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1 **Recent advances in self-healing materials for food packaging**

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12 **Abstract**

13 Food packaging materials are often inevitably and imperceptibly damaged during the
14 transportation, handling, and storage, and the disruption of their integrity poses a
15 challenge to food preservation. Food packaging materials with self-healing capability
16 can automatically repair the damaged areas and reconstruct original properties to avoid
17 degradation of food quality and loss of nutrients. Various self-healing materials based
18 on dynamic covalent bonds and/or dynamic non-covalent interactions have been
19 developed and applied in food packaging in the forms of films and coatings at
20 laboratory scale, and more efforts are required for the commercialization of these novel
21 smart packaging materials. This is the first review to summarize the recent progress in
22 the preparation of self-healing packaging materials through different mechanisms,
23 compare the self-healing efficiency under different conditions, and highlight the
24 potential applications of self-healing films and coatings with recoverable mechanical
25 and barrier properties and other functionalities (e.g. antimicrobial and anti-fogging
26 capacities). Finally, the future opportunities and challenges of applying self-healing
27 materials in food packaging are described.

28 **Keywords:** self-healing, mechanisms, food packaging, functional materials

29

30 **1 Introduction**

31 Food packaging is imperative and a reality of modern living to maintain food quality
32 and safety by protecting food products from influences of the external environment.
33 However, food packaging materials are susceptible to damage during transportation,
34 handling, and storage, posing a huge potential threat to food safety and challenging the
35 future development of the food industry.¹ Therefore, it is necessary to continuously
36 monitor damage and maintain the integrity of packaging material, but the repairing,
37 recoating and replacing are costly and time-consuming.² In addition, pinholes and
38 microcracks in packaging materials that are not observable to the human eye make it
39 difficult to assess the integrity of packaging materials, affecting their barrier properties
40 and further impairing the shelf life of foods.^{3,4}

41 As an innovative type of materials developed in recent years, self-healing materials are
42 naturally inspired by the ability of living systems such as skin, tissue, and bone to repair
43 themselves, recognizing and repairing damage to re-establish their original properties,
44 while extending the function of the material and maintaining its integrity, consistency
45 and effectiveness.⁵⁻⁷ The first attempts at self-healing materials were found in buildings
46 constructed by the ancient Romans 2000 years ago, resisting microcracks in concrete
47 by in situ crystallization of a durable mineral.⁸ Laminates of aluminum foil and
48 polyvinyl acetate fabricated by Malinskii et al. were the first study on self-healing
49 materials.⁹ In 2007, the first international conference on self-healing materials,
50 organized by Sybrand van der Zwaag, provided a concept of self-healing materials with
51 potential applications to all materials.¹⁰ This discussion contributed significantly to the
52 exchange of knowledge and accelerated the development of self-healing materials.
53 Currently, the self-healing ability of packaging materials can be categorized into two
54 types: intrinsic and extrinsic.¹¹ The intrinsic self-healing process is performed based on
55 the reconstruction of chemical bonding of the matrix itself, including dynamic non-
56 covalent interactions such as hydrogen bonds,¹² host-guest interaction,¹³ electrostatic
57 interaction,¹⁴ π - π stacking,¹⁵ and dynamic covalent interactions such as imine bonds,¹⁶
58 acylhydrazone bonds,¹⁷ Diels-Alder reaction,¹⁸ retro-Diels-Alder reactions,¹⁹ etc. The

59 extrinsic self-healing process in food packaging is achieved by capillary action in
60 microcapsules, which repairs damage by releasing the healing agent from the
61 incorporated container into the matrix.¹ Current research directions suggest that the
62 intrinsic self-healing mechanism is more suitable for food packaging due to the absence
63 of external healing agents, avoiding the complex design and extra cost of self-healing
64 systems in the form of microcapsules.²⁰⁻²³ According to the search results of Web of
65 Science, 12,438 research papers on self-healing materials have been published from
66 2016 to 2022, of which 192 papers (1.54%) mentioned the potential applications in the
67 food industry and 12 studies (0.096%) involved real food products. Currently, the real
68 applications of self-healing food packaging materials are mainly restricted by the cost
69 and complexity of materials involved in the manufacturing process and the fact that the
70 required conditions to stimulate the self-healing process, such as high temperature
71 and/or high humidity, may lead to the deterioration of food quality. Overall, self-healing
72 materials for food packaging applications are a relatively new topic but have great
73 potential. Hitherto, different forms of self-healing materials have been reported, such
74 as fibers,^{24,25} hydrogels,²⁶⁻²⁸ microcapsules,²² films,²⁹⁻³¹ and coatings.^{14,32} Among them,
75 self-healable fibers with high aspect ratios are commonly applied for producing
76 batteries.³³ Self-healing hydrogels are 3D networks of hydrophilic polymers and can be
77 developed for human motion detection,³⁴ wound dressing,³⁵ and battery applications,³⁶
78 while self-healable microcapsules contain the encapsulated healing agents in the wall
79 material with a spherical shape and have been applied for dental,³⁷ aerospace coatings,³⁸
80 and paints.³⁹ Films and coatings are the main forms of self-healing materials reported
81 in recent years for food packaging. They can share similar formulations, but the
82 coatings are directly applied on the surface of food and the films are prepared
83 independently by casting or freeze drying. This review is the first attempt to summarize
84 the recently developed self-healing materials and their potential applications in food
85 packaging. The future opportunities, challenges and research directions are described.

86 **2 Self-healing mechanisms**

87 The design of self-healing materials is intimately associated with their applications to

88 the form of food packaging, and is generally governed by the following principles: (1)
89 the applied materials with self-healing ability should be economic and non-toxic, and
90 have an insignificant effect on the taste and texture of the food; (2) the self-healing
91 reactions should occur autonomously or be in response to an external stimulus; and (3)
92 the self-healing reactions should simultaneously repair both morphological and
93 functional properties of the packaging materials.^{40,41} Both dynamic non-covalent and
94 covalent interactions can be used to fabricate self-healing materials for food packaging.
95 Dynamic non-covalent bonds can be effectively repaired with little or no external
96 energy at room temperature, but they are mechanically weaker than dynamic covalent
97 bonds due to their weak intermolecular forces and reversible interactions.^{40,42,43}
98 Simultaneous and continuous reversible assembly and decomposition of dynamic
99 covalent bonding systems usually occur in response to light, heat, and force, and the
100 produced self-healing materials can exhibit high stability even after the removal of
101 external stimuli.^{44,45}

102

103 **2.1 Dynamic non-covalent bonds**

104 **2.1.1 Hydrogen bond**

105 Hydrogen bonds are formed between hydrogen atoms and highly electronegative atoms,
106 such as nitrogen, oxygen, and fluorine. Several studies have reported the fabrication of
107 self-healing materials from chitosan,⁴⁶ starch,⁴⁷ carrageenan,⁴⁸ and protein⁴⁹ by
108 hydrogen bonds. The self-healing efficiency is related to the amount of hydrogen bonds
109 in polymer matrix. For instance, poly(vinyl alcohol) (PVA) is abundant in hydroxyl
110 moieties that contribute to a good gelling and film-forming capacity with self-healing
111 ability.⁵⁰ Liu et al.¹² optimized the weight ratio of PVA and cellulose nanocrystals (CNC)
112 to 60:1 to maximize the self-healing efficiency (37.03%) of PVA-based hydrogels. The
113 inadequate amount of free hydroxyl groups at low PVA content led to weak hydrogen
114 bonding ability, and the excessive PVA reduced the mobility of molecules and
115 decreased the self-healing efficiency.⁵¹ However, hydrogen bonds are relatively weak,
116 so the resultant self-healing materials usually exhibit weak mechanical properties. This

117 could be improved by the addition of cross-linking agents. For example, increasing the
118 concentration of carboxymethyl chitosan (CMCS) from 0% to 1.3% enhanced the
119 tensile strength of cross-linked hydrogels of aluminum ions (Al^{3+}) and polyacrylic acid
120 (PAA) from 83.1 kPa to 149.2 kPa and tensile strain from 1673% to 1722% with no
121 disruption of self-healing ability of hydrogels (92.9% of stress and 98.8% of strain after
122 24 h of healing at 25 °C).⁵² For hydrogels formed by cross-linked networks of 2-ureido-
123 4-pyrimidone monomer, acrylamide and acrylic acid, the increase in acrylic acid
124 content from 10% to 20% improved the tensile strength and elastic modulus from 3.02
125 MPa to 6.54 MPa and 0.53 MPa to 2.93 MPa, respectively.⁵³ It was worth noting that
126 the cross-linking resulted in the enhanced mechanical properties, but additional energy
127 was also required for the breakage and re-construction of polymer networks, leading to
128 the weakened self-healing ability. Recently, a strategy of forming multiple hydrogen
129 bonds has been applied to produce strong self-healing materials. For instance, Zhu et
130 al.⁵⁴ were inspired by nacre and mussels and fabricated an ultrarobust self-healing
131 composite material by establishing quadruple hydrogen bonds between dopamine-
132 modified graphene oxide and polyurethane prepolymer, which showed 90% self-
133 healing efficiency at 25 °C in 1 h. Moreover, the synergistic effects of hydrogen bonds
134 with other dynamic interactions such as acylhydrazone bonds^{55,56} and Diels-Alder
135 reaction^{57,58} could also contribute to the enhancement of mechanical properties while
136 maintaining the self-healing ability.

137 **2.1.2 Hydrophobic interaction**

138 Hydrophobic interaction has been used predominantly as an alternative strategy for the
139 design and fabrication of self-healing hydrogels. For example, hydrophobic monomers
140 were incorporated with hydrophilic monomers to form a hydrogel network through the
141 micellar polymerization technique.⁵⁹ Tang et al.⁶⁰ synthesized amphiphilic
142 metallopolymers to interact with hydrophobic metal coordination units, resulting in
143 micelle aggregation followed by the formation of a stimuli-responsive hydrogel. Zhou
144 et al.⁶¹ proposed that $(\text{NH}_4)_2\text{SO}_4$ solution could improve the hydrophobic interactions
145 among phenylboronic acid-modified gelatin, catechol-modified carboxymethyl

146 chitosan, 3,5-dinitrosalicylic acid, and Eu^{3+} ions, and temporary cross-linked regions
147 could form to stabilize the shape of hydrogels. A simple one-pot method to fabricate
148 dual physically cross-linked hydrogels based on the hydrophobic association and ionic
149 interaction between sodium alginate (SA) and polyacrylamide (PAM) was reported by
150 Zheng et al.⁶² However, although the increased concentration of SA enhanced the
151 fracture stress of composite hydrogels, it also restricted the movement of PAM segment
152 and resulted in the unsatisfactory self-healing ability. Deng et al.⁶³ used the hydrophobic
153 interaction between the sodium dodecyl sulfate (SDS) micelles and the alkyl groups of
154 stearyl methacrylate (SMA) in PAM to construct self-healing hydrogels, which could
155 quickly recover from stretching within 2 min at room temperature. However, the
156 addition of carrageenan to these double cross-linked hydrogels improved the
157 mechanical properties but decreased the self-healing efficiency due to the restricted
158 movement of the molecular chains.

159 **2.1.3 Electrostatic interaction**

160 Reversible electrostatic interactions can be initiated between the charged polymers and
161 ions,⁵² polyelectrolytes,⁶⁴ polyampholytes,⁶⁵ and zwitterionic fusions,⁶⁶ resulting in the
162 formation of self-healing films,³¹ hydrogels,¹² and aerogels.⁶⁷ For instance, a
163 polyelectrolyte multilayer (PEM) film with self-healing ability was constructed from
164 polydopamine and SA by layer-by-layer (LBL) assembly.⁶⁸ Zhao et al.⁶⁷ manufactured
165 an aerogel based on the electrostatic interactions between $-\text{NH}_3^+$ of chitosan and $-\text{COO}^-$
166 of itaconic acid. A similar composite aerogel of chitosan and itaconic acid was
167 constructed by the freeze-drying process, where the structure and mechanical properties
168 of the cut aerogel could recover within 30 s at room temperature via wetting treatment.⁶⁹
169 Maiz-Fernández et al.⁷⁰ fabricated a hydrogel by forming electrostatic interactions
170 between chitosan and hyaluronic acid, which showed complete self-healing after 15
171 min at room conditions. Similarly, electrostatic interactions were also combined with
172 other cross-linking strategies to produce self-healing hydrogels. Fang et al.⁷¹ used β -
173 cyclodextrin (β -CD) and carboxymethyl cellulose (CMC) to prepare pH-sensitive
174 hydrogels based on host-guest and electrostatic interactions. The existence of cationic

175 β -CD effectively increased the self-healing efficiency from 17% to 55% after 6 h,
176 indicating the important role of electrostatic interaction in self-healing process. The
177 antibacterial hydrogels with self-healing ability under acidic conditions were developed
178 from aluminum ions, CMCS, and PAA.⁵² During the self-healing process, hydrogen
179 bonds were firstly reconstructed and Al^{3+} ions gradually diffused to the fracture
180 interface simultaneously with chitosan, promoting the reconstruction of coordination
181 bonds and electrostatic interactions.

182 **2.1.4 Host-guest interaction**

183 Host-guest interaction is formed when the guest moieties are physically inserted and
184 combined with the host moieties. Common host molecules include CD, crown ethers,
185 cucurbiturils, cuproaromatics, etc.,^{72,73} where CD is widely used because it is non-toxic,
186 easy to synthesize, and has a hydrophobic cavity to bind hydrophobic guests.²⁸ Yang et
187 al.²⁸ investigated the host-guest interaction between l-menthol and β -CD. The damaged
188 coating swelled under the stimulus of water, which prompted the contact of free groups
189 and self-healed within 20 min. Peng et al.⁷⁴ disclosed the principle of self-healing by
190 host-guest interaction: upon contact with moisture, the broken interfaces swell and
191 contact with each other, so the free host-guest molecules are reintegrated. A self-healing
192 coating with UV shielding property was produced from β -CD, TiO_2 and 2-hydroxyethyl
193 methacrylate and applied on the surface of food packaging materials to prevent from
194 oxidation. Mohamadhoseini et al.⁷³ provided an in-depth comparison of the interactions
195 between different guest molecules and CD and their effects on the self-healing
196 efficiency and stimulus responsiveness. The association constants of CD and guest
197 molecules determined the self-healing efficiency. Compared with azobenzene (1×10^4),
198 ferrocene (1.7×10^4), and cholic acid (4×10^3), adamantane (AD) possessed the highest
199 association constant (3.5×10^4). β -CD and adamantane were then grafted on the
200 backbone of alginate as host and guest groups to construct self-healing hydrogels, and
201 the self-healing process completed within 5 s, demonstrating the strong interactions
202 between β -CD and adamantane.⁷⁵ Xuan et al.⁷⁶ contrasted the properties of PAA/poly
203 (ethylenimine) (PEI) multilayer films and PAA-AD/PEI- β -CD LBL films, which were

204 formed by electrostatic interaction and host-guest interaction, respectively. The PAA-
205 AD/PEI- β -CD films showed better transparency, mechanical properties, and stability in
206 different pH environments.

207 **2.2. Dynamic covalent bonds**

208 **2.2.1 Imine bond**

209 Imine bond is a reversible covalent bond prepared by the facile condensation of an
210 aldehyde or a ketone with primary amines.⁷⁷ Liu et al.²⁷ used dialdehyde debranched
211 starch to react with chitosan to form Schiff base hydrogels with a gelation time of less
212 than 30 s. The mechanical properties of produced hydrogels could be controlled by
213 varying the molar ratio of amino group and aldehyde group and reaction temperature.
214 Natural vanillin and chitosan were selected to fabricate the self-healing hydrogels,¹⁶
215 and a balance between mechanical properties and self-healing ability was reached when
216 the molecular ratio of aldehyde and amino groups was about 7:1. Further increase in
217 vanillin content resulted in the fixed linkage of chitosan by massive hydrogen bonds,
218 which reduced the imine bonds and self-healing capability. Pettignano et al.⁷⁸ disclosed
219 that the gelation process and self-healing efficiency of hydrogels were related to the
220 concentration, ratio, and type of biopolymers. Specifically, the increase in the weight
221 ratio of oxidized SA and gelatin from 1:1 to 2:1 accelerated the self-healing process and
222 resulted in the high resistance to stretch, while the increased concentration of
223 biopolymers from 5% to 15% led to a more regular and cylindrical hydrogel. At the
224 same time, since type B gelatin had a lower isoelectric point (pI) (4.7-5.2) than type A
225 gelatin (7.0-9.0), type B gelatin could form a more stable hybrid network with faster
226 self-healing process at pH 7.2. Yang et al.⁷⁹ prepared a self-healing hydrogel with a
227 remarkable antioxidant activity using ϵ -poly(L-lysine) carbon dot (PL-CD) and
228 oxidized dextran (ODA). Increasing PL-CD content led to a shorter self-healing process
229 within 32 s, which was ascribed to the prompted Schiff base reaction by higher -NH₂
230 concentration. Wang et al.⁸⁰ synthesized a hydrogel from ABA triblock copolymer and
231 polyethylenimine (PEI), which showed a self-healing process within 30 min and
232 antimicrobial activity against both *Escherichia coli* and *S. aureus*, indicating its

233 potential application as an active food packaging material.

234 **2.2.2 Acylhydrazone bond**

235 Acylhydrazone bond is formed by the reaction of aldehyde and hydrazine groups under
236 mild acidic environment (pH 4-7) or high-temperature catalytic environment,⁸¹ which
237 has similar structure but better stability than imine bond.⁸² Chen et al.¹⁷ produced the
238 pectin achlydrazide based hydrogels with labile phenolic bond and acylhydrazone bond.
239 The pore size of hydrogel was negatively correlated with the concentration of the
240 aldehyde group, and the self-healing capability was limited in the slightly acidic
241 environment (pH 4-6). Qiao et al.⁵⁵ used dialdehyde-terminated polyethylene glycol
242 (PEG-CHO) and adipic dihydrazide-modified alginate (ALG-ADH) to produce self-
243 healing hydrogels. The self-healing process was observed after 6 h without any external
244 stimulation, which was attributed to the dynamic characteristics of acylhydrazone
245 bonds and the reversibility of multiple hydrogen bonds. Huang et al.⁸³ constructed an
246 antimicrobial hydrogel using oxidized cellulose acetoacetate to react with
247 carboxymethyl chitosan and oxalyl dihydrazide, and the enamine linkages and
248 acylhydrazone bonds enabled the self-healing process in 30 min. Zhu et al.⁸⁴ illustrated
249 the effect of metal coordination bond on the self-healing efficiency by the incorporation
250 of Fe₃O₄. The composite hydrogel made of PAM, oxidized alginate, and cellulose
251 nanocrystals showed a fast closing of cracks within 10 min, while the addition of Fe₃O₄
252 accelerated the healing speed to few seconds. Moreover, compared to the Fe₃O₄-free
253 hydrogel, the ionic hydrogels did not exhibit any obvious loss of water after one week,
254 which demonstrated the better stability and ensured self-healing capability.

255 **2.2.3 Diels-Alder (DA) reaction**

256 DA reaction is a click electrocyclic reaction between a conjugated diene and a
257 dienophile with the advantages of high yield, high reaction selectivity, no side reactions,
258 etc.⁸⁵⁻⁸⁷ However, the self-healing process triggered by DA reaction is usually time-
259 consuming and requires a heating condition. Cai et al.⁸⁸ reported the formation of self-
260 healing coatings/films by the DA reaction between poly(lactic acid)-block-poly(2,5-
261 furyl dimethylbutyrate) and bis(maleimide) triethylene glycol. The material with low

262 cross-linking density showed only 50% self-healing efficiency after 5 days at room
263 temperature, while the films with optimized cross-linking density exhibited 96.3% self-
264 healing efficiency in 5 min at a moderate heating condition (30 °C). Recently, DA
265 reaction was commonly combined with other reversible interactions to fabricate the
266 materials with fast self-healing ability. Li et al.¹⁸ prepared the composite hydrogels from
267 furan-modified pectin and maleimide-modified chitosan, where the first stage of the
268 self-healing process was due to the electrostatic interaction, and the network of
269 hydrogels was formed by DA reactions in the second stage. A stiff UV-curable self-
270 healing coating based on DA reaction and multiple hydrogen bonds was disclosed by
271 Liu et al.⁵⁷ DA intermediate was formed by *N,N'*-(4, 4'-methylenediphenyl)
272 bismaleimide (BMI) and furfuryl alcohol (FA), followed by the reaction with
273 methacrylic acid isocyanate (IEM) to obtain the modified DA monomer. The self-
274 healing coating was formulated with polyurethane (PU) pre-polymers, modified
275 monomer, and initiator. By modifying the molecular weight of PU and the ratio of DA
276 monomer to PU, the self-healing efficiency could be changed from 33% to 100% at
277 120 °C in 10 min.

278 **2.2.4 Disulfide bond**

279 Disulfide bond is formed with two mercaptans through either an exchange reaction or
280 a metathesis reaction.⁸⁹ Disulfide exchange reaction is commonly used to produce self-
281 healing hydrogels, since metathesis reaction requires a higher working temperature.⁹⁰
282 The exchange kinetics of disulfide bonds can be promoted under alkaline conditions
283 and inhibited under acidic conditions.⁹¹ Chang et al.⁹² fabricated a transparent PU film
284 by dissolving PU and 4-4-aminophenyl disulfide in *N,N*-dimethylformamide, but the
285 self-healing behavior was only observed when the temperature was above 60 °C. Jian
286 et al.⁹⁰ manufactured a self-healing PU film based on disulfide metathesis reaction, in
287 which 2-hydroxyethyl disulfide reacted with polytetramethylene ether glycol and 3-
288 isocyanatomethyl-3,5,5-trimethylcyclohexyl isocyanate. The self-healing behavior was
289 observed after 12 h at 60 °C. The double cross-linking strategy combining disulfide
290 bonds was also reported. For instance, Xia et al.⁹³ proposed an albumin-based double

291 cross-linked hydrogel with self-healing ability at 25 °C. The exchange and
292 rearrangement of disulfide bonds of albumin were triggered by glutathione, and tetrakis
293 (hydroxymethyl) phosphonium sulfate was grafted to the amino groups of BSA via a
294 Mannich-type reaction to form a double-cross-linked network. Wang et al.⁹⁴ prepared
295 the self-healing hydrogels containing acylhydrazone bonds and disulfide bonds by
296 cross-linking dialdehyde carboxymethyl cellulose (DCMC) and 3,3'-
297 dithiobis(propionohydrazide), and investigated the effect of cellulose nanofibrils (CNF)
298 on the properties of hydrogels. Since the acylhydrazone bonds predominated in the
299 hydrogels, the increase in solid content and oxidation degree of DCMC shortened the
300 gelation period and the healing of cracks happened in 1 h without any external force.
301 Mredha et al.⁹⁵ constructed a multifunctional poly(disulfide)-based hydrogel by
302 copolymerizing 2,3-dimercapto-1-propanol and meso-2,3-dimercaptosuccinic acid.
303 The obtained hydrogels showed a fast self-healing behavior within 5 s at 25 °C in air or
304 under water, which was due to the synergistic effect of dynamic disulfide bonds and the
305 high density of hydrogen bonds and ionic bonds.

306 **3 Potential applications of self-healing materials in food packaging**

307 Self-healing materials provide a new way to design smart packaging.⁹⁶ Recently
308 reported self-healing materials for food packaging are summarized in Table 1 and are
309 usually in the forms of films and coatings. The self-healing process observed in films
310 and coatings were summarized into Figure 1.

311

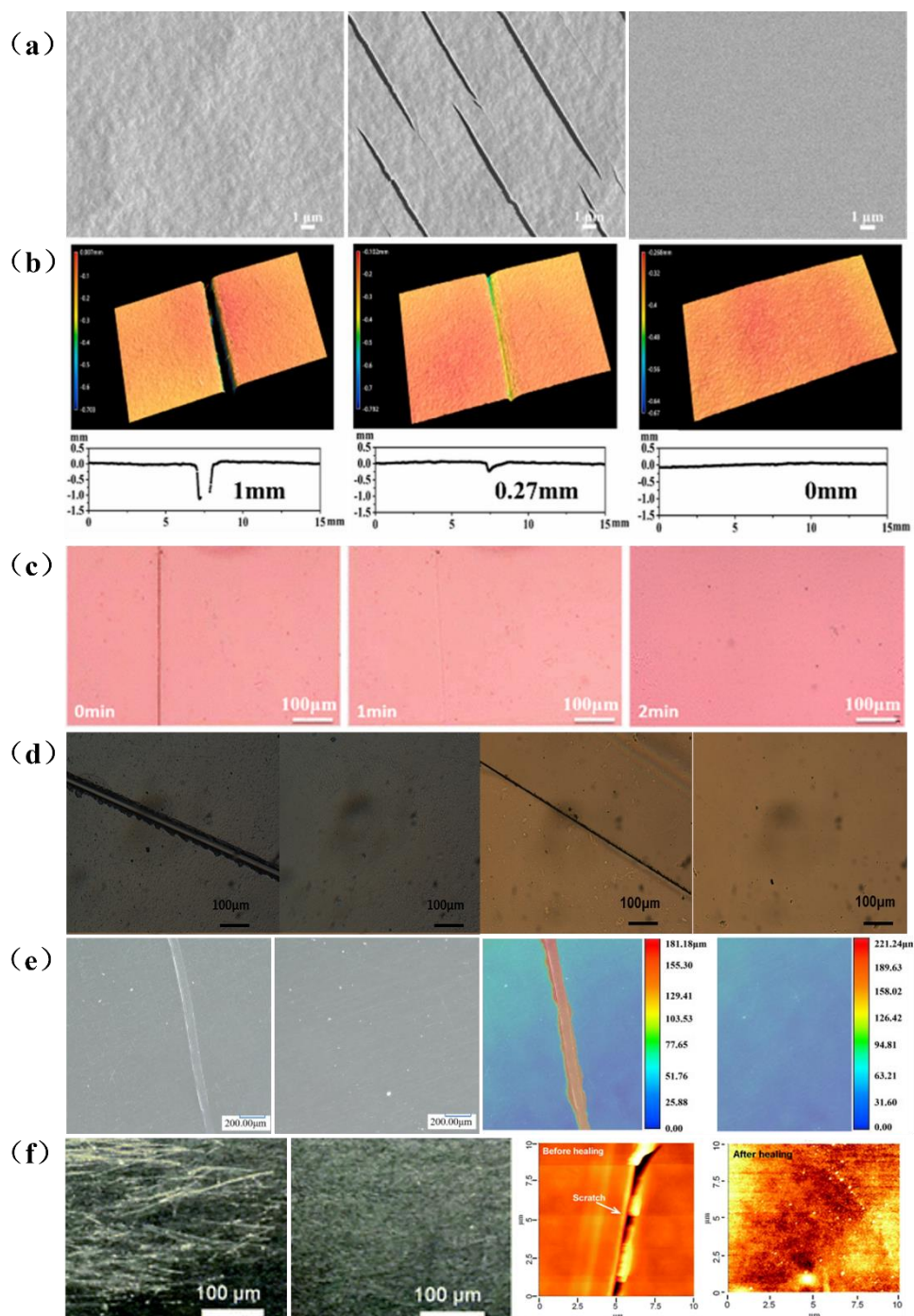
312 Table 1. Recently reported self-healing materials for food packaging applications.

Types	Compositions	Mechanisms	Self-healing efficiency	Recovery rate of functionalities	Refs
Films	CS-CA-CC	Hydrogen bonds	RT for 60 s	E: 56%; TS: 72%	97
	AMCS-ADA	Imine bonds	RT within 2 min	T: 82.1%	29
	MCNC-PVA	Hydrogen bonds	RT within 20 min	TS: 69.54%	98
	Aromatic disulfide based TPU	Disulfide bonds	/	/	30
	β -CD-PEI-adamantane-PAA	Host-guest interaction	RT within 30 min	E: 87.2%	76
	β -CD-PEI-PAA-Fc	Host-guest interaction; hydrogen bonds	RT within 30 min	E: 97.65%; TS: 74.67%	99
	PDA-SA-CAP-CS	Electrostatic interactions	RT for 24 h	/	68
	CMC-CS	Electrostatic interactions	RT for 10 min	/	31
	PEI-PAA	Electrostatic interactions	RT within 10 min	OTR: 100%	100
	PVA-PAA-Ag NPs	Hydrogen bonds	RT within 4 h	T: 97.76%	101
Coating	CS-Beeswax-Pollen grains	Hydrogen bonds	/	TS: 92%	102
	bPEI-PAA-CaCl ₂ -MMT	Electrostatic interactions	RT within 30 s	/	103
	SA-menthol-CD-CS	Host-guest interaction	RT within 20 min	TS: 59.49%	28
	PEI- β -CD-PAA-AD-C ₃ N ₄	Host-guest interaction	RT within 30 min	/	104
	SA-CS	Hydrogen bonds; Electrostatic interactions	RT within 5 min	TS: 97%; OTR: 95%; WVP: 63%	14
	MoS ₂ - β -CD-PEI-AD-PAA	Host-guest interaction	RT within 30 min	/	105

313 Note: RT, room temperature; E, elongation; TS, tensile strength; T, transmittance; OTR,
314 oxygen transmission rate; WVP, water vapor permeability; CS, chitosan; CA, citric acid;
315 CC, choline chloride; AMCS, acrylamide-modified chitosan; ADA, alginate aldehyde;
316 MCNC, multi-branched cellulose nanocrystals; PVA, polyvinyl alcohol; TPU,
317 thermoplastic polyurethane; PEI, polyethyleneimine; Fc, ferrocene; PDA,
318 polydopamine; SA, sodium alginate; CAP, capsaicin; CMC, carboxy methylcellulose
319 sodium; bPEI, branched polyethyleneimine; MMT, sodium montmorillonite nanoclay;

320 CD, cyclodextrin; AD, adamantanamine; Ag NPs, silver nanoparticles

321



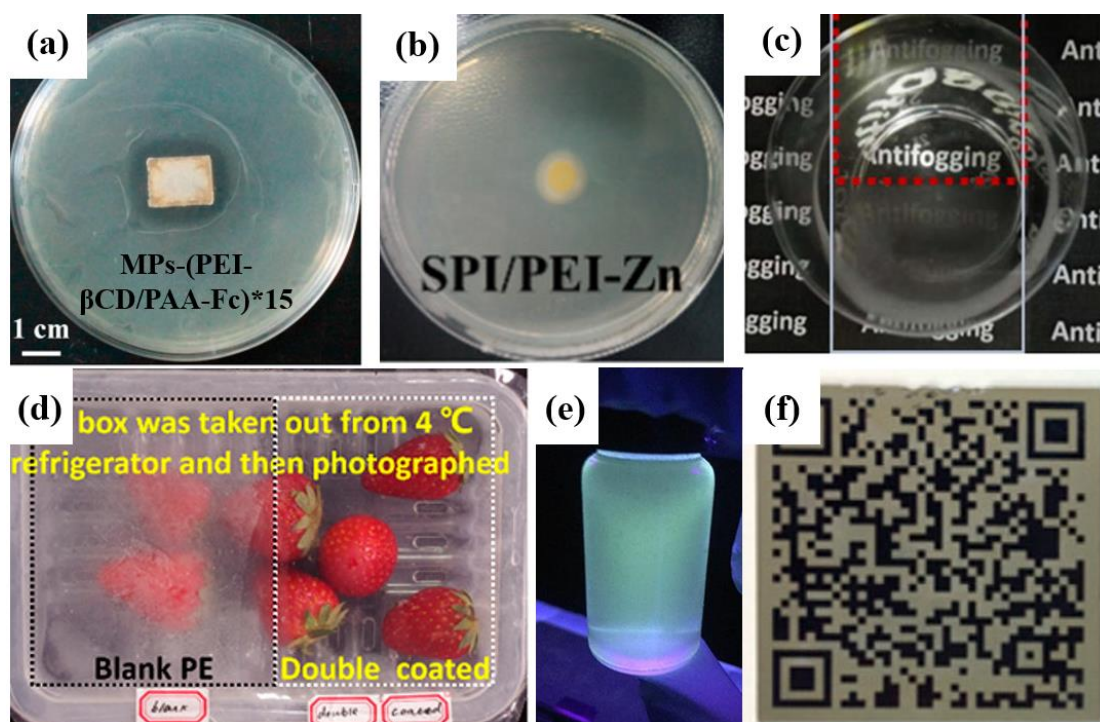
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323 Figure 1. Self-healing of (a) damaged PEI/PAA film under 97% RH for 24h,¹⁰⁰ (b)
324 PBS/LMMT composite film under room condition for 8 h,¹⁰⁶ (c) [ADA50/AMCS3]₆
325 film at room temperature for 2 min,²⁹ (d) scratched [SA/CD-g-CS]₁₀ and [SA/menthol-
326 CD-g-CS]₁₀ in 20 min,²⁸ (e) scratched poly(SBMA-co-IA) coated glass at 25 °C and 55%
327 RH for 6h,¹⁰⁷ (f) scratched PET/Pectin/TA coating over hot water (55 °C and 100% RH)
328 for 60s.¹⁰⁸

329 Note: PEI, polyethyleneimine; PAA, polyacrylic acid; PBS, polyborosiloxane; LMMT,
330 L-cystine modified MMT nanosheets; ADA, alginate aldehyde; AMCS, acrylamide-
331 modified chitosan; SA, sodium alginate; CD, cyclodextrins; CS, chitosan; SBMA,
332 (methacryloyloxy)ethyltrimethyl-(3-sulfopropyl); IA, itaconic acid; PET, poly(ethylene
333 terephthalate); TA, tannic acid.

334 **3.1 Self-healing films for food packaging**

335 Accidental damage during handling and transportation may weaken the protective
336 function of packaging materials. Therefore, self-healing films with recoverable
337 mechanical and barrier properties are necessary to maintain food quality. Smirnov et
338 al.⁹⁷ mixed chitosan with citric acid-based natural deep eutectic solvent (NADES) to
339 prepare a self-healing film using the solution casting method. The damaged films were
340 pressed by finger and healed within 60 seconds at room temperature. The healed films
341 showed a strain of $47 \pm 15\%$ and a stress of 13 ± 2 MPa, with recovery rates of 56%
342 and 72%, respectively. PEM films with super oxygen barrier property were constructed
343 from PEI and PAA by Song et al.¹⁰⁰ The films could heal in a high humidity
344 environment ($>97\%$) within 10 min, but a humidity lower than 50% would not lead to
345 the swelling of the damaged area and the self-healing. The films after ten stretching-
346 healing cycles showed about 92% recovery of elastic modulus (13.09 ± 0.12 GPa) and
347 excellent oxygen barrier property (<0.005 cm³ m⁻² d⁻¹ atm⁻¹). Another self-healing
348 nanocomposite films were manufactured from grafted cellulose nanocrystals (g-CNC),
349 matrix prepolymers from monomers, dibutyltin dilaurate (DBTDL), and
350 hexamethylene diisocyanate (HDI).¹⁰⁹ The dynamic hydrogen bonds and metal-ligand
351 coordinated covalent bonds enabled the self-healing capacity at low temperature. The
352 increase of g-CNC from 0 wt.% to 3 wt.% enhanced the covalent bonds and hydrogen
353 bonds, resulting in 99% self-healing efficiency at 5 °C within 1 h. A self-healable
354 hydrophobic cellulose film was assembled by incorporating natural wax through
355 annealing treatment.¹¹⁰ The annealing process at 150 °C promoted the rearrangement of
356 wax in the cellulose matrix and the wax migrated to the surface of film to achieve self-
357 healing. The incorporation of natural wax also increased the water contact angle (120°)
358 and improved the flexibility (11%) without reducing mechanical strength (122 MPa) of
359 cellulose films.



361

362 Figure 2. Self-healing packaging films with (a,b) antimicrobial, (c,d) anti-fogging, (e)
 363 heavy metal detection, and (f) recoverable labelling capacities.^{29,99,101,111-113}

364 Other functionalities can also be integrated with the self-healing capacity for active
 365 packaging, and some recent examples are shown in Figure 2. Two metal ions (Cu^{2+} and
 366 Zn^{2+}) were added in the solution of soy protein isolate (SPI), PEI, and glycerol to
 367 fabricate the self-healing and antimicrobial films.¹¹¹ The highly branched structure and
 368 abundant amine groups in PEI easily disrupted the ordered structure of SPI and formed
 369 numerous hydrogen bonds. The incorporation of Cu^{2+} in the films led to better self-
 370 healing efficiency and antimicrobial activity than Zn^{2+} , which was attributed to more
 371 coordination bonds and stronger interaction between Cu^{2+} and negatively charged
 372 bacterial cell walls, respectively. The original films containing Cu^{2+} showed the tensile
 373 strain and stress of 81.78% and 10.09 MPa, which were recovered to about 90.17% and
 374 105.57% after self-healing at 25 °C for 10 h. Xuan et al.⁹⁹ assembled dual cross-linked
 375 films consisting of PEI- β -CD and ferrocene-modified PAA with host-guest and
 376 hydrogen bonding interactions. The low water contact angle (31.8°) indicated the
 377 hydrophilic surface that facilitated the spreading and penetration of water into the films
 378 and led to the fast self-healing process. The maximum tensile force of the virgin film

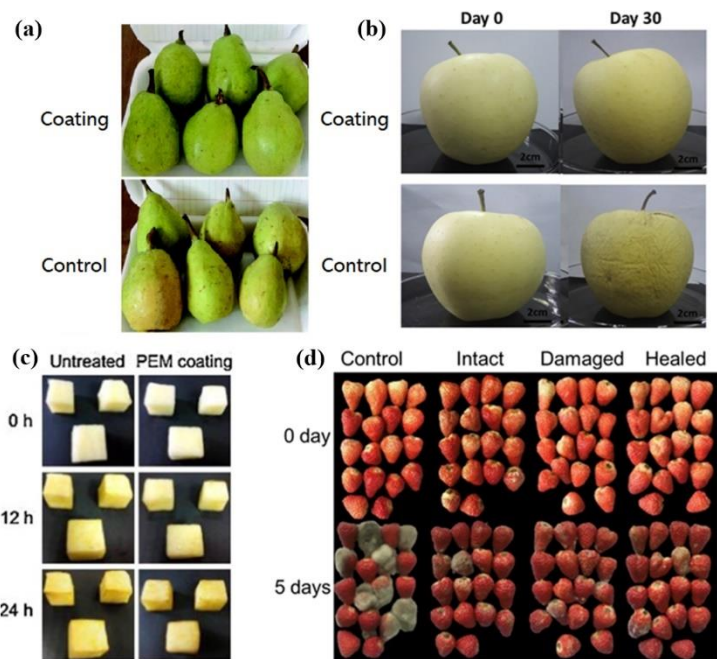
379 was 0.75 N and the stretched length was 0.85 mm, which recovered to 0.56 N and 0.83
380 mm, respectively, after 30 min self-healing process. At the same time, the films loaded
381 with microencapsulated chloramphenicol possessed a controlled release behavior to
382 inhibit the growth of *E. coli*. Anti-fogging packaging films have been developed
383 recently by LBL assembling of acrylamide-modified chitosan and alginate aldehyde
384 with Schiff base reaction to enable the self-healing property and high transparency.²⁹
385 The recovery efficiency of light transmission of the healed film at 600 nm was 82.1%
386 after 20 times of mechanical abrasions with a Taber abramer, indicating its favorable
387 self-healing ability toward mild or heavy scratches. The anti-fogging ability was
388 achieved by absorbing the condensed water in the hydrophilic films and the thicker
389 films could better prevent from fogging. In another case¹⁰¹, a simple one-pot method
390 was used to prepare the PVA/PAA/Ag composite films with self-healing, anti-fogging,
391 and antimicrobial properties. The scratched composite film returned to a smooth and
392 integrated surface after the self-healing process (temperature & time) without any
393 significant effect on its transmittance (slightly changed from 89.1% to 87.1%), which
394 was due to the fluidity of the polymer chains and reforming of hydrogen bonds. The
395 anti-fogging property was attributed to the abundant hydroxyl groups, while the
396 addition of 2% silver nanoparticles results in 99% antibacterial effect against *E. coli*.
397 An intelligent soy protein-based film was designed with self-healing ability and could
398 detect potential contamination of food by heavy metals.¹¹² The addition of
399 polydopamine (PDA) induced the dense cross-linked network with enhanced tensile
400 strength and endowed the film with the capacity to detect heavy metals (>0.0032 mol/L)
401 in food through fluorescent signals. The composite film showed the restorability at 50 °C
402 and RH 60-70% within 30 min, which was attributed to the high flowability of amines
403 of PEI and phenolic hydroxyl groups of dopamine. The healed film had the tensile stress
404 of about 5.76 MPa and the strain of 85.95%, which were 92.46% and 75.43% of the
405 original values, respectively. Interestingly, Chen et al.¹¹³ assembled a photoreactive
406 LBL film from PAA grafted with phenyl amino groups and PEI via a carbodiimide
407 reaction for displaying recoverable QR code. The scratched QR code made from

408 composite film self-healed and became machine-readable in a RH 100% environment
409 after 12 h. It was because the polyelectrolyte chains were highly activated at saturated
410 humidity, and the subsequent interdiffusion of polyelectrolytes allowed the transfer of
411 substances between sites and healed the scratches. This "smart" label material could be
412 used for restorative graphic identification and information retention in food packaging.
413

414 **3.2 Self-healing coatings for food packaging**

415 The self-healing coatings can be applied directly to food surface to decrease water vapor
416 transmission, control oxygen permeability, and prevent oxidation. As shown in Figure
417 3 (a), for instance, a composite coating from chitosan and beeswax–pollen grains was
418 fabricated by Sultan et al.¹⁰² The formation of electrostatic interactions and hydrogen
419 bonds in the coating resulted in an improvement in water vapor barrier property
420 (1038.07 g per m² per day), tensile (0.99 MPa), and self-healing efficiency (96.30%)
421 compared to virgin chitosan films, which were 2065.23 g per m² per day, 0.77 MPa,
422 and 86.70%, respectively. However, the oxygen transmission rate of the composite
423 coating increased due to the addition of beeswax which increased the hydrophobicity
424 of the film and more oxygen diffusion pathways. The coated pear showed extended
425 shelf life of 105 days at 0 °C and one week at 23 °C, maintaining the quality of pears
426 during storage and marketing conditions. A PEM coating was fabricated by LBL
427 assembly of CMC and chitosan, which significantly decreased the weight loss of apple
428 slices and maintained the firmness and quality for 24 h.¹¹⁴ The original coating had a
429 tensile strength of 28.49 MPa, and the strength of the damaged coating restored to 24.9
430 MPa after self-healing for 20 min. This self-healing process was triggered by water and
431 was ascribed to the formation of hydrogen bonds and electrostatic interactions between
432 -NH₃⁺ and -COO⁻. Moreover, the addition of chitosan endowed the fabricated PEM
433 coating with the free radical scavenging activity of 78.62%, and the antibacterial
434 capacity of 100% and 95.48% against *Salmonella* and *E. coli*, respectively.³¹ A similar
435 PEM coating was manufactured from CMC, chitosan, and CaCl₂ for the storage of
436 lemons under refrigerated conditions for up to 60 days.¹¹⁵ It was revealed that the

437 increased thickness of coating contributed to the higher cross-linking degree and led to
438 the enhanced gas barrier, mechanical, and self-healing properties. Specifically, when
439 the number of bilayers increased from 3 to 7, the water vapor permeability decreased
440 from $3.32 \text{ g}\cdot\text{mm}\cdot\text{kPa}^{-1}\cdot\text{h}^{-1}\cdot\text{m}^{-2}$ to $1.15 \text{ g}\cdot\text{mm}\cdot\text{kPa}^{-1}\cdot\text{h}^{-1}\cdot\text{m}^{-2}$, while the tensile strength
441 enhanced from 18.08 MPa to 39.94 MPa and the self-healing efficiency improved from
442 83% to 92%. Du et al.¹⁴ developed an edible coating with self-healing property through
443 LBL assembly of chitosan and SA by hydrogen bonding and electrostatic interactions.
444 The coating formed by three assembly cycles showed > 80% transparency and an
445 insignificant impact on the appearance of packaged fruits and vegetables. The coating
446 healed within 5 min under water stimulation and showed 97%, 95% and 63% recovery
447 of tensile strength (19.89 MPa), oxygen transmission rate ($4.78 \pm 0.82 \text{ cm}^3/(\text{m}^2 \text{ d atm})$),
448 and water vapor transmission rate ($40.56 \pm 7.60 \text{ g}/(\text{m}^2 \text{ d})$) to those of intact coating, so
449 it could diminish the effect of coating damage and remarkably retard the deterioration
450 of packaged strawberries. Manabe et al.¹⁰³ proposed a cephalopods-inspired rapid self-
451 healing nanoclay composite coating with oxygen barrier and super-bubble-phobic
452 properties. The LBL self-assembly of branched PEI and PAA with the addition of CaCl_2
453 and nanoclay could mimic the strong structure of cephalopods and erase the surface
454 scratches in 10 s. Especially, the existence of nanoclay led to the high water-holding
455 capacity, repelled the colliding bubbles in wet state, and improved the oxygen barrier
456 property to less than 1/100 of that of bare polyethylene.



457

458 Figure 3. Applications of self-healing coatings in (a) pear, (b) apple, (c) apple slice, and
 459 (d) strawberry.^{14,102,103,114}

460 Antimicrobial compounds can also be incorporated in self-healing coatings. Yang et
 461 al.¹¹⁵ grafted l-menthol- β -CD with chitosan and assembled them with SA to form an
 462 edible coating. The self-healing ability was enabled by the synergistic effects of
 463 hydrogen bonds and host-guest reaction. The coating of 10 assembled layers had the
 464 transparency of 94% and the 63% recovery rate of the tensile strength (15.89 MPa) after
 465 20 min self-healing process. Besides, the controlled release of l-menthol inhibited the
 466 growth of microorganisms and maintained the firmness of packaged apples, extending
 467 their shelf life to up to 30 days. Another self-healing coating with excellent
 468 antimicrobial activity was prepared from MoS₂ nanosheets, β -CD-modified PEI, and
 469 AD-modified PAA using the LBL self-assembly technique via host-guest interaction.¹⁰⁵
 470 This coating showed a hydrophobic surface with a water contact angle of 98.02° and
 471 could self-heal within 30 min when exposed in a humid environment. The coating
 472 greatly suppressed the bacteria adhesion and inhibited the growth of *E. coli* and could
 473 even measure the concentration of Co²⁺ using fluorescence signal to prevent from
 474 consuming excessive heavy metals.

475 Similar to self-healing films, the fabrication of self-healing coatings with hydrophilic
 476 surfaces was applied as anti-fogging packaging materials. For instance, Yang et al.¹¹⁶

477 manufactured a high-transparency (> 95%) self-healing coating from
478 (methacryloyloxy)ethyltrimethyl(3-sulfopropyl) (SBMA) and itaconic acid (IA). The
479 self-healing process finished in 6 h at 25 °C and 55% RH and was due to the
480 reconstruction of hydrogen bonds in the damaged area. Water droplets could be spread
481 on the surface of the coating in 6 s with a contact angle of less than 10 °C, and the
482 coated glass slide was still transparent after being taken from the refrigerator to room
483 temperature. Another anti-fogging coating was formed with pectin and TA and then
484 applied to the surface of PET film.¹⁰⁸ The abundant -OH groups promoted the diffusion
485 of water droplets, resulting in the excellent anti-fogging capability in both hot-vapor
486 and cold-warm conditions. Moreover, the damaged coating could self-heal in 4 min and
487 was well-retained after rinsing 5 times with a high-speed water faucet, indicating its
488 durability and stability. Xu et al.¹¹⁷ fabricated a robust and transparent (about 90%) self-
489 healing coating from PVA, SA, TiO₂ and glycerin. The cross-linked coating had a water
490 contact angle of 66.8° and was able to absorb the surrounding water vapor quickly,
491 achieving the anti-fogging performance. Besides, after 100 cycles of abrasion by
492 sandpaper at 25 Pa, the coating showed restorable antifogging performance after 1
493 second of water immersion or 5 minutes of solar irradiation treatment, due to the
494 restoration of hydrogen bonds between the absorbed water molecules and the hydroxyl
495 or carboxyl groups of PVA and SA.

496

497 **4 Conclusions and outlook**

498 Recent studies have reported the fabrication of self-healing materials through dynamic
499 non-covalent interactions, dynamic covalent interactions, or their synergistic effects.
500 Several attempts have been made to apply self-healing materials in food packaging in
501 the forms of films or coatings with recoverable mechanical and barrier properties to
502 better protect the packaged food products. Other functionalities such as antimicrobial
503 and anti-fogging properties can also be incorporated to make the food packaging
504 materials versatile. The applications of self-healing materials in food packaging are
505 promising, but future research is required in the following areas:

- 506 1. The self-healing efficiency should be evaluated under different conditions (pH,
507 temperature, humidity, etc.), especially the transportation and storage conditions of
508 food products. Some self-healing processes require external stimuli such as high
509 temperature or high humidity, which may not be suitable for food packaging.
- 510 2. The feasibility of self-healing materials should be considered. The packaging
511 materials with high self-healing efficiency and good mechanical and gas barrier
512 properties are absolutely preferred, but they should also be cost effective and
513 nontoxic, and should not affect the texture and flavor of food products. Compared
514 to the materials that need external stimuli to heal, the intrinsic self-healing films
515 and coatings are more applicable.
- 516 3. Besides the toxicity, the migration of self-healing packaging materials should be
517 carefully evaluated. Chemical modifications are commonly applied during the
518 preparation of self-healing materials, so their fate in human body should be
519 investigated. Moreover, the deterioration of food quality and nutritional value with
520 the application of novel packaging materials should be studied case by case.
- 521 4. The post-consumer waste management of self-healing packaging materials should
522 be investigated. The degradability of materials may be affected by
523 physical/chemical modifications. Therefore, the degradation rate and products of
524 self-healing materials under various conditions should be assessed.

525

526 **Conflict of interest statement**

527 Nothing declared.

528

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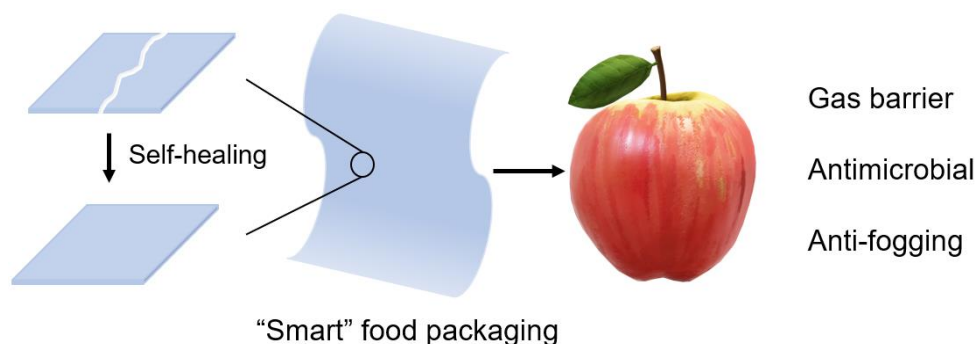
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866 **Graphical Table of Contents**



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869 This review highlights a promising approach to combat damage to food packaging
870 materials during transportation, handling, and storage. We summarized the recent
871 progress in the preparation of self-healing materials through different mechanisms,
872 compared the self-healing efficiency under different conditions, highlighted the
873 potential applications of self-healing films and coatings with recoverable mechanical
874 and barrier properties and other functionalities (e.g. antimicrobial and anti-fogging
875 capacities), and discussed their future opportunities, challenges and research directions.

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