

1 Noncovalently Functionalized Monolayer Graphene for Sensitivity 2 Enhancement of Surface Plasmon Resonance Immunosensors

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10 **S** Supporting Information

11 **ABSTRACT:** A highly efficient surface plasmon resonance (SPR) immunosensor is described using a functionalized single graphene layer on a thin gold film. The aim of this approach was two-fold: first, to amplify the SPR signal by growing graphene through chemical vapor deposition and, second, to control the immobilization of biotinylated cholera toxin antigen on copper coordinated nitrilotriacetic acid (NTA) using graphene as an ultrathin layer. The NTA groups were attached to graphene via pyrene derivatives implying π - π interactions. With this setup, an immunosensor for the specific antibody anticholera toxin with a detection limit of 4 pg mL⁻¹ was obtained. In parallel, NTA polypyrrole films of different thicknesses were electrogenerated on the gold sensing platform where the optimal electropolymerization conditions were determined. For this optimized polypyrrole-NTA setup, the simple presence of a graphene layer between the gold and polymer film led to a significant increase of the SPR signal.

29 **S**urface plasmon resonance (SPR) is an established technique
30 for the study of biomolecular interactions and the trans-
31 duction of biological recognition events in real time without
32 requiring supplemental labeling steps.¹ The principle of this
33 technique is based on light stimulated oscillation of electrons in
34 the conduction band of usually gold films, called resonant surface
35 plasmons. This phenomenon is strongly dependent on the
36 dielectric constant of its environment² and represents a great
37 advantage for biosensing applications since a biological
38 receptor-analyte interaction results in a change of the oscillation
39 frequency which can be recorded by measuring the angle change,
40 intensity, refractive index, or phase of the reflected light.^{1,3}

41 Extensive efforts were invested to improve the sensitivity of
42 SPR signals using, e.g., gold nanoparticles,⁴ quantum dots,⁵ or
43 Au/Ag alloy nanocomposites.⁶ In terms of targeted immobiliza-
44 tion of bioreceptor units on gold surfaces for SPR biosensing,
45 self-assembled (SAM) molecular monolayers⁷ or electrogen-
46 erated functional polymers⁸ are often used. As known, the SPR
47 sensitivity strongly depends on the thickness and dielectric
48 constant of such functional layers on the surface. Therefore, due
49 to a one atom thick sheet of carbon atoms in a hexagonal lattice

and the recent development of its large-scale synthesis and
transfer techniques as well as its functionalization, graphene
should be an excellent candidate for SPR signal enhancement^{9,10}
and label-free monitoring of chemical or biomolecular
interactions.^{11,12} In addition to its high carrier mobility and
zero-band gap characteristics, graphene also exhibits unique and
desirable optical properties, such as broadband and tunable
absorptions.^{13,14} It has been shown that light transmittance
through monolayer graphene is 97.7%,¹⁵ e.g., a one-atom-thick
graphene layer will absorb only 2.3% of incident light.

Theoretical models have predicted that the incorporation of a
single layer of graphene can amplify significantly the optical
sensitivity of SPR sensors.¹⁶ The beneficial optical properties of
graphene monolayers in the visible light range¹⁷ lead to a change
of the propagation constant of surface plasmon polariton (SPP),
thus amplifying the change of the refractive index.¹⁸

Furthermore, biomolecules containing hydrophobic domains
or π -systems like DNA strands or proteins tend to adsorb
spontaneously on graphene.¹⁹ Graphene can also easily be
functionalized²⁰ and thus be modified for targeted immobiliza-
tion of bioreceptor units.

Most of the reported graphene-based SPR biosensors utilize
graphene oxide,²¹ reduced graphene oxide,²² or graphene
decorated metal nanoparticles^{23,24} as a sensing platform.
However, the main limitation of this approach is the lack of
homogeneous and defect-free monolayers on the SPR sensing
platform, hindering the exploitation of the benefits of graphene.

Chemical vapor deposition (CVD) growth that can provide
large scale monolayers of graphene with low defect densities is an
attractive alternative method to the previously reported SPR
substrates.²⁵ However, so far graphene monolayers for SPR
biosensing have been mainly investigated numerically²⁶ and/or
by using DNA²⁷ or protein adsorption²⁸ on the SPR chip surface.

This study reports on the beneficial optical properties of a
single graphene sheet obtained by CVD for SPR biosensing
applications. Graphene was modified for the controlled
immobilization of the receptor unit, antigen cholera toxin either
with a functional polypyrrole film or via noncovalent
functionalization using pyrene derivatives²⁹ where the pyrene

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89 layer was reinforced by electropolymerization.^{30,31} Both, the
 90 polypyrrole and the pyrene groups contain the anchoring
 91 nitrilotriacetic acid (NTA) group for the immobilization of the
 92 specific biotin tagged receptor, cholera toxin as illustrated in
 93 Figure 1.

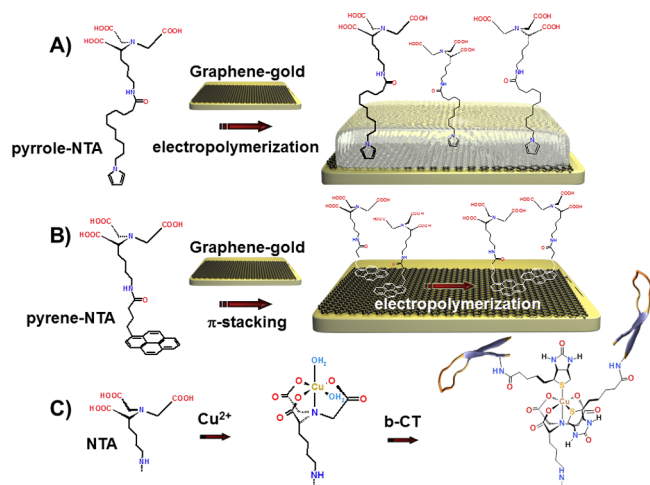


Figure 1. Schematic presentation of the functionalization of the graphene layer via (A) electropolymerization of a polypyrrole-NTA and (B) π stacking of pyrene-NTA followed by electropolymerization for the reinforcement of the layer. (C) Reaction scheme of the conditioning of the NTA group for the immobilization of b-CT.

94 The synthesis of graphene and its transfer to gold is described
 95 in the Supporting Information (SI). Great care was taken to
 96 supply homogeneous graphene with little defect density and
 97 polymer residues,^{32,33} to supply a clean surface for further
 98 functionalization. Raman imaging is a very powerful tool to
 99 evaluate carbon surfaces.³⁴ In Figure 2 averaged Raman spectra

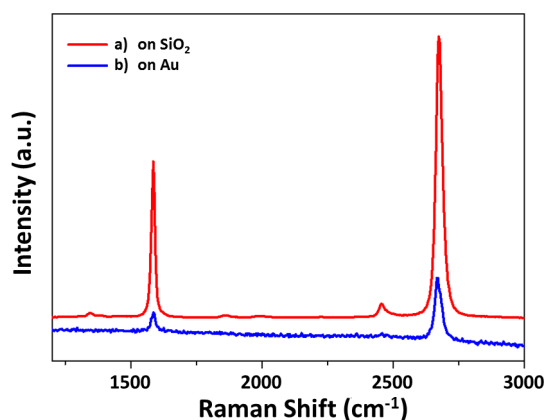


Figure 2. Average Raman spectra of pristine single layer graphene on (a) SiO₂ and (b) gold substrates.

100 of single-layer graphene, composed of 10,000 spectra taken over
 101 an area of 40 × 40 μm are shown. On SiO₂ substrate, D-, G-, and
 102 2D peaks appear at 1350, 1583, and 2680 cm⁻¹, respectively
 103 using 532 nm excitation wavelength. The gold substrate
 104 spectrum is noisier and has a sloped background due to emission
 105 from the metal substrate (Figure 2b)^{35,36} The intensity of the
 106 graphene peaks is enhanced on SiO₂ due to interference on the
 107 dielectric substrate,³⁷ which allows a detailed assessment of the
 108 graphene quality by Raman imaging shown in Figure S1. While

the D band in combination with the D/G band ratio gives insight
 into the quality of graphene in terms of defects in the graphene's
 π -system, the 2D band is representative for the number of
 graphene layers.³⁸ The Raman spectra in Figure 2 clearly indicate
 the presence of high-quality monolayer graphene due to the
 almost negligible presence of a D band and the sharp and
 symmetric 2D band as well as the high 2D/G peak ratio.^{39,40}
 Further Raman analysis, optical images and XPS spectra showing
 the continuity and high quality of the graphene are provided in
 Figures S1 and S3.

Effective immobilization of biomolecules on the sensor surface
 is one of the most challenging steps in the development of high-
 performance biosensors.⁴¹ To secure this requirement, the NTA-
 Cu²⁺/biotin system was used.⁴² Furthermore, noncovalent
 functionalization using π -stacking interactions or electropolymer-
 ization was also an important consideration in order to preserve
 the properties of graphene.

A detailed procedure of the functionalization of the gold
 surface via electropolymerization of pyrrole-NTA as well as the
 electrochemical coating and noncovalent attachment of pyrene-
 NTA can be found in the SI. Three polymer films of different
 thicknesses were electrogenerated giving coatings of 5.66, 16.9,
 and 28.3 nmol·cm⁻² for the electropolymerization conditions at
 1, 3, and 5 mC·cm⁻², respectively. The resulting polypyrrole-
 NTA films thicknesses were estimated using a confocal laser
 microscope and gave around 100 nm for 1 mC·cm⁻², 270 nm for
 3 mC·cm⁻², and 450 nm for 5 mC·cm⁻². These films were used
 for the successive attachment of the bioreceptor unit
 (biotinylated cholera toxin, b-CT) via subsequent coordination
 of copper(II) ions at the NTA chelate and b-CT.

The immunoreaction between the immobilized b-CT and the
 analyte, anticholera toxin from rabbit (anti-CT), was monitored
 in the angle shift mode in real time. The angle shift is correlated
 with the changing thickness and optical properties of the sensing
 layer. The response also depends on the refractive index of the
 bulk solution. There is a linear relationship between the amount
 of bound material (analyte) and the SPR angle shift.⁴³ These
 angle shifts are in the order of millidegrees (mdeg) and are used
 as a response unit to quantify the binding of the analyte to the
 sensor surface. Control experiments were performed without
 immobilization of the receptor antigen cholera toxin (Table S1),
 and all experiments were conducted three times to examine the
 reliability of the assays.

After demonstrating that the thinnest polymer film gives the
 highest SPR angle change (Figure S2), the conditions for the
 electropolymerization of such films of 100 nm thickness were
 applied to graphene-modified gold disks. Figure 3 shows the SPR
 angle change of polypyrrole-NTA/Cu²⁺/b-CT on pure gold films
 and polypyrrole-NTA/Cu²⁺/b-CT graphene-gold films for
 35 and 4 ng·mL⁻¹ injection of anti-CT. The polypyrrole-NTA/
 Cu²⁺/b-CT modified graphene-gold slides showed an angle
 change of 324 mdeg for 35 ng·mL⁻¹ and 70 mdeg for 4 ng·mL⁻¹
 anti-CT, respectively, which is significantly higher than for the
 same configuration without the graphene monolayer (183 and 31
 mdeg, respectively).

For these concentrations of analyte, the simple presence of a
 graphene monolayer led to an almost 2-fold increase of the angle
 shift. Such signal amplification was also observed for the
 reflectivity mode as summarized in Table S1. A 9% increase in
 reflectivity was recorded with graphene-based SPR immnosensor.
 For pure gold surfaces, the reflectivity change was 7%. A
 control experiment was performed with gold polypyrrole-NTA/
 Cu²⁺ and graphene-gold polypyrrole-NTA/Cu²⁺ in absence of

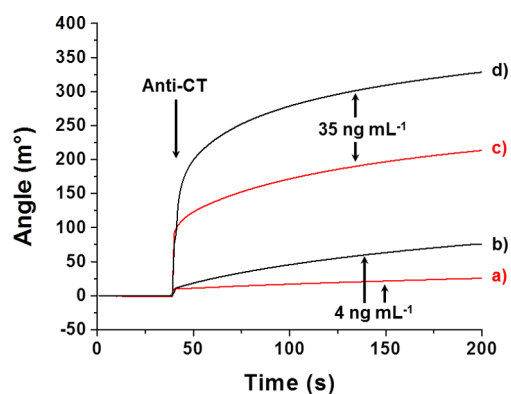


Figure 3. SPR angle shift after anti-CT injection using pure gold (red) and graphene-gold (black) surfaces functionalized with polypyrrole-NTA/Cu²⁺/b-CT at two different anti-CT concentrations (a, b: 4 ng·mL⁻¹ and c, d: 35 ng·mL⁻¹). The polymer film was formed under controlled potential electrolysis (0.95 V, 1 mC·cm⁻²).

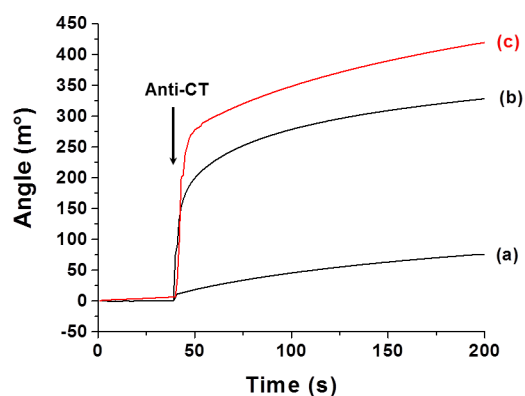


Figure 4. SPR angle change after 35 ng·mL⁻¹ of anti-CT injection using polypyrrole-NTA/Cu²⁺/b-CT on (a) pure gold and (b) graphene-gold. (c) Pyrene-NTA/Cu²⁺/b-CT graphene-gold surface where the pyrene groups were electropolymerized after formation of an ultrathin layer via π -stacking interactions.

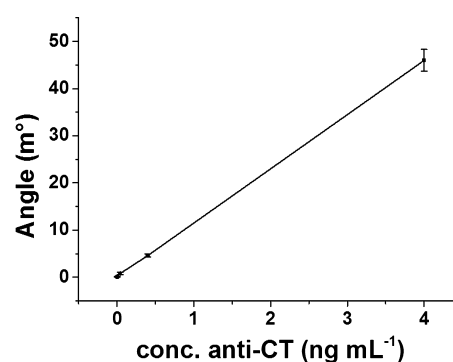


Figure 5. Linear range of the graphene-based SPR immunosensor toward anti-CT.

172 the antigen receptor. For both surface types, a reflectivity change
173 of around 2% with 35 ng·mL⁻¹ of antibody was recorded for the
174 nonspecific binding of the antibody.

175 One potent method, preserving the unique properties of
176 graphene is noncovalent functionalization with molecules
177 containing an extended π -system, such as pyrene and its
178 derivatives. In order to keep the sensing layer as thin as possible,
179 pyrene-NTA was π -stacked onto graphene giving an ultrathin
180 layer. The successful formation of such a layer was confirmed by
181 XPS measurements, revealing C 1s contributions nearly
182 exclusively from graphene and pyrene-NTA and an estimated
183 thickness of 1.4 nm of the graphene/pyrene-NTA stack, as
184 discussed in the SI. This layer was further stabilized by
185 electropolymerization of the pyrene groups under controlled
186 potential electrolysis via radical coupling at the 3-6 or 3-8
187 position.^{30,31} After identical preparation of the immunosensor as
188 for the polypyrrole-NTA setup, the SPR response was measured
189 in angle change and reflectivity mode. An injection of anti-CT at
190 a concentration of 35 ng·mL⁻¹ led to an angle change of 418
191 mdeg. Compared with the best performing polypyrrole-NTA
192 graphene-gold setup (324 mdeg), the reduced sensing layer
193 thickness improved the SPR immunosensor response by ~80%.
194 This angle shift increase is even more significant when compared
195 with the polypyrrole-NTA setup without graphene (183 mdeg).
196 Here, the angle change increased 150% (Figure 4). These clear
197 improvements can be attributed not only to the SPR signal
198 amplifying properties of graphene but also to the possibility to
199 functionalize graphene with SAM techniques using pyrene
200 derivatives.

201 Concerning the measurements in reflectivity mode, a 12%
202 increase in reflectivity was recorded with pyrene functionalized
203 graphene-gold SPR immunosensor (Table S1).

204 The increase is again in the 2-fold range compared to the most
205 optimized polypyrrole setup without graphene. To determine
206 the limit of detection (LOD), the SPR response was recorded for
207 further dilutions of anti-CT in the angle change mode after 50 s.
208 As shown in Figure 5, a linear range could be determined
209 between 0.004 and 4 ng·mL⁻¹ ($R^2 = 0.9999$) with a LOD of 4 pg·
210 mL⁻¹ at a relative standard deviation of 9.2% of three identical
211 experiments. Such performances are orders of magnitude higher
212 than comparable electrochemical setups like amperometry,⁴⁴
213 label-free electrochemical impedance spectroscopy,⁴⁵ or photo-
214 electrochemical transduction.⁴⁶

215 Finally, particular care was taken in characterizing nonspecific
216 binding of anti-CT on the functionalized graphene surface. 216
217 Control experiments were performed under identical conditions
218 by omitting the b-CT immobilization step. In the absence of the
219 antigen receptor unit, a ~2% change in reflectivity and was
220 recorded after injection of 35 ng·mL⁻¹ of antibody. This is a non-
221 negligible value for nonspecific binding of the analyte but
222 represents only 20% of the SPR signal intensity. Other control
223 experiments without receptor unit immobilization using
224 polypyrrole-NTA surfaces or untreated graphene-gold were in
225 the same range. Table S1 represents the data of all SPR responses
226 recorded in angle shift, refractive index, and reflectivity mode as
227 well as captured target biomolecules, including control experi-
228 ments.

229 In conclusion, the exceptional optical properties of monolayer
230 graphene were exploited to amplify the SPR signals for cholera
231 immunosensing as a disease model. NTA functional groups were
232 attached to CVD grown graphene via electropolymerization of a
233 polypyrrole film or via π -stacking of pyrene derivatives, where
234 this layer was further stabilized by electropolymerization. The
235 NTA anchor group served for the controlled immobilization of
236 the biotinylated bioreceptor cholera toxin. The simple presence
237 of a single graphene sheet increased the SPR sensor perform-
238 ances by 80% compared to the graphene-devoid setup. 238
239 Furthermore, the best SPR immunosensor was obtained using
240 π -stacking interactions of pyrene-NTA which led to an ultrathin
241 functional layer after electropolymerization.

242 The clear advantages of the presence of a monolayer graphene
243 and the possibility of ultrathin functional coatings are most likely
244 extensible to other targets for label free SPR immuno- or DNA
245 sensing.

246 ■ ASSOCIATED CONTENT

247 ● Supporting Information

248 Graphene synthesis, transfer and characterization, materials and
249 methods. This material is available free of charge via the Internet
250 at <http://pubs.acs.org>.

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255 Notes

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270 ■ REFERENCES

- 271 (1) Wijaya, E.; Lenaerts, C.; Maricot, S.; Hastanin, J.; Habraken, S.;
272 Vilcot, J.-P.; Boukherroub, R.; Szunerits, S. *Curr. Opin. Solid State Mater.*
273 *Sci.* **2011**, *15*, 208–224.
274 (2) Kelly, K. L.; Coronado, E.; Zhao, L. L.; Schatz, G. C. *J. Phys. Chem.*
275 *B* **2002**, *107*, 668–677.
276 (3) Guo, X. *J. Biophotonics* **2012**, *5*, 483–501.
277 (4) Pedersen, D. B.; Duncan, E. J. S. *Technical Report 2005-109*;
278 Defence R&D Canada: Ottawa, Ontario, 2005.
279 (5) Malic, L.; Sandros, M. G.; Tabrizian, M. *Anal. Chem.* **2011**, *83*,
280 5222–5229.
281 (6) Wang, J.; Song, D.; Wang, L.; Zhang, H.; Zhang, H.; Sun, Y. *Sens.*
282 *Actuators, B* **2011**, *157*, 547–553.
283 (7) Chaki, N. K.; Vijayamohan, K. *Biosens. Bioelectron.* **2002**, *17*, 1–
284 12.
285 (8) Cosnier, S.; Holzinger, M. *Chem. Soc. Rev.* **2011**, *40*, 2146–2156.
286 (9) Geim, A. K. *Science* **2009**, *324*, 1530–1534.
287 (10) Geim, A. K.; Novoselov, K. S. *Nat. Mater.* **2007**, *6*, 183–191.
288 (11) Wang, Y.; Li, Z.; Wang, J.; Li, J.; Lin, Y. *Trends Biotechnol.* **2011**,
289 *29*, 205–212.
290 (12) Salavagione, H. J.; Diez-Pascual, A. M.; Lazaro, E.; Vera, S.;
291 Gomez-Fatou, M. A. *J. Mater. Chem. A* **2014**, *2*, 14289–14328.
292 (13) Bonaccorso, F.; Sun, Z.; Hasan, T.; Ferrari, A. C. *Nat. Photonics*
293 **2010**, *4*, 611–622.
294 (14) Bao, Q.; Loh, K. P. *ACS Nano* **2012**, *6*, 3677–3694.
295 (15) Nair, R. R.; Blake, P.; Grigorenko, A. N.; Novoselov, K. S.; Booth,
296 T. J.; Stauber, T.; Peres, N. M. R.; Geim, A. K. *Science* **2008**, *320*, 1308–
297 1308.
298 (16) Wu, L.; Chu, H. S.; Koh, W. S.; Li, E. P. *Opt. Express* **2010**, *18*,
299 14395–14400.
300 (17) Bruna, M.; Borini, S. *Appl. Phys. Lett.* **2009**, *94*, 031901.
301 (18) Jacek, G.; Dawn, T. H. T. *Nanotechnology* **2013**, *24*, 185202.

- (19) Song, B.; Li, D.; Qi, W.; Elstner, M.; Fan, C.; Fang, H. *ChemPhysChem* **2010**, *11*, 585–589. 302
(20) Georgakilas, V.; Otyepka, M.; Bourlinos, A. B.; Chandra, V.; Kim, 303
N.; Kemp, K. C.; Hobza, P.; Zboril, R.; Kim, K. S. *Chem. Rev.* **2012**, *112*, 304
6156–6214. 305
(21) Chiu, N.-F.; Huang, T.-Y.; Lai, H.-C.; Liu, K.-C. *Nanoscale Res.* 306
Lett. **2014**, *9*, 1–7. 307
(22) Wang, L.; Zhu, C.; Han, L.; Jin, L.; Zhou, M.; Dong, S. *Chem.* 308
Commun. **2011**, *47*, 7794–7796. 309
(23) Zhang, H.; Song, D.; Gao, S.; Zhang, J.; Zhang, H.; Sun, Y. *Sens.* 310
Actuators, B **2013**, *188*, 548–554. 311
(24) Zhang, J.; Sun, Y.; Xu, B.; Zhang, H.; Gao, Y.; Zhang, H.; Song, D. 312
Biosens. Bioelectron. **2013**, *45*, 230–236. 313
(25) Wirtz, C.; Lee, K.; Hallam, T.; Duesberg, G. S. *Chem. Phys. Lett.* 314
2014, *595–596*, 192–196. 315
(26) Choi, S. H.; Kim, Y. L.; Byun, K. M. *Opt. Express* **2011**, *19*, 458– 316
466. 317
(27) Zagorodko, O.; Spadavecchia, J.; Serrano, A. Y.; Larroulet, I.; 318
Pesquera, A.; Zurutuza, A.; Boukherroub, R.; Szunerits, S. *Anal. Chem.* 319
2014, *86*, 11211–11216. 320
(28) Szunerits, S.; Maalouli, N.; Wijaya, E.; Vilcot, J.-P.; Boukherroub, 321
R. Anal. Bioanal. Chem. **2013**, *405*, 1435–1443. 322
(29) Eigler, S.; Hirsch, A. *Angew. Chem., Int. Ed.* **2014**, *53*, 7720–7738. 323
(30) Haddad, R.; Holzinger, M.; Villalonga, R.; Neumann, A.; Roots, J.; 324
Maaref, A.; Cosnier, S. *Carbon* **2011**, *49*, 2571–2578. 325
(31) Waltman, R. J.; Bargon, J. *Can. J. Chem.* **1986**, *64*, 76–95. 326
(32) Hallam, T.; Berner, N. C.; Yim, C.; Duesberg, G. S. *Adv. Mater.* 327
Interfaces **2014**, *1*, 1400115–1400121. 328
(33) Peltekis, N.; Kumar, S.; McEvoy, N.; Lee, K.; Weidlich, A.; 329
Duesberg, G. S. *Carbon* **2012**, *50*, 395–403. 330
(34) Mews, A.; Koberling, F.; Basché, T.; Philipp, G.; Duesberg, G. S.; 331
Roth, S.; Burghard, M. *Adv. Mater.* **2000**, *12*, 1210–1214. 332
(35) Xu, W.; Xiao, J.; Chen, Y.; Chen, Y.; Ling, X.; Zhang, J. *Adv. Mater.* 333
2013, *25*, 928–933. 334
(36) Kim, N.; Oh, M. K.; Park, S.; Kim, S. K.; Hong, B. H. *Bull. Korean* 335
Chem. Soc. **2010**, *31*, 999–1003. 336
(37) Wang, Y. Y.; Ni, Z. H.; Shen, Z. X.; Wang, H. M.; Wu, Y. H. *Appl.* 337
Phys. Lett. **2008**, *92*, 043121. 338
(38) Ferrari, A. C.; Basko, D. M. *Nat. Nano* **2013**, *8*, 235–246. 339
(39) Wink, T.; J. van Zuilen, S.; Bult, A.; P. van Bennekom, W. *Analyst* 340
1997, *122*, 43R–50R. 341
(40) Ferrari, A. C.; Meyer, J. C.; Scardaci, V.; Casiraghi, C.; Lazzeri, M.; 342
Mauri, F.; Piscanec, S.; Jiang, D.; Novoselov, K. S.; Roth, S.; Geim, A. K. 343
Phys. Rev. Lett. **2006**, *97*, 187401. 344
(41) Sicard, C.; Brennan, J. D. *MRS Bull.* **2013**, *38*, 331–334. 345
(42) Baur, J.; Holzinger, M.; Gondran, C.; Cosnier, S. *Electrochem.* 346
Commun. **2010**, *12*, 1287–1290. 347
(43) Stenberg, E.; Persson, B.; Roos, H.; Urbaniczky, C. *J. Colloid* 348
Interface Sci. **1991**, *143*, 513–526. 349
(44) Ionescu, R. E.; Gondran, C.; Cosnier, S.; Gheber, L. A.; Marks, R. 350
S. Talanta **2005**, *66*, 15–20. 351
(45) Haddour, N.; Chauvin, J.; Gondran, C.; Cosnier, S. *J. Am. Chem.* 352
Soc. **2006**, *128*, 9693–9698. 353
(46) Yao, W.; Le Goff, A.; Spinelli, N.; Holzinger, M.; Diaio, G.-W.; 354
Shan, D.; Defranco, E.; Cosnier, S. *Biosens. Bioelectron.* **2013**, *42*, 556– 355
562. 356
357