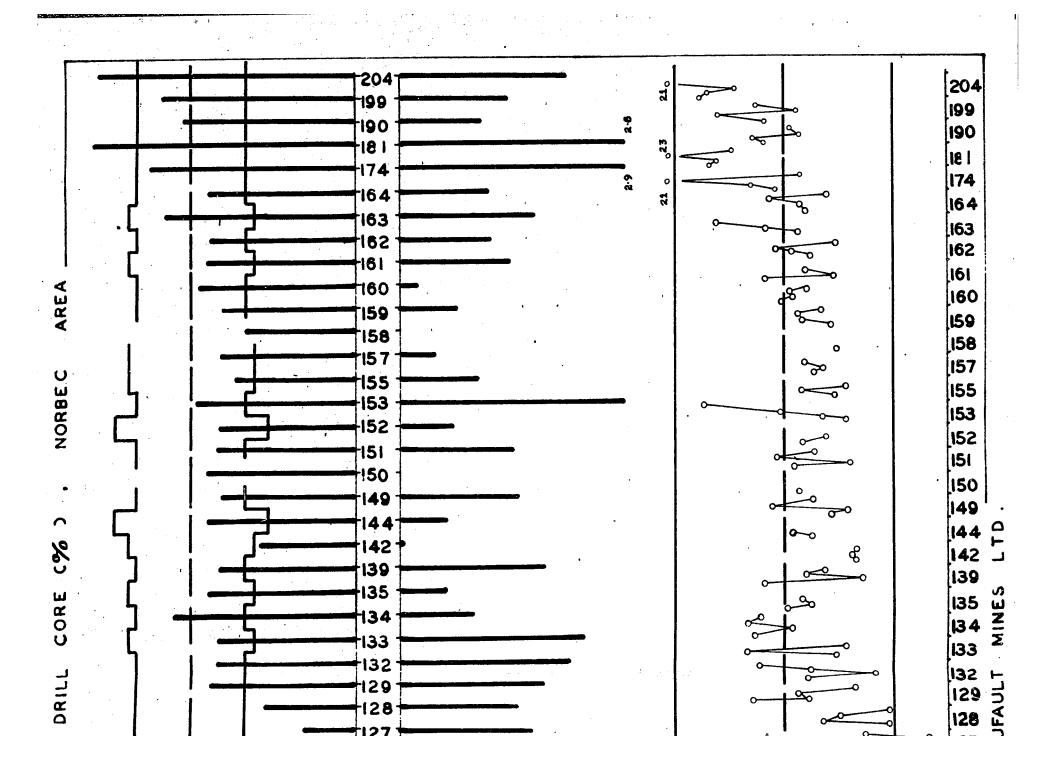
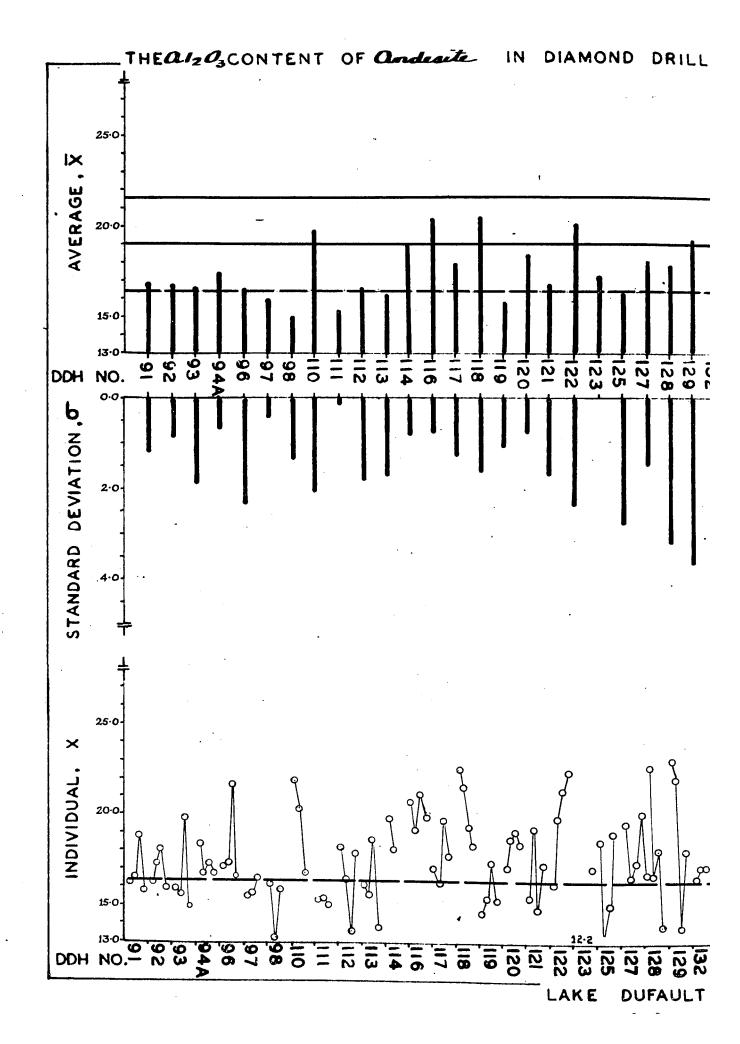
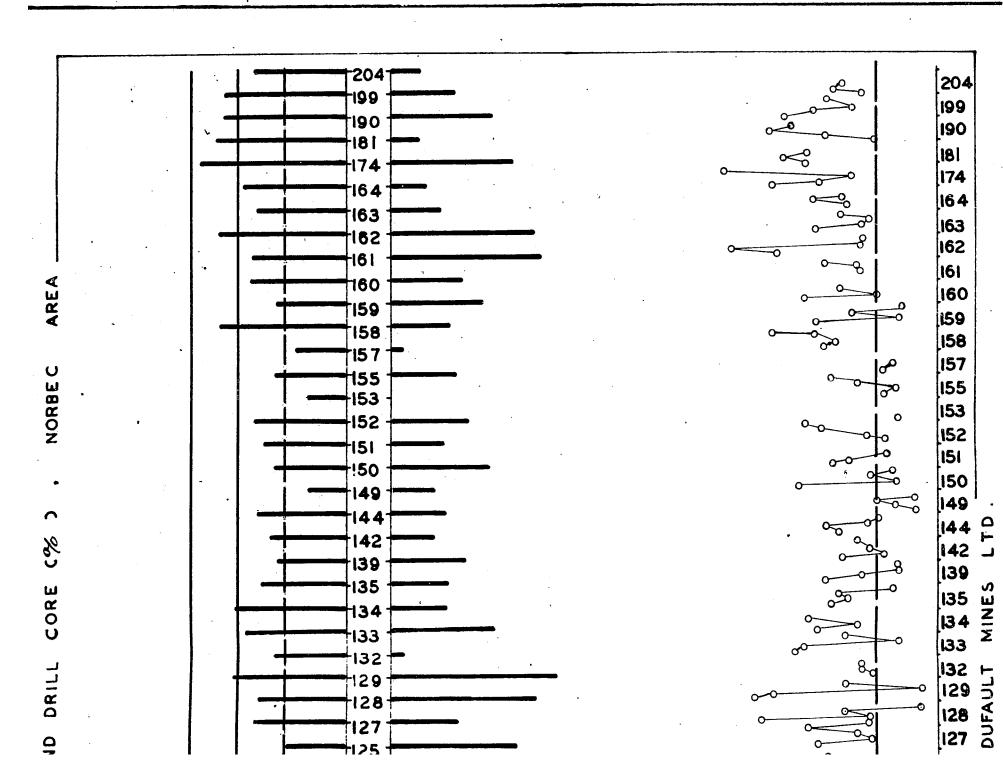


Chart 6
Al₂0₃ andesite





Iron (oxide) Total Fe as Fe₂O₃ Rhyolite chart 7

= X 4.9% Mo 4.5% \$\overline{\text{T}}\$ 1.3% FD Type II a 6.0% b 8.0% CA 2.3% of \$\overline{x}\$ R .09

Only one drill hole away from the immediate vicinity of the ore body cuts rhyolite abnormally high in Fe (97). Of the anomalous holes near the ore body only 162, 129 and 153 do not cut rhyolite which is obviously mineralized.

Total Fe as Fe₂0₃ Andesite chart 8

\$\bar{x}\$ 9.3%

Mo 9.4%

\$\bar{x}\$ 1.4%

FD Type I

ULB 9.3 + 2.1%

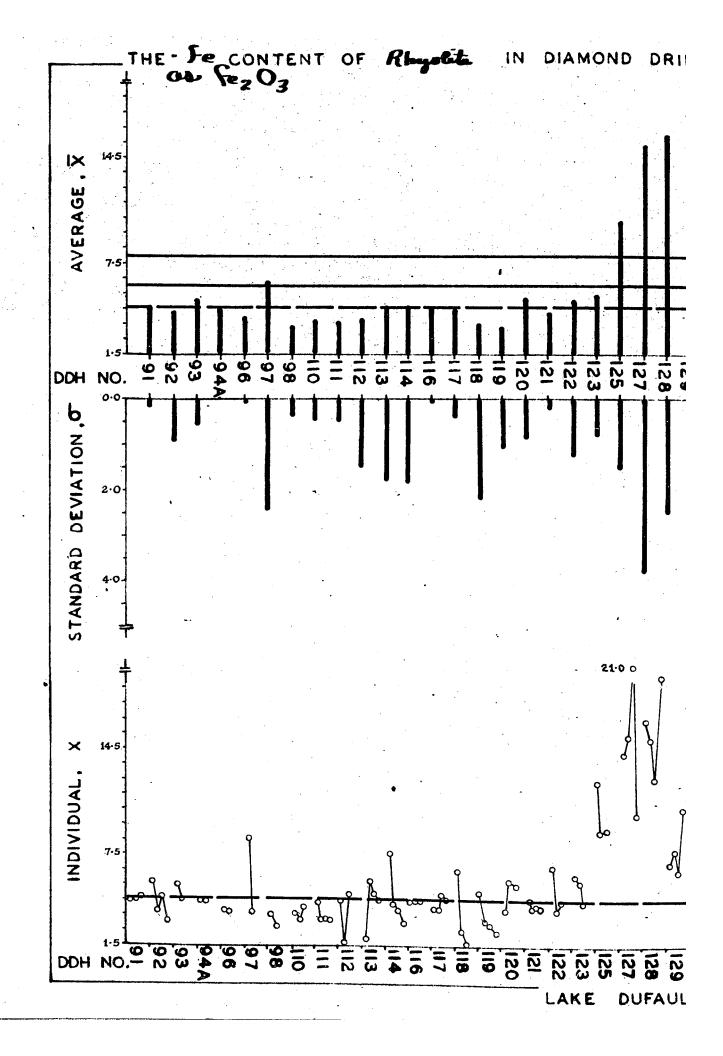
LLB 9.3 - 2.1%

CA 3.0% of \$\bar{x}\$

R .20

Andesite cut by drill hole 92 contains an unusually low amount of Fe. Hole 125 cuts andesite with an unusually high Fe content. One sample of andesite from diamond drill hole 150 contained much more Fe than usual.

Chart 7 total Fe as Fe₂0₃ rhyolite



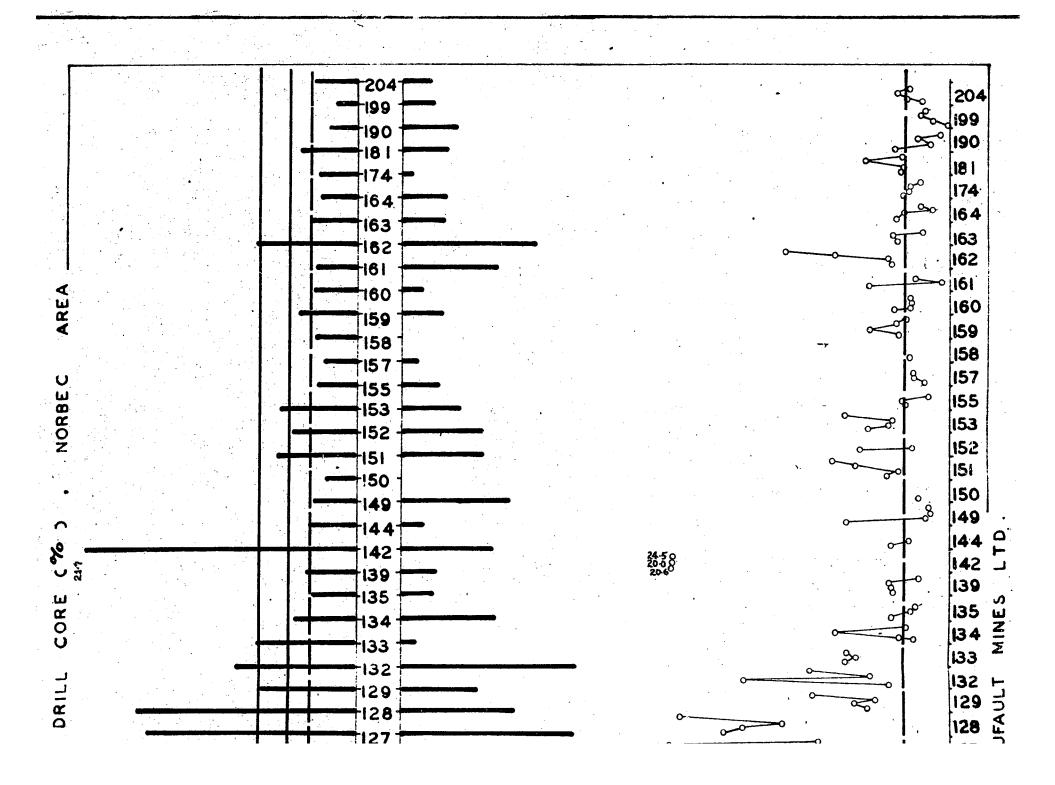
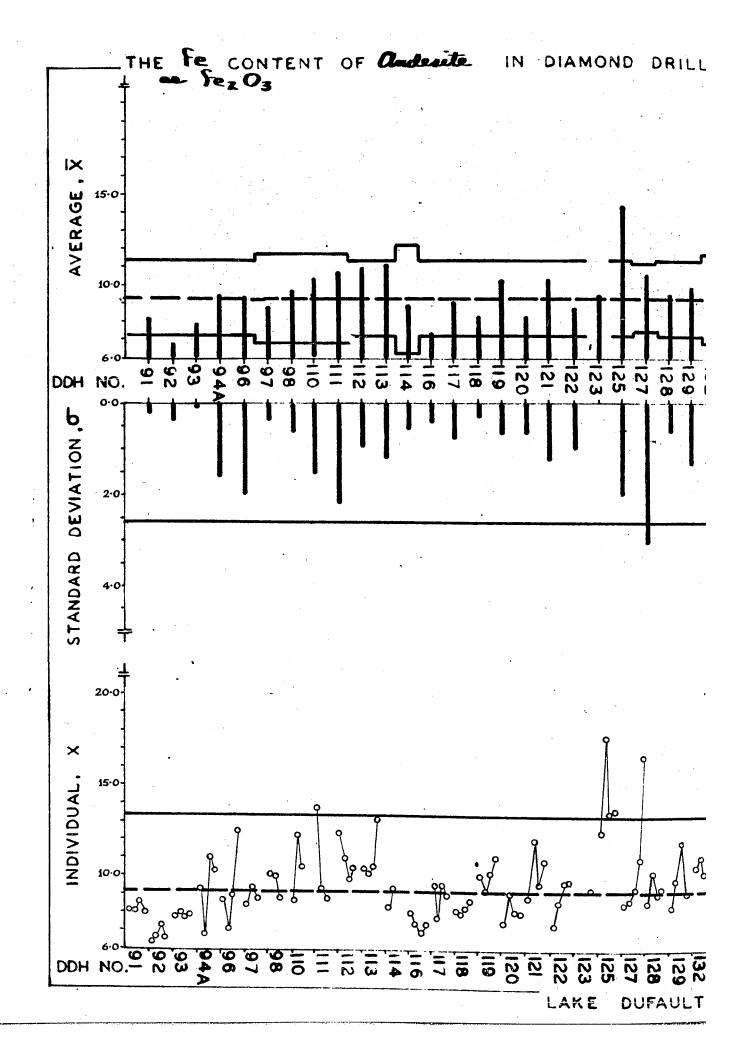
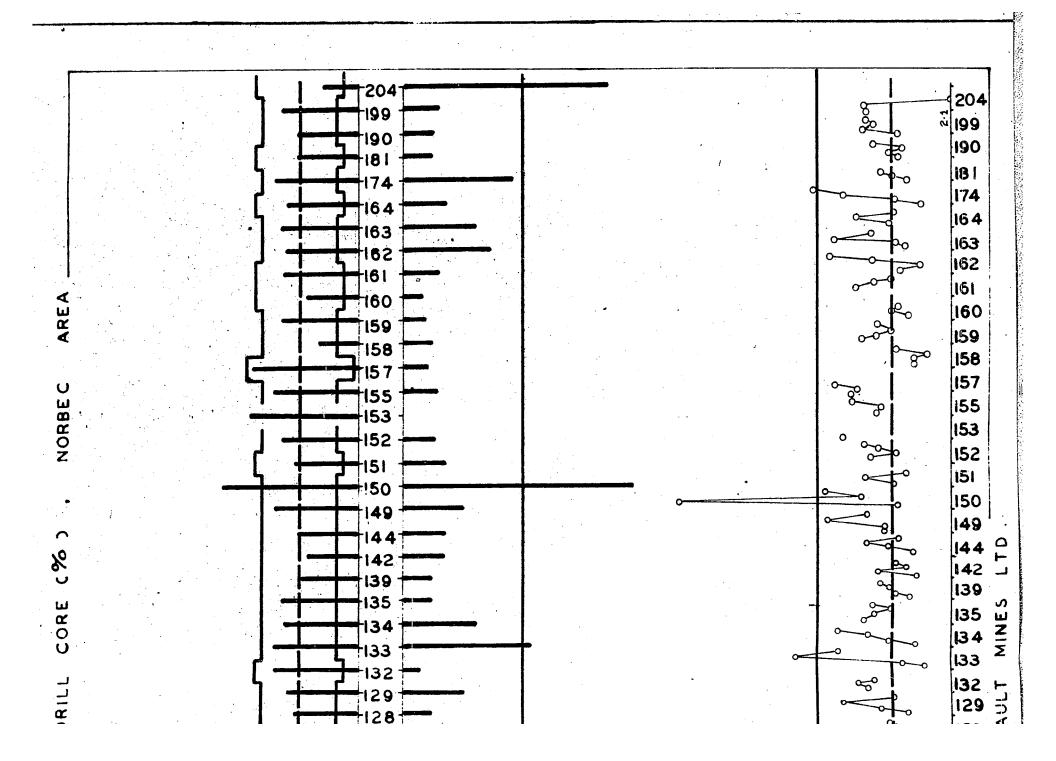


Chart 8 total Fe as Fe203 andesite





Manganese (oxide) (MnD) Rhyolite chart 9

0.156%
Mo 0.10%
0.095%
FD Type IV
a .2%
b .35%
CA 3.0% of \$\overline{x}\$
R .050

Away from the ore body only holes 92 and 113 cut rhyolite containing abnormally high amounts of MnO. Other than those holes cutting mineralized rhyolite only numbers 139, 162 and 129 are anomalous.

MnO Andesite chart 10

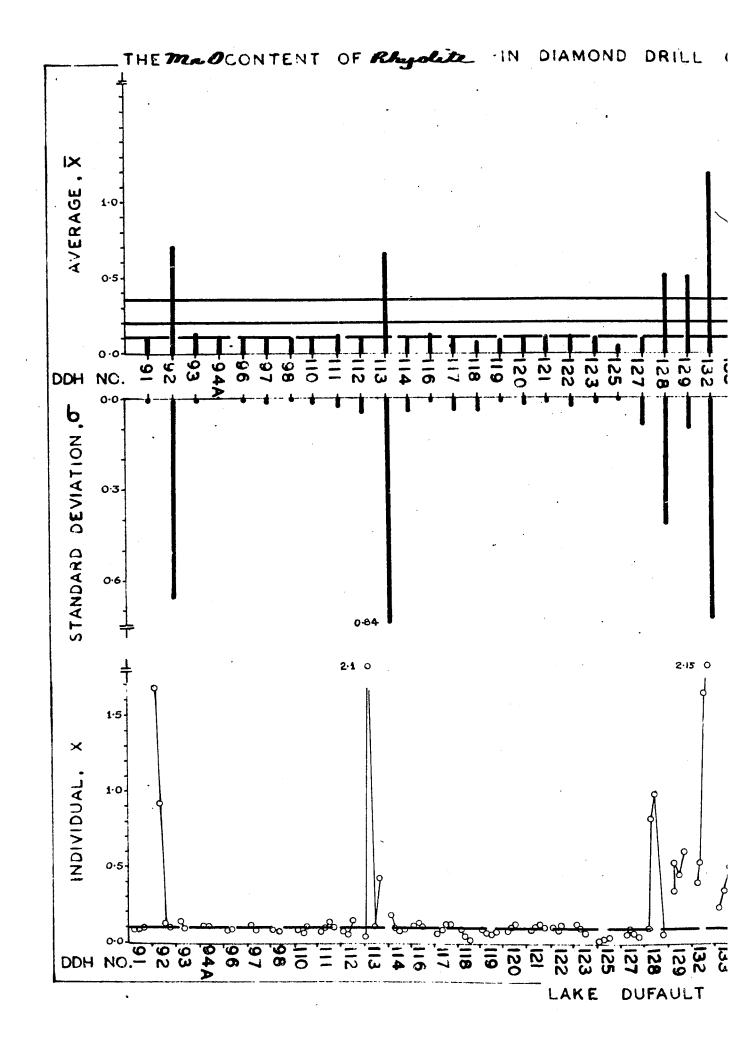
0.19%
Mo 0.16%

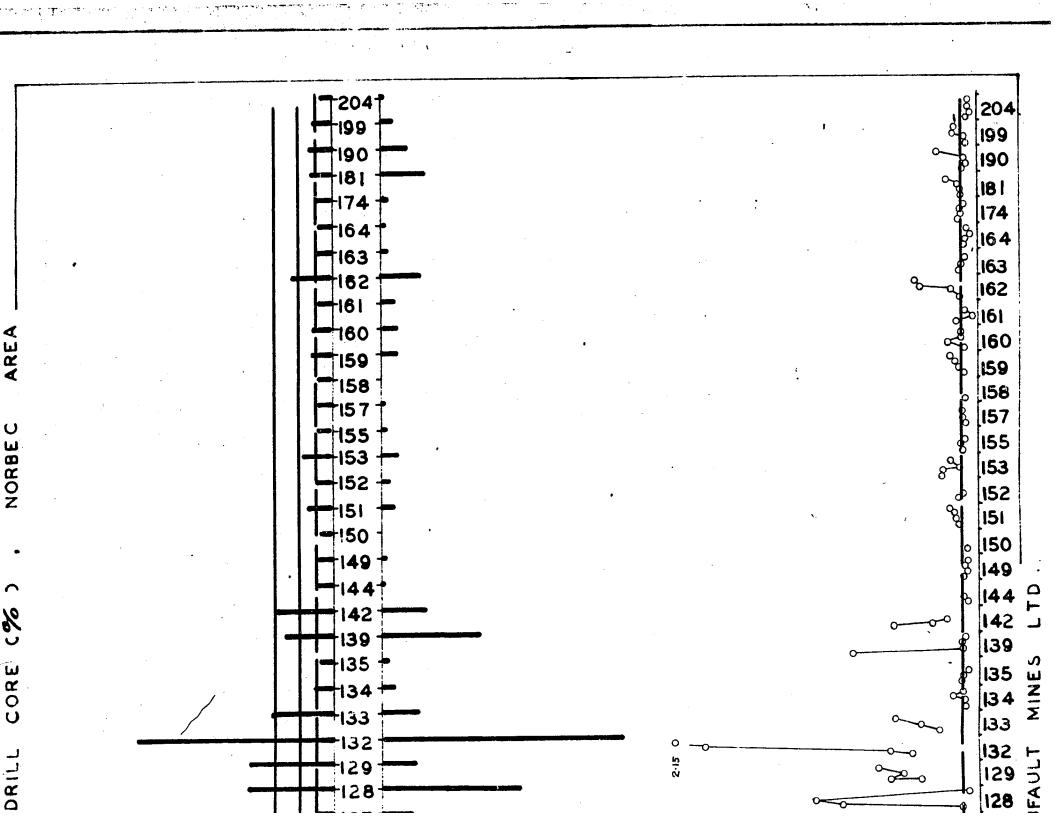
0.035%
FD Type I
ULB 0.19 + .05%
LLB 0.19 - .05%
CA 5.6% of \$\frac{7}{2}\$

Over the ore body, andesite cut by holes 125, 127, 129, 132 and 149 contains unusually high amounts of MnD.

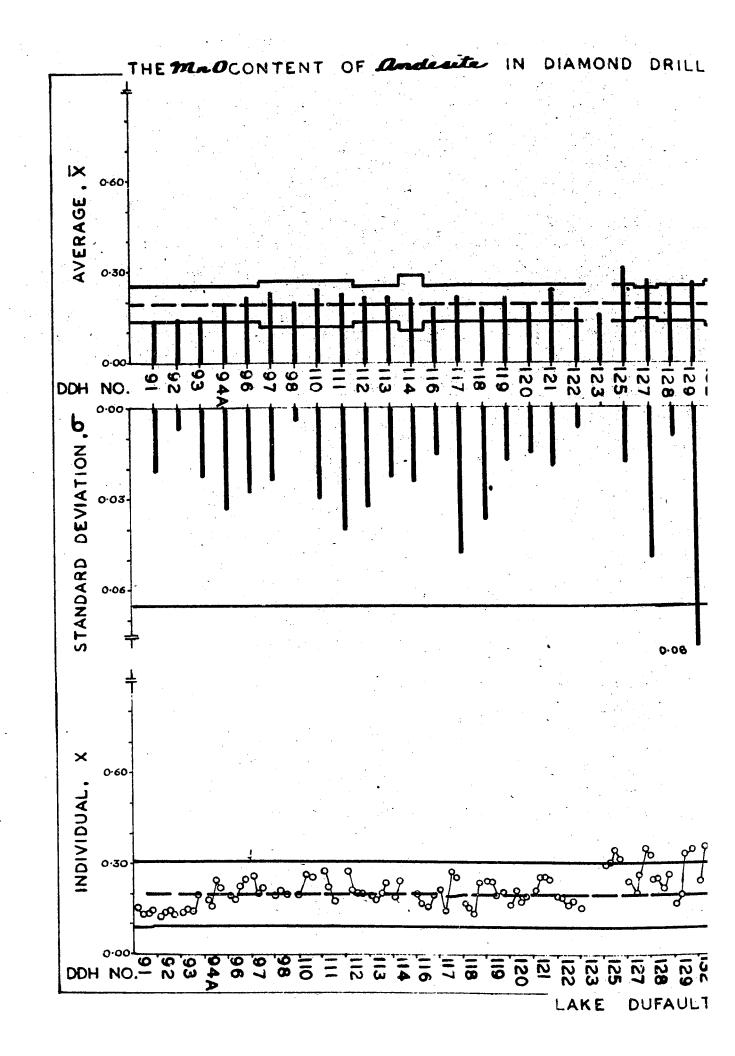
Off to the side of the ore body only hole 164 is anomalous. Far away from the ore body hole 150 cuts andesite containing an extra amount of MnO. The high value for hole 153 is from an individual sample.

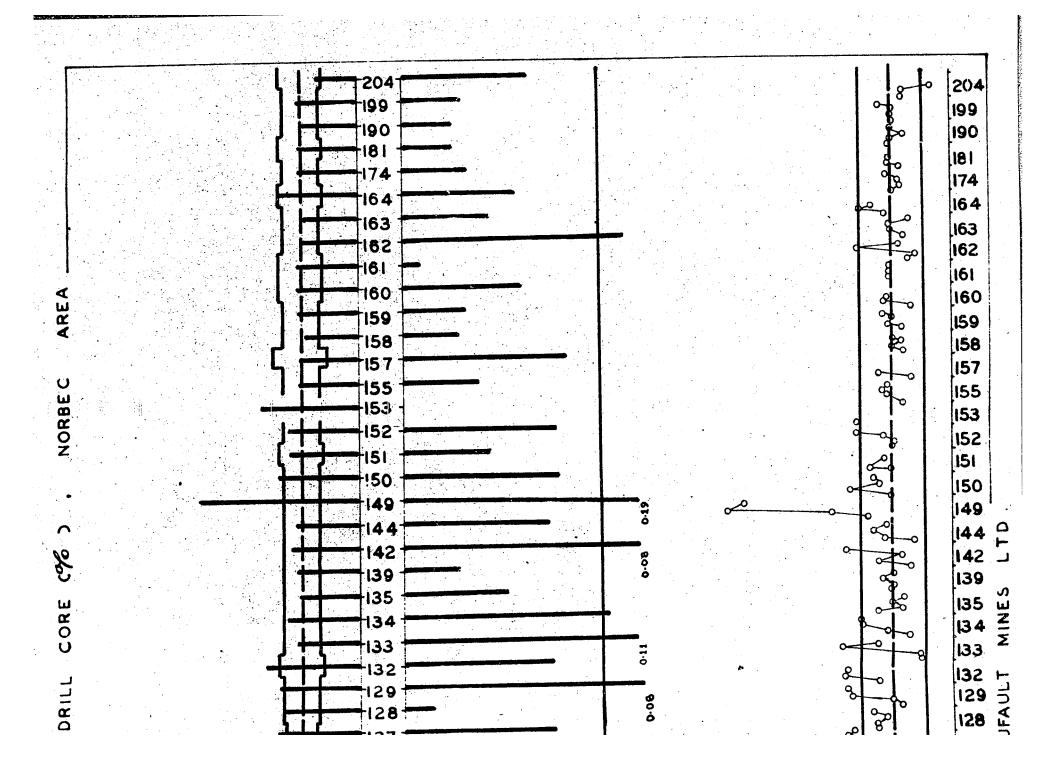
MnO rhyolite





MnO andesite





Calcium (oxide) (CaO) Rhyolite chart 11

2.6%
Mo 2.5%

ō .78%
FD Type I
ULB 2.6 + 1.2%
LLB 2.6 - 1.2%
CA 4.0% of \$\overline{x}\$
R .13

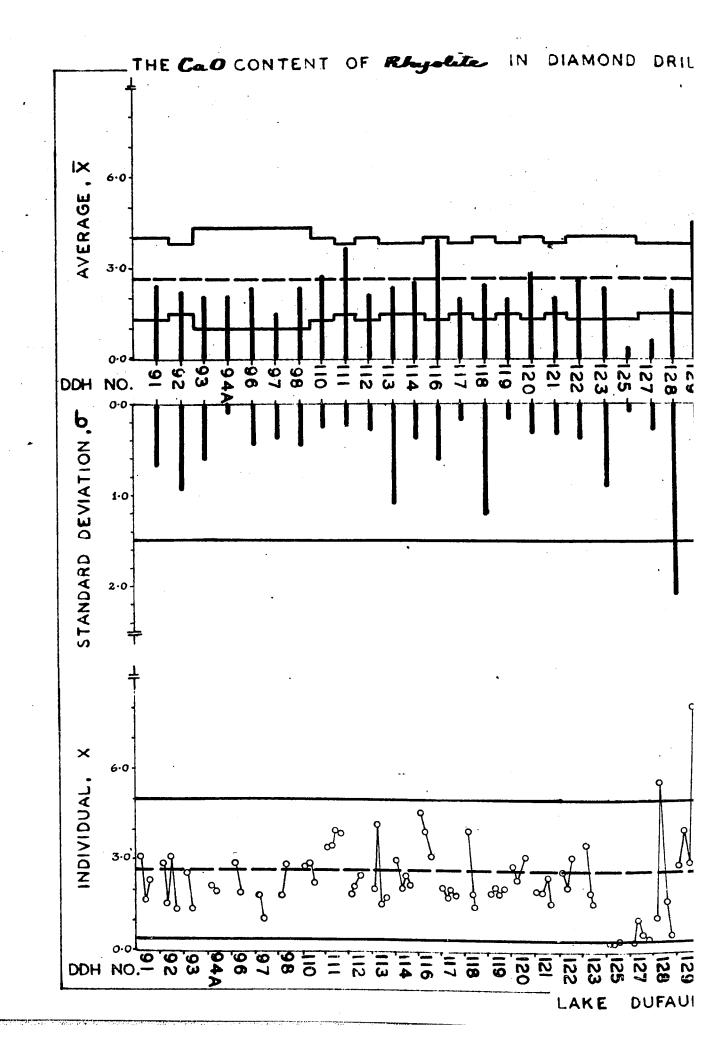
Of the mineralized rhyolite below the ore body, that cut by holes 125 and 127 is unusually low in CaO while that cut by 132 and 133 is high in CaO (142 is on the upper limit line). Rhyolite cut by hole 129 below the ore body is high in CaO. Close to the ore body hole 162 cut rhyolite with a CaO content high enough to be on the border of anomalous. In the general vicinity of ore, holes 181 and 190 cut rhyolite which contains unusually large amounts of CaO.

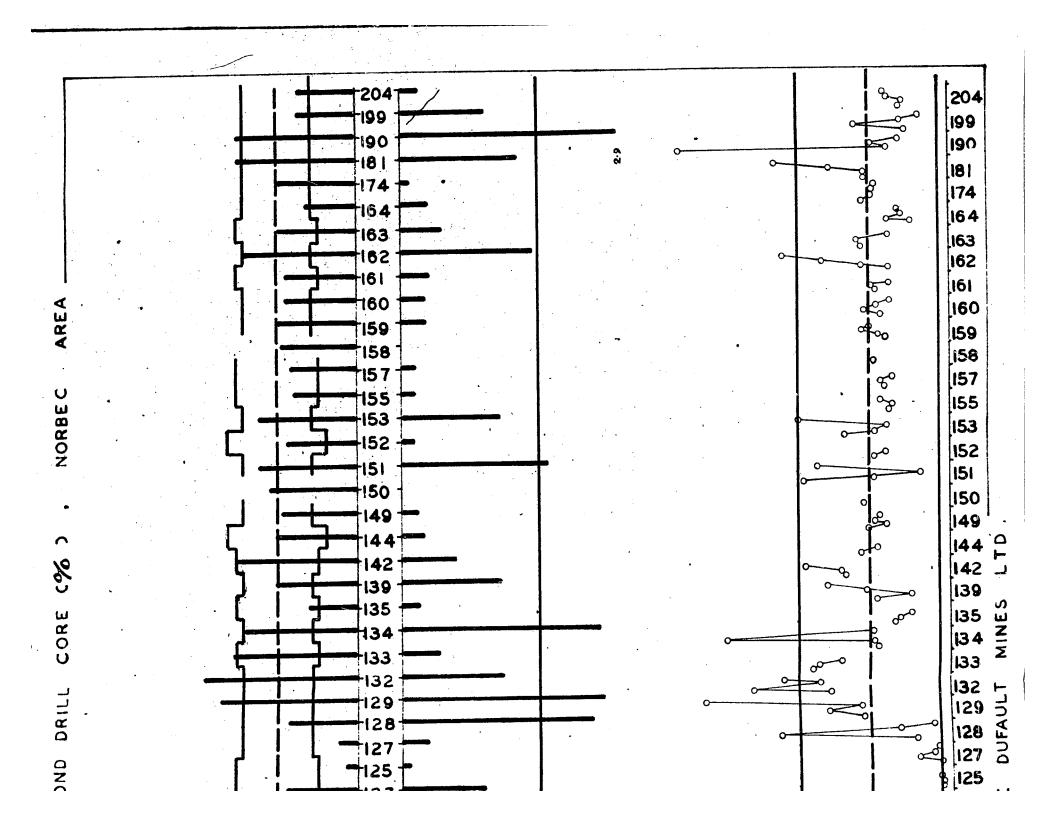
CaO Andesite chart 12

8.9%
Mo 10.3%
To 1.3%
FD Type I
ULB 8.9 + 2.0%
LLB 8.9 - 2.0%
CA 2.9% of \$\bar{x}\$
R .19

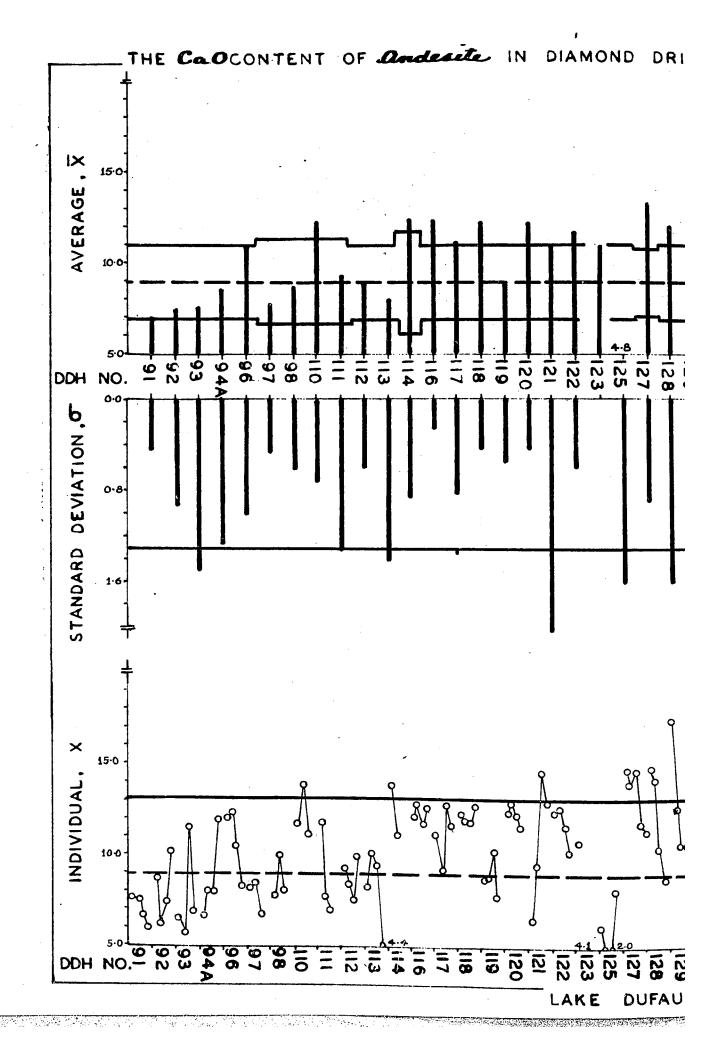
A few holes over and away from the ore body cut andesite with an unusually high content of CaD.

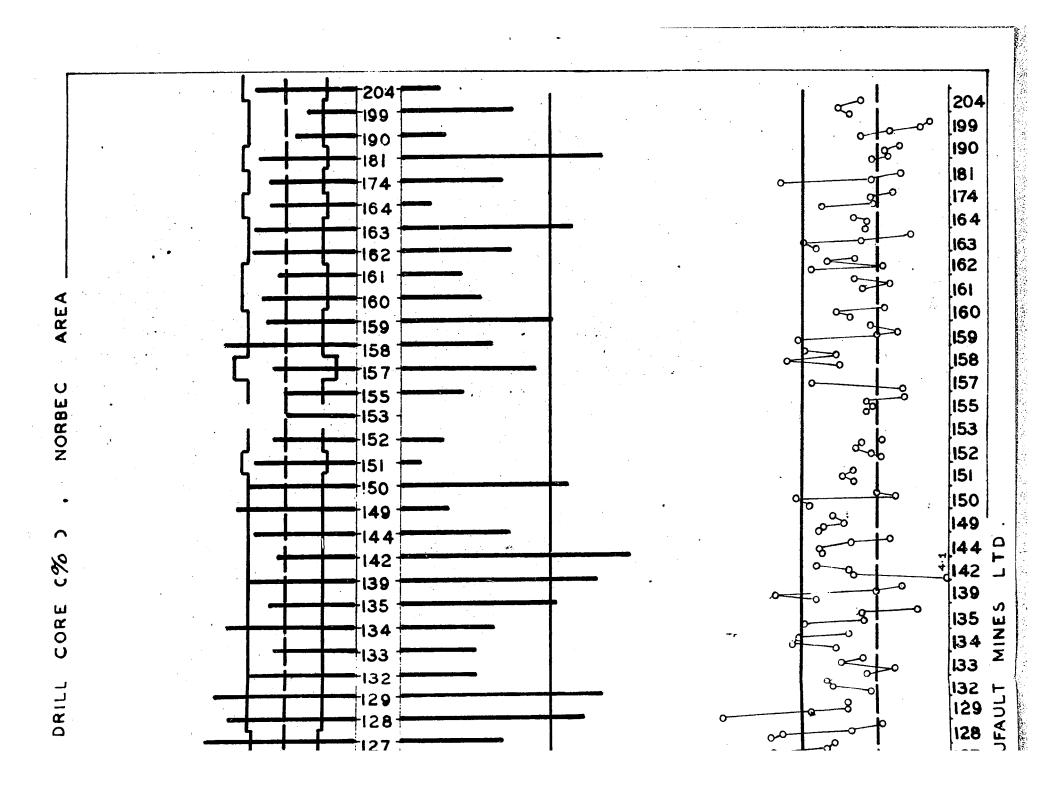
CaO rhyolite





CaO andesite





Potassium (oxide) (K20) Rhyolite chart 13

\$\bar{x}\$ 1.54%

Mo 1.3%

\$\bar{x}\$.50%

FD Type I

ULB 1.54 + .78%

LLB 1.54 - .78%

CA 6.6% of \$\bar{x}\$

R

Diamond drill hole 118 cuts rhyolite with an unusually low $K_2\text{O}$ content. Hole 199 cuts a section of rhyolite containing abnormally large amounts of $K_2\text{O}$.

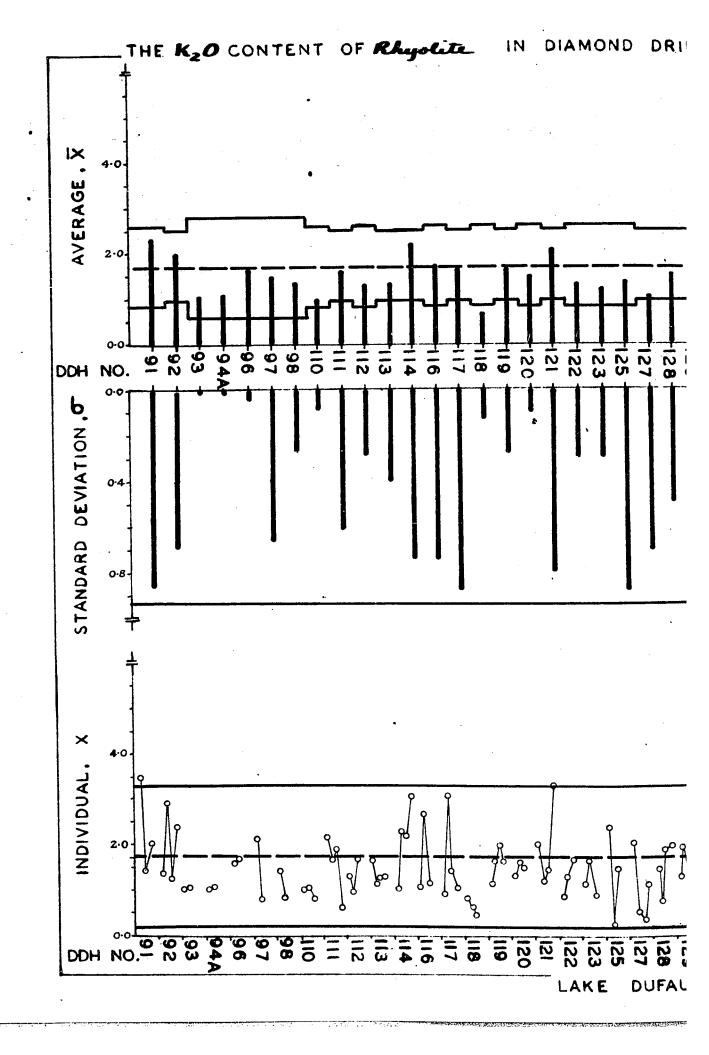
K20 Andesite chart 14

1.11%
Mo 0.5%

0.36%
FD Type II
a 0.75%
b 1.00%
c 1.25%
CA 14.0% of \$\overline{x}\$
R .46

Most of the drill holes show an <u>average K_2O </u> content which is slightly high compared with all andesite samples. This is the case because the samples of andesite closest to the main contact often exhibit high contents of K_2O .

K20 rhyolite



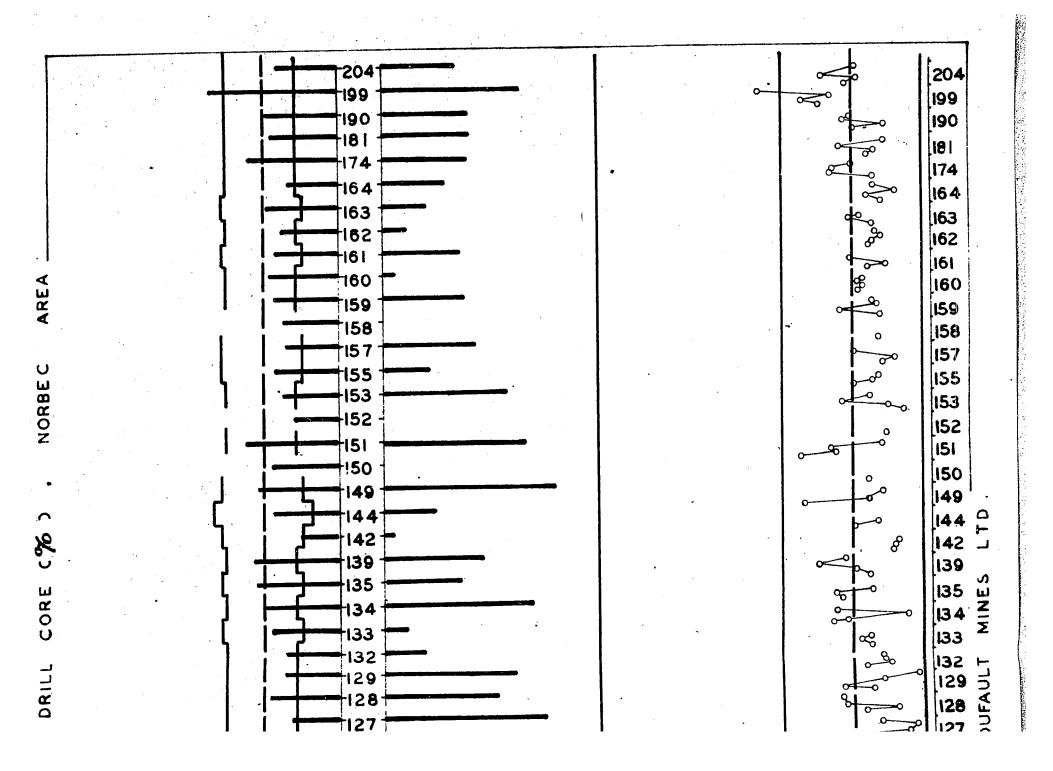
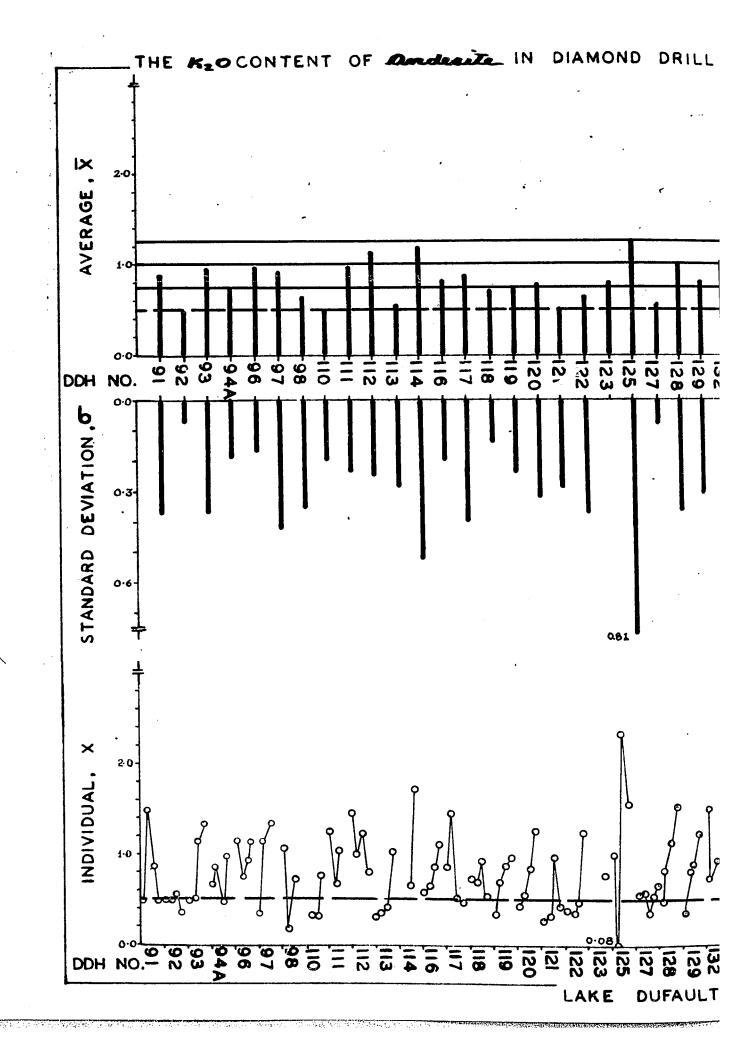
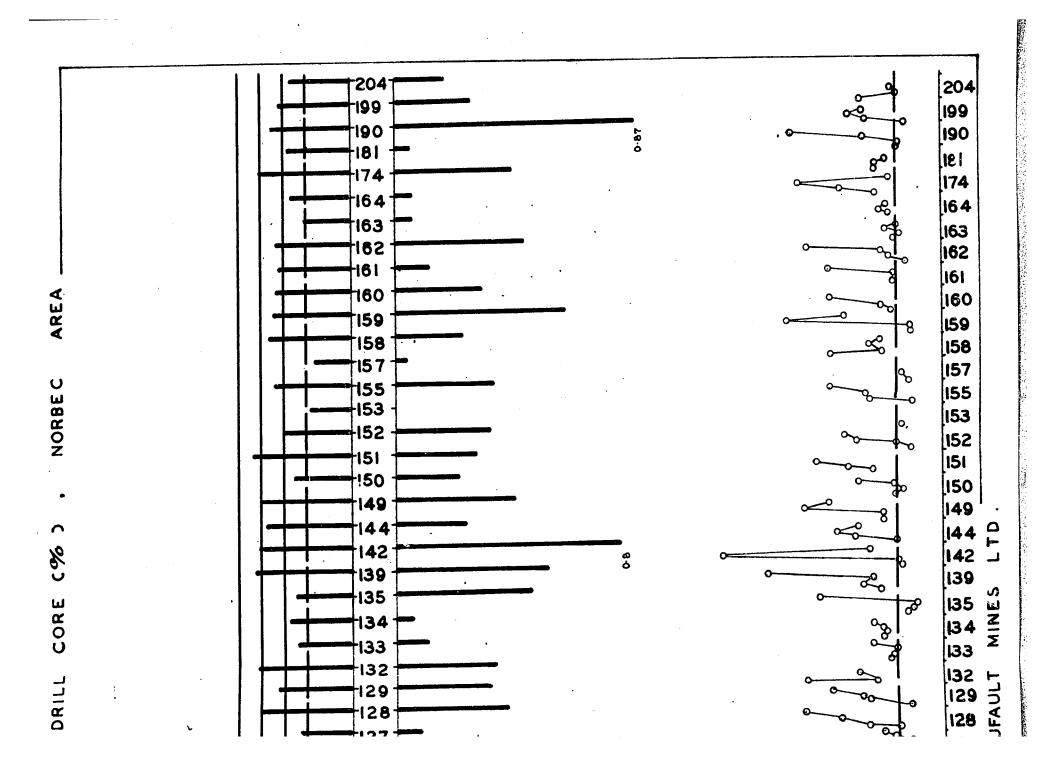


Chart 14 K₂O andesite





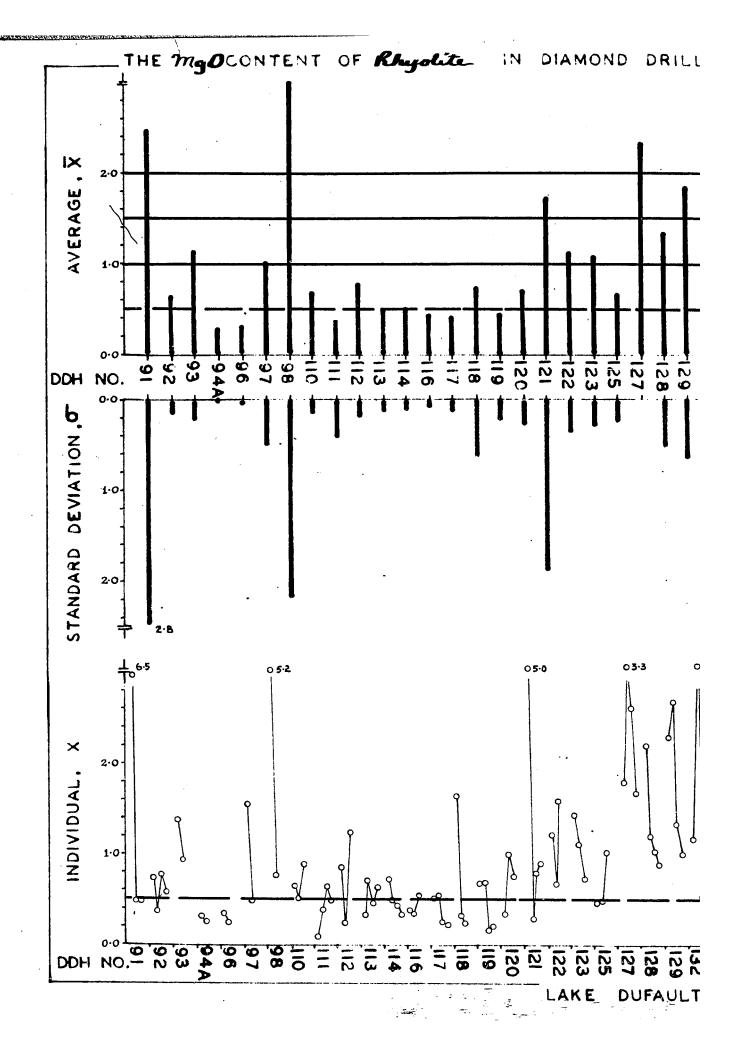
Magnesium (oxide) (MgO) Rhyolite chart 15

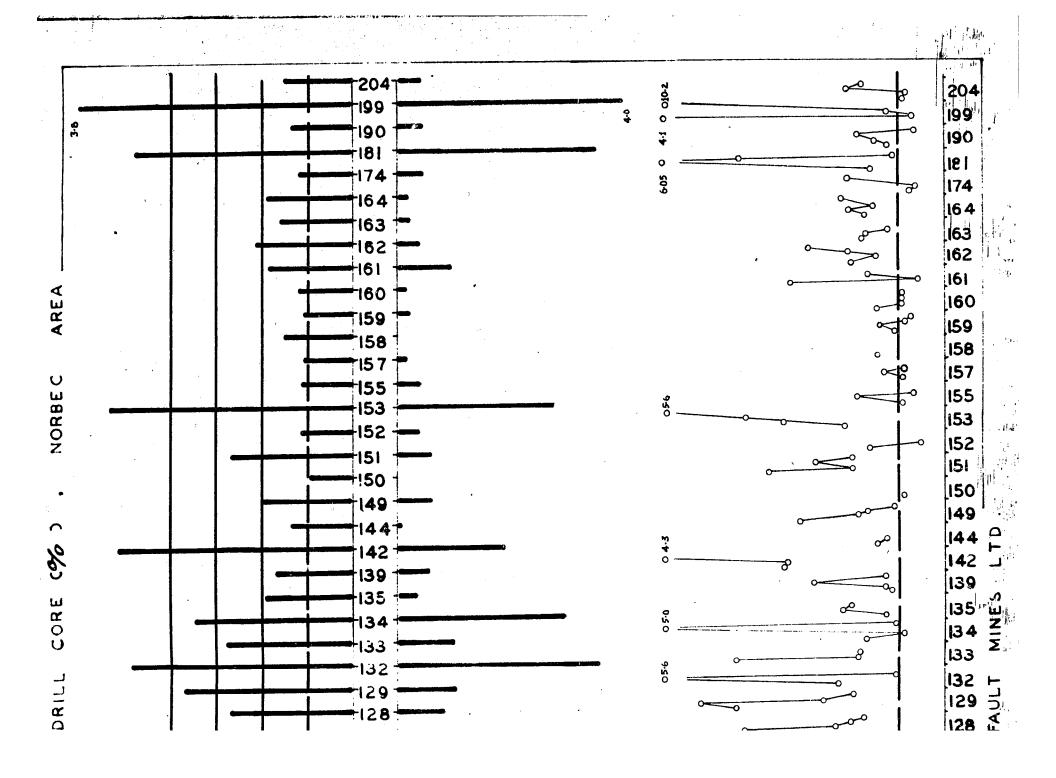
Sections of rhyolite containing unusually high amounts of MgO occur mainly near the ore body and in the northwest portion of the Norbec area.

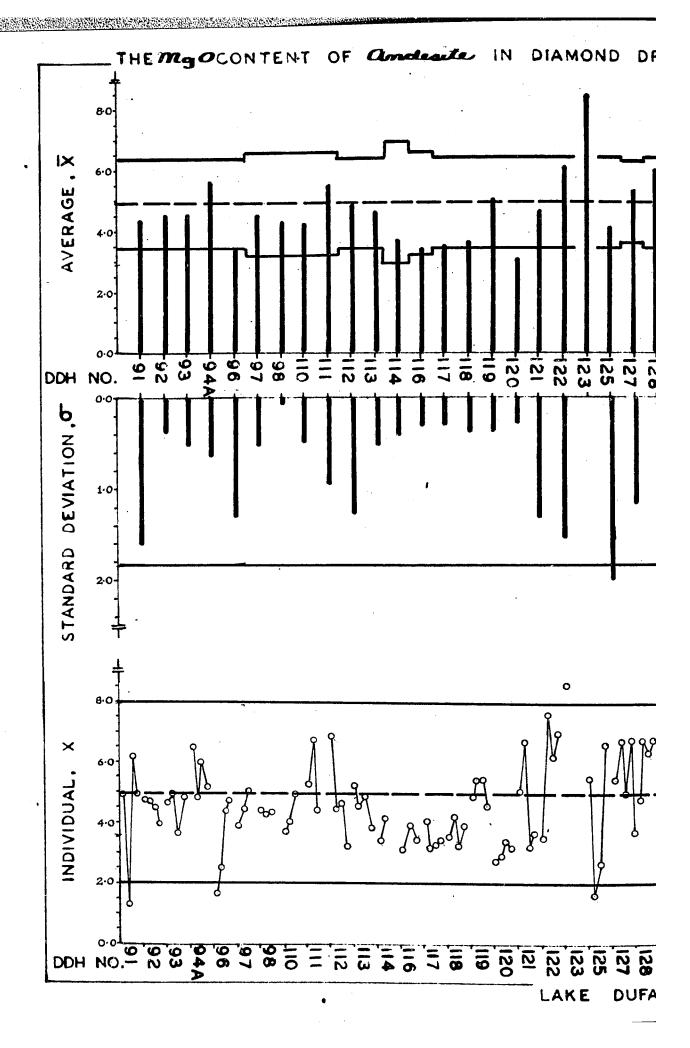
MgO Andesite chart 16

Holes cutting andesite of an unusually high MgO content are 132, 163 and 204, all in the vicinity of the ore body. Andesite of unusually low MgO content was cut by holes 96, 120 and 158, all far away from known ore.

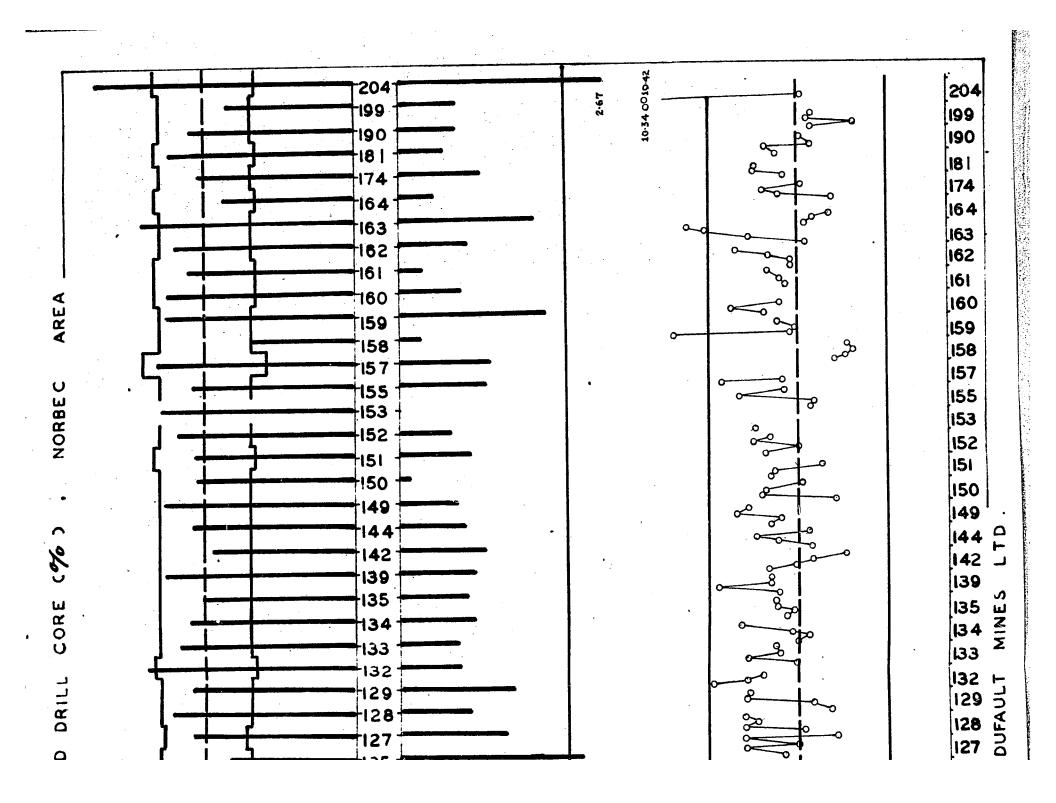
MgO rhyolite







- 🔘



Sodium (oxide) (Na₂O) Rhyolite chart 17

3.13%
Mo 2.70%

\$\overline{G}\$ 0.73%

FD Type III

a 3.3%

b 4.4%

-a 1.7%

-b 0.5%

CA 20.9% of \$\overline{x}\$

R 0.90

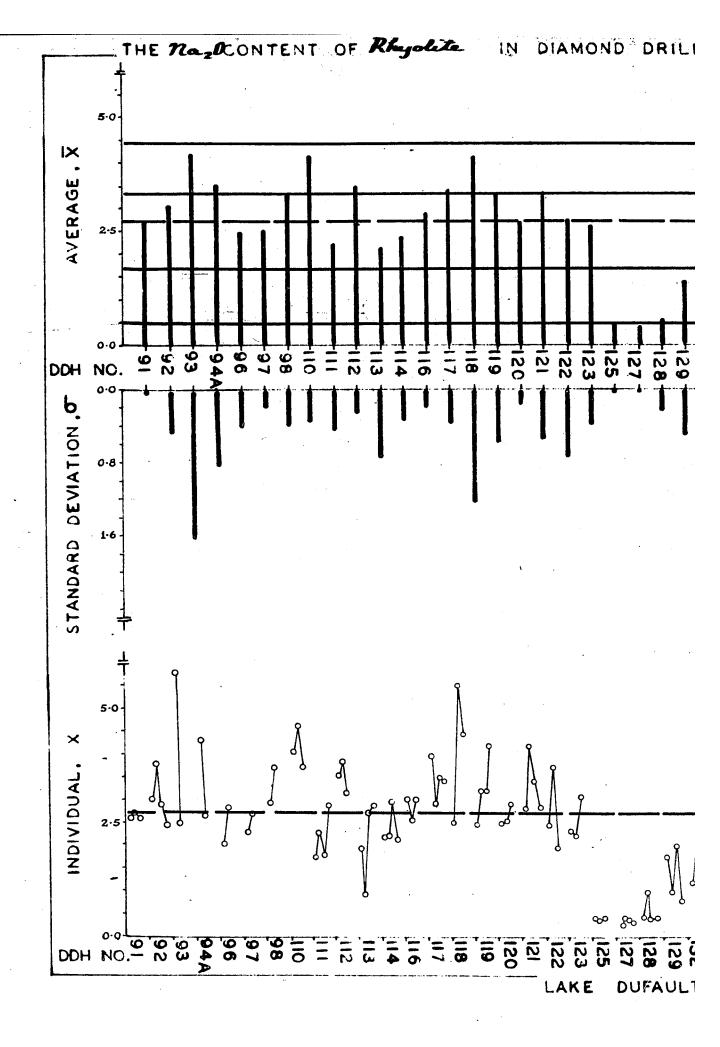
Mineralized rhyolite near known ore contains unusually low amounts of Na_2O . Rhyolite containing amounts of Na_2O slightly greater than usual is cut by several drill holes in the Norbec area.

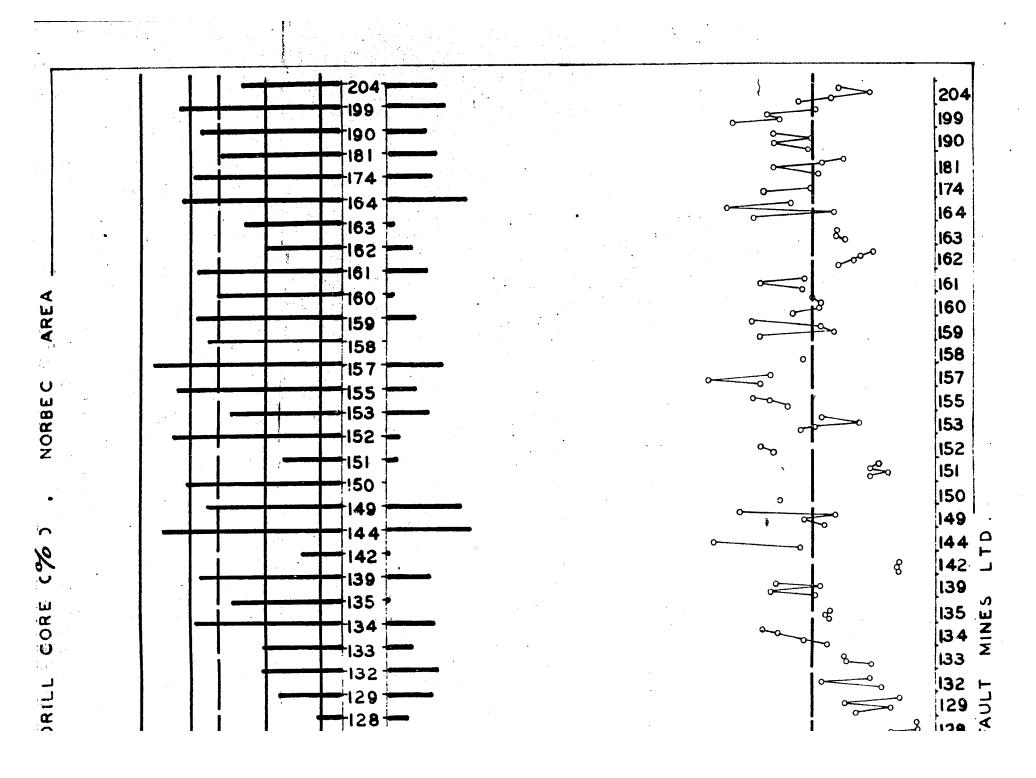
Na₂O Andesite chart 18

2.84% Mo 2.9% 5 .57% FD Type I ULB 2.84 + .85% LLB 2.84 - .85% R .55

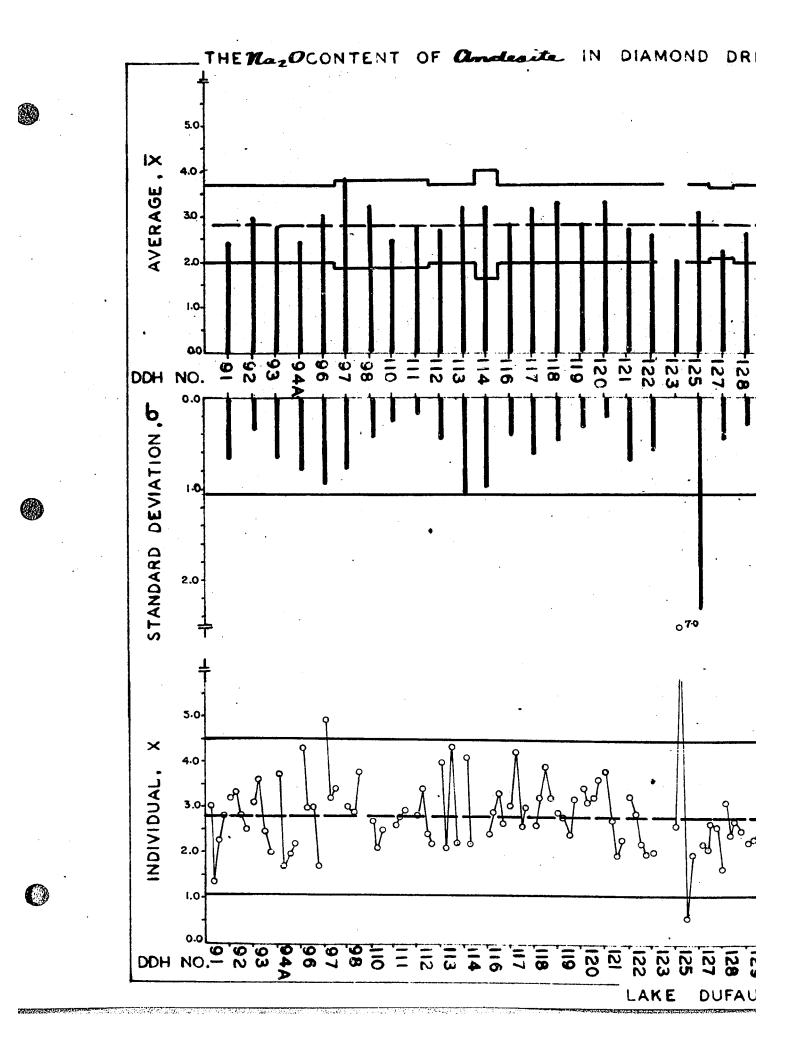
Only hole 97 cuts andesite which contains an unusually large amount of Na₂O.

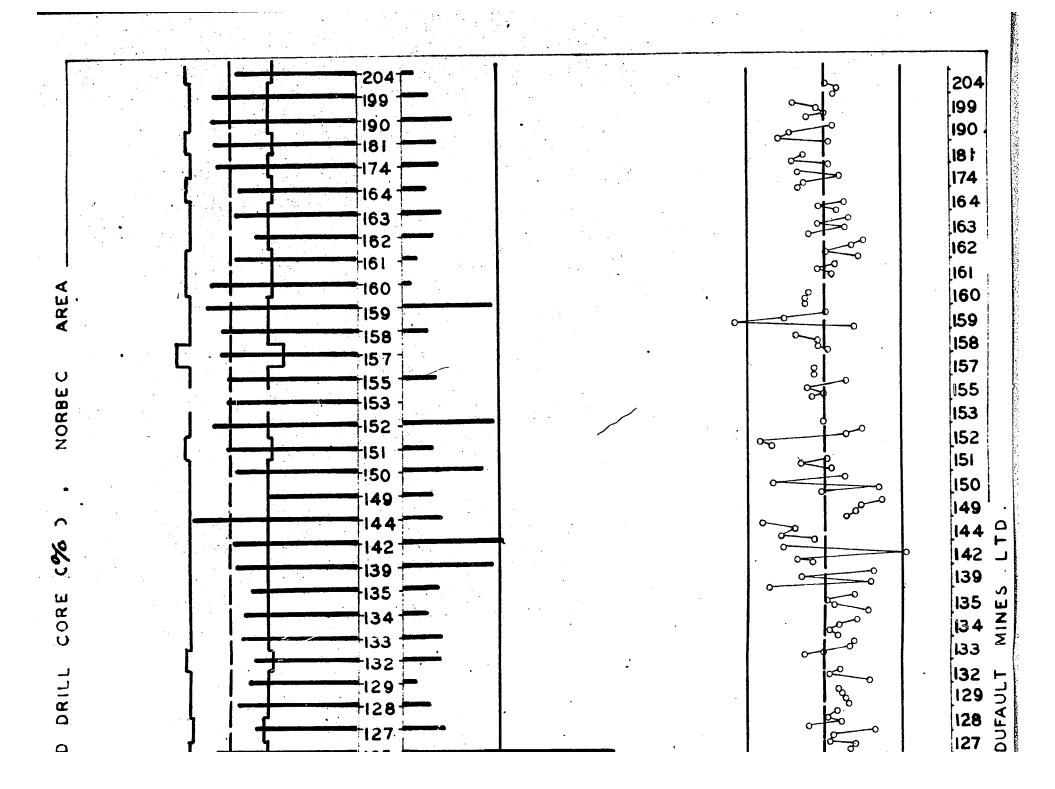
Na₂0 rhyolite





Na₂0 andesite





Silver (Ag) Rhyolite chart 19

Only two drill holes away from the ore body cut rhyolite with an extra high content of Ag (97, 116). All holes at the ore body and to the south except 133, 135 and 174 cut rhyolite containing 0.4ppm Ag or more.

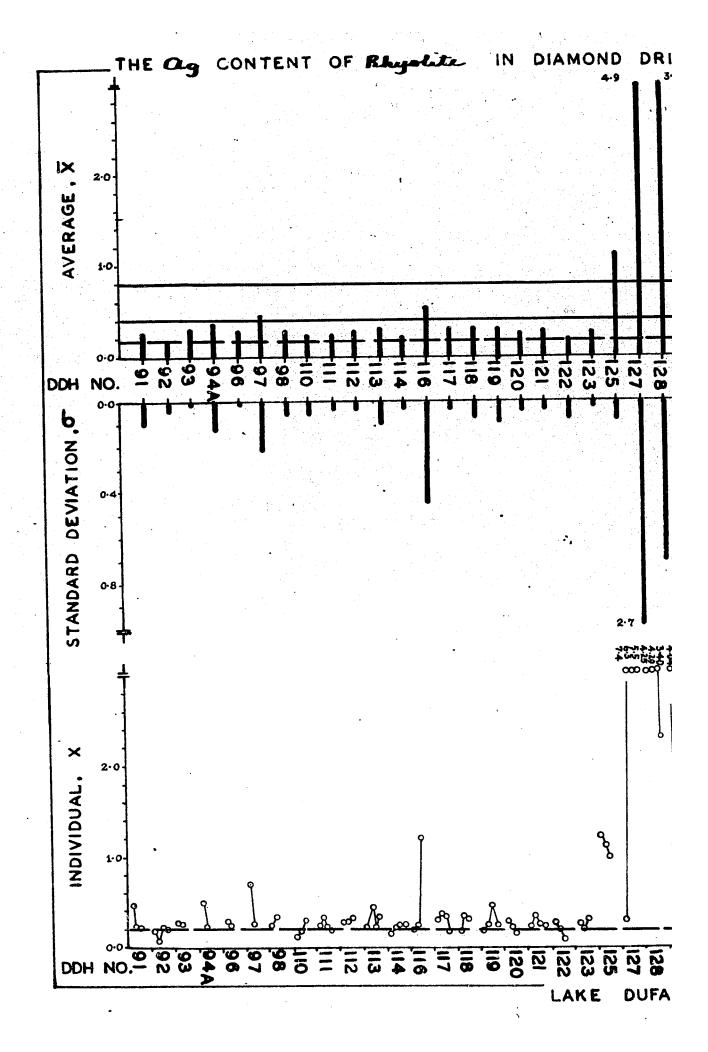
Ag Andesite chart 20

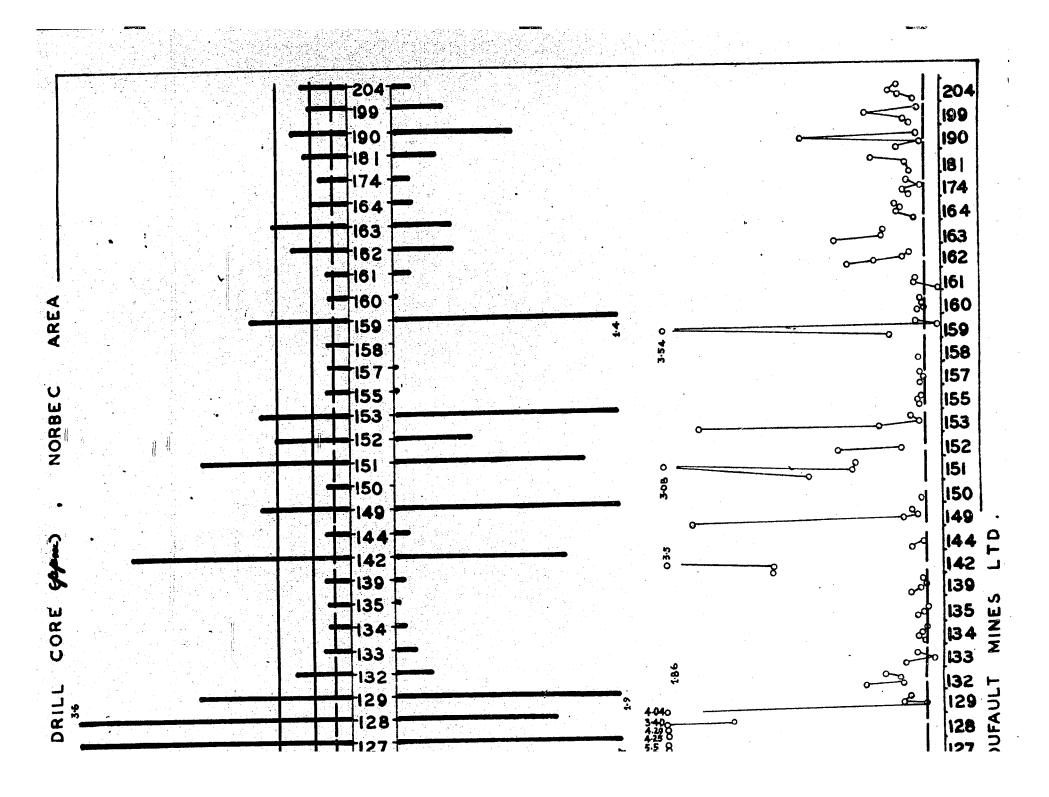
.24ppm
.18ppm
.10ppm
FD Type II
a .35ppm
b .50ppm
c 1.00ppm
CA 53.7% of x
8

Only one hole far away from the ore body (150) cuts andesite high in Ag.

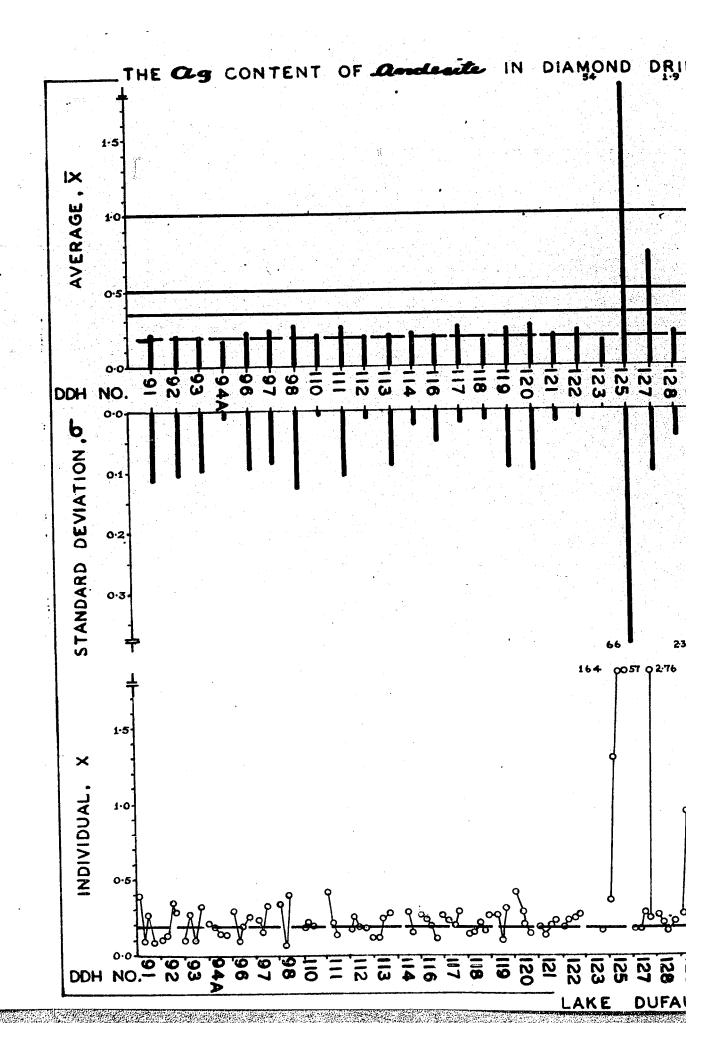
All holes over the ore body except numbers 133 and 128 contain abnormally large amounts of Ag. Here, some andesite samples taken from holes 125 and 129 contain over 2ppm Ag. All other anomalous holes close to the ore body are located to the south.

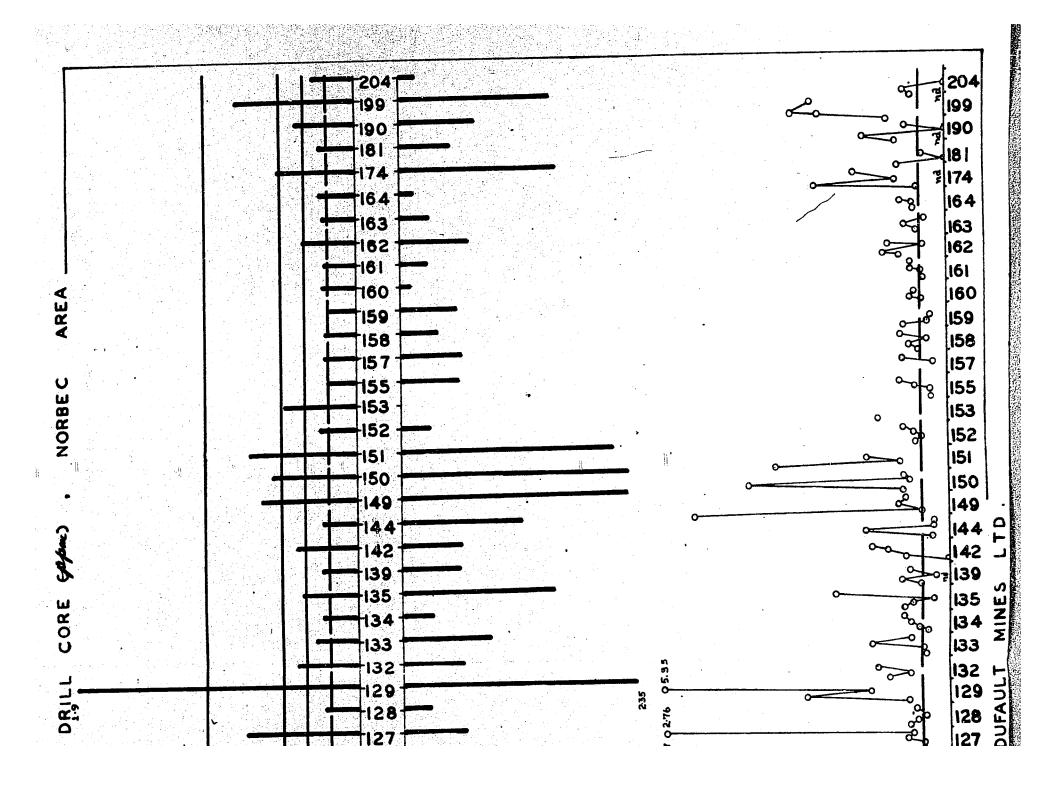
Chart 19
Ag rhyolite





Ag andesite





Cobalt (Co) Rhyolite chart 21

 C_A 20.9% of \bar{x} (Detection limits approximately 1 ppm)

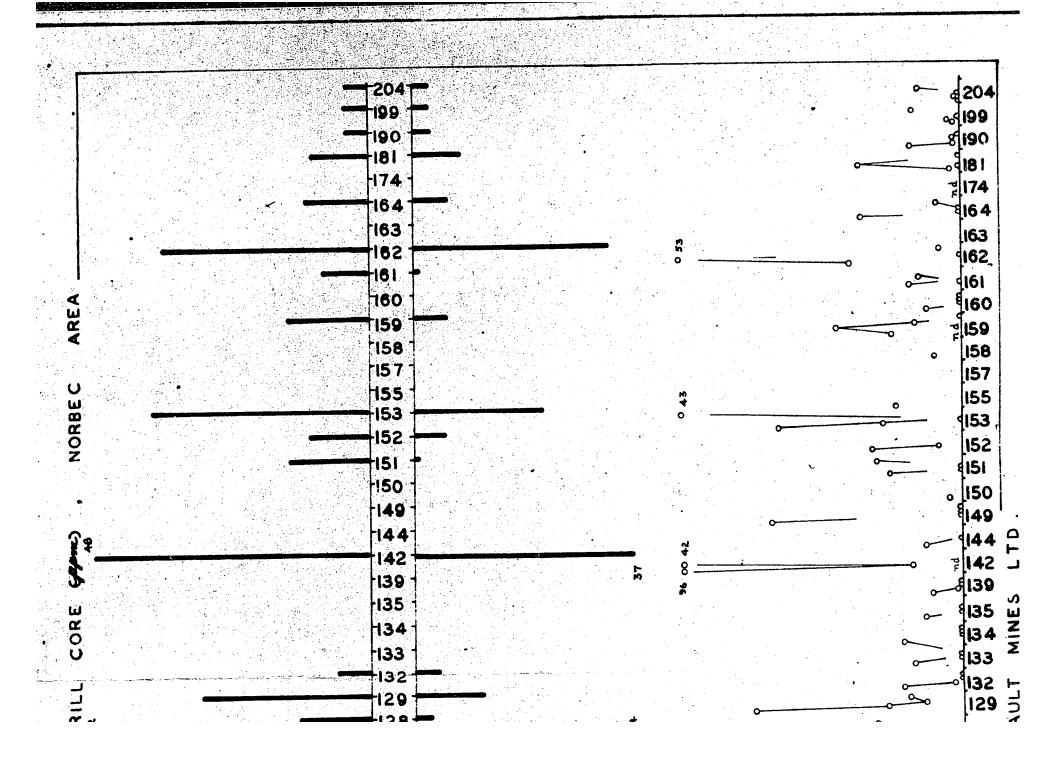
Most of the rhyolite samples contain less than detectable (nd: not detectable) amounts of Co. Those which do contain detectable amounts are anomalous and generally occur next to the main contact and near the ore body.

Co Andesite chart 22

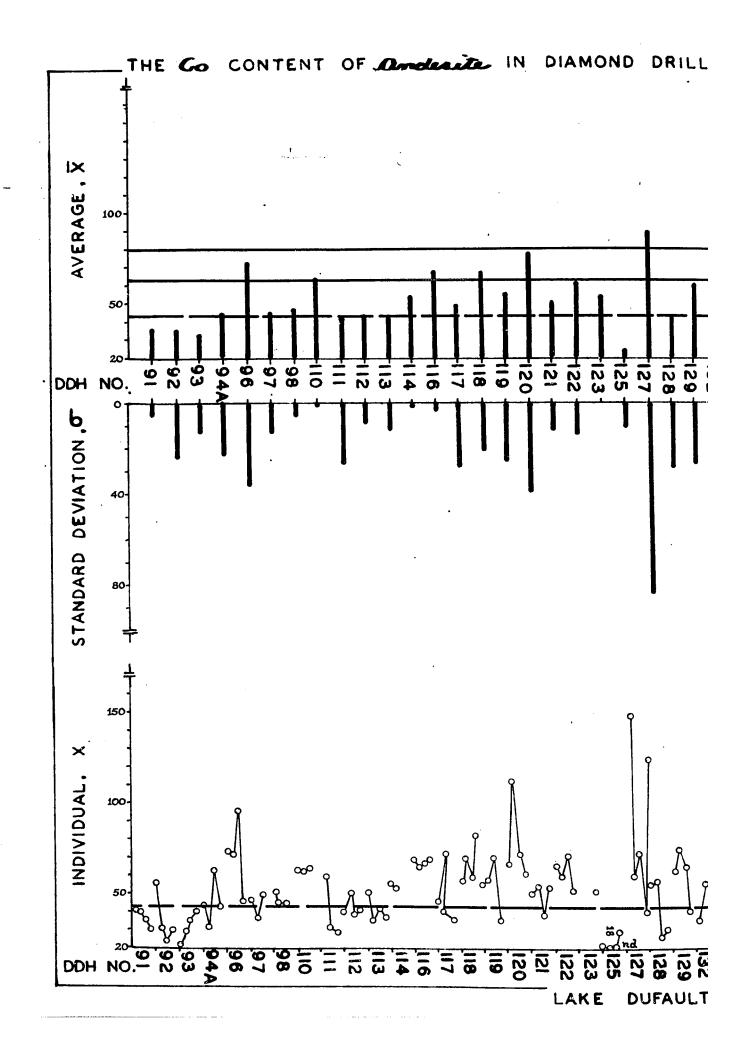
= x 53.8ppm Mo 43.0ppm To 15.8ppm FD Type II a 63ppm b 80ppm CA 13.9% of x R .48

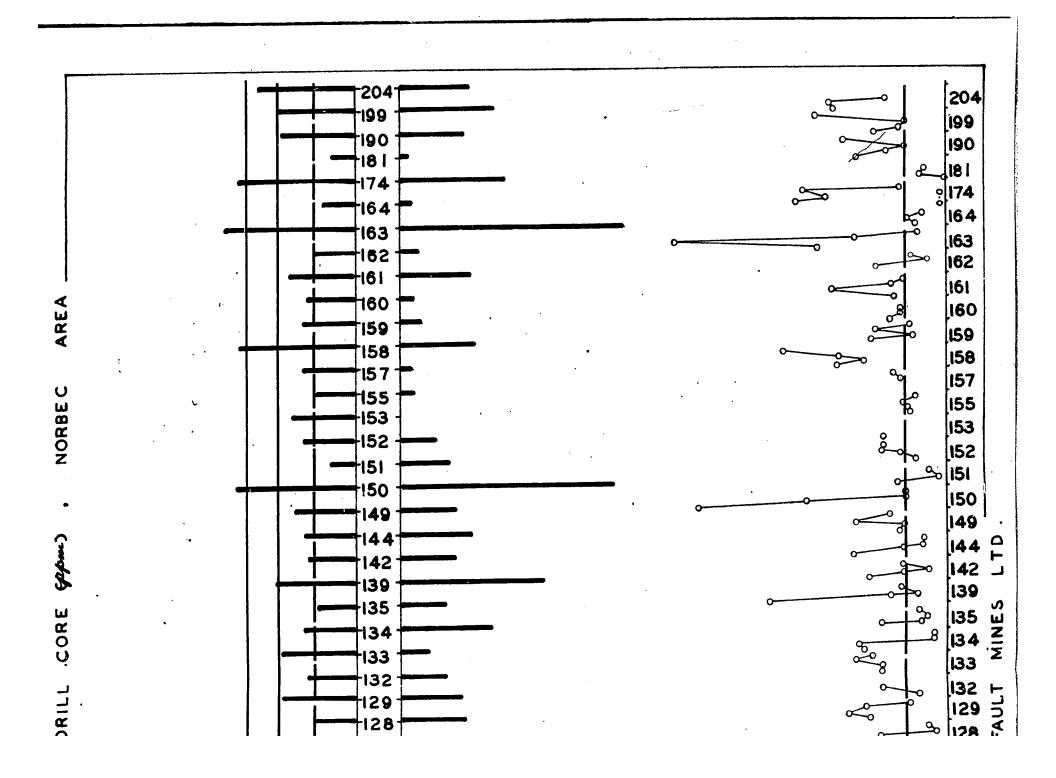
Andesite samples from some scattered drill holes away from and over the ore body have an unusually high Co content.

Co rhyolite



Co andesite





Chromium (Cr) Rhyolite chart 23

 C_A 27.5% of \bar{x} (Detection limit approximately 40ppm)

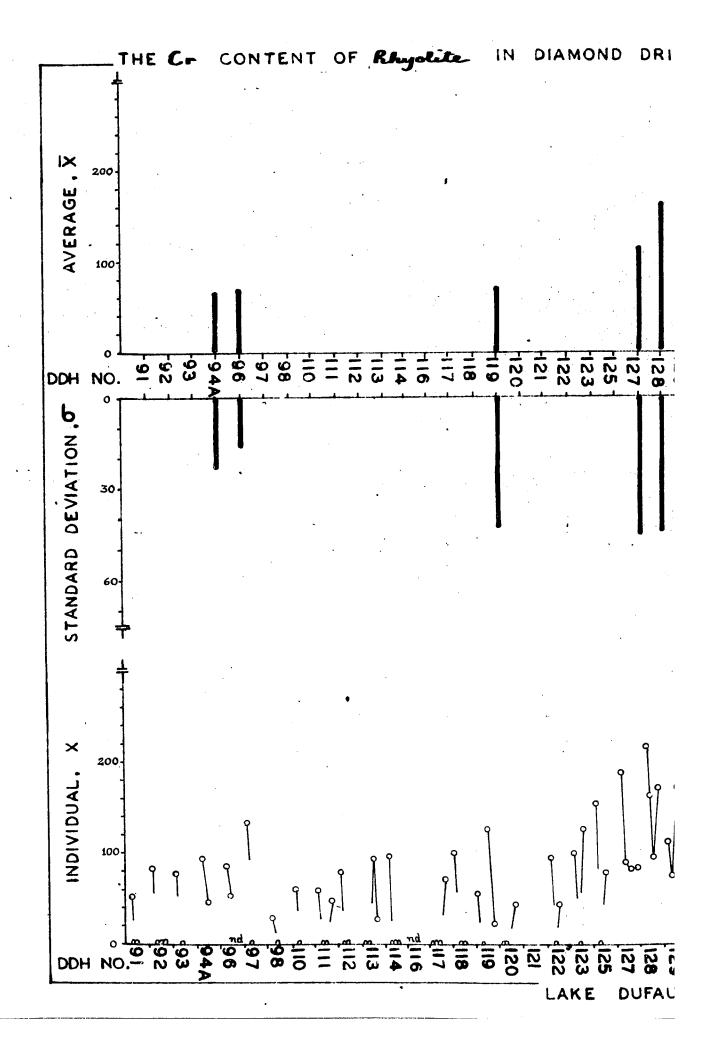
Most of the rhyolite samples contain less than detectable amounts of Cr. Those that contain greater than 40ppm are often the closest to the main contact and the ore body.

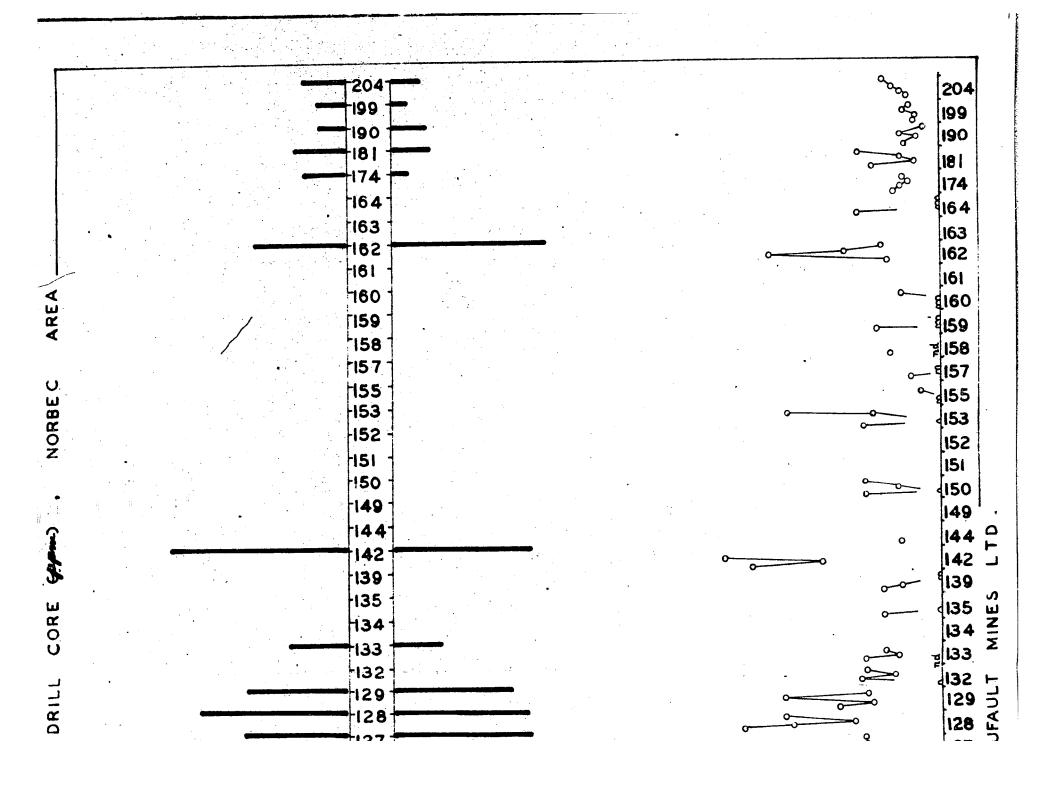
Cr Andesite chart 24

197ppm Mo 155ppm 5 58ppm FD Type II a 240ppm b 300ppm CA 13.6% of X R .46

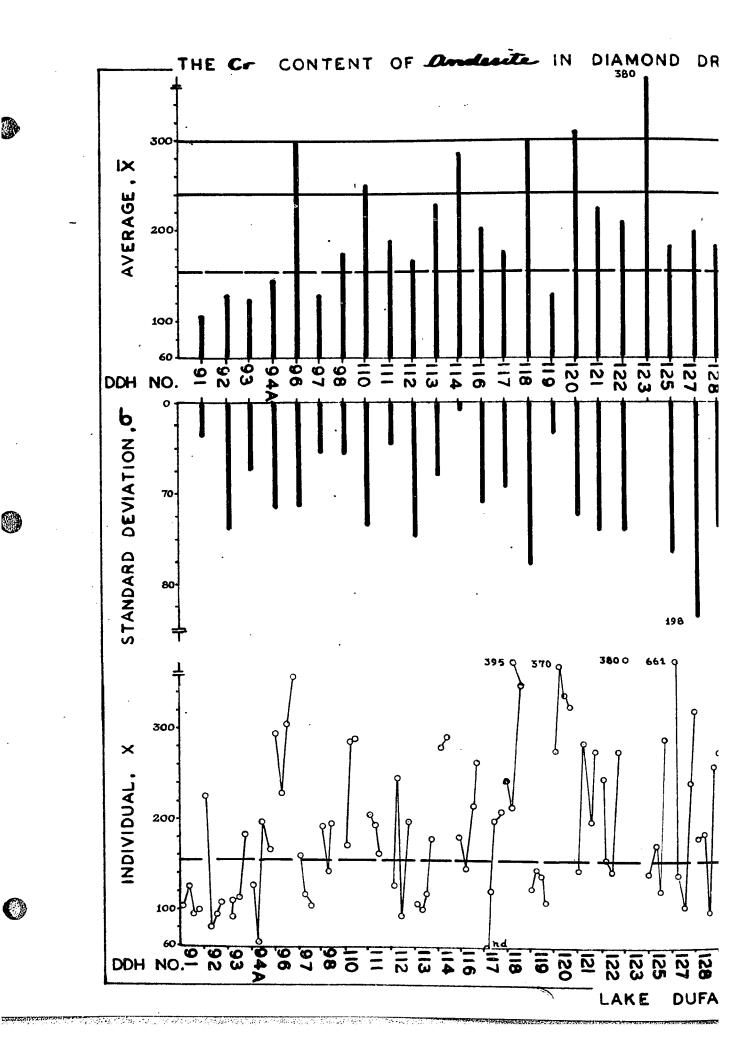
Only scattered sections of andesite contain abnormally large amounts of Cr.

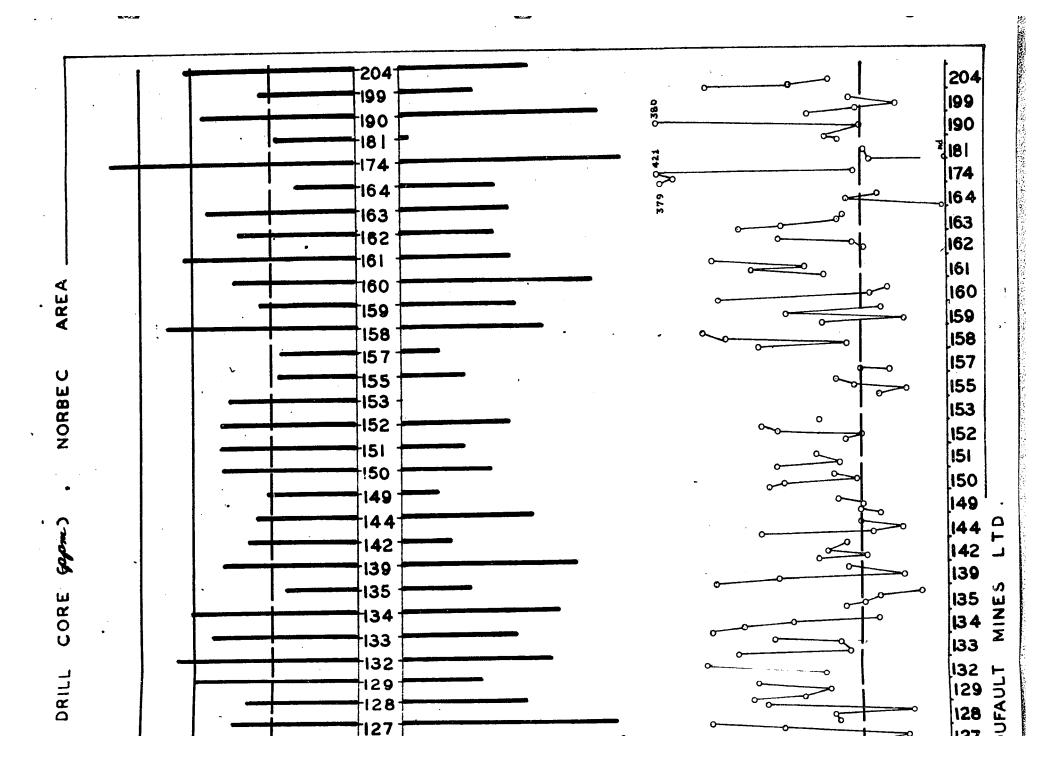
Cr rhyolite





Cr andesite





Copper (Cu) Rhyolite chart 25

104ppm
20ppm
73ppm
73ppm
FD Type V
80ppm
5 150ppm
6 225ppm
CA 20.4% of x
R .29

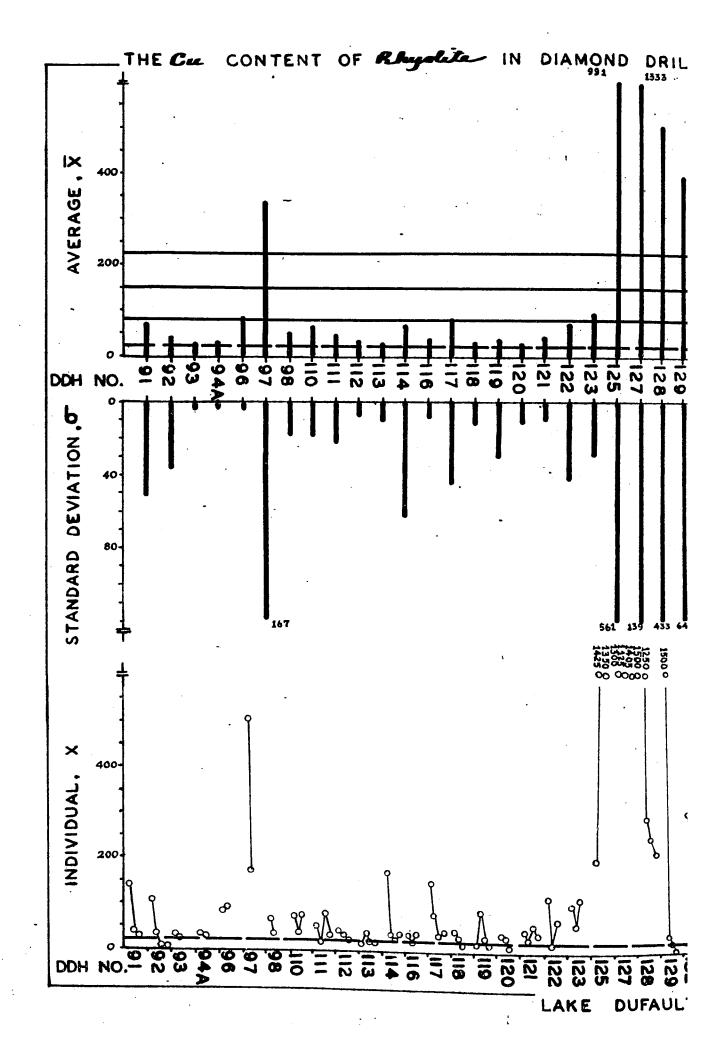
Holes 96 and 97 in the northwest corner of the property cut rhyolite containing unusually large amounts of Cu. Number 123 cuts rhyolite with a high Cu content. Some holes in the vicinity of the ore body cut rhyolite containing high amounts of Cu. Of these the most significant are numbers 134, 135, 139, 144, 152, 153, 159 and 162.

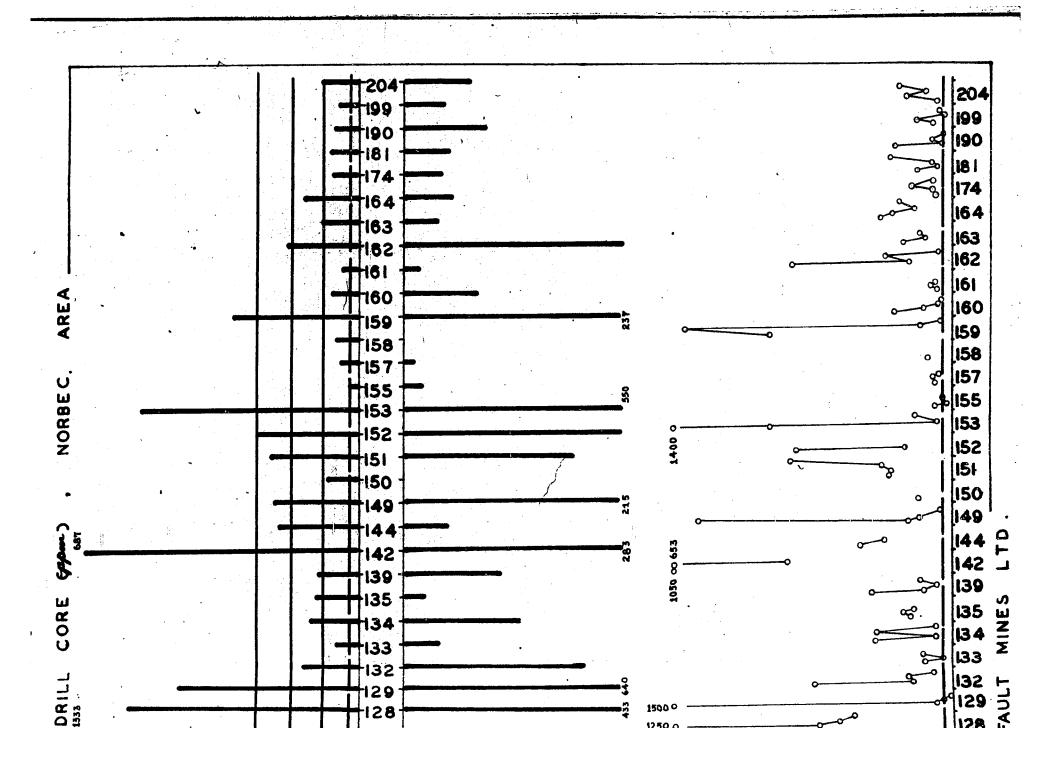
Cu Andesite chart 26

X 119ppm Mo 70ppm õ 101ppm FD Type IV а 100ppm -150ppm ь 250ppm C C_A 30.5% of x .34

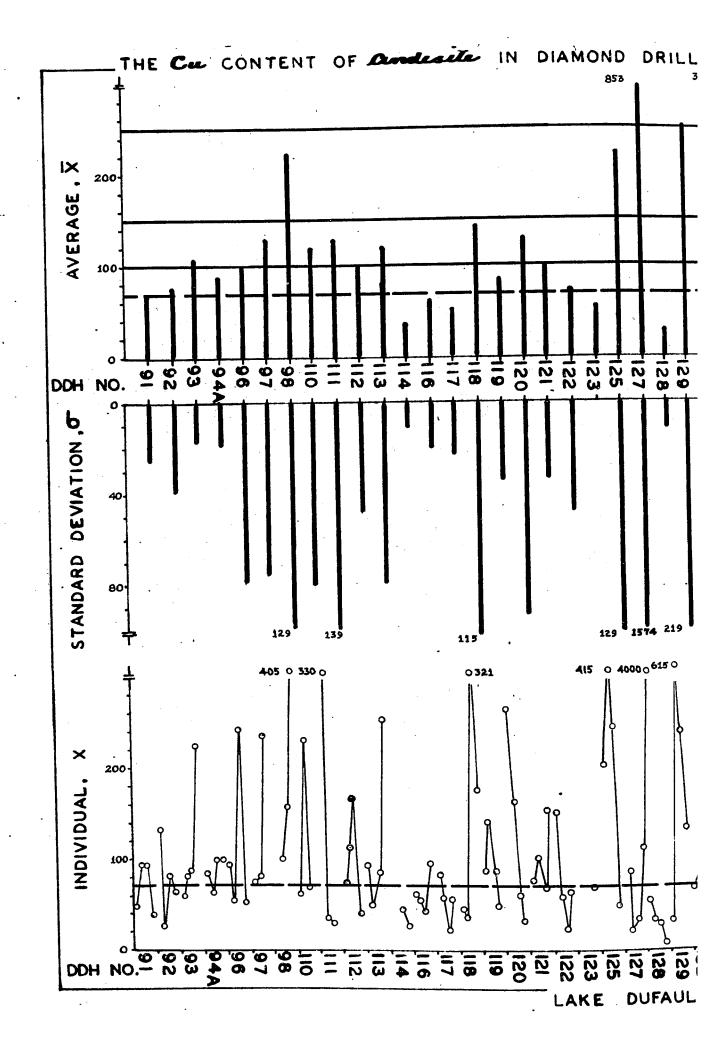
Most sections of andesite averaging greater than 250ppm occur over or near the ore body. Sampled sections of andesite averaging between 100 and 150ppm are scattered throughout the area. Not all the andesite sampled over the ore body has an exceptionally high Cu content. Hole 150 in the Mid-Norbec area cut andesite with a very high Cu content.

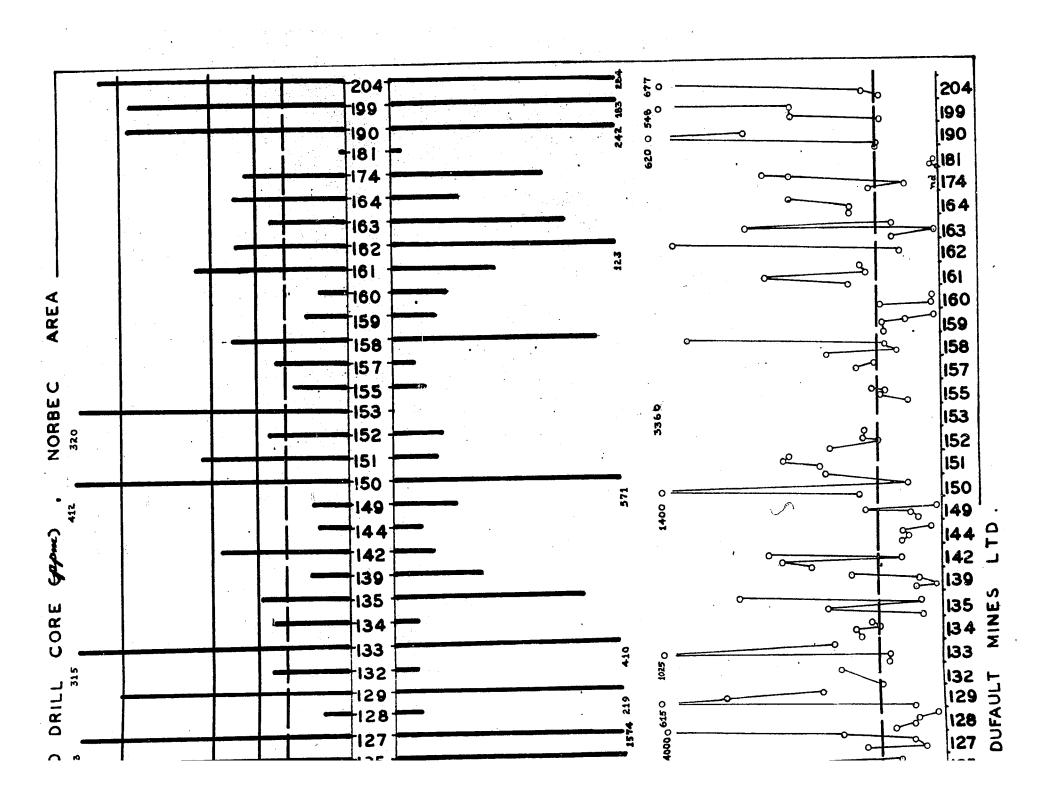
Cu rhyolite





Cu andesite





Gallium (Ga) Rhyolite chart 27

30ppm Mo 27ppm ō 4.6ppm FD Type I ULB 30 + 7ppm LLB 30 - 7ppm CA 9.3% of x R .60

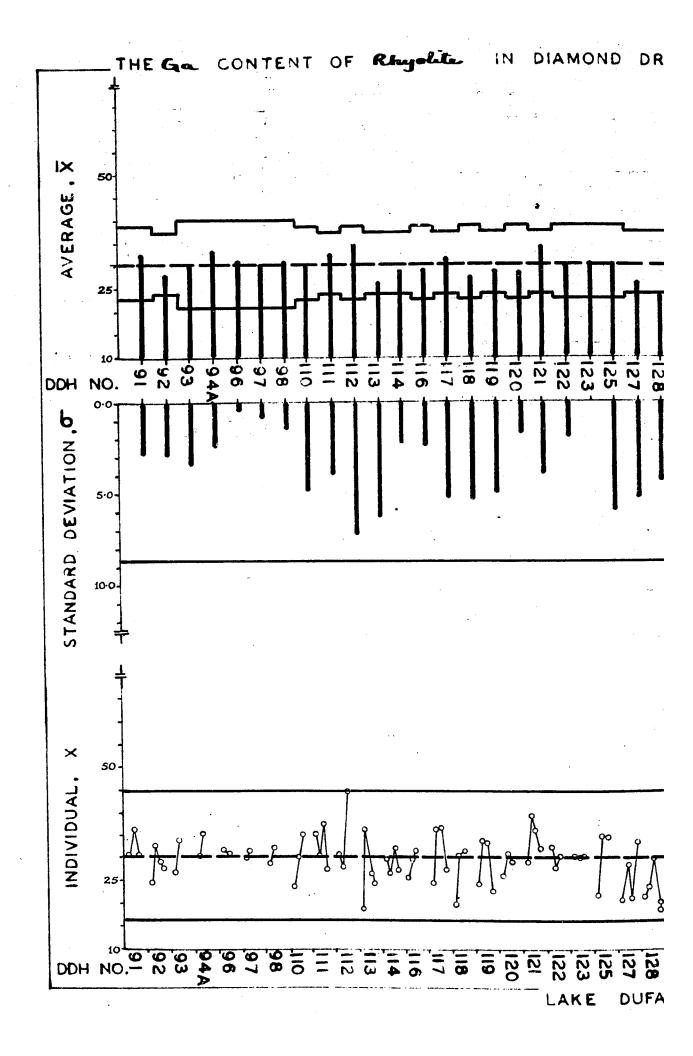
There are no anomalous values.

Ga Andesite chart 28

24.5ppm
Mo 24.5ppm
Type I
ULB 24.5 + 6.1ppm
LLB 24.5 - 6.1ppm
LLB 24.5 - 6.1ppm
CA 10.5% of \$\bar{x}\$
R .63

Holes 174 and 190 cut andesite with an unusually high Ga content.

Ga rhyolite



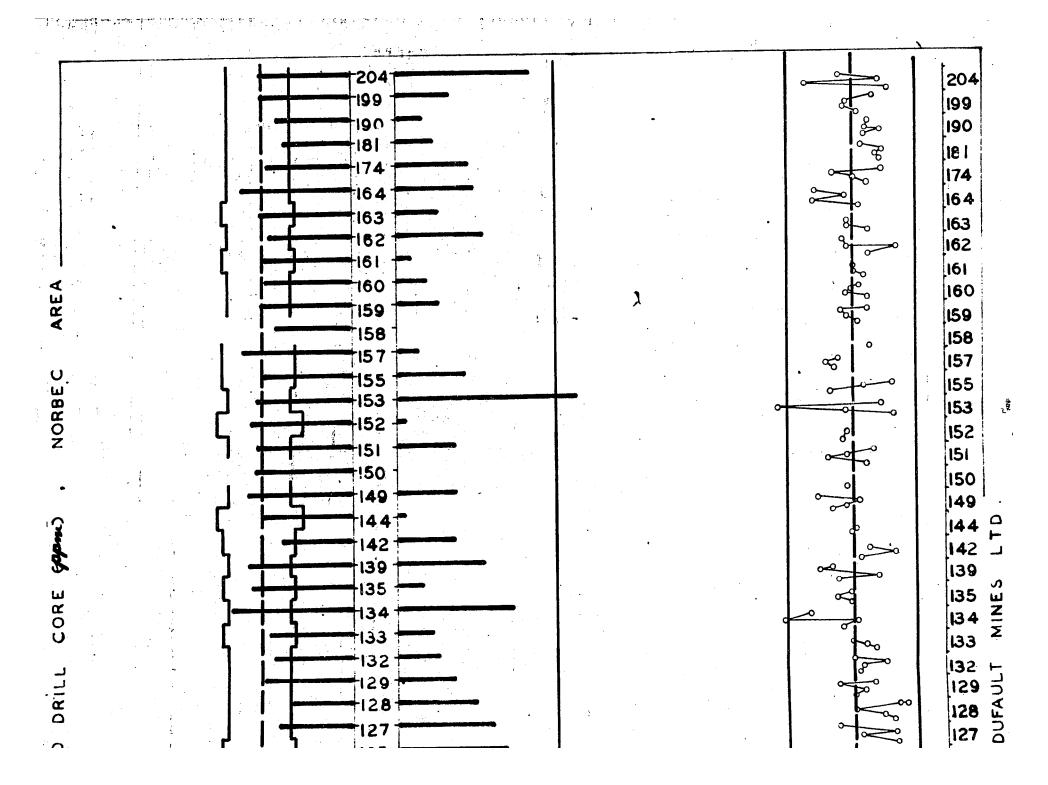
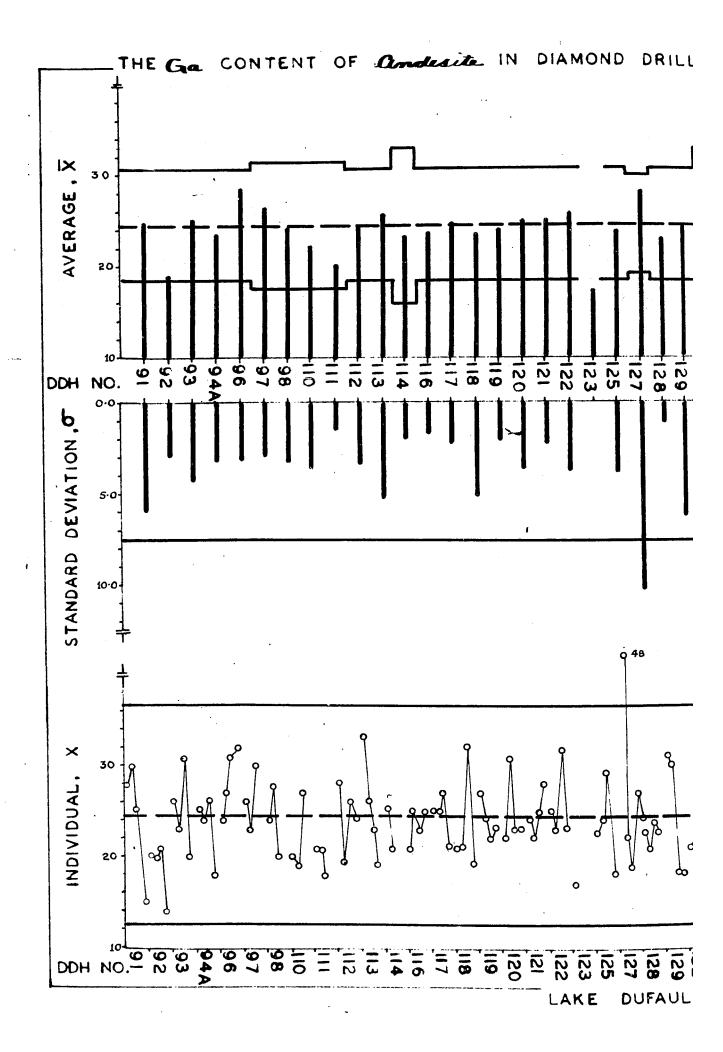
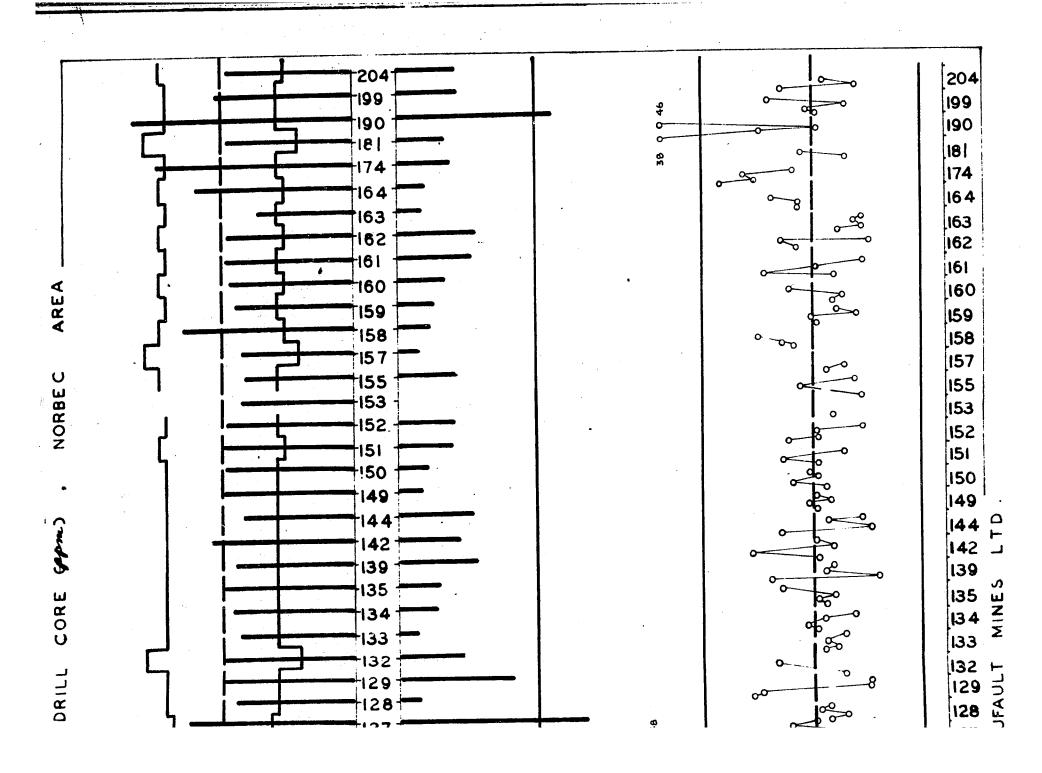


Chart 28
Ga andesite





Nickel (Ni) Rhyolite chart 29

Ξ× 7.8ppm 1.0ppm Mo 5 7.3ppm FD Type V 2.5ppm а 5.0ppm 10.0ppm C 78.7% of X C_A .84

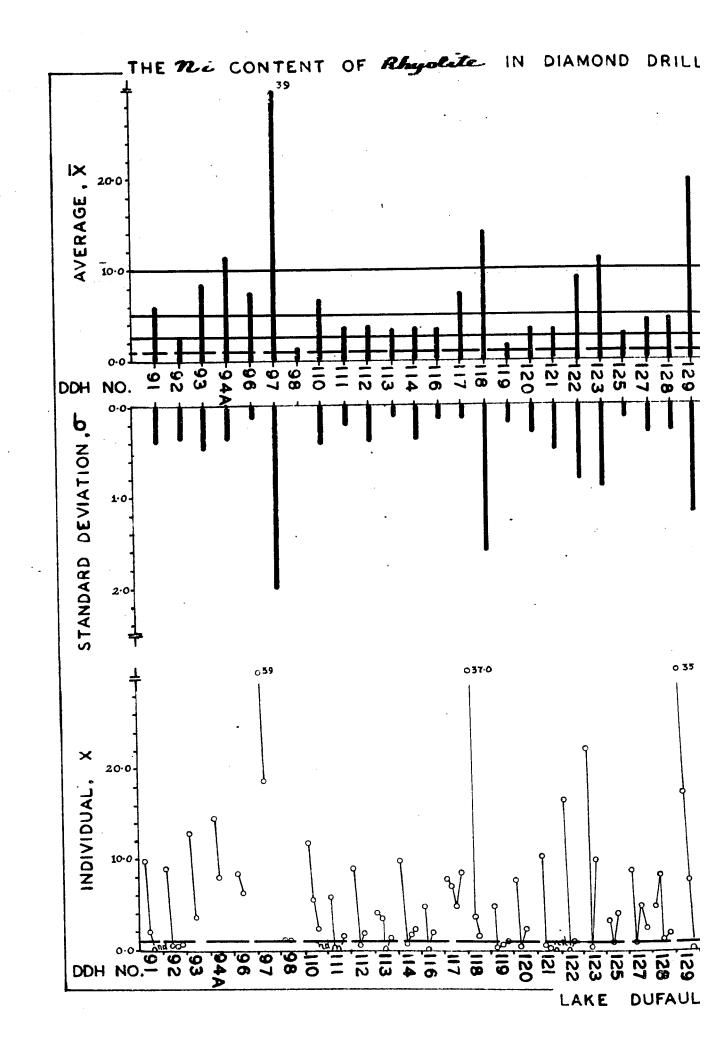
Exceptionally high values for Ni in rhyolite are scattered among all sampled drill holes in the area. One cause of this is that the highest individual values often occur closest to the main contact rather than near the ore body. Also, the analytical error is large.

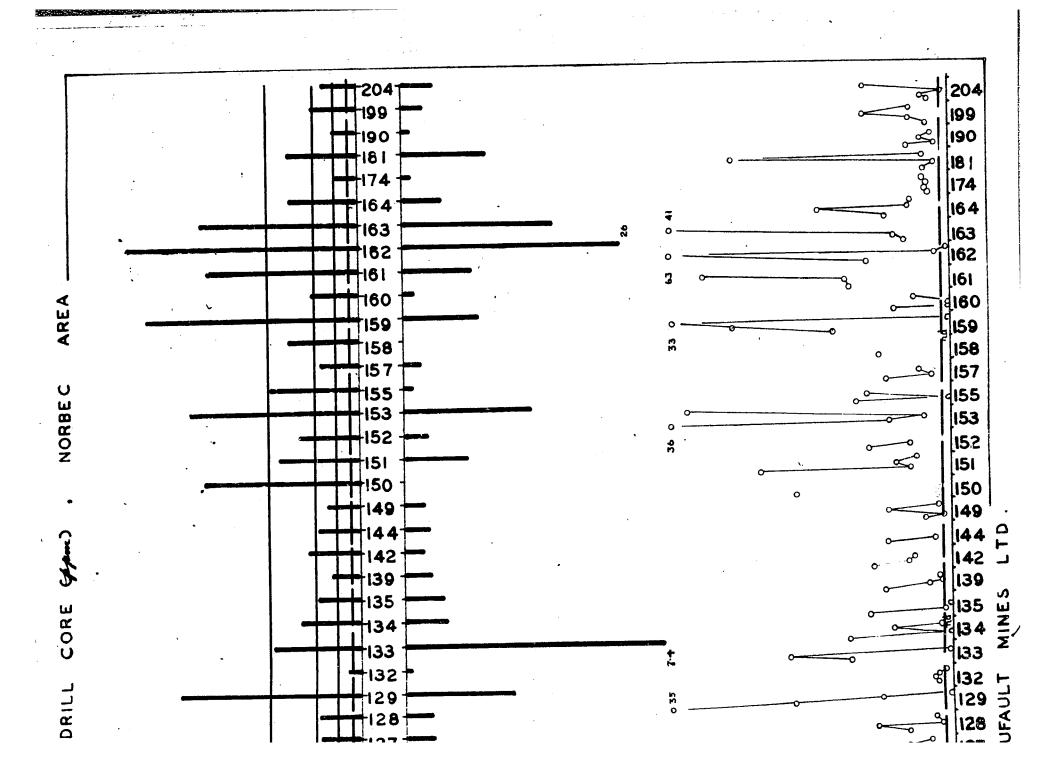
Ni Andesite chart 30

104ppm Mo 50ppm 5 44ppm FD Type V a 100ppm b 150ppm c 200ppm CA 30.3% of x R .72

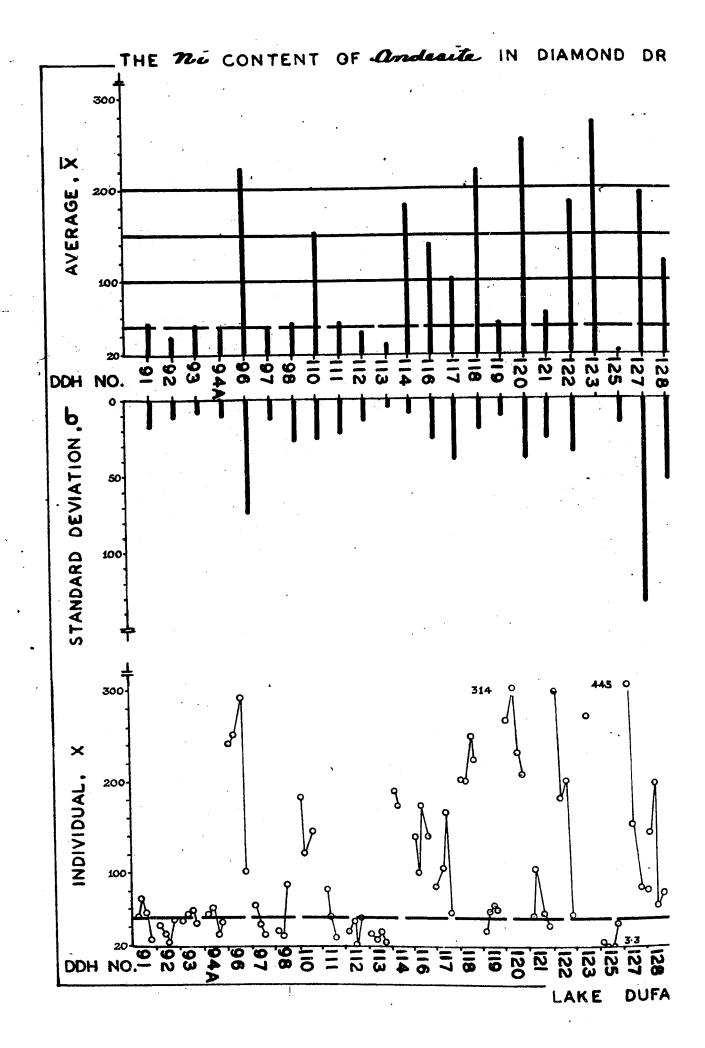
Anomalous values for Ni in andesite do not seem to bear any relation to the presence of ore. However, the lowest individual values are often the closest to the main contact.

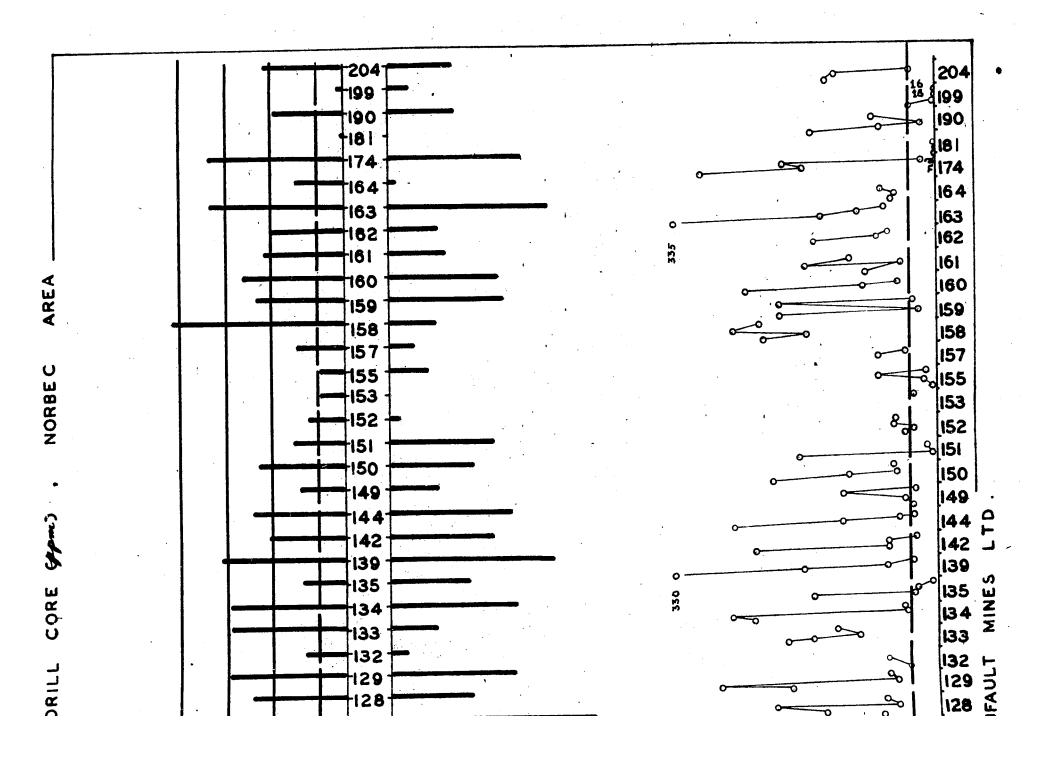
Ni rhyolite





Ni andesite





Lead (Pb) Rhyolite chart 31

Z.4ppm
Mo 0.7ppm
Type V
a 4ppm
b 7ppm
CA 16.4% of X
R

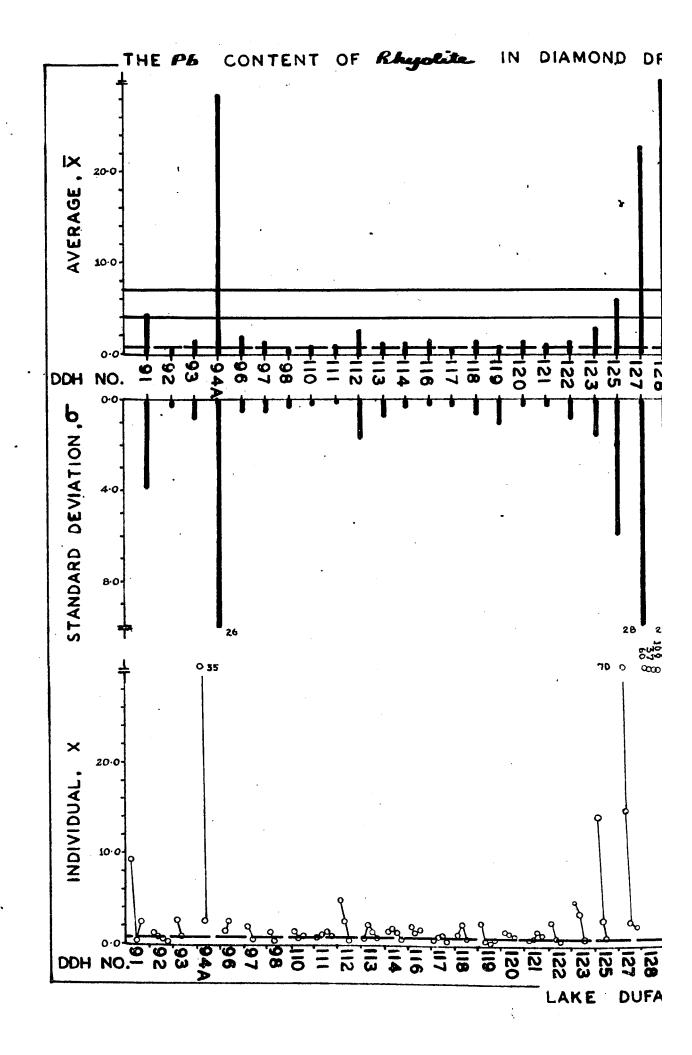
Abnormally high values for Pb in rhyolite occur only around the ore body and in the northwest corner of the property. Some holes which are very close to the ore body (152, 133, 162, 129) are not anomalous.

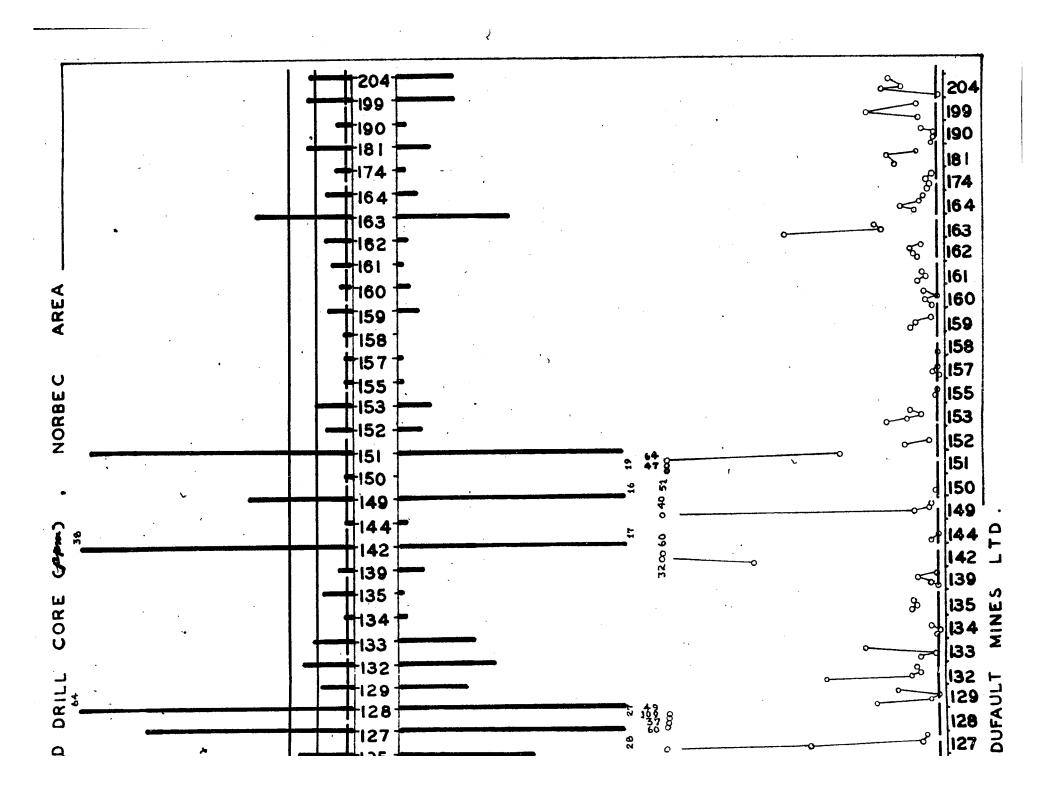
Pb Andesite chart 32

1.0ppm
Mo 0.3ppm
5 .9ppm
FD Type V
a 1.5ppm
b 3.0ppm
c 6.0ppm
CA 66.4% of x
R .76

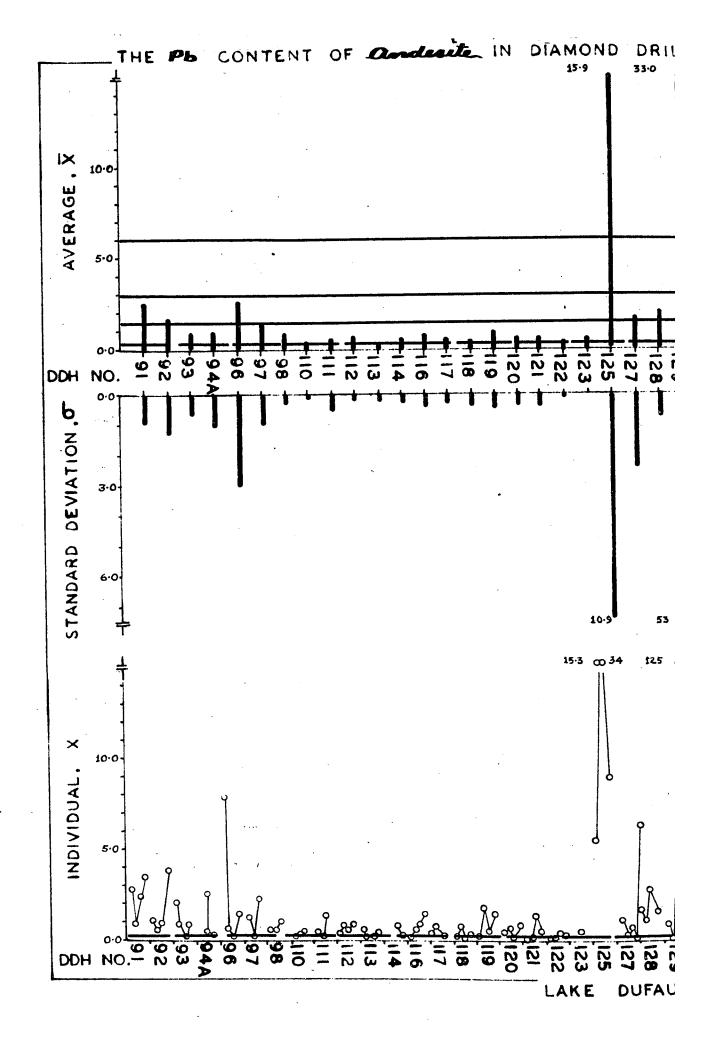
The only unusually high values for Pb in andesite occur in sections from drill holes over or close to the ore body or in the northwest corner of the property.

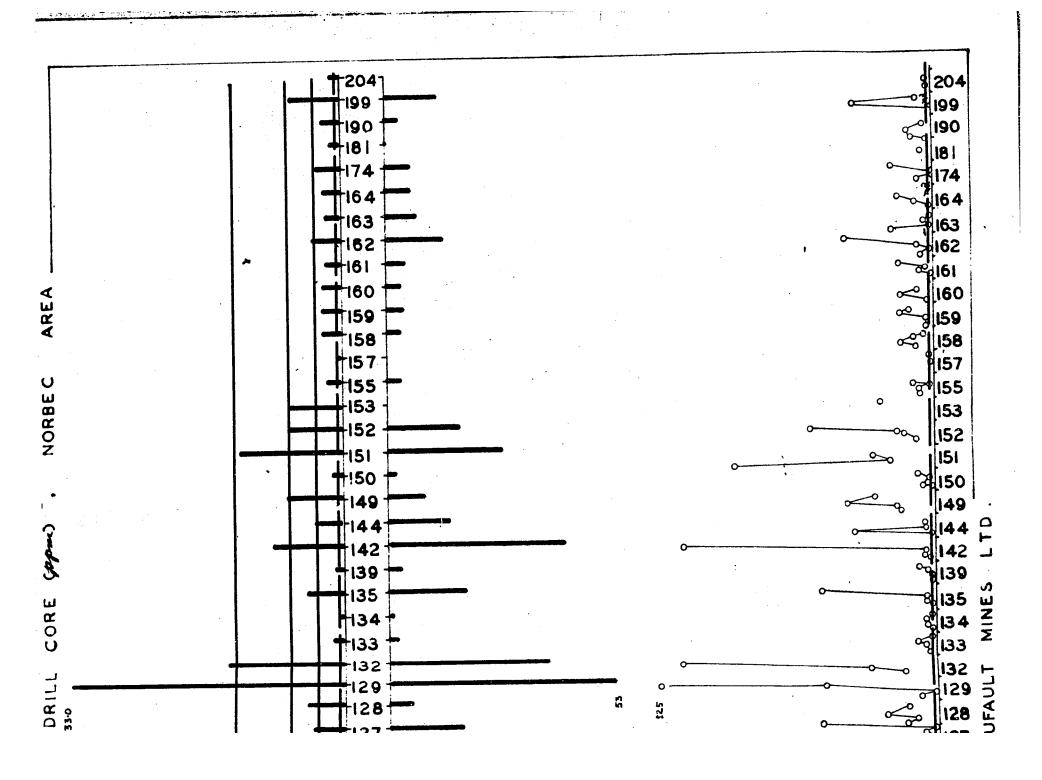
Pb rhyolite





Pb andesite





Tin (Sn) Rhyolite chart 33

1.0ppm
0.5ppm
0.5ppm
Type IV
a 1.0ppm
b 1.5ppm
c 2.5ppm
CA 27.8% of x
R .31

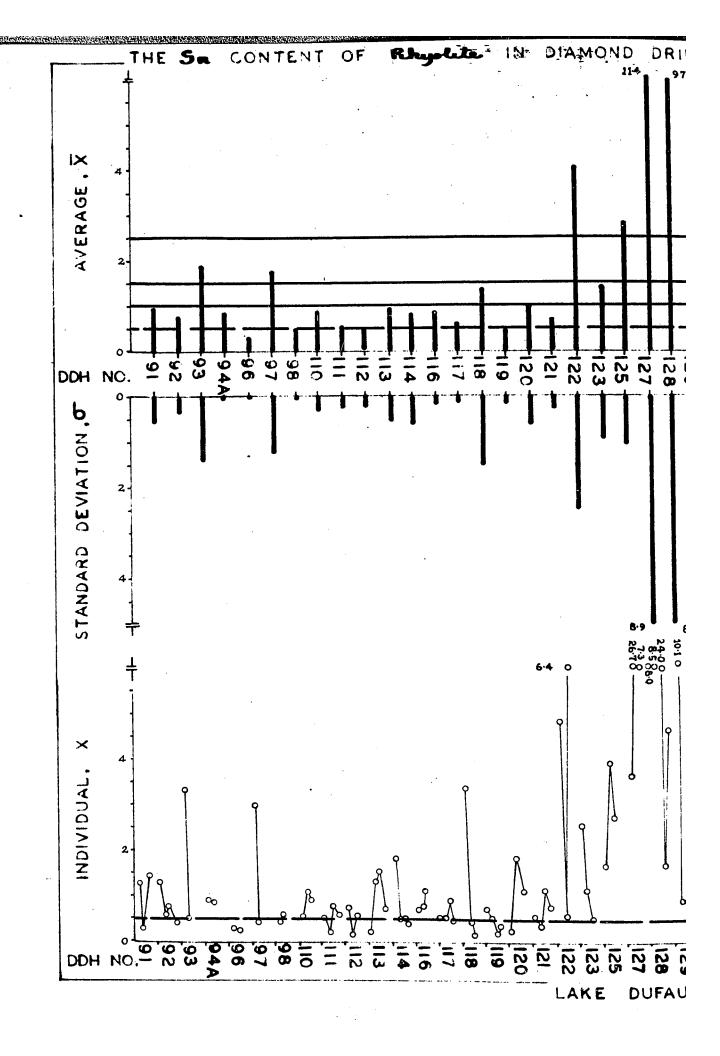
Holes which cut rhyolite with anomalous tin enrichment occur scattered over the whole property, as well as very near the ore body.

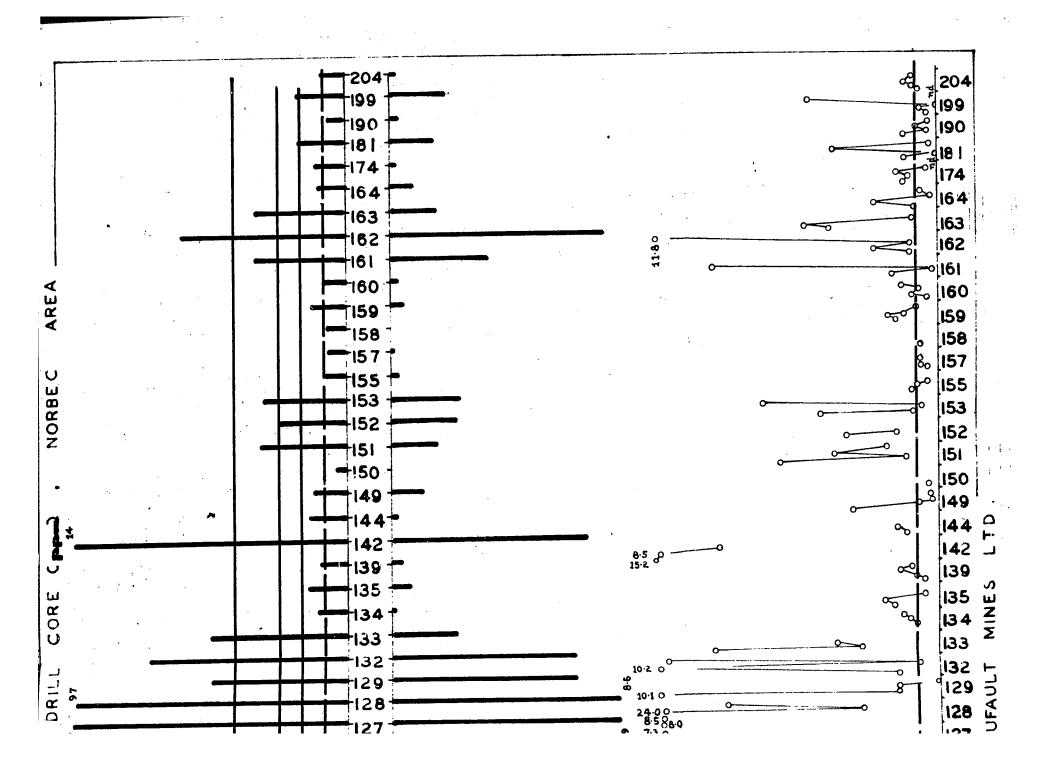
Sn Andesite chart 34

4.4ppm
Mo 2.5ppm
To 2.8ppm
Type IV
Typ

Abnormally high values for Sn in andesite occur scattered throughout the sampled drill holes. Also, the precision of the analyses is low.

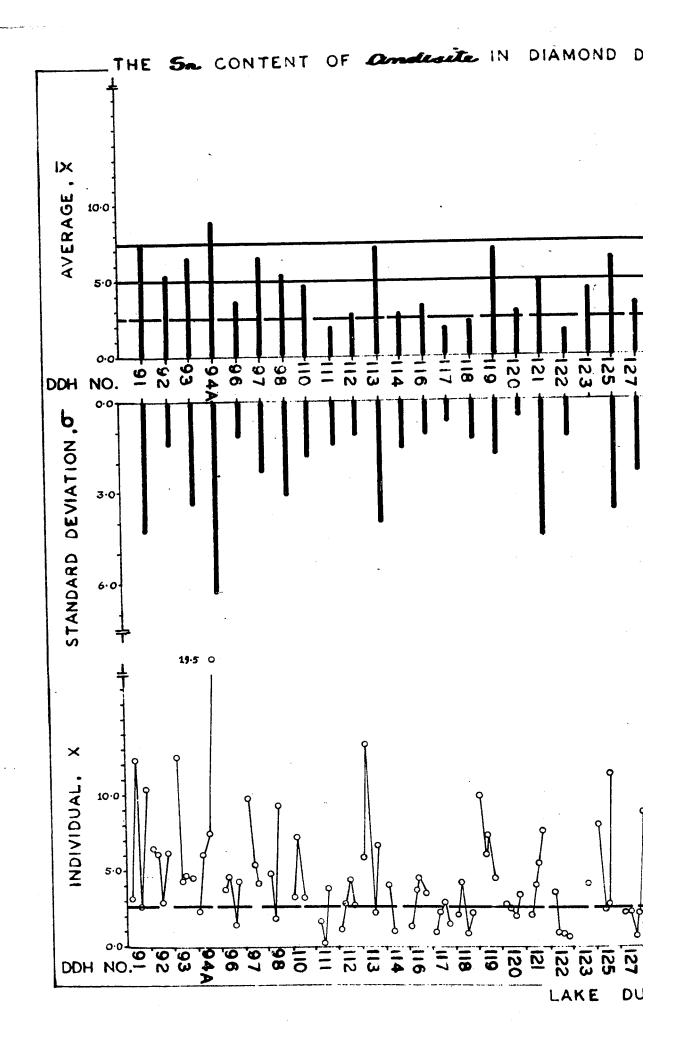
Chart 33 Sn rhyolite

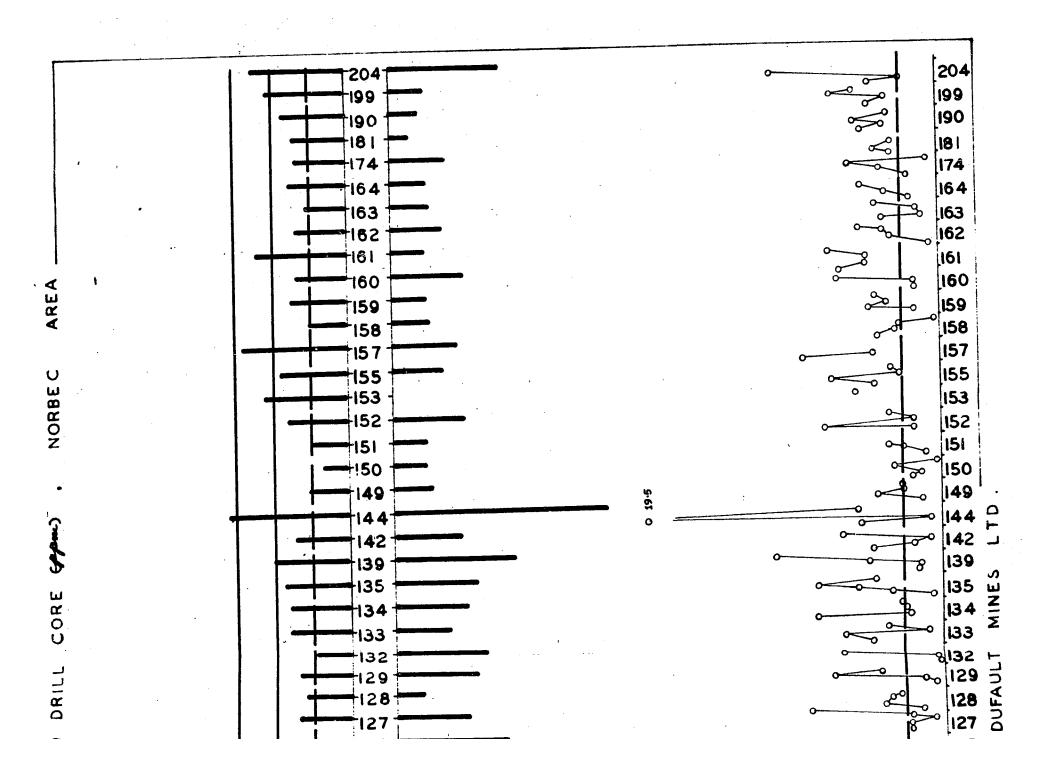




Sn andesite

Sn andesite





Strontium (Sr) Rhyolite chart 35

\$\frac{\frac{\pi}{2}}{2} 418ppm\$
\$\text{Mo} 370ppm\$
\$\frac{\pi}{\pi} 107ppm\$
\$\frac{\pi}{2} 1 07ppm\$
\$\text{ULB} 418 + 107ppm\$
\$\text{LLB} 418 - 107ppm\$
\$\text{CA} 25.3% of \$\frac{\pi}{x}\$
\$\text{R} .98\$

Two mineralized sections of rhyolite beneath the massive ore body in holes 127 and 128 have unusually low contents of Sr.

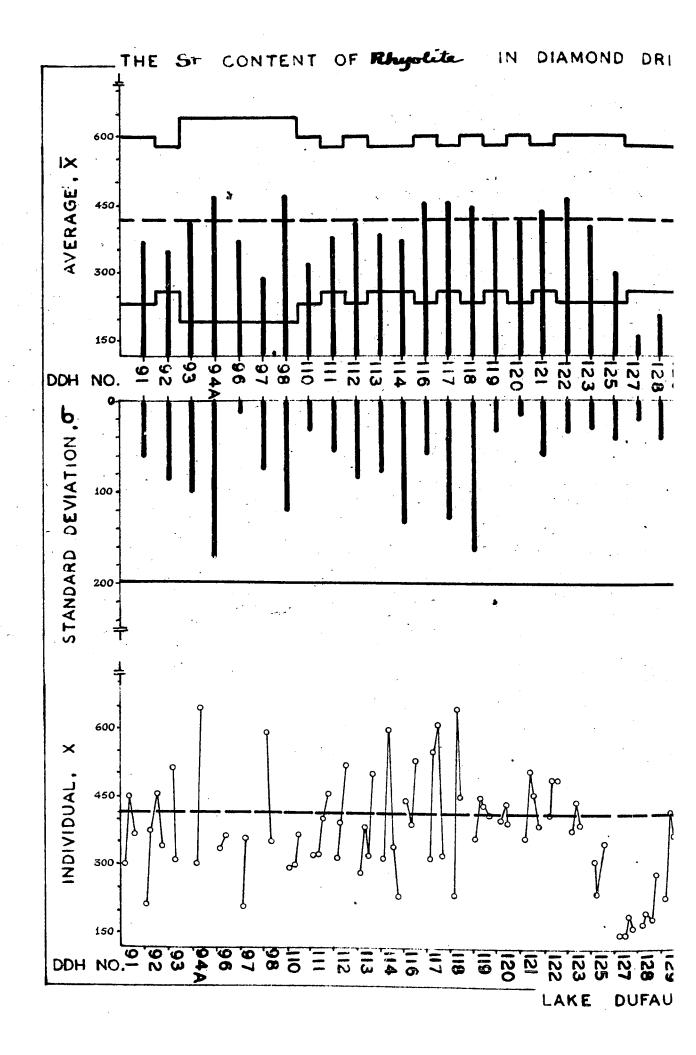
Sr Andesite chart 36

197ppm
Mo 150ppm
G 64ppm
FD Type III ?
CA 30% of X
R •90

The analyses for Sr at this low level are strongly suspect.

No conclusions were drawn from the lateral distribution of Sr in andesite.

Chart 35 Sr rhyolite



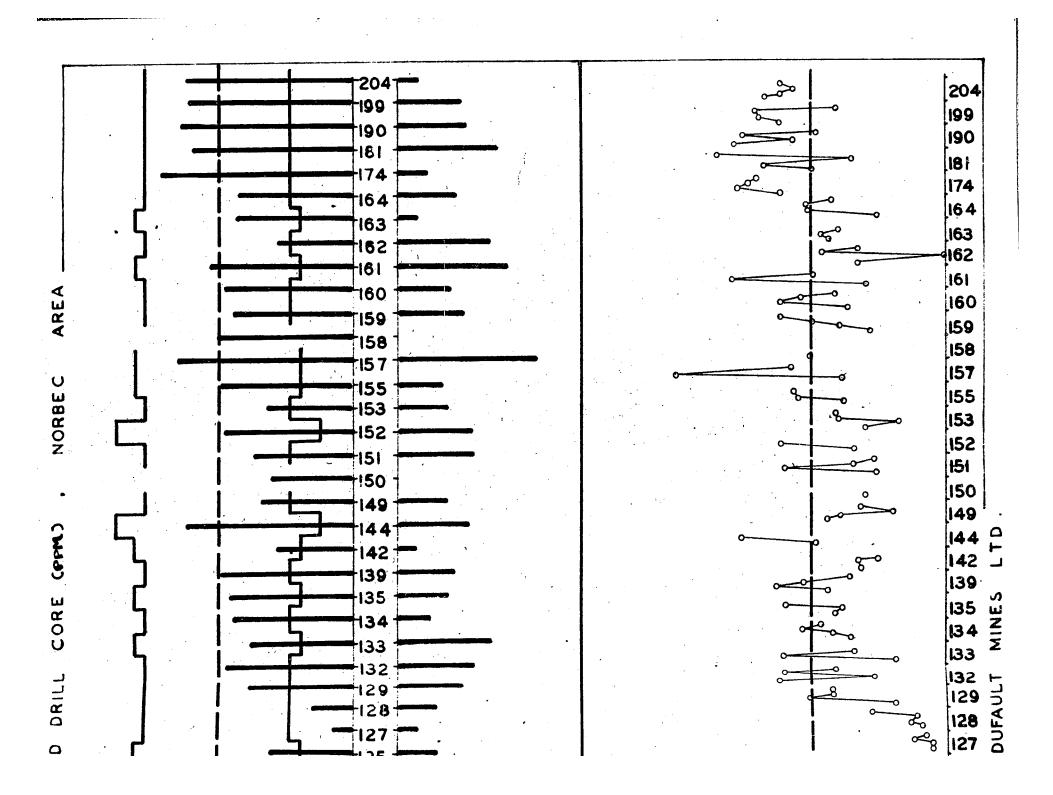
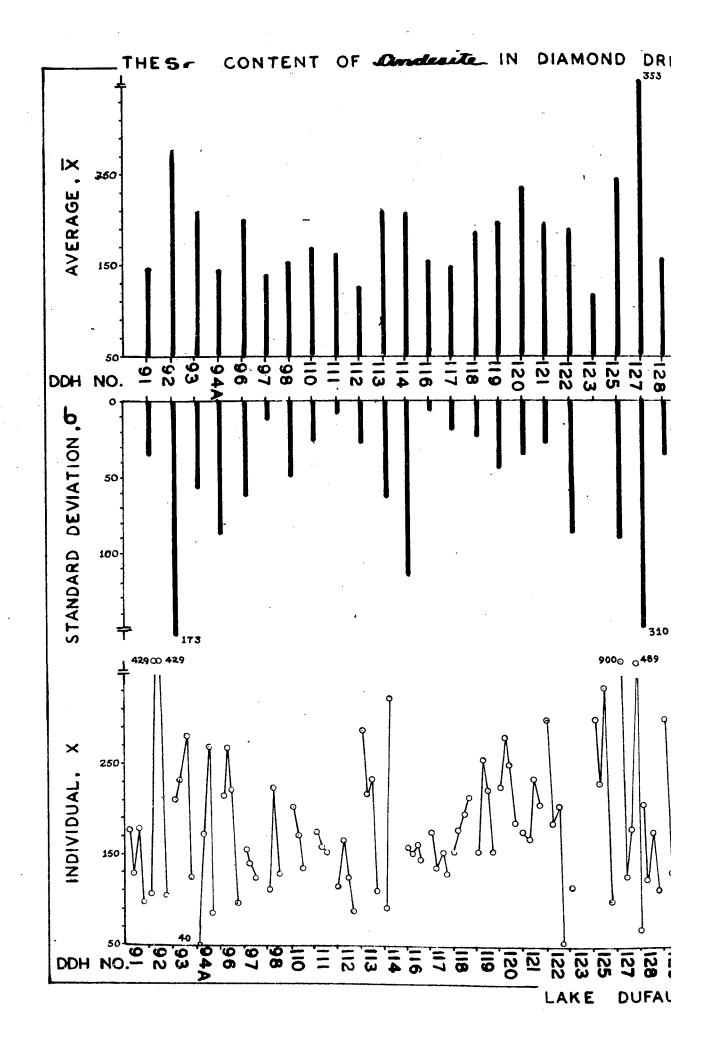
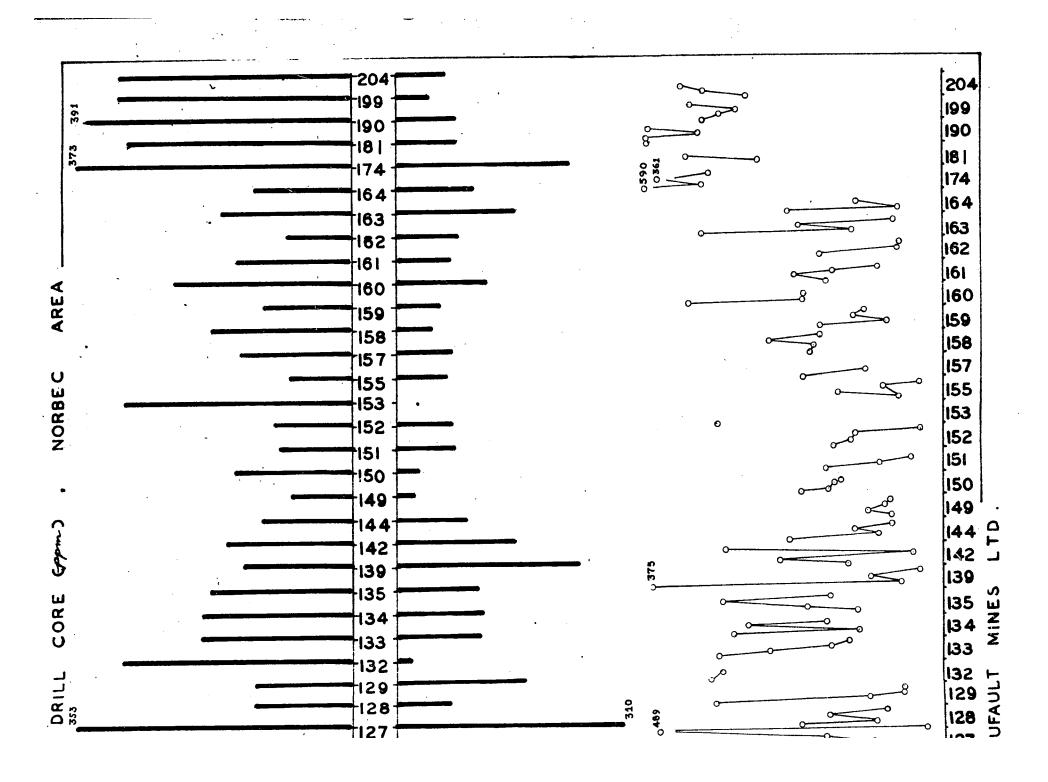


Chart 36 Sr andesite





Yittrium (Y) Rhyolite chart 37

70ppm Mo 50ppm 50ppm 15ppm FD Type I ULB 70 + 23ppm LLB 70 - 23ppm CA 19.3% of x R .90

Holes 96 and 164 cut rhyolite which has an unusually high content of Y. Holes 127 and 181 cut rhyolite which has an exceptionally low Y content.

Y Andesite chart 38

24ppm

Mo 20ppm

G 6.3ppm

FD Type I

ULB 24 + 9.5ppm

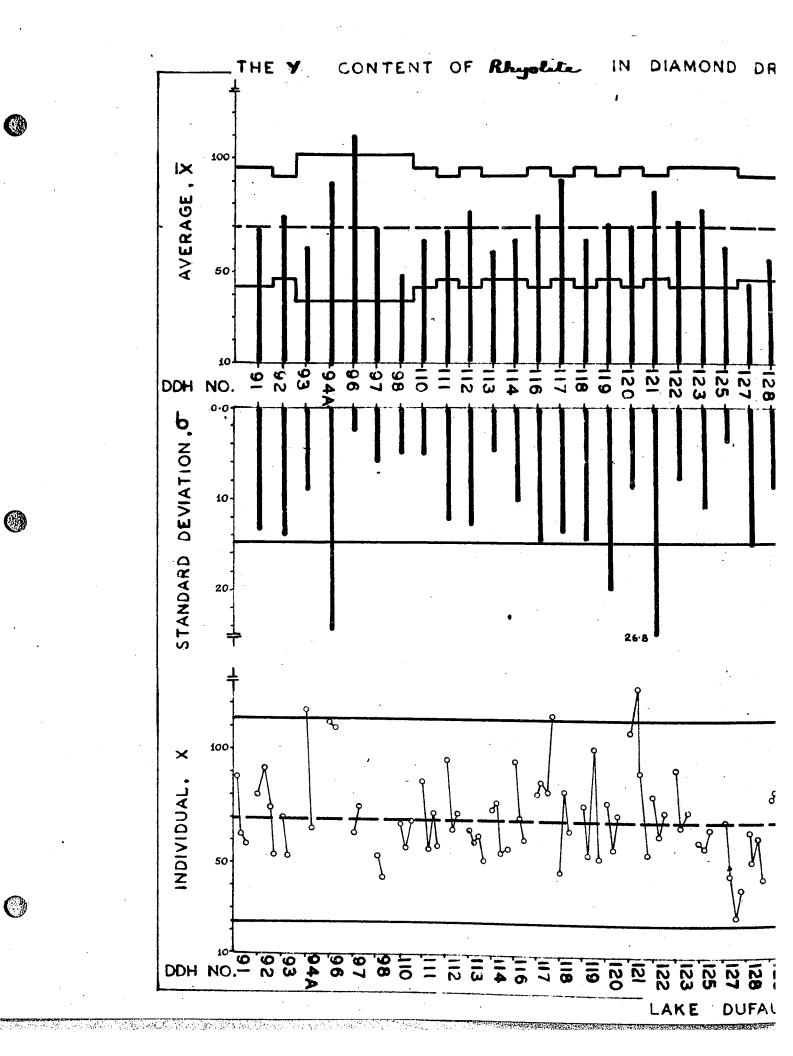
LLB 24 - 9.5ppm

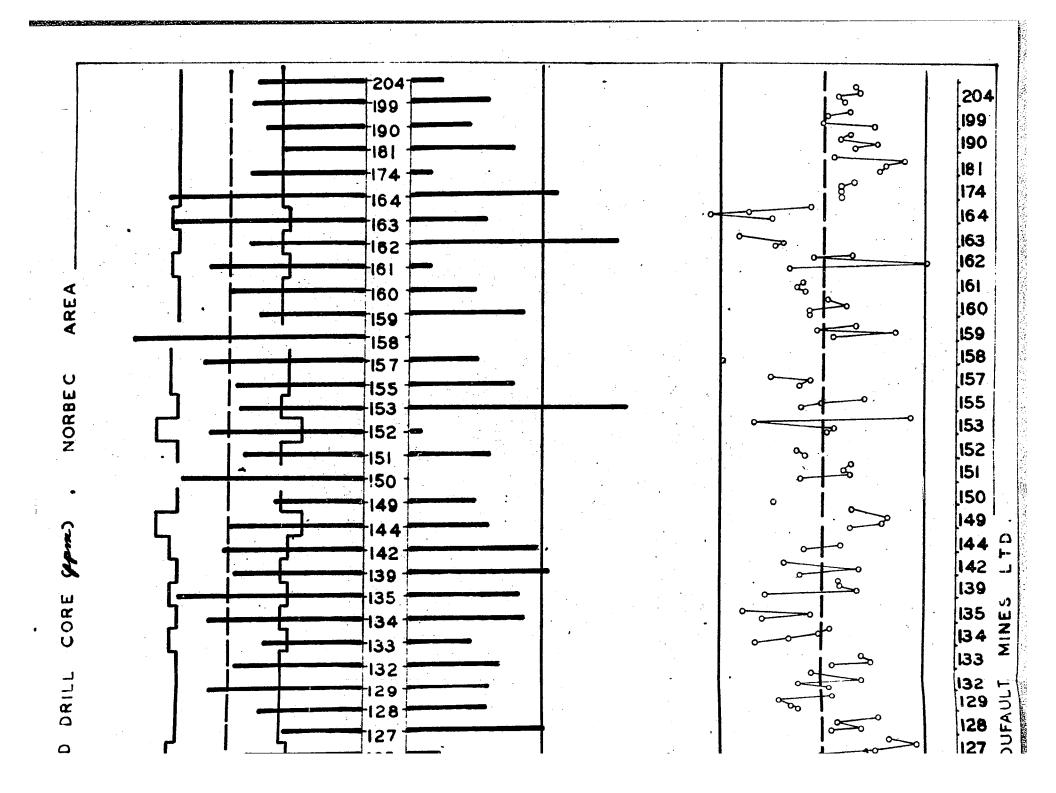
CA 17.1% of X

R .65

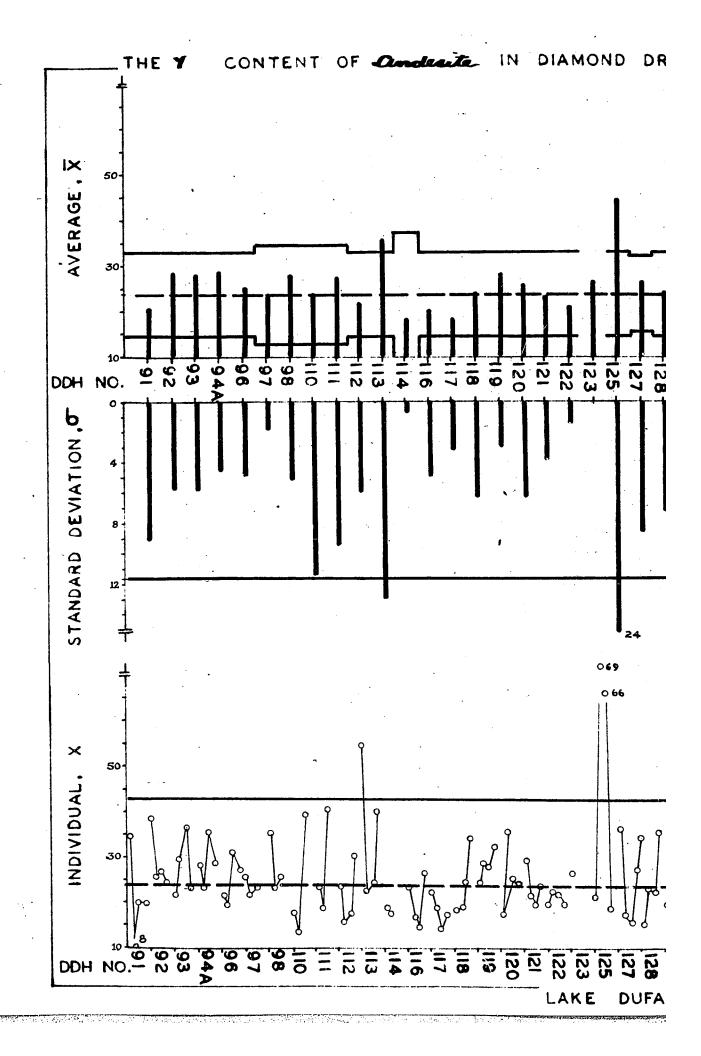
A section logged as andesite in hole 125 is unusually high in Y, probably because at least two of the samples are actually rhyolite. Andesite cut by hole 113 also has an unusually high Y content.

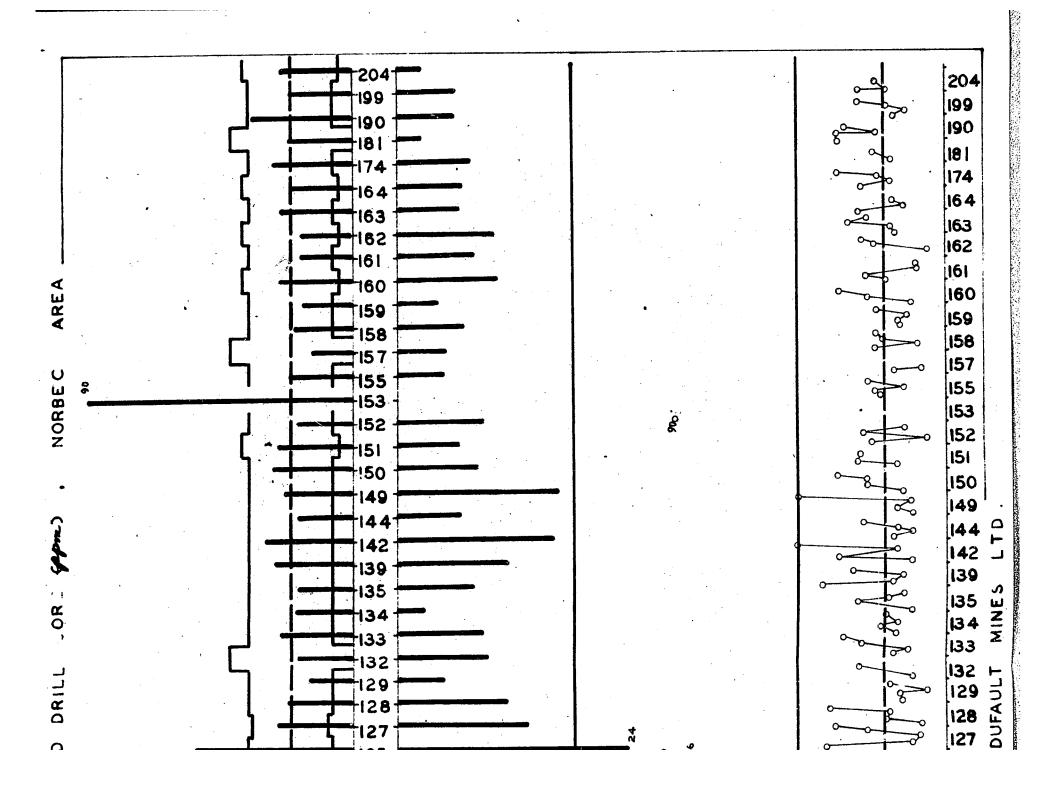
Y rhyolite





Y andesite





Zinc (Zn) Rhyolite chart 39

```
394ppm
180ppm
349ppm
FD Type II
a 350ppm
b 500ppm
c 1000ppm
c 10.3% of x
R .12
```

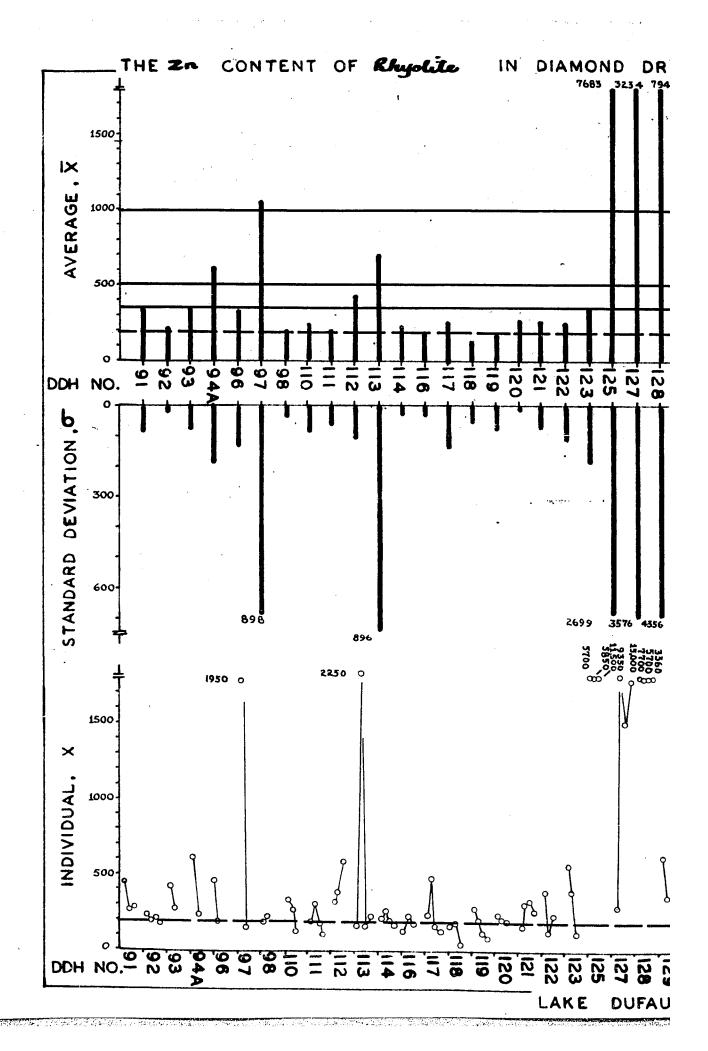
Rhyolite near the ore body contains unusually high amounts of Zn, whether or not it is obviously mineralized. Two holes in the northwest Norbec area (94A, 97) also cut rhyolite with an unusually high Zn content.

Zn Andesite chart 40

240ppm
 160ppm
 103ppm
 Type IV
 250ppm
 500ppm
 19.5% of x
 46

Most sections of andesite which contain abnormally high amounts of Zn come from drill holes located over or near the ore body. Other similar sections are located in the northwest corner of the property (98), north of the ore body (110) and in the Mid-Norbec area (150).

Zn rhyolite



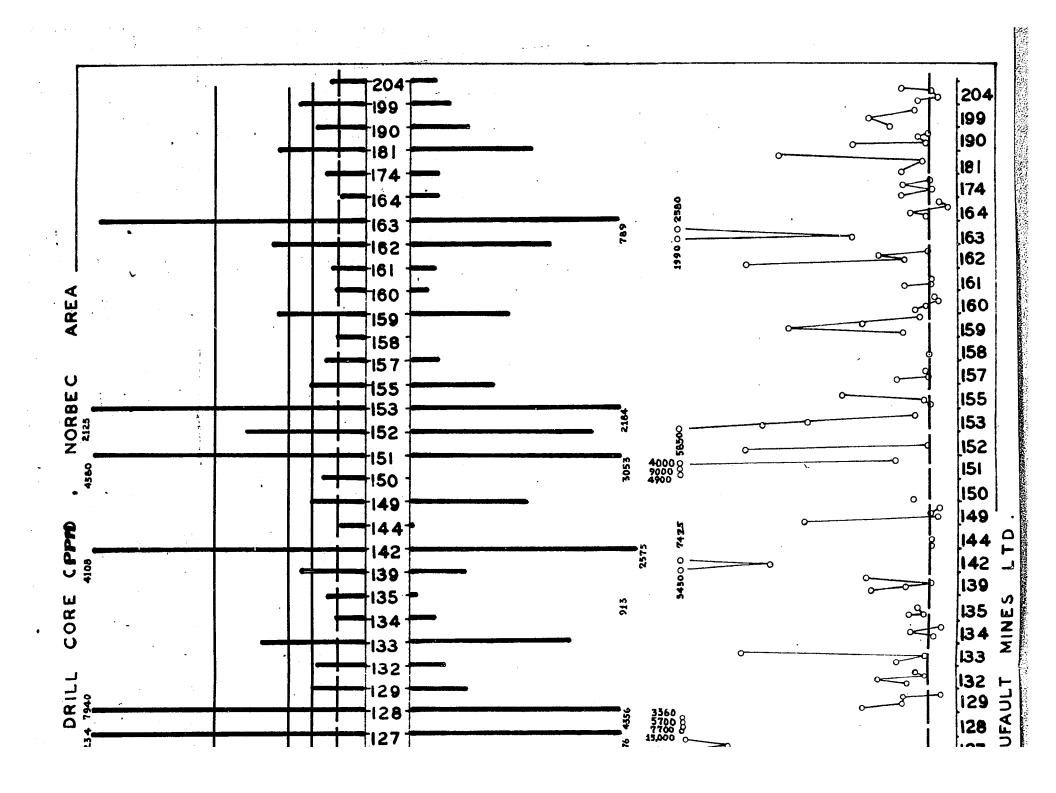
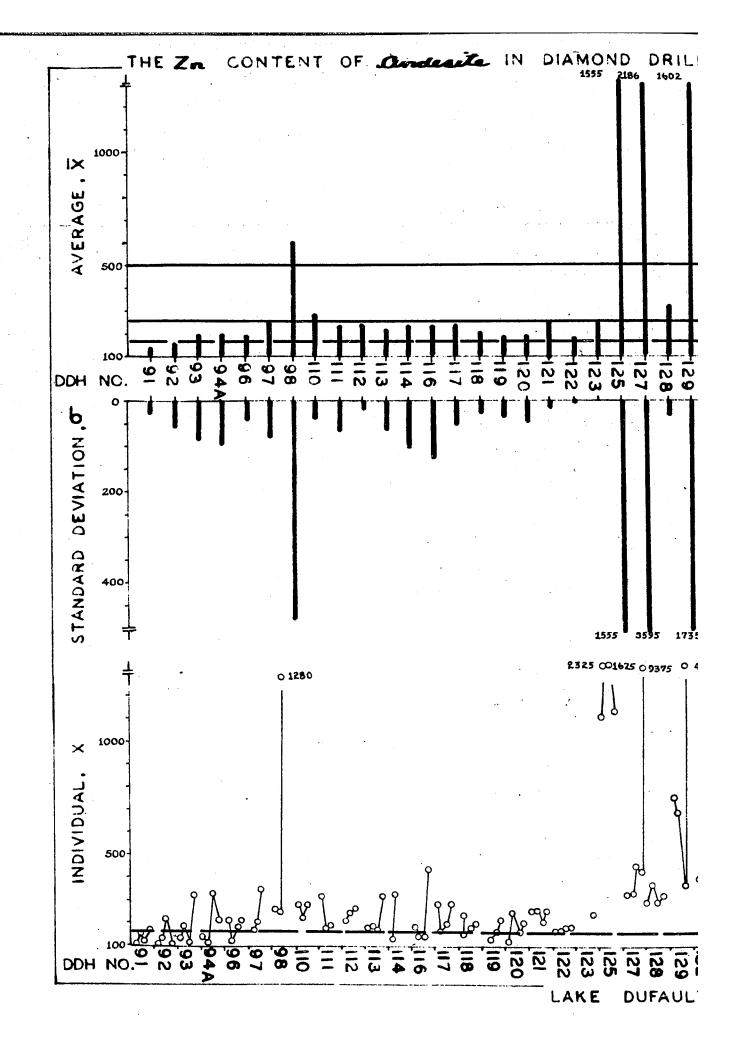
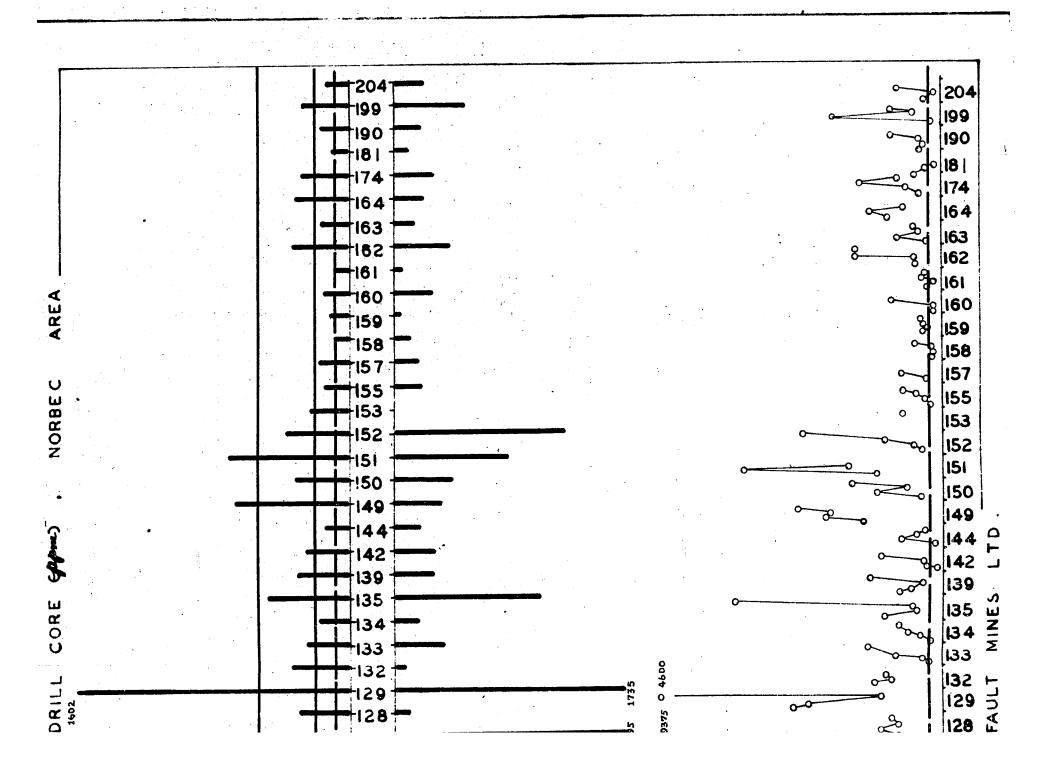


Chart 40

Zn andesite





Zirconium (Zr) Rhyolite chart 41

X 405ppm
Mo 260ppm
To 149ppm
Type II
A 450ppm
B 600ppm
CA 27.9% of X
R .76

Holes 121 and 142 cut rhyolite containing unusually high amounts of Zr. Rhyolite with a Zr content slightly higher than usual was cut by several scattered drill holes.

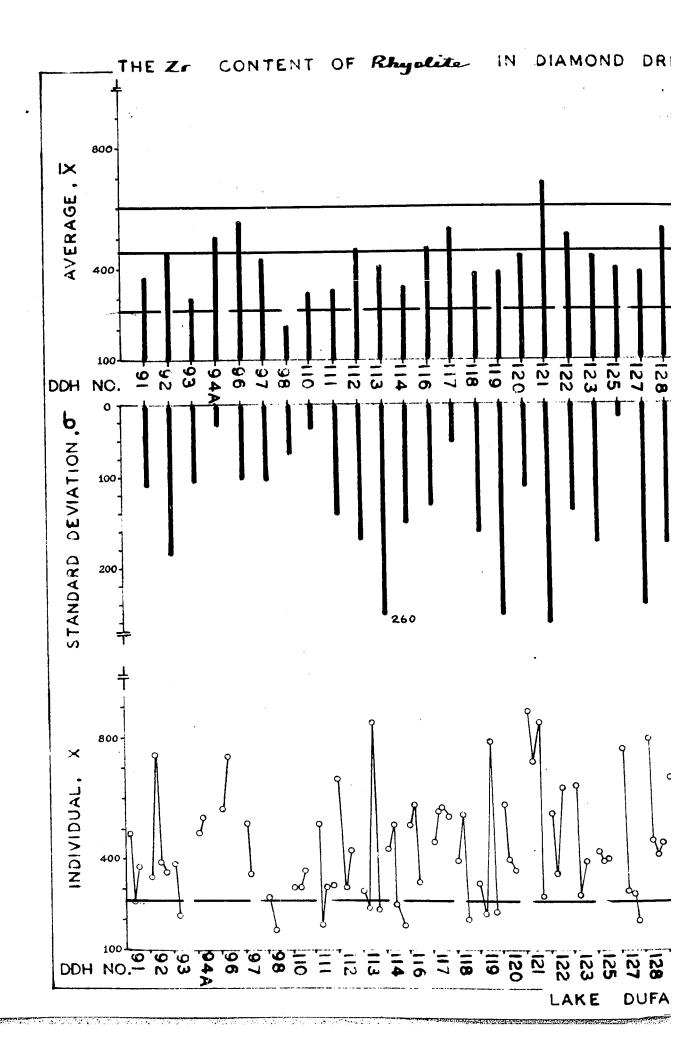
Zr Andesite chart 42

180ppm Mo 180ppm 5 43ppm FD Type I ULB 180 + 65ppm LLB 180 - 65ppm C_A 17.5% of \bar{x} R 1.0

A few scattered drill holes cut andesite with an unusually high Zr content. The two anomalous samples from hole 125 are probably rhyolite.

Chart 41

Zr rhyolite



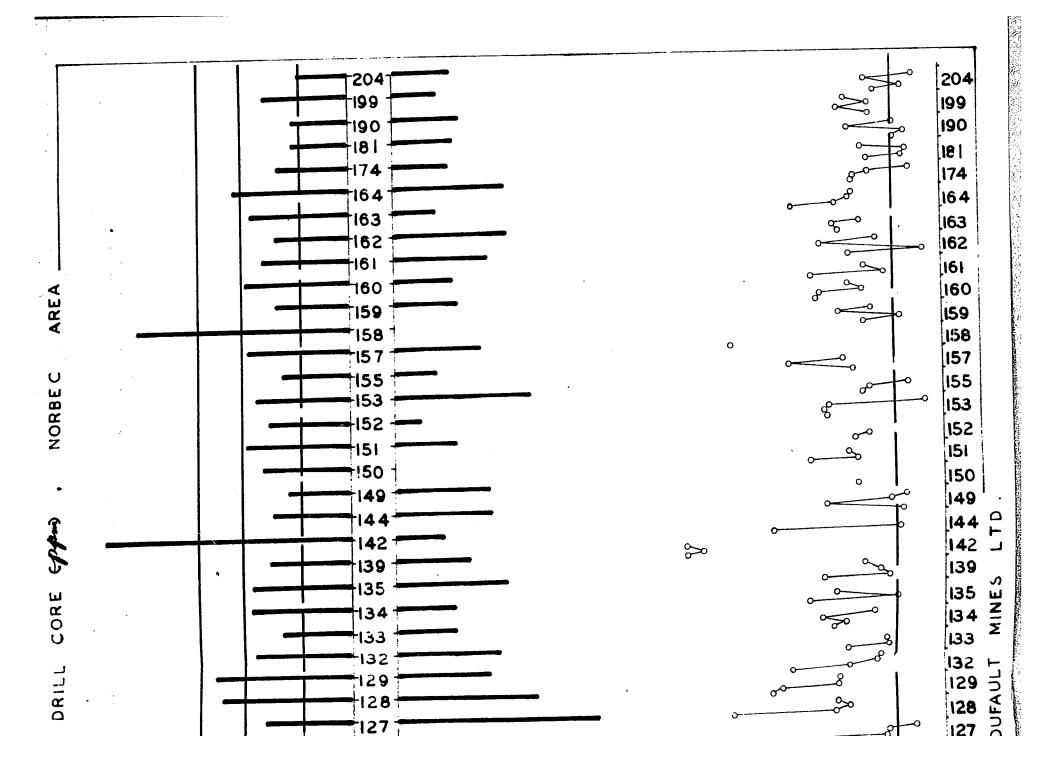
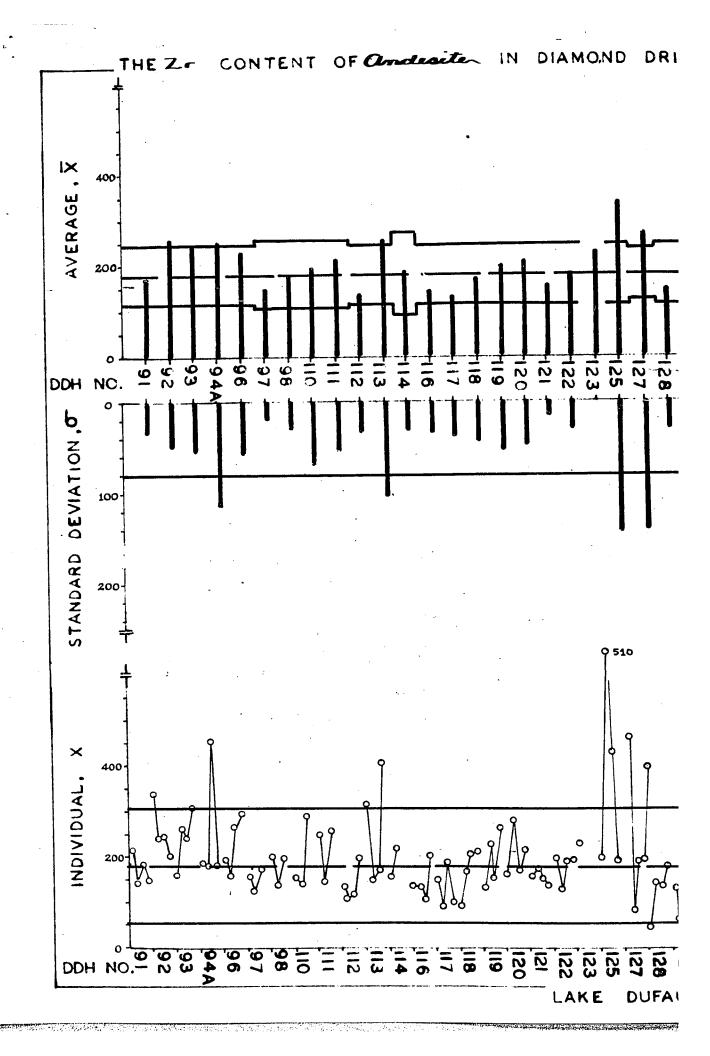
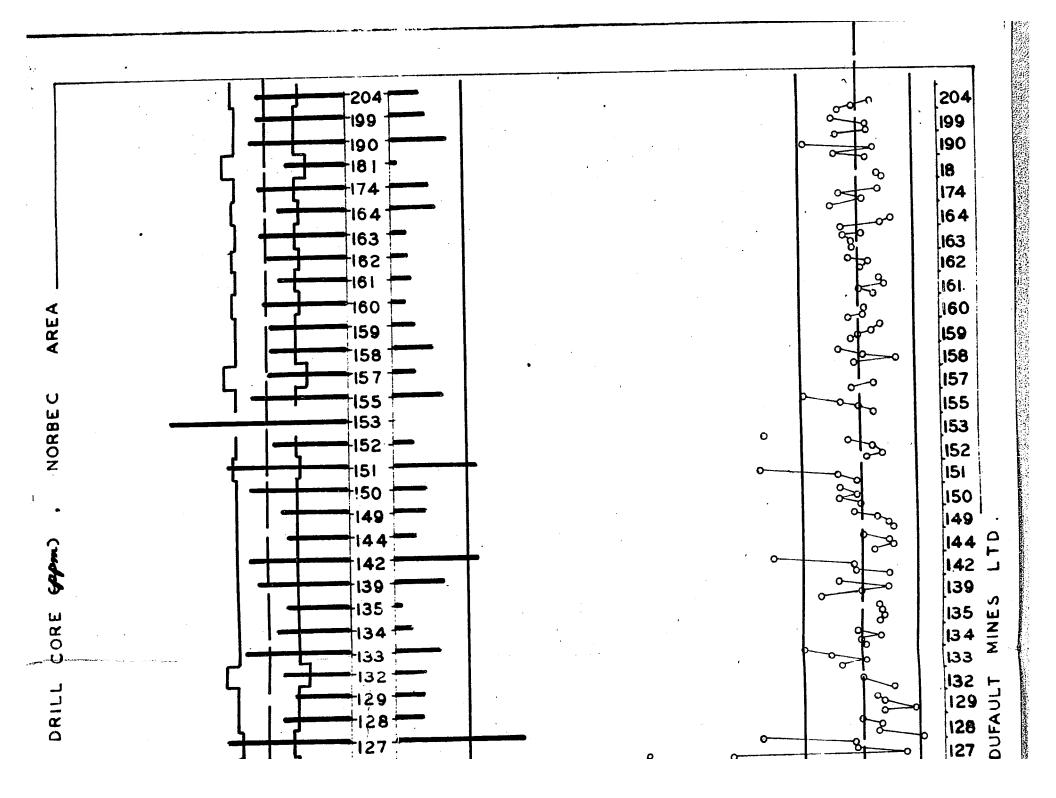


Chart 42

Zr andesite





APPENDIX C

Analytical Results for Bedded, Cherty Tuff

Sio ₂	,		Tio2		•
*159 ¹ *135 *139 *163 119 118 *164 121	51.4% 53.3 53.6 54.0 56.4 57.5 59.0 60.7		*164 *139 *174 119 150 *181 158 157	.40% .50 .52 .54 .59 .63 .65	
*174 *144 155 157 116 *181 150 158	61.0 61.3 61.8 62.6 62.7 64.3 66.0 66.8		121 155 118 *144 *190 *159 116 *163 *135	.68 .71 .75 .78 .79 .88 .96 1.02	
	: 15.4%			0.68% e: 0.68% /5	
A1203	3		Fe ₂ 0	5	
*164 *174 155 *163 118 158 150 *135	11.8% 13.3 13.9 14.1 14.1 14.5 15.3		157 *135 *181 *159 *190 *163 121 155	8.8% 8.98 10.6 10.7 11.2 11.7 12.0 12.7	
*139 119 121 *181 *159 157 *144 116	15.85 16.8 16.9 18.2 18.4 19.0 20.0 20.2	Mdn: 15.8% Range: 814% r: 4/4	*164 150 *144 116 119 *174 118 *139	13.8 14.4 14.5 17.0 17.5 17.7 19.0 24.6	Mdn: 13.2% Range: 15.6% r: 5/4

*samples near the orebody l diamond drill hole

Table C1

MnO	•		CaO		
121 ¹ 157 *190 *181 *174 *164 *139 116	.09% .10 .10 .11 .12 .13 .14		*139 *190 119 *144 116 *174 *181	1.54% 2.36 3.88 4.43 4.8 4.83 4.84 5.14	
和 44 155 和 35 和 63 158 118 150 和 59 119	.15 .16 .16 .17 .18 .18 .20 .23 7.47	Mdm: 0.15% Range: 0.14% (except 7.47) r: 5/4	157 150 158 155 *163 *135 *164 118 *159	5.57 5.6 6.4 6.45 7.0 7.35 8.0 8.25 9.5	Mdn: 5.5% Range: 7.96% r: 5/4
K ₂ 0 *139 118 150 *159 *135 *164 116 158	.84% .94 1.06 1.07 1.18 1.52 1.62		Mg0 *135 *181 *164 119 150 *174 155 157 *190	.19% 1.21 1.46 1.75 1.77 2.3 2.41 2.66 2.79	
157 155 *144 119 *163 *174 *181 121 *190	1.7 1.73 1.86 1.93 1.95 2.12 2.2 2.69 2.8	Mdn: 1.7% Range: 1.96% r: 4/5	158 *139 121 116 *159 118 *144 *163	2.86 2.97 3.17 3.2 3.45 3.46 3.49 3.82	Mdn: 2.8% Range: 2.62 r: 5/4

^{*} samples near the orebody l diamond drill hole

Na20			Ag		
*174 ¹	.93%		*190	.34ppm	
*181	.95		*135	.64	
#164	1.00		*159	.69	
119	1.09		*144	.83	
*190	1.24	•	119	.98	
155	1.33		157	1.02	
*163	1.42		116	1.17	
158	1.44		118	1.17	
136	1 56	•	*101		
116	1.56		*181	1.25	
118 150	1.85		*139 121	1.33	•
	1.94			1.53	
*139	2.0		*163	1.54	
121	2.14		150	1.57	
*144	2.18	Man 1 Ed	158	2.16	Mån. 1 OFmm
157	3.06	Mdn: 1.5%	155	2.21	Mdn: 1.25ppm
*159	3.32	Range: 2.57%	*164	4.23	Range: 4.36ppm
*135	3 . 5	r: 5/4	*174	4.70	r: 4/5
Bi			Cd		
*135	1.3		118	.4ppm	
157	1.4		*135	. 4	
121	1.47		15 7	.7	
*181	1.5		155	.7	•
*15 9	1.5		*144	•9	
118	1.7		158	1.1	
150	0.4		×3.07	2 7	
150	2.4		*163	1.3	
#144 #174	2.5		*139	1.4	
*174 *170	3.5		*174	1.6	Waddan I O
#139	3.6	Maddan . O day	*190	1.6	Median: 1.2ppm
158	4.9	Mediam: 2.4ppm	*181	2.0	Range: 2.0ppm
*1 90	6.0	Range: 6.1ppm	*164	2.4	r: 2/6
155	7.4	r: 3/4			

^{*} samples near the orebody 1 diamond drill hole

G⊕			N1		
138 ¹			*159	27.3ppm	
	.7		*135	38.0	
118	1.0		*164	44.0	•
*1 44	1.2 1.2		155	47.5	
157 *181	1.2		*190	49.0	
*TOT	1.6		119	51.0	
			158	53.9	
121	1.3		*144	57.3	
119	1.4				
#164	1.6	Median: 1.25ppm		·	
150	2.3	Range: 1.7ppm	*163	67.3	
*135	2.4	r: 2/2	*139	70.0	
		•	150	72.0	
			157	76.0	
			116	77.0	
•			118	86.8	Median: 60ppm
			*174	88.0	Range: 77.9ppm
			121	105.17	r: 5/3
Pb			Sn		
			3.68	•	
118	.Sppm		157	.1	
*144	•7	•	*159 *135	.2 1.5	
*164	.9		*135 *181	2.2	
#190	1.7	·	150	2.3	
150	1.8		121	2.37	
116	2.8 2.9		*144	2.8	
119 121	3.47		119	3.5	
121	U+ 11				
*159	4.6		118	4.0	
*181	5.0		*164	4.2	
157	5.3		*163	4.7	
*139	6.7		158	5.3	
*135	7.0		*139	5 .5	
158	9.3		116	8.2	
*163	14.4	Median: 4.6ppm	*190	8.8	Median: 4.Oppm
155	25.3	Renge: 24.8ppm	*174	12.4	Range: 21.1ppm
*174	48.0	r: 3/6	155	21.2	r: 4/5

^{*} samples near the orebody 1 diamond drill hole

Sr		¥		
1161	228ppm	118	22.0ppm	
*144	254	*163	22.0ppm 24.7	
*1 39	283	*135	26.0	
119	284	150	27.5	
#164	294	*144	27.6	
*163	305	158	29.7	
157	317	*164	30.0	
158	340	121	31.27	
121	345.67	*159	35.4	
150	347	116	36.0	
#159	357.5	158	36.6	
155	371	*139	37.0	
*190	372	119	37.0	
118	398.5 Median: 340ppm	157	38.0	Median: 32ppm
#135	461 Range: 285ppm	*174	45.0	Range: 23ppm
#174	513 r: 4/4	*190	45.0	r: 4/4
Zn		Zr		
	•	150	192ppm	
*1 64	245.Oppm	*164	194	
* 159	270 ₀ 0	*163	217	
121	270.67	121	222.67	
*1 90	280.0	155	225	
150	438.0	*144	231.5	
119	513.0	157	250	
116	1225.0	158	255.5	
118	1635.0	-		
*174	1015.0	119	270	
*174 *135	1815.0	*174	273.5	
155	1915.0	* 159	288	
*181	1945.0	139	301.5	
157	2040.0 2140.0	*135 *300	334 350	V-31 000
158	2280.0	*190	358	Median: 260ppm
¥144	2605.0 Median: 1800ppm	118	387.5	Range: 223ppm
*139	3035.0 Range: 3925ppm	116	415	r: 3/4
*163	4170.0 r: 3/6			
	オサエハサハ アニ ハヘク			

^{*} samples near the orebody 1 diamond drill hole

APPENDIX C

Summary of Anomalous Values of the Elements in Rhyolite and Andesite

Table Dl Cummary of Anomalous Values for Major Elements in Rhyolite

DDH	s10 ₂	Tio ₂	A1203	CaO	K20	Na ₂ O
91 92 93 94 96 97 91 112 113 113 114 115 115 115 115 115 115 115 115 115	111101100000000000000000000000000000000	00000000000000000000000000000000000000	000000000000000000000000000000000000000	0 % % 0 % 0 % 0 % 0 % 0 0 0 0 0 0 0 0 0	00000000000000000000000000000000000000	00-10000-10-100000-1-1000-1-100-1-0000-0-10-0-0-10-0-0-10-0-0-0-10-0-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-10-0-0-0-10-0-0-0-10-0-0-0-10-0-0-0-10-

^{1.} l = a-b; 2 = b for frequency distributions of Types II to V
2 > ULB, Type I; -2 < LLB, Type 1
for Na₂O -1 = 3.3 - 4.4%, 1 = 1.7 - 0.5%, 2 < .5%
O = background values</pre>

Summary of Anomalous Values for Trace Elements in Rhyolite¹

DDH	Ga.	Ni	Sr	Y	Zr
91 92 93 94 96 97 98 111 113 113 114 115 115 115 115 115 115 115	000000000000000000000000000000000000000	2122202111112201122111202221210222210222202121	000000000000000000000000000000000000000	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	00000000000000000000000000000000000000

^{1. 1 =} a-b; & = b for frequency distributions of Types II to V
2 > ULB, TypeI; -2 < LLB, Type I
0 = background values</pre>

Table D3 Summary of Anomalous Values for "Ore-Major" Elements in Rhyolite

DDH	Fe ₂ 0 ₃	Mno	Mgo	Sum
91 92 93 94A 96 97 91 112 113 114 117 118 119 121 123 125 127 128 129 121 121 123 133 144 144 150 161 163 164 163 164 174 119 119 119 119 119 119 119 119 119 11	000001000000000000000000000000000000000	02000000000000000000000000000000000000	201001200000000001102122120020101020000010002020	22100220002000000112456652015010203000004000020

 ^{1 =} a-b; 2 = b for frequency distributions of Types II to V
 2 > ULB, Type 1; -2 < LLB, Type 1
 0 = background values

Summary of Anomalous Values for "Ore-Trace" Elements in Rhyolite

^{1.} l = a-b; 2 = b for frequency distributions of Types II to V
2 > ULB, Type 1; -2 < LLB, Type 1
0 = background values</pre>

Table D5 Summary of Anomalous Values for Major Elements in Andesite1

DDH	sio ₂	Tio2	Fe ₂ 0 ₃	MnO	CaO	K ₂ o	MgO	Na ^S O
9999678111111111111111111111111111111111	00000000000000000000000000000000000000	พพพพ 000000000000000000000000000000000	0 NOOOOOOOOOOOOOOOOOOOOOOOOOOOOOOOOOOOO	00000000000000000000000000000000000000	00000000000000000000000000000000000000	1011110012021100100120012000221202011011	000000000000000000000000000000000000000	000 0 00000000000000000000000000000000

^{1.} l = a-b; 2 = b for frequency distributions of Types II to V
2 > ULB, Type I; -2 < LLB, Type I
0 = background values</pre>

Summary of Anomalous Values for Trace Elements in Andesite1

DDH 91 92 94 96 97 91112 113 114 115 115 115 115 115 115 115 115 115	00000100100000010100000000000000000000	C00001001000100102002000010100000000000	N000020020002112020120211011021101000002111020	&ooooooooooooooooooooooooooooooo	\$11120110001000010000100000000000000000	x0000000000000000000000000000000000000	x 0 x 0 x 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
158 159 160 161 162 163 164 174 181 190 199 204	80000000000000000000000000000000000000	1 0 0 1 0 0 0 2 0 0	2 1 1 0 2 0 2 0 0 0	0 0 0 0 0 0 0 0 2 1	0000000000	000000000	00000000000

^{1.} l = a-b; 2 = b for frequency distributions of Types II to V
2 > ULB, Type I; -2 < LLB, Type I
0 = background values</pre>

Table D7 Summary of Anomalous Values for "Ore-Trace" Elements in Andesite

22	Ag	, . (1)-	nh.	72	Q ₁₂ m
DDH 91 92 94A 96 98 91 112 113 114 115 115 115 115 115 115 115 115 115	000000000000000000000000000000000000000	000101121111000101100220202020010022020202020202020202020202020202020202	P10020000000000000000000000000000000000	x00000021000000000000000000000000000000	Sum10121142111000101100784644042606682700100231351373

^{1.} l = a = b; 2 = b for frequency distributions of Types II to V
2 > ULB, Type 1; -2 < LLB, Type 1
0 = background values</pre>

APPENDIX E

Test for Systematic Vertical Variation of Elements in Rhyolite and Andesite

Table E-1

NUMBER OF DIAMOND DRILL HOLES SHOWING SYSTEMATIC VERTICAL VARIATION IN WAITE RHYOLITE

(HOLES AWAY FROM ORE)

Number of	Element "Added" to top		Element "Removed" from top		No Change
Number of Samples	→ ≥3	=2	53	=2	
SiO ₂ TiO ₂ Al ₂ O ₃	14 9 12	3 4 2	16 10 18	1 1 4	3 13 1
Fe ₂ 0 ₃ MnO CaO	16 13 14	6 1 6	13 14 12	0 3 1	1 5 4
K ₂ O M ₉ O Na ₂ O	13 16 14	3 7 3	16 13 11	3 0 5	1 1 3
Ag Co Cr	14 22 12	6 4 7	14 · 4 12	1 1 0	2 3 0
Cu Ga Pb	21 7 17	5 3 7	8 19 7	1 4 1	4 3
Ni Sn Sr	24 11 7	6 5 2	6 20 22	0 1 5	5 1 0 1
Y Zn Zr	19 24 20	5 . 4 4	11 7 9	2 1 2	0 1 2

Table E-2

NUMBER OF DIAMOND DRILL HOLES SHOWING SYSTEMATIC VERTICAL VARIATION IN WAITE RHYOLITE

(HOLES NEAR TO AND UNDER MASSIVE DRE)

	Element "Added" to top		Element "Removed" from top		No change	
Number of Samples	≥3	=2	≥3	=2		
SiO ₂ TiO ₂ Al ₂ O ₃ Fe ₂ O ₃ MnO CaO K ₂ O MgO Na ₂ O Ag Co Cr Cu Ga Pb Ni Sn Y Zn	563546871954737884760		6 4 5 4 6 4 3 4 6 2 5 3 2 6 3 3 3 7 4 5 0		0 1 3 2 1 1 0 0 4 0 1 3 2 2 1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	

Diamond drill holes 135, 151, 163, 132, 133, 142, 149, 127, 128, 129, 125.

Table E-3

NUMBER OF DIAMOND DRILL HOLES SHOWING SYSTEMATIC VERTICAL VARIATION IN AMULET ANDESITE

(HOLES AWAY FROM ORE)

	Element "Added" to base		Element "Removed" from base		No change	
Number of Samples —	> ≥3	=2	≥3	=2		
SiO ₂ TiO ₂ Al ₂ O ₃ Fe ₂ O ₃ MnO CaO K ₂ O MgO Na ₂ O Cr Cu Ga Ni Pb Sr Y Zn	22 10 20 23 24 12 24 18 13 21 21 20 13 10 21 17 8 17 26	22021121001211010212	12 11 15 9 7 22 9 16 21 15 24 14 18 20 7 13 25 14 5	0320010111022212120	2 11 0 3 5 1 2 1 2 2 2 2 2 2 2 2 3 6 6 5 1 3 4	
Zr	19	2	12	1	3	

Table E-4

NUMBER OF DIAMOND DRILL HOLES SHOWING SYSTEMATIC VERTICAL VARIATION IN AMULET ANDESITE

(HOLES NEAR TO AND OVER MASSIVE ORE)

Number of	Element "Added" to base		Element "Removed" from base		No change	
Samples	. ≥3	=2	≥3	=2		
SiO ₂ TiO ₂ Al ₂ O ₃ Fe ₂ O ₃ MnO CaO K ₂ O MgO Na ₂ O Ag Co Cr Cu Ga Ni Pb Sn Sr Y Zn Zr	444584954734722901776	0 0 0 0 0 0 0 0 0 0 1 1 1 1 0 0 1 0	7 7 6 6 3 6 1 6 6 4 7 6 3 7 8 2 1 8 3 2 1	000000000000000000000000000000000000000		

Diamond drill holes 135, 151, 163, 132, 133, 142, 149, 127, 128, 129, 125.

APPENDIX F

Use of the I.B.M. 7040 Computer in Handling Data

Values for emulsion calibration curves and relative intensity ratios were obtained using an I.B.M. 7040 computer at the McGill University Computer Center. Programs for these two procedures are described by Lickus (1965).

An additional program was designed to manipulate the analytical results obtained in this report. While it is unlikely that the specific program described here will ever be useful, certain parts of it could be adapted to other projects. Of special interest should be the form in which data were stored and the "format statement" by which the data were removed from storage.

Figure F1 shows the Fortran program which was used in this project. Cards were punched from special data sheets which were filled out according to the example shown in figure F2. 1 Codes and conventions dealing with the use of these data sheets are filed in the Department of Geological Sciences, McGill University.

The Fortran program instructs the computer:

1. To read into storage, the project number (PN), lab sample number (ALN), and analytical results (K) for twenty two elements and J samples per data group 2 (ISN 2, 4, 5, 22).

The general form was not followed in three places:

Ag, the decimal point is two spaced from the right side of the column,

^{2.} In, there is no decimal point in the column, and

^{3.} Na20 and MgO, obtained by a different analytical procedure, were punched on a separate card which was placed last in the group of cards.

^{2.} A data group is all the samples taken from one drill hole for one rock type.

ISN, internal statement number, refers to the statements shown in fig. F1.

Values for emulsion calibration curves and relative intensity ratios were obtained using an I.B.M. 7040 computer at the McGill University Computer Center. Programs for these two procedures are described by Lickus (1965).

An additional program was designed to manipulate the analytical results obtained in this report. While it is unlikely that the specific program described here will ever be useful, certain parts of it could be adapted to other projects. Of special interest should be the form in which data were stored and the "format statement" by which the data were removed from storage.

Figure F1 shows the Fortran program which was used in this project. Cards were punched from special data sheets which were filled out according to the example shown in figure F2. Codes and conventions dealing with the use of these data sheets are filed in the Department of Geological Sciences, McGill University.

The Fortran program instructs the computer:

1. To read into storage, the project number (PN), lab sample number (ALN), and analytical results (K) for twenty two elements and J samples per data group 2 (ISN 2, 4, 5, 22).

^{1.} The general form was not followed in three places:

^{1.} $\bar{\text{Ag}}$, the decimal point is two spaced from the right side of the column,

^{2.} In, there is no decimal point in the column, and

Na₂D and MgD, obtained by a different analytical procedure, were punched on a separate card which was placed last in the group of cards.

A data group is all the samples taken from one drill hole for one rock type.

ISN, internal statement number, refers to the statements shown in fig. F1.

```
Fig. F1 FORTRAN SOURCE LIST
SAKRESON
          STAT TEST
     ISN
                SOURCE STATEMENT
        O SIBFTC MAIN
                DIMENSION ALN(5), FN(5), XMD(5), DHN(5), EL(5), FT(5), X(5,24), SD(5),
        1
               1PN(5)
           4000 READ31,M
       2
        4
             31 FORMAT(111)
        5
                READ 32, ((PN(J), ALN(J), (X(J,K),K=1,22)),J=1,M)
       22
             32 FORMAT(2X,F6.0,F8.0/27X,4F4.2,4X,F4.2,4X
               F5.1/47X,2F5.1/42X,F5.1,F5.0,F5.1/51X,F4.2,4X,F4.2)
               2
       23
                L=0
       24
             39 DO40K=1,22
       25
            310 V=0.
       26
                SUMX=0.
       27
                xsqs=0.
                D0320J=1,M
       30
       31
                IF(X(J,K))30,320,30
       32
             30 V=V+1.
       33
                SUMX=SUMX+X(J,K)
       34
                XSQ=X\{J,K\}**2
       35
                XSQS=XSQS+XSQ
       36
            320 CONTINUE
       40
                [f(v.EQ.0.)G0T040
       43
                XM=SUMX/V
       44
                SXSQ=SUNX**2
       45
                SDX=(V*XSQS-SXSQ)**.5/V
       46
                VARX=SDX++2
       47
                CVARX=SDX/XM*100.
       50
                SEXM=SDX/(V-1.)**.5
                SESDX=SDX/(2.*(V-2.))**.5
       51
       52
                IF(V-2.)50,600,50
       53
            600 SSEXM=SEXM+12.706
       54
                SSESDX=SESDX#12.706
       55
                GOT01200
       56
             50 [F(V-3.)60,800,60
       57
            800 SSEXM=SEXM*4.303
       60
                SSESDX=SESDX+4.303
                GOT01200
       61
       62
             60 [F(V-4.)70,900,70
            900 SSEXM=SEXM*3.182
       63
       64
                SSESDX=SESDX*3.182
                GOT01200
       65
             70 IF(V-5.180,1000,80
       66
           1000 SSEXM=SEXM+2.776
       67
       70
                SSESDX=SESDX+2.776
                GOT01200
       71
       72
             80 CONTINUE
           1200 PRINT33
       73
             33 FORMAT(1H1,10X,14HLAB SAMPLE NO.)
       74
                PRINT35, (ALN(J), J=1, M)
       75
      102
             35 FORMAT(1H0,5(/F19.0))
                GO TO(51,52,53,54,55,57,59,61,62,63,64,65,66,67,68,69,71,72,
      103
                  73,74,56,581,K
             51 PRINT1
      104
              1 FORMAT(1HO,15X,12HSID2 PERCENT/)
      105
                GU TO 140
      106
```

```
Fig. F1 (continued)
FORTRAN SOURCE LIST MAIN
SAKRISON
          STAT TEST
     ISN
                SOURCE STATEMENT
     107
             52 PRINT2
     110
              2 FORMAT(1H0,15x,12HTIO2 PERCENT/)
                GO TO 140
     111
     112
             53 PRINTS
     113
              3 FORMAT(1H0,15X,13HAL2O3 PERCENT/)
     114
                GO TO 140
     115
             54 PRINT4
     116
               FORMAT(1HO,15X,13HFE2O3 PERCENT/)
     117
                GO TO 140
     120
             55 PRINTS
     121
              5 FORMAT(1H0,15X,11HMND PERCENT/)
     122
                GO TO 140
     123
             57 PRINT7
     124
              7 FORMAT(1HO,15X,11HCAO PERCENT/)
     125
                GO TO 140
     126
             59 PRINT9
     127
              9 FORMAT(1H0,15X,11HK2D PERCENT/)
     130
                GO TO 140
     131
             61 PRINTIO
     132
             10 FORMAT(1HO,15X, 6HAG PPM/)
     133
                GO TO 140
     134
             62 PRINT11
     135
             11 FORMAT(1H0,15X, 6HC0 PPM/)
     136
                GO TO 140
     137
             63 PRINT12
     140
             12 FORMAT(1HO,15X, 6HCR PPM/)
     141
                GO TO 140
     142
             64 PRINT13
     143
            13 FORMAT(1H0.15X. 6HCU PPH/)
     144
                GO TO 140
     145
             65 PRINT14
     146
             14 FORMAT(1HO,15X, 6HGA PPM/)
     147
                GO TO 140
     150
            66 PRINT15
             15 FORMAT(1H0,15X, 6HGE PPM/)
     151
     152
                GO TO 140
     153
            67 PRINT16
             16 FORMAT(1HO,15X, 6HNI PPM/)
     154
     155
                GO TO 140
     156
            68 PRINT17
     157
            17 FORMAT(1HO,15X, 6HP8 PPM/)
     160
                GO TO 140
            69 PRINTI8
     161
     162
            18 FORMAT(1HO,15X, 6HSN PPM/)
     163
                GO TO 140
     164
            71 PRINT19
     165
            19 FORMAT(1HO,15X, 6HSR PPM/)
     166
                GO TO 140
     167
            72 PRINT20
            20 FORMAT(1H0,15X, 5HY PPM/)
     170
     171
                GO TO 140
     172
            73 PRINT21
     173
            21 FORMAT(1HO,15X, 6HZN PPM/)
     174
                GO TO 140
     175
            74 PRINT22
```

```
Fig. F1 (continued)
FORTRAN SOURCE LIST MAIN
SAKRISON
          STAT TEST
     ISN
                SOURCE STATEMENT
     176
             22 FORMAT(1HO,15X, 6HZR PPM/)
     177
                GO TO 140
     200
             56 PRINT6
     201
             6 FURMAT(1HO,15X,11HMGO PERCENT/)
     202
               GO TO 140
     203
             58 PRINTS
     204
              8 FORMAT(1HO,15X,12HNA2O PERCENT/)
     205
           140 CONTINUE
     206
                IF(L-1)3000,36,3000
     207
             36 PRINT37
     210
             37 FORMAT(1H ,15x,42HTHE NATURAL LOGARITHM OF THE CONCENTRATION)
          3000 PRINT38, (X(J,K), J=1,M)
     211
            38 FORMAT(1H ,F23.2)
     216
     217
                PRINT151
     220
            151 FORMAT(1H0,25X,3HSUM)
     221
                PRINT41, SUMX
             41 FORMAT(1H ,F24.4)
     222
     223
                PRINT42
             42 FORMAT(1HO,15X,15HARITHMETIC MEAN)
     224
     225
                PRINT41,XM
     226
                PRINT43
             43 FORMAT(1HO,15X,25HSAMPLE STANDARD DEVIATION)
     227
                PRINT41,SDX
     230
     231
                PRINT44
     232
             44 FORMAT(1HO, 15X, 8HVARIANCE)
     233
                PRINT41, VARX
     234
                PRINT45
             45 FORMAT(1H0,15X,24HCOEFFICIENT OF VARIATION)
     235
     236
                PRINT41,CVARX
     237
                PRINT46
             46 FORMAT(1HO, 15X, 26HSTANDARD ERROR OF THE MEAN)
     240
                PRINT41, SEXM
     241
     242
                PRINT47
           . 47 FORMAT(1HO,15x,40HSTANDARD ERROR OF THE STANDARD DEVIATION)
     243
     244
                PRINT41, SESDX
     245
                PRINT48
      246
             48 FORMAT(1HO, 15X, 31HS.E. OF THE MEAN SMALL SAMPLES)
                PRINT41, SSEXM
     247
      250
                PRINT49
      251
             49 FORMAT(1HO,15X,36HS.E. OF THE STD. DEV. SMALL SAMPLES)
     252
                PRINT41, SSESDX
             40 CONTINUE
      253
      255
                L=L+1
      256
                IF(L.GE.2)G0TU4000
      261
                D05000J=1.M
                D05000K=1,22
      262
                1F(X(J,K))5000,5000,6000
      263
      264
           6000 X(J,K)=ALOG(X(J,K))
           5000 CONTINUE
      265
                GOTO39
      270
                END
      271
```

Fig. F-2 Special Data Sheets Used in Data Handling. LAD SAMPLE FIELD SAMP MATRIX STATE + CO SC TWP RG YEA NO PROJECT NO SAMPLE NO DESC DESC FOOTAGE MISC et⊊. LAT LONG LAT DEP DDH NO ELEV 17563 220001 DWA QUE 940GE METPRC DATE S:02 Ti 02 A1203 Fe203 FeΟ MnO Naso K20 P205 H20 82 $(C_2, Y)_2 O_2 | C_{P_2} O_3 |$ V2O3 BaO SrO BeO CuO LizO B203 B Ba Вe Bi Ce Co 38 OSTNJUNGSHS Cs Cu G_{a} Ge 28 103 H Nb Nd Pb Sb Sc 75 10

LAB SAMPLE NO.

35163.

36863.

32063.

35863.

SIO2 PERCENT

76.50

79.00

63.60

78.00

SUM

297.1000

ARITHMETIC MEAN 74.2750

SAMPLE STANDARD DEVIATION 632271

VARIANCE 38.7770

COEFFICIENT OF VARIATION 8.3839

STANDARD ERROR OF THE MEAN 345952

STANDARD ERROR OF THE STANDARD DEVIATION 3-1136

S.E. OF THE MEAN SMALL SAMPLES. 11:44400

S.E. OF THE STD. DEV. SMALL SAMPLES 9.9073

Fig. F3 Example of Output

- 3. To calculate the arithmetic mean (ISN 43).
- 4. To calculate the sample standard deviation (ISN 34, 35, 44, 45).
- 5. To calculate the variance (SN 46).
- 6. To calculate the coefficient of variance (ISN 47).
- 7. To calculate the standard error of the mean (ISN 50).
- 8. To calculate the standard error of the standard deviation (ISN 51).
- 9. To correct the standard error of the mean for small samples (ISN 52, 53, 56, 57, 62, 63, 66, 67).
- 10. To correct the standard error of the standard deviation for small samples (ISN 52, 54, 56, 60, 62, 64, 66, 70).
- 11. To print titles e.g. LAB SAMPLE NO. (ISN 73, 74), SiO₂
 PER CENT (ISN 103, 104, 105).
- 12. To place under the appropriate title those figures which the computer has calculated and stored, e.g., the lab sample number (ISN 75, 102) or per cent SiO₂ (ISN 3000).
- 13. When all the preceding has been completed for all the elements in all the samples of one data group (J), to convert all the analytical results to natural logarithms and repeat the whole program (ISN 206, 264).

A special card, an "M" card, must be inserted before each data group. This card is punched 2, 3, 4, or 5 in column one in order to tell the computer how many samples there are in one data group (ISN 2, 4). The number of data groups which can be handled in one job is unlimited. The computer will simply stop storing data after the last data card has

been read. The number of samples per data group (J) is limited to five but could be increased with minor changes to the program.

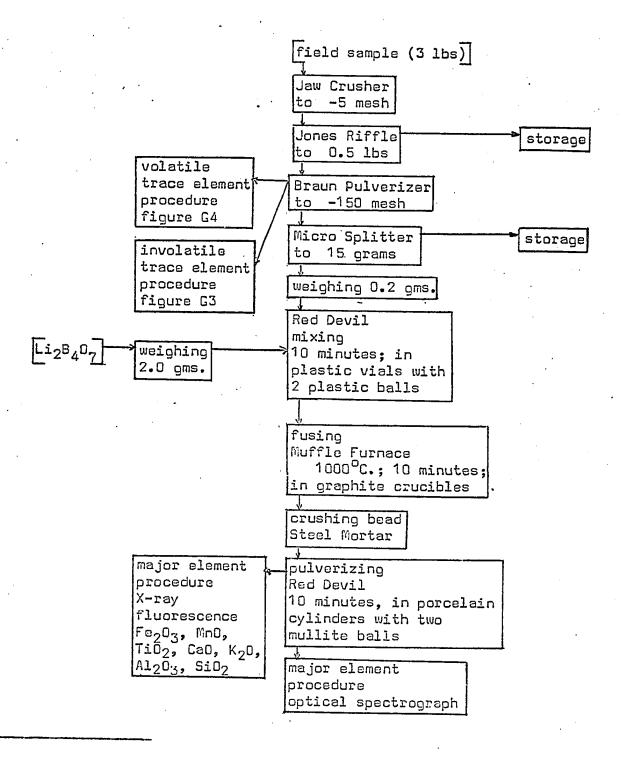
An example of the output of this program for one element in one data group of four samples is given in fig. F3. The program was designed to produce neat pages of figures but it uses up a great deal of paper and computer time and is therefore not at all economical. This fault should be changed before this program is used again.

APPENDIX G

Description of Sample Preparation and Analytical Procedures

Figure G1

SAMPLE REDUCTION AND SAMPLE PREPARATION FOR THE MAJOR ELEMENT PROCEDURE, X-RAY FLUORESCENCE 1



Based on a procedure designed by J. Jellema and G. R. Webber in the Department of Geological Sciences, McGill University.

SAMPLE PREPARATION, MAJOR ELEMENT PROCEDURE OPTICAL SPECTROGRAPH

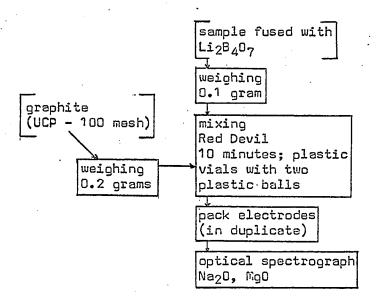


Figure G3

SAMPLE PREPARATION, INVOLATILE TRACE ELEMENT PROCEDURE, OPTICAL SPECTROGRAPH

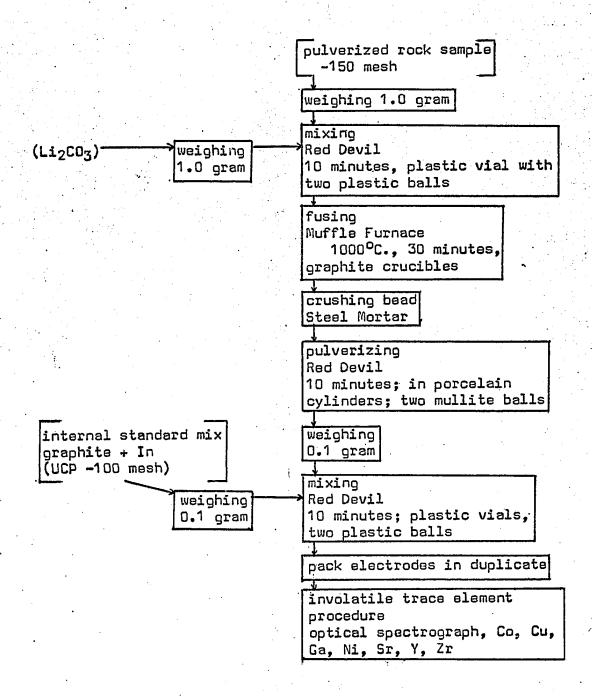


Figure G4

SAMPLE PREPARATION, VOLATILE TRACE ELEMENT PROCEDURE, OPTICAL SPECTROGRAPH

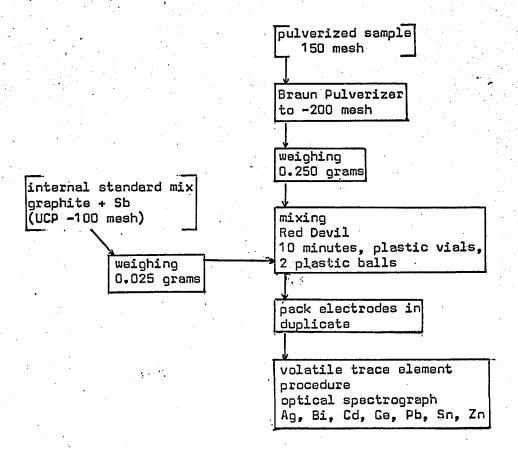


Table G-1

TIME REQUIRED FOR AN INEXPERIENCED TECHNICIAN TO PREPARE APPROXIMATELY SIXTY SAMPLES FOR ALL PROCEDURES

(NOT INCLUDING CRUSHING AND PULVERIZING)

MAJOR ELEMENT PROCEDURES

				Hours		
4.				8 9 3 3.5 12 9		
. •	•		Total		48.5	hours
		INVOLATILE	TRACE ELEMEN	T PROCEDURE		
1. 2. 3. 4. 5. 6.	Crushing	Li ₂ Co ₃ paperwork) and pulverize	ing bead and graphite	9 3 10 12 10 5		
			Total		58	hours
		VOLATILE	TRACE ELÉMENT	PROCEDURE		
1. 2.	Weighing Mixing	sample and gr	raphite	7.5 3		
,		·	Total		10.5	hours
			Grand Tota	1	117.0	hours

DETAILS OF ANALYTICAL PROCEDURES USING THE OPTICAL SPECTROGRAPH

	I MAJOR ELEMENTS ¹ (Na ₂ O, MgO)	II INVOLATILE TRACE ¹ ELEMENTS	III VOLATILE TRACE ELEMENTS
Apparatus (ancillary)	JACO console microphotometer	as I	as I
Standard samples	N.B.S. 1A, 69A, 8B, 97, 104 and mixtures of these	prepared by adding compounds of the metals to an artificially prepared base	as II
Flux	Li ₂ 8 ₄ 0 ₇ (10 parts flux; 1 part sample)	Li ₂ Co ₃ (1:1)	none
Internal standard	Li added as flux	0.01% In in graphite + 20% NaCl; mix 1:1 with fluxed sample	0.08% Sb ₂ 0 ₃ in graphite; mix 1 part to 10 parts sample
Electrodes	Cathode 1/8" dia. graphite (ST-40) Anode 3/16" dia. graphite (ST-45) cup in anode, 3 mm x 5 mm	as I	cathode as I anode as I cup; 3 mm x 3/16" boiler caps; 3/16" i.d., graphite, National AGKSPL 3916

^{1.} The major element and involatile trace element procedures were designed by J. Jellema and G. R. Webber in the Department of Geological Sciences, McGill University.

DETAILS OF ANALYTICAL PROCEDURES USING THE OPTICAL SPECTROGRAPH

	I MAJOR ELEMENTS (Na ₂ O, M ₉ O)	II INVOLATILE TRACE ELEMENTS	III VOLATILE TRACE ELEMENTS
Excitation and exposure	11 amps to completion; 15 seconds initial burn at 4 amps;	as II	12 amps for 90 seconds; no preburn
TARA	preburn ½ second		
Spectral range control setting	9.87 (second order)	6.10	as II
Slit width	10u	as I	as I
Slit height	5mm	as I	6mm
Filter steps	3,4,5,6	as I	as I
Focus	16	15	as II
Initial racking setting	7	as I	as I
Racking units	5	as I	6
Number of exposures/plate	16	as I	13

able G-2 (cont'd)

DETAILS OF ANALYTICAL PROCEDURES USING THE OPTICAL SPECTROGRAPH

	I MAJOR ELEMENTS (Na ₂ O, M ₉ O)	II INVOLATILE TRACE ELEMENTS	III VOLATILE TRACE ELEMENTS
Emulsion	Kodak SA-2, K-33	as I	Kodak K-33
Developing	3 min. at 20 ^o C.; Kodak D−19	as I	as I
Fixing	10 min., Kodak Fixer	as I	as I
Element and internal standard analytical lines (wave length, A)	Na (3302), Mg (2781); Li (2741)	Co (3453), Cu (3273), Ga (2943), Ni (3414), Sr (3464), Y (3327), Zr (3391), In (3039)	Ag (3381), Bi (3067), Cd (3261), Ge (3039), Pb (2833), Sn (3262), Zn (3345), Sb (2877)

Photographs of the Host Rocks



Fig. 1 - <u>Waite Rhyolite</u>, <u>pyroclastic</u>. Note the broken plagioclase crystals in a fine grained matrix. Leucoxene is opaque. The section is from DDH N-115, about 80 feet below the top of the rhyolite. (crossed nicols)

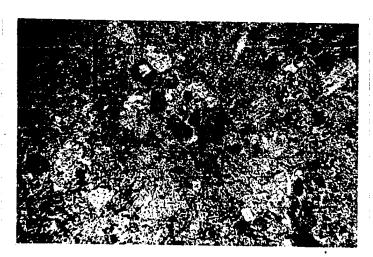


Fig. 2 - <u>Same as Fig. 1</u>. Chlorite and sericite appear gray. (plane polarised light)



Fig. 1 - <u>Waite Rhyolite</u>, <u>deformed chloritic</u>
<u>fragment in dense siliceous matrix</u>. The
mineral which appears light gray in the small
fragment is epidote. The dark areas are
mostly fine grained chlorite and sericite.
The specimen is from the same locality as
that of Plate 1. (plane polarised light)

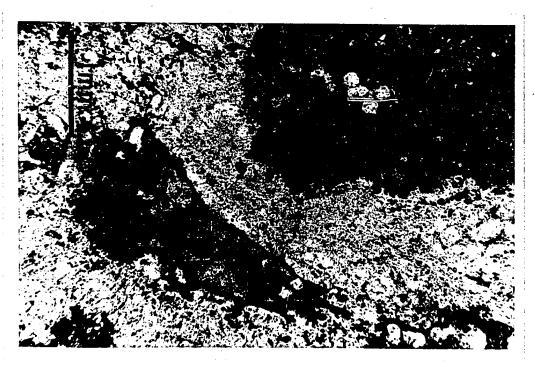


Fig. 1 - Waite Rhyolite, deformed chloritic fragment in dense siliceous matrix. The mineral which appears light gray in the small fragment is epidote. The dark areas are mostly fine grained chlorite and sericite. The specimen is from the same locality as that of Plate 1. (plane polarised light)

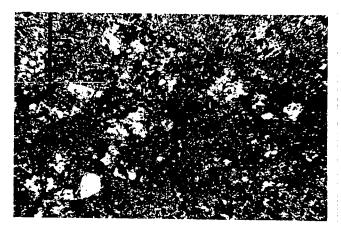


Fig. 1 - <u>Waite Rhyolite</u>, <u>pyroclastic</u>. This photograph is of the same thin section shown in Plate 3. Note the many fragments of plagioclase feldspar. (crossed nicols)



Fig. 2 - <u>Same as Fig. 1</u>. Note the slight suggestion of layering in the alignment of feldspar fragments and leucoxene. (plane polarised light)



Fig. 1 - Waite Rhyolite, scoriaceous fragment. Note the aligned quartz-filled vesicles. The specimen is from DDH N-114 about 100 feet below the top of the rhyolite. (plane polarised light)

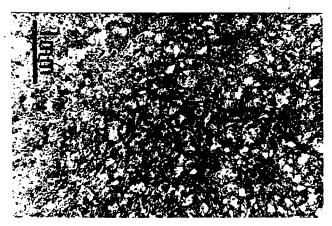


Fig. 2 - Waite Rhyolite, tuffaceous matrix of scoria. Note the irregularly shaped quartz grains in a very fine grained mat of sericite, chlorite and epidote. A faint suggestion of layering is present at 45 degrees across the photograph. The specimen is from the same location as that of Fig. 1. (plane polarised light)

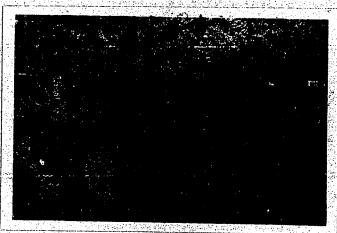


Fig. 1 - Waite Rhyolite, altered, from the fractured and mineralized zone below massive ore. (c, altered cordierite (pinite); b, biotite; ch, chlorite; q, quartz mosaic; a, anthophyllite) The specimen is from DDH N-124 at 1458 feet. (plane polarised light)



Fig. 2 - <u>Same as Fig. 1</u>
Note that the areas of pinite have uniform extinction. (crossed nicols)

PLATE 6

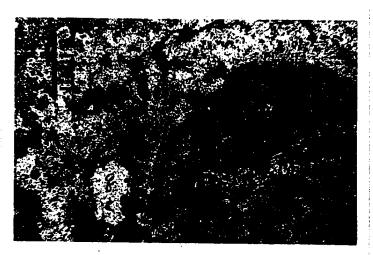


Fig. 1 - Waite Rhyolite, altered, from the fractured and mineralized zone below massive ore. (c, altered cordierite (pinite); b, biotite; ch, chlorite; q, quartz mosaic; a, anthophyllite) The specimen is from DDH N-124 at 1458 feet. (plane polarised light)

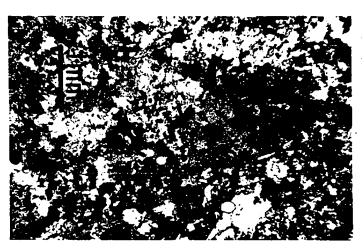


Fig. 2 - <u>Same as Fig. 1</u>
Note that the areas of pinite have uniform extinction. (crossed nicols)

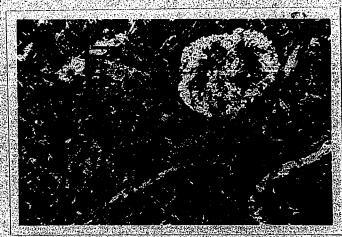


Fig. 1 - Amulet Andesite

Note the amygdales and the altered laths of plagicolase feldspar. The white rim of the amygdale is actinolite, the dark center is chlorite and sericite. Epidote shows up as dark, 0.2 mm, round patches. The speciman is from DDH N-160 about 75 feet above the base of the andesite. (plane polarised light)



Fig. 2 - Altered Amulet Andesite
The andesite is almost totally converted to a mat of sericite, chlorite, and leucoxene yet the original texture is still slightly apparent. The section is from DDH N-114, about 10 feet from the base of the andesite. (plane polarised light)



Fig. 1 - Amulet Andesite
Note the amygdales and the altered laths of
plagioclase feldspar. The white rim of the
amygdale is actinolite, the dark center is
chlorite and sericite. Epidote shows up as
dark, 0.2 mm, round patches. The speciman
is from DDH N-160 about 75 feet above the
base of the andesite. (plane polarised light)

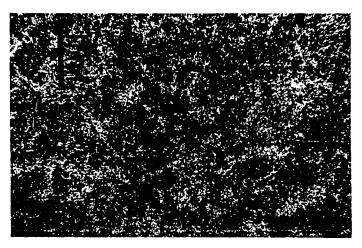


Fig. 2 - Altered Amulet Andesite
The andesite is almost totally converted to a mat of sericite, chlorite, and leucoxene yet the original toxture is still slightly apparent. The section is from DDH N-114, about 10 feet from the base of the andesite. (plane polarised light)

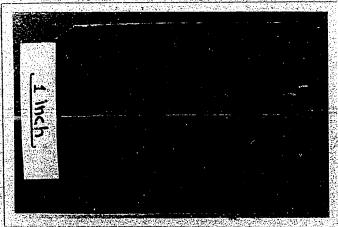


Fig. 1a (above) - <u>Waite Rhyolite</u>, <u>pyroclastic</u>.

Note the one inch long fragment with the narrow bleached rim in the left half of the section of drill core. Numerous grains of pyrrhotite are present in the matrix but the fragment is barren. Some pyrrhotite, but not as much as in the surrounding matrix, is present in the fragment in the right half of the core section. The specimen is from DDH N-97, 15 feet below the top of the rhyolite.

Fig. 1b (below) - <u>Waite Rhyolite</u>, <u>pyroclastic</u>.

Dark fragments with sharp corners, very rarely containing sulphide grains, are poorly sorted in a matrix which contains abundant pyrrhotite and some sphalerite. The specimen is from DDH N-97, 90 feet below the top of the rhyolite.

PLATE 8

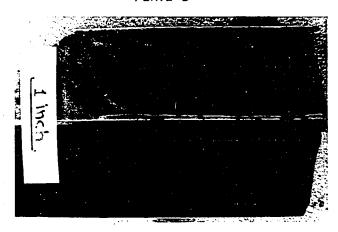


Fig. 1a (above) - <u>Waite Rhyolite</u>, <u>pyroclastic</u>.

Note the one inch long fragment with the narrow bleached rim in the left half of the section of drill core. Numerous grains of pyrrhotite are present in the matrix but the fragment is barren. Some pyrrhotite, but not as much as in the surrounding matrix, is present in the fragment in the right half of the core section. The specimen is from DDH N-97, 15 feet below the top of the rhyolite.

Fig. 1b (below) - <u>Waite Rhyolite. pyroclastic.</u>
Dark fragments with sharp corners, very rarely containing sulphide grains, are poorly sorted in a matrix which contains abundant pyrrhotite and some sphalerite. The specimen is from DDH N-97, 90 feet below the top of the rhyolite.

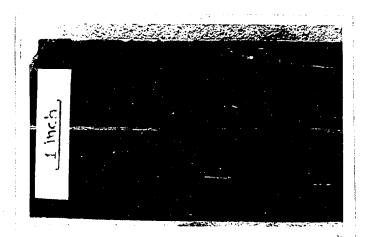


Fig. 1a (above) - <u>Waite Rhyolite</u>, <u>finely pyroclastic</u>.

The rhyolite contains about 0.5 per cent of disseminated, 0.1mm, pyrite cubes and is cut by many quartz-carbonate-chlorite veinlets which contain only very rare pyrite grains. The specimen is from DDH N-113, 2 feet below the top of the rhyolite and a 7 foot wide shear zone.

Fig. 1b (below) - <u>Waite Rhyolite</u>, <u>pyroclastic</u>. White fragments which contain no sulphide grains are scattered in a matrix which contains very small, disseminated grains of pyrrhotite. The specimen is from DDH N-113, 50 feet below the top of the rhyolite.



Fig. 1a (above) - <u>Waite Rhyolite, finely pyroclastic</u>.

This specimen is typical of much rhyolite in that fragments are not visable although the rhyolite has a mottled appearance. The sample comes from 375 feet below the top of the rhyolite and 150 feet below more obvious breccia. DDH N-121

Fig. 1b (below) - <u>Waite Rhyolite</u>, finely pyroclastic.
This specimen comes from the same locality as that of Fig. 1a but shows one quarter inch diameter, light colored, vesicular fragments concentrated in the right half of the section of core.

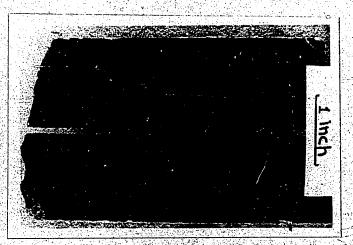


Fig. 1a (above) - <u>Waite Rhyolite, pyroclastic</u>.
This specimen shows dark, deformed fragments in a light colored dense matrix. It is from DDH N-115, 60 feet below the top of the rhyolite.

Fig. 1b (below) - <u>Waite Rhyolite, pyroclastic</u>.

This specimen is from the same locality as that of Fig. 1a.

It shows how the same fragmental rhyolite grades into an apparently non-descript, mottled rock. A photograph of a thin section of this material is presented in Plate 2, Fig. 1.

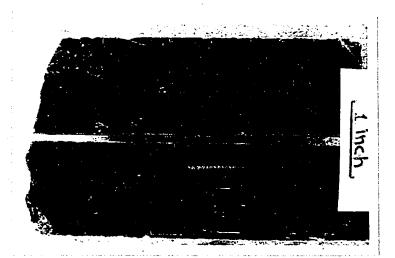


Fig. 1a (above) - <u>Waite Rhyolite, pyroclastic</u>.
This specimen shows dark, deformed fragments in a light colored dense matrix. It is from DDH N-115, 60 feet below the top of the rhyolite.

Fig. 1b (below) - <u>Waite Rhyolite</u>, <u>pyroclastic</u>.

This specimen is from the same locality as that of Fig. 1a.

It shows how the same fragmental rhyolite grades into an apparently non-descript, mottled rock. A photograph of a thin section of this material is presented in Plate 2, Fig. 1.

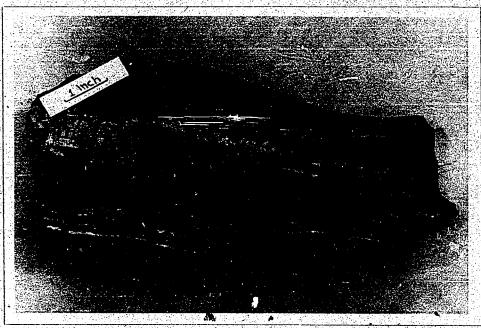


Fig. 1 - Amulet Andesite in contact with Bedded,
Cherty Tuff. Relatively fresh andesite is dark and
is altered only in a one inch wide zone at the tuff
contact. The white bleaching in the andesite is
caused by the development of sericite.

(403 drive, station 417 + 20 feet north,
west wall)

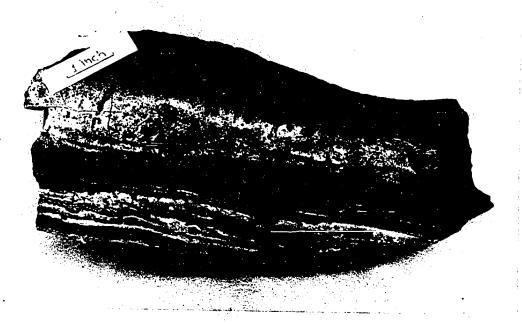


Fig. 1 - Amulet Andesite in contact with Bedded, Cherty Tuff. Relatively fresh andesite is dark and is eltered only in a one inch wide zone at the tuff contact. The white bleaching in the andesite is caused by the development of sericite.

(403 drive, station 417 + 20 feet north, west wall)

APPENDIX I

Sample Locations

Table I-1

APPENDIX I

SAMPLE LOCATIONS

(See diamond drill hole plans for drill hole locations.)

DDH (elev. at collar, feet)	Field Samplo Number	Lab. Samplo Numbe r	Footage (feet)	Rock Type1
N-91 1070	710 711 712 713 714 715 716	17563 17263 15963 15763 32663 18163 16463	75-107 125-150 150-168.7 180.5-211 212-225 225-250 250-267.5	A A A A A
N-92 1050	717 718 719 720 721 722 723 724	18463 15863 16163 20163 24963 16563 15463	100-125 125-150 150-175 175-207.5 259-275 275-300 300-325 325-350	A A A R R R R
N-93 1050	725 726 727 728 729 730 731	24463 17363 26163 17663 18163 26563 13063	195-200 200-225 225-250 250-275 275-296 298-325 325-357.5	A A A A R R
N-94A 1070	732 733 734 735 736 737	55963 22263 26463 25963 20863 39163	225-250 250-275 275-300 300-331 333-350 350-375	A A A R R

^{1.} R: Rhyolite; A: Andesite; T: Tuff

DDH (elev. at collar, feet)	Field Sample Number	Lab. Sample Number	Footage (feet)	Rock Type
N-96 (1154)	738 739 740 741 742 743	13463 18063 26263 26063 20663 16363	150-167 180-200 200-225 225-251 279-300 300-336	A A A R R
N-97 (1169)	744 745 746 747 748 749 750 751	20763 17763 7763 25463 25063 20363 33263 16663	450-475 475-494 502-532 532-548 548-575 575-607 607-632 632-650	A A A R1 A1 A1 R R
N-98 (1079)	752 753 754 755 756	23063 14263 14563 13363 15263	332-425 425-450 450-468 471-486 486-524	A A A R R
N-110 (1216)	757 758 759 760 761 762	22563 21863 8363 8263 7863 22063	1200-1223 1245-1251 1261-1273 1288-1319 1458-1489 1489-1525 1550-1575	A A R R R
N-111 (1170)	935 936 937 938 939 940 941	25263 28463 30163 30463 36763 29163 36063	800-825 825-840.5 856.5-888 888-925 925-950 950-975 975-1000	A A A R R R R

rock type questionable; these samples sere not included in the compilation.

DDH (elev. at collar, feet)	Field Sample Number	Lab. Sample Number	Footage (feet)	Rock Type
N-112 (1153)	763 764 765 766 767 768 769	13563 21763 14463 18363 19663 22463 21463	925-947 954-975 975-1000 1000-1029 1029-1044 1044-1093 1093-1125	A A A R R R
N-113 (1150)	942 943 944 945 946 947 948 949	37863 37763 37463 31163 36963 9863 29663 39263	1325-1350 1350-1375 1375-1410 1410-1418 1425-1441 1441-1475 1475-1500 1500-1525	A A A R R R R
N-114 (1163)	950 951 952 953 954 955A	32763 25863- 28063 28363 7363 20063	1150-1175 1175-1200 1200-1225 1225-1250 1250-1275 1275-1300	A A R R R
N-116 (1115)	770 771 772 773 1343 774 775	22663 8563 19763 22363 51563 7663 29363 6363	550-575 575-600 600-625 625-658 660.5-662.5 662-700 700-725 725-750	A A A T R R
N-117 (1191)	777 778 779 780 781 782 783 784	23863 20463 8763 21563 20963 30863 15563 29763	1150-1175 1175-1200 1200-1225 1225-1257.5 1257.5-1272 1282-1300 1300-1325 1325-1350	A A A R R R

DDH (elev. at collar, feet)	Field Sample Number	Lab. Sample Number	Footage (feet)	Rock Type
N-118 (1130)	785 786	28963 26363	450-475 475-491.5	A A
	787 788A 1342 788 789 790	28563 20563 51463 31563 29063 30563	497.5-500 500-525 525-557 556.5-560 558.5-576 576-600 600-625	A A T R R R R
N-119 (1018)	791 792 793 794 1341 795 796 797	22763 22463 27763 30363 51364 31763 24863 29263 55963	900-925 925-950 950-975 975-992 992-994.5 994-1025 1025-1050 1050-1075	A A A T R R R R
N-120 (1232)	799 800 801 802 803 804 805	33663 29463 27663 6563 26963 27563 27463 (lost)	800-825 825-850 850-875 875-897 898.5-925 925-950 950-975	A A A R R R R
N-121 (1029)	807 808 809 810 1340A 1340B 1340C 811 812 813	28763 16263 24563 14963 51063 51163 51263 23663 7463 19563 16963	675-700 700-725 725-750 750-772.5 773-774 786-787.5 798.5-799.5 794.5-825 825-850 850-875 875-900	A A A T T R R R

DDH (elev. at collar, feet)	Field Sample Number	Lab. Samplo Number	Footage (feet)	Rock Type
N-122 (1315)	815 816 817 818 819 820 821	21063 19963 26863 16763 32563 34063 33163	925-950 950-975 975-1000 1000-1019 1030-1050 1050-1075	A A A A R R R
N-123 (1107)	822 823 824 825	24263 24163 23563 23963	409-439 444-475 475-500 500-516	A R R R
N-125 (1169)	1022 1023 1024 1025 1026 1027 1028	39063 35963 11563 10463 37963 35563 37563	1250-1275 1275-1325 1325-1335 1344-1366 1335-1344 1366-1409 1409-1425 1425-1450	A A A A R R R R
N-127 (1146)	997 998 999 1000 1001 1002 1003 1004 1005	11263 9563 9263 9363 34263 3063 35463 35763 34363	1000-1025 1025-1050 1050-1075 1075-1100 1100-1116 1273-1300 1300-1325 1325-1350	A A A A R R R R
N-128 (1178)	1006 1007 1008 1009 1010 1011 1012	9963 10263 9163 10663 34463 38963 36463 37663	1050-1075 1075-1100 1100-1125 1125-1155 1248-1277 1277-1303 1303-1325 1325-1350	A A A R R R R

DDH (elev. at collar, feet)	Field Sample Number	Lab. Sample Number	Footage (feet)	Rock Type
N-129 (1157)	1014 1015 1016 1017 1018 1019 1020	11363 9763 8863 9663 34163 10163 33763 27263	1075-1100 1100-1124.5 1169-1200 1200-1225 1305-1323 1346-1375 1375-1406 1406-1425	A A A A R R R R
N-132 (1162)	962 963 964 965 966 967 968	10963 10063 11663 40363 11063 56063 10363	1250-1276 1302.5-1325 1325-1347 1352-1362.5 1362.5-1396 1396-1425 1425-1450	A A A R R R
N-133 (1185)	969 970 971 972 973 974 975	27863 12363 37063 35663 9463 34863 31063	1050-1075 1075-1093 1110-1125 1125-1157 1199-1225 1400-1425 1425-1450	A A A R R R
N-134 (1107)	927 928 929 930 931 932 933	32963 30263 32163 35163 36863 32063 35863	1000-1025 1025-1050 1050-1075 1075-1098 1111.5-1132 1132-1162.5 1162.5-1177 1177-1200	A A A R R R
N-135 (1143)	826 827 828 829 1339 830 831 832	36363 36563 37263 36263 50963 15063 12163 15363	1000-1023 1023-1050 1050-1083 1083-1114 1091-1092 1114-1150 1150-1175 1175-1200	A A A T R R P

DDH (elev. at collar, feet)	Field Sample Number	Lab. Sample Numbe r	Footage (feet)	Rock Type
N-139 (1128)	833 834 835 836 1338 837 838 839 840	19863 17063 6763 22863 50863 24663 13863 25363 7163	950-969 969-988 988-1025 1025-1046 1046-1047 1047-1075 1075-1100 1100-1125 1125-1150	A A A T R R R
N-142 (1177)	976 977 978 979 980 981 982	20263 38363 18963 19163 17863 21363 25663	1025-1050 1050-1067 1081-1096 1096-1109 1121-1144.5 1144.5-1175 1175-1200 1200-1225	A A A R R R
N-144 (1112)	841 842 843 844 1335 845 846 847 848	30063 19063 38263 38163 50563 21663 38863 (lost) (lost)	925-950 950-975 975-1000 1000-1023.5 1081-1082 1091.5-1093 1036-1050 1050-1075 1075-1100 1100-1125	A A A T R R R R
N-149 (1147)	983 984 985 986 987 988 989	11863 14163 13263 13163 6963 12063 34963 18863	1300-1325 1325-1345 1348-1375 1375-1391 1412.5-1425 1425-1450 1450-1475 1475-1496	A A A R R R R
N-150 (1198)	849 850 851 852 1336 853	26763 28863 31263 28663 50663	1014-1050 1050-1075 1075-1094 1193.5-1234 1233.5-1235 1237.5-1243 1240-1275	A A A T R

		•		
DDH (elev. at collar, feet)	Field Sample Number	Lab. Sample Number	Footage (feet)	Rock Type
N-151 (1190)	859 860 861 862 863 864 865	32363 24763 8063 33363 13663 14663 24063	1050-1075 1075-1100 1100-1122 1123-1150 1150-1175 1175-1200 1200-1225	A A R R R
N-152 (1140)	866 867 868 869 870 871	19463 7263 33463 16063 15163 30663	1300-1325 1325-1350 1350-1375 1375-1398.5 1405-1425 1425-1465	A A A R R
N-153 (1122)	991 992 993 994 995 996	10563 11763 10763 9063 8963 11163	1089-1130 1130-1139 1139-1175 1175-1200 1200-1225 1225-1250	A T R R R
N-155 (1180)	872 873 874 875 1335 876 877 878	16863 17963 23263 23363 50563 25563 23463 8163	1000-1025 1025-1055 1055-1075 1076-1090 1081-1082 1091.5-1093 1094-1125 1125-1150 1150-1168	A A A T R R R
N-157 (1198)	879 880 1334 881 882 883	23163 25763 50463 30763 29863 25163	1175-1200 1200-1234 1257-1260 1266-1300 1300-1325 1325-1350	A A T R R
N-158 (1080)	884 885 886 887 1133 888	19363 13763 31863 21263 50363 21963	1075-1100 1100-1125 1125-1150 1150-1175 1180-1185 1185-1199	A A A T R

DDH (elev. at collar, feet)	Field Sample Number	Lab. Sample Number	Footage (feet)	Rock Type
N-159 (1119)	889 890 891 892 1132 893 894 895 896	12863 14363 17163 11263 50263 18563 38363 38063 18763	950-988 988-1025 1025-1050 1050-1066 1061-1066 1067-1086 1086-1100 1100-1125 1125-1150	A A A T R R R R
N-160 (1112)	955B 956 957 958 959 960 961	32863 33963 27963 32463 22163 28163 29563	975-1000 1000-1025 1025-1057.5 1070-1100 1100-1125 1125-1150 1150-1175	A A R R R
N-161 (1026)	897 898 899 900 901 902 903	31363 31963 32263 26663 28263 9663 29963	950-975 975-1000 1000-1025 1025-1044.5 1052.5-1075 1075-1100 1100-1113 1140-1144	A A A R R
N-162	904 905 906 907 908 910 911	36663 35263 34763 24363 35363 34563 14763 12563	1050-1075 1075-1100 1100-1125 1125-1152.5 1157-1175 1175-1200 1200-1225 1225-1250	A A A R R R
N-163	913 914 915 916 1131 917 918 919	27163 33563 17463 23763 50163 27363 31663 33063	975-1000 1000-1026.5 1026.5-1050 1050-1080.5 1080-1087 1088.5-1125 1125-1150 1150-1175	A A A T R R

DDH (elev. at collar, feet)	Field Sample Number	Lab. Sample Number	Footage (feet)	Rock Type
N-164 (1136)	920 921 922 1130 923 924 925 926	36163 34663 37163 50063 14863 12463 21163 11963	1300-1332.5 1346.5-1375 1375-1398 1404-1407.5 1412-1425 1425-1450 1450-1475 1475-1500	A A T R R R
N-174 (1157)	1088 1089 1090 1091	51663 51763 51863 51963	800-825 825-850 850-875 875-890 894-912	A A A
	1092 1093 1094 1095 1096	52063 52163 52263 52363 52463	942.8-943.4 954-975 975-1000 1000-1025 1025-1050	TRRRRRRRRRRRRRRRRRRRRRRRRRRRRRRRRRRRRR
N-181 (1119)	1097 1098 1099 1100 1101 1102 1103 1104 1105	52563 52663 52763 52863 52963 53063 53163 53263 53363	925-950 950-975 975-1000 1000-1023 1028.5-1034.5 1034-1060 1060-1085 1085-1109 1109-1134	A A A T R R R R
N-190 (1151)	1106 1107 1108 1109 1110 1111 1112 1113	53463 53563 53663 53763 53863 53963 54063 54163 54263	925-950 950-975 975-1000 1000-1023 1041-1049 1049-1075 1075-1100 1100-1125 1125-1150	A A A T R R R

DDH (elev. at collar, feet)	Field Sample Number	Lab. Sample Number	Footage (feet)	Rock Type
N-199 (1110)	1115 1116 1117 1118 1119 1120 1121 1122	54363 54463 54563 54663 54763 54863 54963 55063	1325-1350 1350-1375 1375-1400 1400-1431 1502-1525 1525-1550 1550-1575 1575-1600	A A A R R R
N-204 (1110)	1123 1124 1125 1126 1127 1128 1129	55263 55363 55463 55563 55663 55763 55863	1025-1050 1128-1150 1150-1168 1200-1225 1225-1250 1250-1275 1275-1300	A A A R R R