High Temperature Friction and Wear Behavior of Cold-Sprayed Ti6Al4V and Ti6Al4V-TiC Composite Coatings

Venkata Naga Vamsi Munagala¹, Tyler B. Torgerson², Thomas. W. Scharf², and Richard. R. Chromik¹

¹Department of Mining and Materials Engineering, McGill University, Montreal, QC H3A 0C5,

Canada

² Department of Materials Science and Engineering, University of North Texas, Denton, TX 76203, USA

Abstract

The poor wear resistance of titanium and its alloys often results in galling and high wear rates. One method to improve their wear resistance is by hard secondary phase reinforcement to create a metal matrix composite, which can be utilized in the bulk or as a coating. In the present study, Ti6Al4V coatings and Ti6Al4V-TiC metal matrix composite coatings were deposited on mild steel substrates using cold spray process and their dry sliding wear behaviour was studied over a static temperature range of 25–575°C. Tests were performed using a unidirectional ball-on-disc tribometer with WC-Co sphere as the counterface at a load of 2.5 N and sliding velocity of 2.1 cm.s⁻¹. Wear rate and coefficient of friction (CoF) decreased with an increase in temperature and composite coatings exhibited higher wear resistance at all temperatures. Below 200°C, abrasive wear characterized by ploughing by wear debris resulted in high wear of Ti6Al4V coatings, whereas formation of tribolayers led to lower wear of composite coatings. At elevated temperatures (> 200°C), oxide glaze layers formed on both coatings were composed of WO₃, TiO₂ and CoWO₄. Electron channel contrast imaging of the wear track cross-sections showed no splat debonding, less extent of recrystallization and larger grain sizes (at 575°C) in composite coatings compared to Ti6Al4V.

Keywords: Cold spray, Metal matrix composites, Sliding wear, Ti6Al4V, Tribo-oxides

1. Introduction

During wear, metal matrix composite (MMC) coatings form tribolayers that are often beneficial and improve tribological properties in comparison to a pure metal coating of the same material as the matrix [1-6]. Tribolayers form due to the agglomeration of third bodies that are generated from one or both of the mating surfaces (first bodies) during wear [7–10]. The presence of tribolayers, which often have different properties than the first bodies, alter the contact conditions and tribology of the system [3,7]. Alidokht et al. and Torgerson et al. studied the sliding wear of cold sprayed Ni and Ni-WC composite coatings and found that the reinforcement of WC in Ni matrix facilitated the formation of tribolayers during dry sliding wear which improved the wear resistance compared to pure Ni coatings [6,8]. To enhance the tribological properties of Ti6Al4V, Zhang et al. artificially sprinkled TiO₂ and Fe₂O₃ nanoparticles during the sliding wear tests to accelerate the formation of tribolayers [11]. They concluded that a minimum mixture of TiO₂-50 vol.% Fe₂O₃ was required to form tribolayers that increase the load bearing capacity and thus reduce the wear rate. Studies performed by Du et al. on wear behavior of NiCr/Cr₃C₂ - NiCr/hBN coatings show that presence of hBN improved the wear resistance and reduced the CoF compared to NiCr/Cr₃C₂ [12]. They reported that formation of a surface layer with embedded hBN particles, reduced the adhesion between the mating surfaces and improved tribological properties. Based on the above studies, the hard phase or solid lubricant changed the kinetics of tribolayer formation and also imparted better properties and wear resistance.

The nature of tribolayers formed on metals and MMCs will change with temperature. Other than mechanical mixing, tribo-oxidation with the atmosphere and tribochemical reactions between the two first bodies are the main mechanisms for tribolayer formation. At elevated temperatures, tribolayers are often referred as glaze layers since they are compacted layers of tribo-oxides comprising of third bodies from either one or both the mating surfaces. In terms of Ti alloys, Cui et al. studied the sliding wear behavior of Ti6Al4V coatings in a temperature and load range of 20–400°C and 50–250 N, respectively [13]. They found that an increase in temperature up to 200°C increased the wear rate with abrasive and adhesive wear resulting in severe wear. Beyond 200°C, the wear rate decreased due to the formation of protective tribolayers resulting in mild wear. Wang et al. performed sliding wear test on Ti alloys within a range of 25–600°C with steel as counterface [14]. They reported a transition in wear mechanisms between 300–400°C from predominantly

abrasive to oxidative wear, which significantly decreased the wear rate. The change in wear mechanism and improvement in wear resistance was attributed to the continuous formation of hard tribolayers (or oxide glaze layers) comprising of Ti and Fe-based oxides. In addition, similar results of improvement in wear resistance of a Ti alloy sliding against steel counterface due to the formation of tribolayer beyond 200°C was reported by Chen et al. [15] The studies mentioned above illustrate that below ~ 300°C, Ti alloys exhibit severe abrasive wear due to the inability of their tribo-oxides to coalesce and form protective tribolayers, while the tribolayer formation occurs only beyond ~ 300° C.

Thus, Ti alloys show different friction and wear mechanisms with changes in temperature due to the formation of tribolayers of different chemical compositions based on the elements present at the tribo-pair interface. Furthermore, the sliding wear of Ti6Al4V based MMC coatings at elevated temperatures is not extensively studied in the literature. The aim of the present study is to understand the tribological behavior of cold sprayed Ti6Al4V and Ti6Al4V-TiC MMC coatings over a temperature range of 25-575°C. The effects of temperature and TiC reinforcement in modifying the friction and wear mechanisms was investigated in detail. Electron channel contrast imaging (ECCI) was done on the wear track cross-sections to characterize the third bodies and subsurface microstructures. In addition, nanoindentation was performed on the third bodies to determine their hardness.

2. Experimental procedure

2.1. Cold spray of coatings and characterization

Irregular Ti6Al4V (Cristal Metals, USA) and spherical TiC (Tekna, Canada) powders of size range: $0 - 45 \ \mu\text{m}$ and $15 - 45 \ \mu\text{m}$ were cold sprayed to deposit Ti6Al4V and Ti6Al4V-TiC MMC coatings. The size distribution, average particle diameter and microstructure of the two feedstock powders were reported in previous publications [9,16]. Coatings were deposited on to mild steel plates (substrate) of dimensions $75 \times 75 \times 3 \ \text{mm}^3$ using a high pressure cold spray system PCS 800 (Plasma Giken, Japan). Prior to deposition the substrates were de-greased and grit blasted using alumina to enhance the coating-substrate bonding. Nitrogen (N₂) was used as propellant gas with pressure and temperature fixed at 4 MPa and 800°C for all the coatings. The standoff distance between the nozzle and substrate was 40 mm and the gun traverse speed was 0.2 m.s⁻¹. The CS system used was equipped with a dual feeding system in which Ti6Al4V and TiC powders were fed into two separate hoppers. During the deposition of Ti6Al4V coatings, Ti6Al4V powder was fed through the nozzle whereas for the deposition of composite coatings both hoppers (i.e. dual feeding system) were activated. The feed rate of the hoppers was adjusted separately to get the required initial metal-ceramic mixture at the inlet of the nozzle.

Deposition efficiency (DE) of the coatings was determined using Eq. 1-3. After CS deposition, coatings were cut perpendicular to the gun traverse direction, cold mounted and polished down till 0.05 µm using colloidal silica suspension. The polished samples were characterized using a scanning electron microscope (SEM) SU3500 (Hitachi, Japan) equipped with a back scattered electron (BSE) detector. BSE images were taken at different locations across the sample cross-section and were uploaded into ImageJ software and the pixels associated with their respective contrast was calculated to determine the porosity and ceramic retention percentages. At least 15 images taken at various locations were analysed to calculate the final average values in each case.

$$DE (\%) = \frac{Weight of the powder deposited (W_d)}{Weight of the powder sprayed (W_s)}$$
(1)

$$W_d = Weight of the sample (after coating) - Weight of the substrate$$
 (2)

$$W_{\rm s} = \frac{Powder \, feedrate \, \times Distance \, travelled \, by \, the \, gun \, on \, the \, sample}{Gun \, traverse \, speed} \tag{3}$$

Microhardness tests were performed on the polished coating surface at random locations using a Vickers microhardness tester (Buehler, USA) at 5 kg load and 15 s holding time. This indentation load was chosen to have the indenter interact with the reinforced carbide particles along with the matrix. An average of 10 indents was done to calculate the average hardness.

2.2. Sliding wear tests

Dry sliding wear tests on the coatings were performed using a unidirectional ball-on disc tribometer Falex ISC-450 (Falex Corporation, USA). The tribometer was equipped with a heating stage (on which the sample was mounted), a temperature controller (Honeywell, USA) and a thermocouple. The thermocouple, temperature controller and the heating stage were connected through a feedback loop wherein the temperature controller collects the data through the thermocouple and activates or de-activates the heating coil placed under the heating stage to maintain the desired static test temperature. WC-Co balls of diameter 6.35 mm were used as counterface for all the tests. The sliding wear tests were conducted in ambient air (~40% RH) under a normal load of 2.5 N (~0.8 GPa initial maximum Hertzian contact stress), which was chosen to be below the ~0.9 GPa yield strength of Ti6Al4V alloy. The sliding speed was fixed at 2.1 cm.s⁻¹ and all the tests were run for a total sliding distance of 200 m to observe the steady-state friction behavior. At least two sliding tests were performed on each coating at every temperature up to 400°C while for test at 575°C only one test was done due to constraints from the experimental setup. Coatings used for the wear tests were polished using 400 to 1200 grit SiC abrasive paper, subsequently using 9 µm, 3 µm, 1 µm diamond suspension solution and finally using 0.05 µm colloidal silica suspension. Sliding wear tests were performed at 25°C, 200°C, 400°C and 575°C static test temperatures for both Ti6Al4V and Ti6Al4V-TiC composite coatings. The above temperature range was choosen since Ti and its alloys are widely used in aerospace applications where they are subjected to elevated temperatures up to ~ 600° C [14]. The coefficient of friction (CoF) for the entire test duration was recorded using the software supplied by the manufacturer. Generally, in a sliding wear test there is a running-in period characterized by either sharp rise or fall in CoF followed by a steady state regime where the CoF converges approximately to a stable value. In the present study, the average CoF was calculated by averaging the CoF after it had reached steady state. The steady state regime was chosen based on the criteria that the CoF did not abruptly rise or fall beyond a value of ± 0.05 for the subsequent sliding distance till the end of the

test. For all the coatings the steady state was found to be lasting for at least 150 m after the initial running-in period. To calculate the wear rate, wear tracks were analyzed under a non-contact optical profilometer (Zygo Corporation, USA) and profiles were generated across the wear tracks. A minimum of 35 profiles were used to calculate the wear area (by integrating the profiles above and below the surface), that was subsequently averaged and multiplied with the total track length to determine the final wear volume (mm³). The wear volume was normalized with load (N) and total sliding distance (m) to calculate the wear rate (mm³.N⁻¹.m⁻¹). Profilometry was performed on the worn counterface using a non-contact optical profilometer to calculate the wear volume. The worn counterface was fitted with a sphere of same radius and was flattened to determine the volume below the unworn region (reference surface). The volume below the unworn region was considered as wear volume.

The worn surface and counterface after the wear tests were characterized using an SEM SU3500 equipped with an energy-dispersive X-ray spectroscope (EDS) and BSE detectors. Raman analysis was also performed on the wear tracks and counterface using an inVia Raman spectrometer (Renishaw, UK) with Ar⁺ laser source of $\lambda = 514.5$ nm. To characterize the subsurface microstructure, ECCI was performed on the wear track cross-sections using a FE-SEM SU8230 (Hitachi, Japan) equipped with a BSE detector. Nanoindentation was performed on the wear track cross-sections using a Triboindentor (Hysitron Corporation, USA) at a peak load of 5 mN using a Berkovich tip. The hardness of the tribolayer and the subsurface was determined from the load-displacement curves using Oliver and Pharr method and each hardness value represents an average of 5 to 10 indents [17]. Indentation was done in a grid pattern on the wear track cross-sections from the unworn coating up to the worn surface. Subsequently, the residual indent images were matched with the corresponding pre-indent images and load-displacement curves to obtain the required features of interest.

3. Results

3.1. Cold spray, characterization and properties of the coatings

Ti6Al4V and Ti6Al4V-TiC coatings were cold sprayed and the DE, porosity, ceramic retention, hardness and thickness values are summarized in Table 1. From Table 1, the DE of composite coating was significantly less than Ti6Al4V coatings mainly due to the non-bonding nature of the ceramic particles. Fig. 1a and 1b show the BSE images of the cross-section of the Ti6Al4V and

Ti6Al4V-TiC composite coatings. In the BSE images of the composite coatings (Fig. 1b), the grey contrast of the spherical and fragmented particles corresponds to the ceramic particles whereas the dark spots correspond to porosity (shown by white arrows in Fig. 1b). BSE images reveal that the Ti6Al4V coatings were dense with negligible porosity whereas the composite coatings showed evidence of some porosity due to the fragmentation of ceramic particles. Ceramic retention in the composite coatings was found to be lower compared to the initial feedstock. Initially, a mixture of 50 vol.% of metal and ceramic were cold sprayed and only 23 vol.% was retained in the coatings that correspond to 46% of ceramic in the initial feedstock. The hardness of the composite coatings was higher compared to pure Ti6Al4V coatings (Table 1) due to the presence of ceramic particles that increase the load-bearing capacity.

Coating	Deposition efficiency (%)	Porosity (%)	Ceramic in the initial feedstock (vol.%)	Ceramic retention (vol.%)	Hardness (HV)	Thickness (mm)
Ti6Al4V	92	0.3 ± 0.1			191 ± 12	1.7 ± 0.04
Ti6Al4V-TiC	42	1.0 ± 0.4	50	23	274 ± 19	1.0 ± 0.06

Table 1: Coating characteristics



Fig. 1: SEM images of (a) Ti6Al4V (b) Ti6A4V-TiC composite coating. White arrows indicate porosity due to ceramic fragmentation.

3.2. Sliding wear of Ti6Al4V and Ti6Al4V-TiC coatings

3.2.1. Friction and wear behavior of the coatings

Figure 2a shows the wear rates of Ti6Al4V and Ti6Al4V-TiC composite coatings at different test temperatures. There was a trend of decreasing wear rate with increased temperature in both the

coatings. At all test temperatures, the composite coating had a lower wear rate compared to Ti6Al4V coating (see Fig. 2a). For example, composite coating exhibited approximately 7 times lower wear rate compared to Ti6Al4V coating at 25°C (room temperature). At higher temperatures (400°C and 575°C), composite coating showed negative wear rate (i.e. material gain on the wear track was recorded due to the transfer from the counterface). For the Ti6Al4V coatings, there was some amount of wear that occurred irrespective of the temperature. However, the difference in the wear rates between Ti6Al4V and the Ti6Al4V-TiC composite coating became less as the test temperature increased (see Fig. 2a).

Figure 2b shows the average CoF of the Ti6Al4V and Ti6Al4V-TiC composite coatings. The average CoF was between 0.25 – 0.45 for both coatings at all test temperatures. To check the hypothesis that the average CoF of Ti6Al4V and composite coatings at all the temperatures was statistically different, an independent samples t-test (i.e. student's t-test) was performed with the probability (p) value to accept null hypothesis (i.e. there is no significant difference between the means) greater than 0.05. T Test results showed that the average CoF was different both in pure Ti6Al4V and composite coatings at all temperatures. For the case of Ti6Al4V coatings, the average CoF increased initially and then decreased with temperature, whereas in the composite coatings the average CoF decreased with an increase in temperature. Above 200°C, the Ti6Al4V-TiC coating exhibits lower CoF than Ti6Al4V coating.



Fig. 2: (a) Wear rate (b) average CoF of the coatings.

3.2.2. Wear track morphology

The worn surface morphology of the coatings after sliding wear tests at different temperatures are shown in Fig. 3 and 4. Worn surface of Ti6Al4V coating at 25°C showed abrasive wear with grooves aligned in the sliding direction (Fig. 3a). These grooves were formed due to the scoring action caused by the entrapped wear debris and the counterface during sliding. Additionally, wear debris were found to be displaced on either side of the track along with few patches inside the track as seen in Fig. 3a. A similar abrasive wear mechanism was found at 200°C (Fig. 3c). However, at 200°C the grooves were found to be deeper (see Fig. 3c) compared to 25°C. At 400°C and 575°C, the wear tracks are characterized by different features compared to 25°C and 200°C. These wear tracks are covered with a compact tribolayer (or glaze layer) indicating a change in wear mechanisms (see Fig. 3e and g). However, apart from tribolayers, abrasive grooves in the sliding direction were still visible on the wear track at 400°C, but to a lesser extent as compared to 200°C. Furthermore, a uniform coverage of the wear track with tribolayer was found at 575°C (Fig. 3g) with no evidence of abrasive wear.

Figure 4 shows the wear track morphology of the composite coatings at different test temperatures. At 25°C, tribolayers were found to be covering the entire wear track (see Fig. 4a). Additionally, minor abrasive grooves formed due to ploughing by the wear debris were found on the wear track, which is a characteristic of abrasive wear. Similarly, at 200°C, tribolayers along with abrasive grooves were found on the wear track, however the area covered by the tribolayer at this temperature was less compared to 25°C (Fig. 4c). At 400°C and 575°C, the wear track was covered with a tribolayer (or glaze layer) similar to Ti6Al4V coatings (Fig. 4e and 4g). Along with tribolayers, a few shallow wear scars along the direction of sliding were found on the wear track. At 400°C, the tribolayer was partially covering the wear track (Fig. 4e), whereas at 575°C, the tribolayer was found to be covering the entire wear track (Fig. 4g).



Fig. 3: SEM images of wear track of Ti6Al4V coatings tested at (a) $25^{\circ}C$ (c) $200^{\circ}C$ (e) $400^{\circ}C$ (g) $575^{\circ}C$ and (b), (d), (f), (h) EDS spectra taken at respective positions on the wear tracks.



Fig. 4: SEM images of wear track of Ti6Al4V-TiC coatings tested at (a) 25° C (c) 200° C (e) 400° C (g) 575° C and (b), (d), (f), (h) EDS spectra taken at respective positions on the wear tracks. Inset images in (a) and (e) show BSE image of tribolayers with arrows showing fragmented TiC particles.

3.2.3. Wear track chemical analysis

EDS analysis on the wear track of Ti6Al4V coating at 25°C and 200°C revealed the presence of Ti, Al, V and O both on the wear track as well as on wear debris (Fig. 3b and 3d). However, the amount of oxygen content in the wear debris and wear track increased with an increase in temperature due to greater oxidation. At 400°C, W was also found along with Ti, Al, V and O in the tribolayers (see Fig. 3f). The W in the tribolayer was due to the transfer of the WC-Co ball material during the wear process. Furthermore, as the test temperature increased to 575°C, W, C and Co were also found along with Ti, Al, V and O on the wear track that was covered with tribolayer (see Fig. 3h).

In contrast, EDS analysis on the wear track of composite coatings showed the presence of different elemental compositions with increasing test temperatures. EDS of the worn surface of the composite coatings at 25°C and 200°C showed the presence of Ti, Al, V, O and C on the tribolayers and wear track (Fig. 4b and 4d). However, the amount of C and O detected was higher on the tribolayers as compared to the wear track (see Fig. 4b and 4d). At 400°C and 575°C, EDS analysis on the tribolayer also revealed the presence of W, and additionally Co (only at 575°C), due to WC-Co ball transfer, along with Ti, Al, V, O and C (Fig. 4f and 4h). Furthermore, C content on the wear track decreased as the temperature increased due to limited exposure of TiC carbide particles as they were masked by the oxide-rich tribolayers. However, at 575°C, the higher C content could be due to the transfer of material from the counterface. Compared to Ti6Al4V coatings, composite coatings showed higher amount of W at 400°C and Co at 575°C correspondingly, indicating more material transfer from the ball.

To identify the tribochemical phases formed, Raman analysis was performed on the wear track and tribolayers at the respective test temperatures and the resultant peaks corresponding to the Raman shift were matched with those reported in the literature [9,18–20]. Raman analysis on the wear debris of the Ti6Al4V coatings tested at 25°C and 200°C showed characteristic peaks at 143 cm⁻¹, 270 cm⁻¹, 405 cm⁻¹ and 609 cm⁻¹ corresponding to rutile and anatase phases of TiO₂ (Fig. 5a and 5b). However, on the wear track at 25°C and 200 °C, no peaks were detected (Fig. 5a and 5b) and was similar to the Raman spectra of as-sprayed coating (Fig. 5a), whereas at 400°C, peaks corresponding to TiO₂ were observed. Tribolayers formed at 400°C revealed peaks at 870 cm⁻¹ and 1583 cm⁻¹ corresponding to WO₃ and C (Fig. 5c) respectively, along with TiO₂. At 575 $^{\circ}$ C, a sharp peak at 885 cm⁻¹ corresponding to CoWO₄ (Fig. 5c) was observed on the tribolayers in addition to TiO₂ and WO₃.



Fig. 5: Raman spectra of Ti6Al4V coatings corresponding to (a) as-sprayed coating, wear track and wear debris at 25° C (b) wear track and wear debris at 200° C (c) tribolayer at 400° C and 575° C

Raman analysis on the tribolayer and wear track of composite coatings revealed distinct phases with an increase in test temperatures. At 25°C and 200°C, tribolayers on the track showed peaks corresponding to TiO₂ along with TiC (Fig. 6b). However, the carbon peaks corresponding to a Raman shift of 1335 cm⁻¹ (D-band) and 1583 cm⁻¹ (G-band) were significantly broad and different compared to the peaks observed in the initial TiC particle (see Fig. 6a). Wear debris on the side of the 25°C wear tracks showed the presence of TiC and TiO₂ phases indicating that the debris were predominantly the oxide particles of Ti and exposed TiC particles (Fig. 6b). At 400°C, the wear debris showed peaks corresponding to TiO₂ whereas the tribolayer showed additional peaks of WO₃ along with TiC (Fig. 6c). The tribolayer covering the wear track at 575°C showed an additional high-intensity sharp peak of CoWO₄ along with TiO₂, TiC and WO₃ peaks (Fig. 6c).



Fig. 6: Raman spectra of (a) as-sprayed composite coating (TiC particle and Ti6Al4V matrix); worn composite coatings corresponding to (b) tribolayers at 25° C, 200° C and wear debris (c) tribolayer at 400° C and 575° C

3.2.4. Counterface analysis

The WC-Co counterfaces after the sliding test were examined under SEM and optical profilometer to characterize the worn morphology and to determine the wear volume. At 25°C and 200°C, there was no observable ball wear. However, at 400°C and 575°C, there was ball wear for counterfaces used in wear tests on both coatings. The volume of material worn from the ball after the sliding wear tests at different temperatures are summarized in Table 2. For tests on both coatings, the wear volume of the counterface was higher at 400°C compared to 575°C.

Counterfaces sliding on the Ti6Al4V coatings for tests at 25°C (Fig. 7a) and 200°C (not shown here) showed the coverage of the contact surface with the wear debris and transfer layer. Raman analysis on the wear debris showed characteristic peaks corresponding to TiO₂ (Fig. 7d) whereas on the transfer layer, no active Raman peak was found indicating the transfer of Ti6Al4V on to the ball. As the test temperature was increased to 400°C, wear scar was found on the counterface with wear debris at its circumference as seen in Fig. 7b. Raman analysis on the wear debris revealed

peaks corresponding to TiO₂ along with WO₃ as shown in Fig. 7e, similar to the phases found on the tribolayer of the wear track. At 575°C, minimal wear was observed on the ball (Fig. 7c) corresponding to the low ball wear volume reported in Table 2. Furthermore, scattered wear debris were found at the contact interface, with few at the point of contact, which could have occurred during the wear process. Raman peaks corresponding to CoWO₄, WO₃ and TiO₂ were observed on the wear debris as shown in Fig. 7f.

Coating	Temperature (°C)	Wear volume (µm ³) x 10 ⁴
Ti6Al4V and Ti6Al4V-TiC	25°C and 200°C	No wear
Ti6Al4V	400°C	92 ± 6
Ti6Al4V-TiC	400°C	120 ± 13
Ti6Al4V	575°C	0.3 ± 0.1
Ti6Al4V-TiC	575°C	1.2 ± 0.5

 Table 2: Wear volume of the counterface



Fig. 7: SEM images of the counterfaces used on Ti6Al4V coatings at (a) 25° C (b) 400° C (c) 575° C; (d), (e) and (f) Raman spectra at corresponding places shown with a box.

Figure 8 shows the SEM images of the counterfaces used for sliding tests on composite coatings at different temperatures. At 25°C (see Fig. 8a) and 200°C (not shown here), formation of transfer film and wear debris at the contact surface was found on the counterface. Raman analysis on the transfer film formed at the contact interface showed characteristic peaks corresponding to TiO₂ and TiC (see Fig. 8d) whereas scans on the wear debris showed peaks corresponding to TiO₂. As the temperature increased to 400°C, the wear scar (see Fig. 8b) showed significant wear of the counterface surrounded by wear debris. Raman analysis on the wear debris revealed characteristic peaks corresponding to TiO₂, TiC and WO₃ (Fig. 8e). At 575°C, the ball wear significantly reduced compared to that of 400°C where no visible wear scar was found in the SEM image. Additionally, wear debris showed an additional phase of CoWO₄ along with TiO₂, TiC and WO₃ (Fig. 8f) similar to Ti6Al4V coatings at 575°C. However, the volume fraction of wear debris found were low compared to previous temperatures indicating less transfer of material from the coating.



Fig. 8: SEM images of the counterfaces used on composite coatings at (a) 25° C (b) 400° C (c) 575° C; (d), (e) and (f) Raman spectra at corresponding places shown with a box.

3.2.5. Subsurface analysis

Figures 9 and 10 show the ECCI of the cross-sectional wear track subsurface of the Ti6Al4V and composite coatings tested at different temperatures. For the Ti6Al4V coatings at 25°C, there were no observable tribolayers, while underneath the wear track, debonding of splats (see Fig. 9a) was seen up to $\sim 600 \ \mu m$ into the coating due to the transfer of stress to the subsurface during wear. The initial static maximum Hertzian contact stress in the case of Ti6Al4V was ~ 0.8 GPa and the maximum stress point was ~ 19 μ m below the contact surface. At 200°C, discontinuous tribolayers were observed due to compaction of wear debris. Below the tribolayers, ultrafine-grained region up to a depth of $\sim 2.5 \,\mu\text{m}$ and recrystallized regions were seen (Fig. 9b and 9c). As the temperature was increased to 400°C and 575°C, a more continuous tribolayer was found on wear track (Fig. 9d and 9g). Tribolayer was of $\sim 10 \pm 1.8 \ \mu\text{m}$ at 400°C and $\sim 3.5 \pm 0.5 \ \mu\text{m}$ at 575°C with cracks inside them (see Fig. 9d and 9g). At 400°C, underneath the tribolayer ultrafine-grained region of ~ 1.5 um thickness was observed followed by a recrystallized region (Fig. 9f). In contrast, at 575°C below the tribolayer ultrafine grains were not seen, however, microcrystalline grains of grain size $\sim 0.8 \ \mu m$ (see Fig. 9h), similar to the grain size in the thermally cycled region (i.e. region away from the wear track) were observed. Additionally, de-bonding of the splats were observed at all test temperatures below the wear track and tribolayers, however the extent of debonding decreased as the temperature increased (Fig. 9).

In the case of the composite coatings, tribolayers were observed at all test temperatures due to the presence of stabilizing TiC particles. At 25°C, tribolayer of ~ $8.2 \pm 2.7 \,\mu\text{m}$ was found on the wear track with no evidence of splat debonding beneath the tribolayers. Composite coatings at 200°C showed discontinuous tribolayers of thickness ~ $7.5 \pm 2.7 \,\mu\text{m}$ as seen in Fig. 10c, that coincides with the SEM image of the wear track morphology. This was followed by ultrafine grains up to a depth of ~ 2 μ m and subsequently by recrystallized region (Fig. 10d). At 400°C, a tribolayer of thickness ~ $2.5 \pm 0.4 \,\mu\text{m}$ (Fig. 10e) was observed and below it similar to Ti6Al4V coatings, ultrafine-grained region was found up to a thickness of ~ 2 μ m with few fragments of carbide particles, followed by recrystallized grains that extended into the coating (see Fig. 10f). At a test temperature of 575°C, a thin tribolayer (Fig. 10g) was observed with recrystallized grains of size ~ 1.5 μ m underneath as seen in Fig. 10h.



Fig. 9: SEM images of tribolayers of Ti6Al4V coatings at (a) 25° C (b) 200° C (d) 400° C (g) 575° C and (c), (e), (f) and (h) ECCI of the subsurface respectively. The red box in the SEM images shows the sub-surface region where the ECCI was done. \bigotimes – indicates sliding direction



Fig. 10: SEM images of tribolayers of composite coatings at (a) 25° C (c) 200° C (e) 400° C (g) 575° C and (b), (d) and (f) ECCI of the subsurface respectively. The red box in the SEM images shows the sub-surface region where the ECCI was done. \bigotimes – indicates sliding direction

Hardness of the tribolayers and the unworn coating at different test temperatures is shown in Fig. 11. Tribolayers were formed on the wear tracks of Ti6Al4V-TiC composite coatings at all test temperatures and in Ti6Al4V coatings at elevated temperatures (> 200° C). Below 200° C, no stable and continuous tribolayers were formed on the wear tracks on the Ti6Al4V coatings, hence the hardness of the subsurface underneath the wear track (i.e. ultra fine grain region) was plotted in Fig. 11. The hardness of the tribolayers for both coatings was higher than the unworn surfaces at all test temperatures. At the lower two test temperatures (25 and 200° C), the tribolayer hardness was similar, with the tribolayers on the composite coating being harder than for Ti6Al4V coatings. This was also true for 400° C and at all three of the lower test temperatures there was sufficient coating wear that fragmented TiC particles were likely incorporated into the tribofilms, leading to higher hardness. Also of note, at 400 °C and 575°C the hardness of tribofilms for both coatings (see

Fig. 11) become harder than at lower temperatures. In both cases, this is due to oxidation and glaze layer formation. At 575°C, the tribolayers of both Ti6Al4V and composite coatings showed similar hardness, which was the highest measured, and was connected to oxide glaze layer formation with minimal wear of the coatings.



Fig. 11: Hardness of the tribolayer and coating at different static test temperatures

4. Discussion

MMCs, used as bulk materials or coatings, are designed to resist mechanical stresses better than the pure metal or alloy used as the matrix. From a tribological standpoint, the hard particles provide load support to resist the hydrostatic and shearing stresses at the interface of a sliding contact. However, recent research has also shown that the hard particles modify local stress states or accelerate the formation of wear-resistant tribofilms [5,9,21,22]. These tribofilms are often formed by oxidative wear of the matrix material but can also incorporate some of the hard particles. The chemical and mechanical mixing that forms these tribofilms may be modified by changes in ambient temperature. In this study of Ti6Al4V and Ti6Al4V-TiC coatings, the hard particles were effective in the manner described above and changes in the tribofilm composition with temperature were linked to changes in the wear mechanisms and wear rates. As will be discussed below, the wear mechanisms and the properties of third bodies were changed by temperature, with the TiC in the composite coating making the processes distinctly different from the pure Ti6Al4V coating.

A schematic of the wear mechanisms occurring in Ti6Al4V and Ti6Al4V-TiC composite coatings at lower (< 200°C) and higher (> 200°C) test temperatures is shown in Fig. 12. Up to 200°C, Ti6Al4V coatings showed high wear rates compared to composite coatings due to abrasive wear caused by the trapped wear debris at the interface (see Fig. 3a and 3c). The wear debris were composed of oxides of Ti that acted like abrasive particles during sliding wear, resulting in material ploughing until getting displaced to either side of the wear track (Fig. 12a and Fig. 3a). Additionally, the formation of an adhesive transfer film of Ti6Al4V on the counterface was observed at both the test temperatures (Fig. 7a), due to the low d-bond character of Ti and its alloys [9,23]. Along with the transfer film, wear debris were found around the transfer film (on the counterface). At 25°C, no tribolayers were formed, whereas at 200°C, minor evidence of tribolayer formation due to compaction of wear debris was observed on the wear track. However, the tribolayers were non-protective and their poor resistance to the localized deformation induced by the counterface during sliding led to the transfer of stress to the subsurface [9]. This resulted in debonding of particle-particle interfaces underneath the wear track (see Fig. 9a). Similar evidence of loose, non-adherent and non-protective tribolayer formation, which led to high wear of Ti and its alloys up to 200°C was reported by Straffelini et al., Cui et al., Molinari et al., Chen et al. and Mao et al. [13,15,24–26] Furthermore, the high affinity of Ti and its alloys towards oxygen results in the early formation of oxides. However, its high "Pilling-Bedworth ratio" results in internal stresses in the oxide layers (that are further enhanced during the wear process) and the mismatch for the coefficient of thermal expansion between the oxide layer and subsurface material result in their breakage [13,25,27]. These fragmented oxide debris further act as abrasive particles and accelerate the wear process resulting in high wear rates.

Reinforcement of TiC ceramic particles into Ti6Al4V matrix improved the wear resistance. This was due to the formation of tribolayers on the wear track (Fig. 12b) and could also be due to the improvement in the hardness of the coating making it resistant to plastic deformation. Previously, Alidokht et al. reported the formation of tribolayers around TiC particles in Ti-TiC MMC coatings and illustrated that their reinforcement in the metallic matrix offered resistance to plastic deformation during sliding [9]. Candel et al. also found that the presence of TiC particles in Ti6Al4V matrix resulted in the formation of tribolayers that decreased the friction and wear of the MMC coatings [28]. Similar tribolayers composed of fragmented oxide debris of Ti and TiC particles were found on the wear track of composite coatings at both 25°C and 200°C (see Fig. 4a).

Formation of tribolayers on the wear track offered resistance to localized plastic shearing and thus improved the wear resistance of the coatings. Studies performed by Venkataraman et al. illustrated that the presence of hard tribolayer resulted in mild wear with lower wear rates in Al based composites compared to the pure Al where there was no evidence of tribolayer [3]. Similar improvements in wear resistance due to the formation of tribolayers were reported by Torgerson et al., Shockley et al., Alidokht et al., Zhang et al. and Cui et al. [6,8,11,13,29] Thus, reinforcing TiC ceramic particles in the Ti6Al4V matrix facilitated the formation of the protective tribolayers and their presence in the form of fragments in the tribolayers (Fig. 12b) offered resistance to localized plastic deformation during sliding wear, leading to improved wear resistance [9,28].

Beyond 200°C, oxide glaze layers formed on the wear tracks of both coatings were composed of TiO₂, WO₃ and CoWO₄ (only at 575°C) and significantly improved the wear resistance compared to lower temperatures. At elevated temperatures (> 200°C), a similar mechanism for the formation of tribolayers in Ti and its alloys were reported by Cui et al., Wang et al. and Chen et al. [13–15] Furthermore, composite coating showed almost no wear in addition to material gain on the wear track by material transfer from the counterface, which resulted in negative wear rates. Additionally, the glaze layers of composite coatings showed relatively uniform coverage compared to Ti6Al4V coatings. The formation of WO₃ and CoWO₄ (only at 575°C) phases in the glaze layers was due to the oxidation and wear of the WC-Co counterface, resulting in the transfer of material onto the wear track (see Fig. 7 and 8). Prior studies have shown that WC-Co composites exhibit high wear resistance up to 200°C, while beyond this temperature, oxidation results in significant wear along with the formation of oxides of W and Co [19,20,30]. This was consistent with the observations made in this study, where ball wear occurred only beyond 200°C and hence no oxides of W and Co were detected on the wear tracks at 25°C and 200°C (Fig. 5 and 6).

The tribolayer hardness of composite coatings was found to be higher compared to Ti6Al4V due to the presence of fragmented TiC particles; however, at 575°C, tribo-oxidation lead to limited pulling out and subsequent fragmentation resulting in similar hardness. Ultrafine recrystallized grains were found beneath the glaze layers in both coatings up to 400°C. The formation of ultrafine region was due to dynamic recrystallization that occurs due to stress transfer during the wear process. In contrast, at 575°C, micron sized grains were observed underneath the glaze layers that were similar to the thermally cycled region (i.e. region away from the wear track) indicating

recrystallization had occurred prior to the start of wear tests at this temperature. However, the average grain size beneath the glaze layer was higher in the composite coatings compared to Ti6Al4V indicating these coatings were more resistant to subsurface deformation.

The average CoF in Ti6Al4V and Ti6Al4V-TiC coatings decreased with an increase in temperature. Below 200°C, the average CoF of Ti6Al4V and composite coatings did not vary significantly with temperature. However, as the test temperature was increased to 400°C and 575°C, the formation of new phases changed the tribo-chemistry resulting in a decrease in CoF. Erdemir et al. proposed that higher ionic potential of the oxides result in greater screening of the cations by the anions and prevent their additional reactivity with the other available atoms which reduces the adhesion and thereby the CoF [31,32]. At 400°C, WO₃ was detected on the wear track of both coatings, which has a high ionic potential of ~ 8.3 and exhibits CoF within a range of 0.25 -0.33 [32]. However, the decrease in CoF was more significant in the case of composite coatings due to a higher amount of W transferred from the counterface with more uniform coverage over the entire wear track (Fig. 4d). Additionally, abrasive grooves were observed on the wear track of Ti6Al4V coatings even at 400°C which could have led to a higher CoF despite the presence of WO_3 (see Fig. 3c). Similar trends of a decrease in CoF due to the presence of WO_3 on the wear track was reported by Geng et al. and Mi et al. [19,20,30]. With a further increase in test temperature to 575°C, CoWO4 was additionally found on the wear track along with TiO2 and WO3. Previously, Erdemir et al. illustrated that the large differences in the ionic potentials of the chemical species result in lower CoF [32]. Their observations revealed that CoWO4 (that exists as CoO-WO₃) had an ionic potential difference of 6.9 and exhibited CoF in the range of 0.2 - 0.47 at temperatures between 573 K - 1072 K. This coincides with the low CoF values observed in our study where the formation of CoWO₄ significantly decreased the CoF in both coatings at 575°C.

Thus, reinforcement of TiC ceramic particles favoured the formation of wear resistant tribolayers at lower temperatures unlike the pure Ti6Al4V coating, whereas, tribo-oxidation led to the formation of oxide glaze layers comprising of WO₃ and CoWO₄ at elevated temperatures that reduced the CoF and wear for both coatings.



Fig. 12: Schematic showing the third bodies chemical compositions and wear mechanisms in (a) Ti6Al4V and (b) Ti6Al4V-TiC MMC coatings at different static test temperatures

5. Conclusions

Ti6Al4V and Ti6Al4V-TiC MMC coatings were deposited using CS process and their high temperature sliding friction, wear behavior, tribochemical phase formation and wear mechanisms were studied in detail. Based on the results following conclusions were drawn:

- At lower test temperatures, abrasive grooves along the sliding direction were observed in Ti6Al4V coatings whereas tribolayers were found on the wear track of composite coatings. This indicated that the reinforced TiC ceramic particles favoured the formation of tribolayers that resist localized deformation and result in low wear of composite coatings compared to Ti6Al4V.
- 2. The wear rate of Ti6Al4V and Ti6Al4V-TiC composite coatings was lower at elevated temperatures. This was due to a transition in wear mechanisms to tribo-oxidation that led to oxide glaze layers composed of TiO₂, WO₃ and CoWO₄ phases on the wear track. Glaze layers were partially formed by wear and oxidation of the WC-Co counterface followed by adhesive transfer to the wear track. Even with this material transfer, the Ti6Al4V coatings showed a measurable positive wear rate. The composite coating, with reinforced TiC, showed a negative wear rate, indicating minimal true wear of the coating and mostly glaze layer formation.
- Interparticle de-bonding was seen underneath the tribolayers of Ti6Al4V coatings due to the transfer of stress to the subsurface, while the formation of tribolayers comprising of oxides of Ti and TiC particles increased their load bearing capacity and prevented debonding in composite coatings.
- 4. Average CoF decreased with increasing temperature in both coatings. Below 200°C, the CoF of Ti6Al4V and Ti6Al4V-TiC composite coatings did not vary significantly with temperature. However, at elevated temperatures, formation of glaze layers composed of oxides of W and Co reduced the adhesion and decreased the CoF.

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